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ORIGINAL RESEARCH

Synergistic Nanocomposites of Different Antibiotics Coupled with Green Synthesized Chitosan-Based Silver Nanoparticles: Characterization, Antibacterial, in vivo Toxicological and Biodistribution Studies

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Purpose: The present study reports chiter of functionalized grand synthesized CS-AgNPs, conjugated with amoxicillin (AMX), or time (CEF), and lengthoxacin (LVX) for safe and enhanced antibacterial activity.

Methods: The CS-AgNPs and conjugates CX AgNPs+AMX CS-AgNPs+CEF, and CS-Led by UV-Vis, FTL, SEM, TEM, EDX spectroscopy. The AgNPs+LVX were character size distribution and zeta referitial were dasured using the dynamic light scattering (DLS) technique. The interaction tween CS-A NPs and antibiotic molecules was also investigated using UV–Vis spectrose wat the oncentrations of 5, 50, 500, and 5000 μ M for each stivity and synergism were assessed by the Fractional Inhibitory antibiotic. Antib. ten. inder the echanism for synergistic activity was investigated by the Concentration (FIC detection pecies based on the chemiluminescence of luminol. The biocompathydrox alculated from IC₅₀ using the HeLa cell line. In vivo toxicity and ibil index BI) wa. ue distr of silver ions were evaluated on Sprague Dawley rats. Physical interactibiotics and significant (P<0.05) antibacterial activity were observed after loading tion on CS-A NPs surfaces.

Results: It spherical shape nanocomposites of CS-AgNPs with different antibiotics were pared with mean size ranges of 80–120 nm. IC_{50} of antibiotics-conjugated CS-AgNPs decreased compared to CS-AgNPs. The biocompatibility (BI) index showed that antibiotics-conjugated CS-AgNPs have high antibacterial potential and low toxicity. Highly significant (P<0.005) increase in the generation of hydroxyl species indicated the radical scavenging mechanism for synergistic activity of CS-AgNPs after combined with different antibiotics. Biochemical analysis and histopathological examinations confirmed low toxicity with minor hepatotoxicity at higher doses. After oral administration, extensive distribution of Ag ion was observed in spleen and liver.

Conclusion: The study demonstrates positive attributes of antibiotics-conjugated CS-AgNPs, as a promising antibacterial agent with low toxicity.

Keywords: chitosan functionalized silver nanoparticles, antibiotic resistance, synergistic antibacterial activity, in vivo toxicity, tissue distribution

Introduction

The convergence of nanotechnology with nanomedicine has brought new hope to the pharmaceutical and therapeutic fields. Today, nanoparticles (NPs) have been used as diagnostic agents, fluorescent labels, antimicrobial agents, and transfection

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labels.^{1,2} Silver nanoparticles (AgNPs) have unique physicochemical and biological characteristics with the high surface area due to the variable size ranges from 1 to 100 nm.³ These unique characteristics of AgNPs have attracted globally and are considered a potent antimicrobial agent. It has been suggested that silver NPs has a significant antimicrobial activity due to multiple reasons including their high affinity with the surface active groups of microbial strains, released of silver ions can rapture the bacterial cell wall or distortion of bacterial DNA helical structure.^{4,5}

At recent times, natural polysaccharides have been widely used in numerous medical fields such as drug delivery, electrochemical devices, cell imaging, energy storage and biosensors.⁶⁻⁸ Especially, chitosan (CS) a carbon-based natural polymer, is one of the non-toxic, biocompatible, and cost-effective raw biomaterial, majorly found in crustaceans.⁹ CS has been a widely used biological macromolecule as a drug carrier in the pharmaceusector. Previously, researchers reported the tical antimicrobial activity of CS as nanocarrier.^{10,11} CS is non-toxic, a biodegradable natural polysaccharide with intrinsic antibacterial activity. It has an NH³⁺ group in its structure can adsorb on a cell wall by electrosta interaction.¹² It also reduces the need for additional reduc tant while conjugation with silver nanoparticles ¹³

The worldwide increase in mortality and moleidity rates due to multi-drug resistant patherenic betterial strains poses a severe challenge in the reld charedicines. The World Health Organization (Web) reported that more than 64% of patients die die to tethicillin-resistant *Staphylococcus aureus* (MESA) infection compared to non-resistant bacterial infected patients.¹⁴ The irrational use of antibiotics is the poor clase of the development of resistance by bacterial strate against various antibiotics.¹⁵

the ulti-di sistance crisis, novel To comb approaches hould complied to bring improvements in current treath, options. Many researchers worked on the synergism of tibiotics using different non-toxic and eco-friendly techniques. Recently, several studies reported the synergistic antibacterial activities of different antibiotics after combinations with different metallic nanoparticles.¹⁶⁻²⁰ Some studies have described the reduction in the cytotoxicity of antimicrobial compounds after conjugations.^{21,22} However, the potential in vivo toxicity and tissue distribution of these synthesized and conjugated silver nanoparticles have not vet been investigated. Similarly, no report has been found for the assessment of synergism, in vivo toxicity and tissue distribution of chitosan functionalized silver nanoparticles (CS-AgNPs), synthesized from an extract of *Syzygium aromaticum* (clove bud).

The exponential rise in antimicrobial resistance is a big everyday challenge faced by clinicians. To combat the crisis, novel combinations of green synthesized silver nanocomposites, functionalized with chitosan, and conjugated with antibiotics, have been developed in the current study. The work is focused on evaluating the toxicity and antibacterial effectiveness of these supersistic combinations. Moreover, histopathological evaluation and bio-distribution of silver ion in different loses of CS-ugNPs, by atomic absorption technique were an operformed.

Materials and Nethods Antibiotics loading in Synthesized Chitosar Based Silver Manoparticles (CS-AgNP Surfaces

Great synthesis of AgNPs using Syzygium aromaticum etha olic extract SAEE) and functionalized with chitosan (CS-NPs) we described in our previous study.²³ were synthesized by adding SAEE (10 Briefly, A dropwise to the AgNO₃ (1 mmol/L) solution in m e ratio of 9:1. At room temperature, the mixture was ontinuously stirred and left in a darkroom for 24 h to event photodegradation of silver ions. Then, the obtained suspension was centrifuged at 10,000 rpm for 40 min. After centrifugation, supernatant was drawn, washed with distilled water and assess on UV spectrophotometer for AgNPs synthesis. Moreover, the change in color from silver to brownish-black was also indicated the completion of AgNPs synthesis. Finally, the synthesized AgNPs were dried at 65°C for 3 h in a hot air oven. However, functionalization of chitosan with AgNPs was done by dissolving 0.5 g of CS in an acetic solution, and then 0.1 g of AgNPs was added in the prepared CS solution with constant stirring for 45 min. The synthesized precipitate of CS-AgNPs was washed, centrifuged at 10,000 rpm for 10 min, and then dried at 65°C for 12 h in a hot air oven to obtain a purified nanocomposite of CS-AgNPs.

Ten milliliters of 10 μ g/mL of synthesized CS-AgNPs was added to the selected widely prescribed antibiotics, ie, amoxicillin (AMX) (25 μ g/mL), cefixime (CEF) (5 μ g/mL) and levofloxacin (LVX) (5 μ g/mL) in the presence of 2-(N-morpholino) ethane sulfonic acid buffer. The solution

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was stirred using a magnetic stirrer at ambient temperature for 45 min, and surface plasmon resonance (SPR) fluctuation was monitored using a UV spectrophotometer. The nanocomposites of CS-AgNPs after conjugation with different antibiotics were washed with distilled water, filtered, and then dried at 65°C in an oven for 12 h. The efficiency of antibiotics loading on CS-AgNPs was evaluated by measuring the supernatant residues of antibiotics using a UV spectrophotometer after centrifugation at 14,000 rpm for 15 min.²⁴ The antibiotics loading were calculated by Eq. 1:

Antibiotic (AB) loading efficiency (%) = $\frac{Total \ amount \ of \ AB - Amount \ of \ sup \ erna \ tan \ tAB}{Total \ amount \ of \ AB} x \ 100 \ (Eq. 1)$

Characterization and Interaction of Antibiotics Loaded on CS-AgNPs Surfaces

UV visible spectrophotometer (Shimadzu, Model no: UV-1712, Japan) was used to evaluate the optical properties of antibiotics-loaded CS-AgNPs at the wavelength range of 200-700 nm. Field emission scanning electron microscope (FESEM; Joel, Model no: JSM 4380B, Japan) and mission electron microscope (TEM; Joel, Model no. EM 300F, Japan) were used to assess the size and sur antibiotics-loaded morphology of anoc nposit Fourier-transform spe oscope infrared (FTIR Shimadzu, Model no: IR-100, Ja t1 operational n) wavelength of 500-4000/cm 2 energy dersion x-ray spectroscope (EDX; Joel, Jodek o: JSM 6.0, Japan) within 0 and 10 kV ge, were ed to identify the presence of function, groups and silve, ions in antibiotics-loaded CS-Ag, Ps, rejectively. The size distribution and stability of green thesized rgNPs, CS-AgNPs, and antibiotic conjugated how mosites were analyzed using mamic obt scattering (DLS) technique along with a price size analyzer (Brookhaven, USA).^{24,25} The interaction of CS-AgNPs with the studied antibiotics was investigated by the addition of each antibiotic separately to CS-AgNPs solution (50 µM). The CS-AgNPs/ antibiotic ratio varied with the concentration of the antibiotic from 5 to 5000 µM. The UV/Vis spectrums were recorded after mixing for 2 h at different concentrations.²⁶ Also, the mass of each nanocomposite was estimated after dried in a hot air oven at 65°C while proper concentrations of AgNPs, CS, and different antibiotics in each nanocomposite were determined by dissolving each nanocomposite

in deionized water and estimated on UV spectrophotometer.

Antibacterial Activity of AgNPs, CS-AgNPs and Antibiotics-Loaded NPs Zone of Inhibitions (ZIs)

Pathogenic bacterial strains of Escherichia coli (LT 01253). Klebsiella pneumoniae (LT 0471). Staphylococcus aureus (LT 3512), Salmonella typhi (LT 01057), and Pseudomonas aeruginosa (LT 0261) were obtained from Kutiyana Mer n hital Laboratory Karachi – Pakistan. The ing tro antibacit ial activity of SAEE, CS, AgNPs, CS AgN and antoiotics-loaded NPs were analyzed ying the Ox of well diffusion method.²⁷ The contentration of each acterial strain was made equivale to FU/mL s McFarland turbidity standard in utrient bro. The oacterial suspension was spread, d 6 m of stern. Oxford diffusion cups were placed on agar press. Then, each antibacterial agent was pensed in Oxford iffusion cups individually with the oncentration of 10,000 µg/mL of SAEE, 1000 µg/mL of S, 50 μg/m of both AgNPs and CS-AgNPs, 25 μg/mL of MX 5 +25 μg/mL of CS-AgNPs+AMX, 5 μg/mL of CEF, 50+5 µg/mL CS-AgNPs+CEF, 5 µg/mL of LVX and 50, µg/mL of CS-AgNPs+LVX. Plates were incubated for 24 h at $37 \pm 2^{\circ}$ C after 30 min diffusion period. The ZIs of each test solution were measured using a digital Vernier caliper and expressed in millimeter. ZIs determination was performed in triplicate.

The increase in antibacterial activity of each antibiotic (AB) after combination with synthesized CS-AgNPs was calculated using the following equation:

Fold increase of
$$ZI = \frac{ZI \text{ of } CS - AgNPs + AB - ZI \text{ of } AB}{ZI \text{ of } AB}$$
 (Eq. 2)

Minimum Inhibitory Concentrations (MICs), Fractional Inhibitory Concentrations (FICs) Index, and Minimum Bactericidal Concentrations (MBCs)

In the current study, MICs of SAEE, CS, AgNPs, CS-AgNPs, and antibiotics-loaded CS-AgNPs were estimated using the broth dilution method.²⁸ The assay was performed at the concentration ranges of SAEE (1000 to 30,000 μ g/mL), CS, AgNPs and CS-AgNPs (1 to 512 μ g/mL) and antibiotics alone (0.016 to 1024 μ g/mL) in nutrient broth. However, the concentrations of CS-AgNPs in combinations with antibiotics were tested using the checkerboard titration method.²¹ Basically, a checkerboard titration method is normally used to assess two variables at

once: in our case, CS-AgNPs concentration and antibiotics concentration. By plating each strain with a different ratio of CS-AgNPs to an antibiotic, we found not only the optimal concentration of each but the optimal ratio of concentrations as well. All culture strains were adjusted to Mcfarland standard concentrations of 10^6 CFU/mL. An optical density (ODs) of each plate was determined after incubation using the ELISA reader (Infinite 200; USA) at 600 nm. Each assay was performed three times and values are expressed in mean \pm SD.

FIC index determination is an important tool for the assessment of synergism of different antibacterial agents. The concentrations of CS-AgNPs and antibiotics were used in the range of 2 x MICs to 1/16 x MICs for the determination of FICs. For calculation of the FIC index, the comparison was made between the MICs of the antibacterial agent alone and in combination derived MIC. FICs index equal to ≤ 0.5 demonstrates the synergistic activity of the combination used, while FICs in the ranges of 0.5–1, 1–4, and >4 are considered as an additive, indifferent and antagonistic combinations respectively.²⁹ FIC index was calculated using the following equation:

 $FIC = \frac{MIC \ of \ CS - AgNPs \ combined \ with \ AB}{MIC \ of \ CS - AgNPs \ alone} + \frac{MIC \ of \ AB \ combined \ with \ CS - AgNPs}{MIC \ of \ AB \ alone}$

Minimum Bactericidal Concentrations (cellecs) were determined by spreading the 100 photolution from already incubated tubes of each test cample from nutrient agar plates and incubated for 24 that $37 \pm 2^{\circ}$ Cellen, bacterial colonies were observed other the incubation period.²¹ The MBC is defined as the subject of incentration of test solution exhibiting complete (>9.04%) killing of a bacterial strain.³⁰

Mechanism of Conergistic Effects of CS-AgNPs in Combination w. Different Antibiotics

The mechanism of $\sqrt{10}$ pergistic activity of CS-AgNPs in combination with different antibiotics was evaluated by the assessment of hydroxyl radical generation using the luminol chemiluminescence method.³¹ The production of hydroxyl radicals was evaluated by mixing 100 µL of luminol solution with 100 µL AgNPs and 2 µL of antibiotic water solution (40 mg/mL) to obtain a final concentration of 0.4 mg/mL for each antibiotic while combinations of these agents were used at their FICs. The measurement of chemiluminescence was done on a luminometer (LUMAT LB 9507) (Berthold Technologies) and noted in $\sec^{-1}(\sec \text{ Figure 5})$.

Toxicity Studies

In vitro Cytotoxic Evaluation of Antibiotics and Antibiotics-Loaded CS-AgNPs

The cytotoxic potential of antibiotics-loaded CS-AgNPs was evaluated in triplicate using the HeLa cell line (ATCC, Virginia, USA) assay. The percentage viability of cells was determined using MTT (3 (4,5 dimethyl thiazol 2) 2.5 diphenyltetrazoliu bromk assay, as reported in previous studies.^{21,22} or the determination of IC_{50} , all antibiotics in the concentrations range of 0.016– 1024 µg/mL were used whereas CS-X NV conjugated with AMX, CEF, and X, we applied in the concentrathen espective FICs (defined in tions according Table 3). Brief prior to the involution for 24 h, each test solution was a red separately in an adherent culture medium Then, the centwere washed using phosphatebuff ed saline (PBS) and again incubated for 30 min at root temperature in MTT reagent (1 mg/mL). After incubatic cell vial ity was assessed using a differential interfere. trast microscope also with the help of ectrophotometer at 570 nm and the growth inhibi- \mathbf{U} on was expressed in percentages.

iocompatibility Index (BI) of CS-AgNPs, Antibiotics and Antibiotics-Loaded CS-AgNPs

The biocompatibility index is defined as the ratio of IC_{50} values of test solution determined on HeLa cell line and concentration of test solution produced 3 log_{10} reductions of bacterial growth (99%). The BI values higher than "1" demonstrates that test solution has more bactericidal potential and low cytotoxicity, while lower than "1" indicates low bactericidal potential with high cytotoxic activity.³² In this study, BI was determined using MBC values at which a 99.9% reduction in bacterial growth was observed.

In vivo Acute- and Sub-Toxicity Studies of CS-AgNPs Study Animals

For in vivo toxicity studies on an animal model, ethical approvals were obtained from the Institutional Bioethics Committee (IBC) of the University of Karachi, Karachi, Pakistan, and also from Institutional Review Board (IRB) of the Jinnah Sindh Medical University with the reference number of JSMU/IRB/2019/286. Healthy Sprague dawley rats (10–12 weeks old) of both genders were purchased

from the animal house of Dow University of Health Sciences. Animals were kept in a standard environment (45–55% humidity and $25^{\circ}C \pm 2^{\circ}C$) with a 12 h light-dark cycle.³³ They were divided into four groups, with 10 rats in each. All administrations were performed orally, and tested solutions were given once daily. Normal saline was given to 1st group (control) while 2nd, 3rd, and 4th test groups received 30 mg/kg, 60 mg/kg, and 90 mg/kg CS-AgNPs for 28 days, respectively. The test doses were adapted based on a preliminary acute toxicity study where a lethal dose (LD₅₀) was found to be >200 mg/kgusing a staged approach to the dosing method.³⁴ T-61 agent was administered IV as euthanasia, while medetomidine was used for animal sedation.³⁵ Animal handling was performed according to the guidelines of the National Advisory Committee for Laboratory Animal Research (NACLAR). The toxicity studies were performed according to the Organization for Economic Corporation (OECD) guidelines no 407 and 425.³⁶

Clinical Examinations, Body Weight, and Relative Organ Weights

Initially, all test animals were evaluated for generalized well-being. After the dosing of CS-AgNP, the vitation of any sign of toxic effects were recorded twice willy. Treatment effects on animal general health, behaver hairs, and skin were observed. The body weight of ear animal was noted on initial then after ontinued resing for 28 days. Organ weights such as the brancheart, kidney, liver, and lungs were measured and relative sign weights were calculated based on that tota body weight.

Hematological and Prochemical Analysis

Animals were and hetize and then sacrificed after the last dosing. A plume 5 mL of lood samples from each rom the fertoral artery into the 20 mg/ rat was c rected mL et lenediz instetraacetic acid (EDTA), which is used as a a accoagulant for hematological and biochemical analysis. lood samples were analyzed for the counts of red blood cells (RBCs), white blood cells (WBCs), and differential WBCs using an automated blood sample analyzer (Beckman Coulter, U.S). Hemoglobin levels (Hb levels), serum electrolytes, erythrocyte sedimentation rate (ESR), and cholesterol levels were also estimated. The auto analyzer (7600-110, Hitachi, Japan) was used for biochemical analysis related to enzymatic levels of heart, liver, and kidney. Different inflammatory biomarkers, including C-reactive proteins, interleukins (IL)-1, IL-2,

IL-6, IL-10, and tumor necrosis factor-alpha (TNF- α) were determined by MAGPIX (R&D Systems) in accordance with the manufacturer's instructions provided on commercial assay kits.^{33,37}

Histopathological Examination

Tissues of heart, kidney, and liver were fixed in formalin (10%), and tissues were handled using standard laboratory techniques recommended for histopathological examinations. Tissue cleaning was done using a xylene solution (1%) for 1 h. Then, 3 to 5 μ m sections were sliced using a microtome and stained with banatoxy be-eosin (H&E) stain according to the standard protocol for taining.³⁸

Determination of Silver Ica Concentration in Zissues

Silver ion concentration was determined in different organs after we administed in of CS-AgNPs in similar dosing schedulet used for toxicity studies according to the management by use et al in 2013.³⁷ The animals were acrificed after 28 days, and tissues were digested in nitric cid using a multiwave microwave digester (Anton Paar, UnA). The atomic absorption (AA) spectrophotometer (AA-7000, Shimadzu, Japan) equipped with a graphite respective (Elmer, USA) was used for the Ag ion determination. The Ag ion concentration in each tissue was calculated in $\mu g/g$ wet weight. The limit of detection (LOD) and the limit of quantification (LOQ) were found to be 0.29 $\mu g/kg$ and 0.88 $\mu g/kg$, respectively, in the quantification method.

Statistical Analysis

All results are presented as their mean \pm S.D values. The synergistic antibacterial activity, in vitro and in vivo toxicity findings, and silver ion determination were subjected to analysis of variance (ANOVA), two-tailed *t*-test, and Tukey post hoc tests using SPSS software (version 23). P< 0.05 and P< 0.005 were considered statistically significant and highly statistically significant results, respectively.

Results and Discussion Antibiotics Loading on Synthesized Chitosan-Loaded Silver Nanoparticles (CS-AgNPs) Surfaces

Amount and loading efficiencies of AMX, CEF, and LVX to CS-AgNPs were found to be 2.5 μ g/ μ g (93.3%), 0.5 μ g/ μ g (85.8%), and 0.5 μ g/ μ g (82.0%) respectively, using

UV-Vis spectrometric analysis. UV-vis absorption peaks for AMX were obtained at 337 nm, for CEF at 321 nm and LVX at 292 nm with CS-AgNPs (Figure S1). The peaks of these antibiotics at similar wavelengths have also been reported by other researchers.^{39–41}

Characterization and Interaction of Antibiotics Loaded on CS-AgNPs Surfaces

The size, shape, and agglomeration of synthesized CS-AgNPs conjugated with antibiotics were evaluated by SEM and TEM analysis. SEM and TEM images are presented in Figure S2 and S3, respectively. Images showed more agglomeration in antibiotics-loaded NPs compared to AgNPs and CS-AgNPs alone, which may be due to the antibiotics conjugation on NPs surfaces.²⁵ For AgNPs and CS-AgNPs, absorption peaks at 2922/cm (hydrocarbon chains), 1647/cm (C-O), and 1382/cm (C-O-H stretching vibrations) correspond to the presence of organic compounds that prevent agglomeration.42 FTIR spectra also confirmed the antibiotics were loading on CS-AgNPs surfaces, as reflected in Figure S4. Major absorption peaks observed at 3463/cm (N-H) and 3167/cm (O-H) wh stretching vibrations of carboxylic group and amid groups were observed at 1775/cm and 1685/cm espectively, in CS-AgNPs+AMX spectra, indicating the njugation of AMX with CS-AgNPs. How voofa reported similar absorption peaks ju ation and the se e.⁴³ IR s characterization study of AMX stra of CS-AgNPs+CEF displays the absorption peak of NH₂ at 3260/cm and at 1665/cm for the amide group while the absorption band for the carboxylic group at 1760/cm confirms the presence of CEF on CS-AgNPs surfaces.44 Characteristics peaks at 1730/cm (C=O), 1888/cm (quinolone moiety), and 2940/cm (aromatic stretching) were observed for LVX in CS-AgNPs+LVX spectra corresponded with the absorption peaks of LVX reported by Khan et al in 2016.45 EDX technique was used for the localized detection of silver metal in antibiotics-loaded CS-AgNPs. In Figure S5 absorption peaks at 3 KeV, confirmed the presence of sirer ion hantibioticsloaded NPs in all EDX spectra.⁴ be size distribution and zeta potential of the AgNP CS-Ag Ps, and htibioticsloaded CS-AgNPs were retermined by The particle size distribution curve how that obtained AgNPs and CS-AgNPs were conodisp, sed with the mean size ranges of 10–20 av 9–40 nm, ectively (Figure S6). However, polydisperion was observed in antibiotics-conjugate composite with an increase in mean size ranges from 80 to 120 nm, which might be due to the presence of non-secific binding and aggregation between antibutics and $\sum_{n=1}^{46}$ The measurements of zeta potential evealed the negative charge of green synthesized AgNPs AgNPs with the mean potential of -23.6 and ap 16.1 mV, respectively (Figure S7). The shifting of surace charges toward positivity was observed after conjugaon with AMX and LVX with the mean zeta potential of -7.9 and -19.4 mV, respectively. Aggregation between antibiotics and AgNPs might be observed due to these

Antibacterial Agents	Zue of Inhibitio				
	çoʻ	K. pneumoniae	S. aureus	S. typhi	P. aeruginosa
Control (D. Ved wate	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00
SAEE	7.1 ± 0.57	6.3 ± 0.38	6.2 ± 0.17	6.2 ± 0.36	8.3 ± 0.74
CS	9.6 ± 0.32	9.4 ± 0.47	8.8 ± 0.26	7.1 ± 0.20	11.4 ± 0.39
AgNPs	13.2 ± 0.41*	12.3 ± 0.52*	11.9 ± 0.22	9.4 ± 0.47	13.1 ± 0.68*
CS-AgNPs	16.1 ± 0.83*	15.2 ± 0.16*	16.5 ± 0.54*	15.3 ± 0.23*	18.4 ± 0.30*
AMX	11.4 ± 0.23	6.3 ± 0.36	9.8 ± 0.19	12.8 ± 0.61*	11.8 ± 0.57
CS-AgNPs + AMX	21.5 ± 1.03**	19.7 ± 0.67**	23.6 ± 1.57**	24.5 ± 0.90**	26.2 ± 1.53**
CEF	14.5 ± 0.54*	17.6 ± 0.71*	7.1 ± 0.15	14.8 ± 0.77*	10.5 ± 0.24
CS-AgNPs + CEF	23.4 ± 0.35**	26.8 ± 1.32**	19.7 ± 0.97**	23.5 ± 1.57**	21.9 ± 1.07**
LVX	16.0 ± 0.64*	13.0 ± 0.74*	15.3 ± 0.83*	16.0 ± 0.24*	17.9 ± 0.62*
CS-AgNPs + LVX	25.5 ± 0.49**	27.7 ± 1.26**	29.4 ± 2.18**	31.7 ± 2.80**	32.5 ± 2.60**

Table I Zone of Inhibitions (as) of Different bibacterial Agents Against Clinical Isolates

Notes: All experiments were performed in triplicates and reported as mean \pm SD. * $p \leq 0.05$ significant as compared to control, ** $p \leq 0.005$ highly significant as compared to control.

Abbreviations: S.D, standard deviation; SAEE, Syzygium aromaticum ethanolic extract; CS, chitosan; AMX, amoxicillin; CEF, cefixime; LVX, levofloxacin.

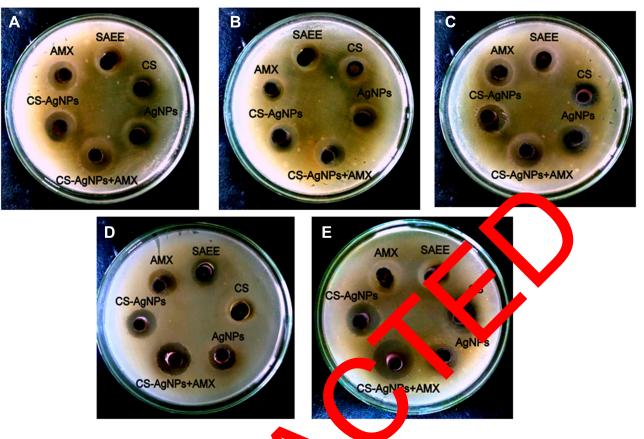


Figure I Antibacterial activities of CS-AgNP with amoxicillin against (A Ese Pseudomonas aeruginosa.

positive-negative charges attractions. However, the negative potential value after final conjugation 4.4 mV supports good consider name, long-term stability and high dispersity on pocomposite 4^{46}

UV/Vis spectroscopy was used to prestigate the interaction between CS-A Ps and individe antibiotic molecules. Upon addition of AV, a slightly decreased in the extinction at 410 n. S-AgN in the entire concen-5 to 900 A was observed, without tration ran signific a aggingation synthesized CS-AgNPs. However, the proadour, dominates the spectrum when CEF or LV concentrations gradually increased, and the aggregation of AgNPs was observed at 500 and 5000 μ M after the addition of LVX and CEF, respectively, as shown in Figure S8. This means that both antibiotics formed complexes physically with CS-AgNPs resulted in aggregation of the CS-AgNPs. This spectral information from UV-Vis spectrum clearly demonstrate that CEF and LVX interacted physically with CS-AgNPs strongly, replacing the citrate molecules on surface and forming antibiotic-CS-AgNPs complexes.²⁶ Moreover, these antibiotics

a coli; (**B**) Neosiella pneumoniae; (**C**) Staphylococcus aureus; (**D**) Salmonella typhi; (**E**)

readily caused aggregation of CS-AgNPs at higher concentrations, while no such aggregation was observed for AMX. However, the molecular nature of the interaction between the CS-AgNPs and the different antibiotics needs further investigation.

Antibacterial Activity of AgNPs, CS-AgNPs and Antibiotic-Loaded CS-AgNPs

The antibacterial activity of SAEE, CS, AgNPs, CS-AgNPs, and antibiotics-loaded CS-AgNPs were evaluated against five clinical pathogens using the Oxford cup diffusion method. Table 1 and Figures 1–3 show the zone of inhibitions produced by tested antibacterial solutions against the selected clinical isolates. Amongst all pathogenic strains, CS-AgNPs exhibited a maximum antibacterial effect against *Pseudomonas aeruginosa* (18.4 \pm 0.30 mm), and its conjugation with LVX, produced more augmented response (32.5 \pm 2.60 mm). A similar trend was observed with other antibiotics-conjugated with CS-AgNPs. The antibacterial mechanism of metal NPs is still not identified, but few studies have reported the plausible

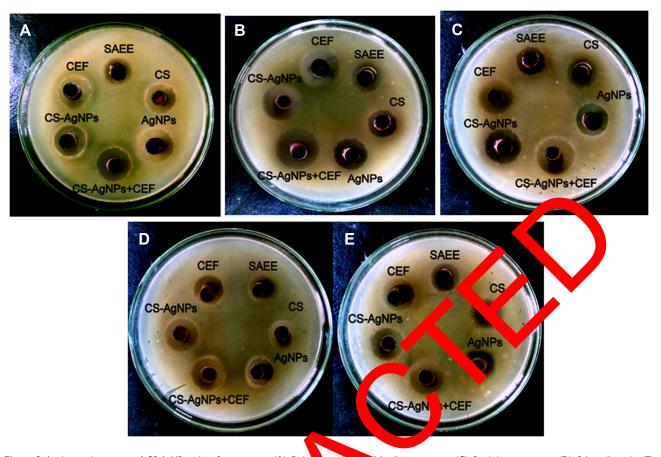


Figure 2 Antibacterial activities of CS-AgNP with cefixime against (A) Escherenia coli; the bsiella pneumoniae; (C) Staphylococcus aureus; (D) Salmonella typhi; (E) Pseudomonas aeruginosa.

mechanism like Reidy et al (2013) have suggested that metal NPs have more significant antibiotent, effects due to their high affinity with the subace-active broups of microbial strains.⁴⁷ Wypij et al. (2018) and Raceet al (2012) have reported that the released betal ions from NPs can rapture the bacterial cell wall and our cause cell death.^{21,48} Moreover, the distortion of bacterial DNA helical structure by a metal houses also been reported.^{49,50} In addition, chiterian has been beggined to react with both the bacteria cell wall and the cell membrane.⁵¹

All the analysis accombined with CS-AgNPs showed enhanced antibact vial potential against all tested pathogens ($p \le 0.003$). The total increase in zones is reflected in Figure 4. Interestingly, antibiotics like AMX and CEF showed minimal susceptibility, when conjugated with CS-AgNPs exhibited potential antibacterial activity against *Klebsiella pneumoniae* and *Staphylococcus aureus* respectively after conjugation as reflected in Table 1. CEF showed the highest fold increase in its activity against *Staphylococcus aureus* (1.8) while AMX and LVX showed 2.2 and 1.2 fold increase in their activity, respectively, AgNPs. Different researchers have also been reported the synergistic activity of antibiotics when they were combined with AgNPs, synthesized by different methods.^{18,22,52}

CS-AgNPs exhibited low values of MICs against all tested isolates (32 μ g/mL) compared to AgNPs (64 μ g/ mL) as presented in Table 2. The results of MIC assays after conjugation are given in Table 3. The MICs of all antibiotics reduced considerably when combined with CS-AgNPs against all pathogenic strains, and maximum reduction in MIC was observed for AMX from 1024 to 64 and 32 µg/mL against Escherichia coli and Klebsiella pneumoniae, respectively. The FIC values ranging from 0.12 to 0.25 for all antibiotics conjugations against the tested isolates demonstrated their synergistic activity (Table 3). The previous study reported the comparable synergistic response of synthesized AgNPs with different antibiotics.²¹ In the present work, the MBC value of CS-AgNPs was found to be 32 µg/mL against all tested isolates (Table 4). However, these values reduced drastically after conjugation with different antibiotics ranging from 32 to 4 µg/mL.

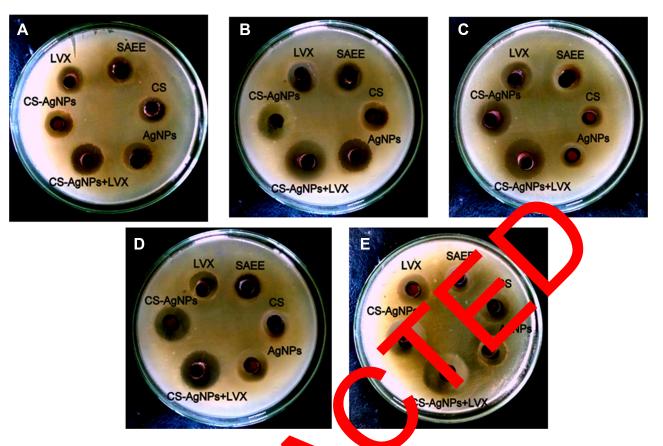


Figure 3 Antibacterial activities of CS-AgNP with levofloxacin against Pseudomonas aeruginosa.

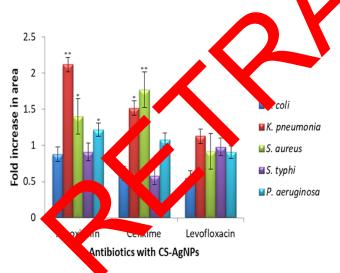


Figure 4 Fold increase in inhibition zones of antibiotics after conjugation with CS-AgNPs against tested isolates. All experiments were performed in triplicates and reported as mean ± SD. * $p \le 0.05$ significant as compared to control (considered as no increase in activity), ** $p \le 0.005$ highly significant as compared to control.

Mechanism of Synergistic Effects

Silver nanoparticles (AgNPs) exhibit very distinctive physicochemical characteristics, and tremendous antibacterial activity, which highly recommend them as an alternative treatment against multi-drug resistant bacteria.^{53,54}

hia coli; (B) Neosiella pneumoniae; (C) Staphylococcus aureus; (D) Salmonella typhi; (E)

Biogenic silver nanoparticles reported showing greater superbugs.55 activity against these antibiofilm Nanocomposites of silver with other metals observed the disruption of biofilm structure and penetration of metal ions into under layers of bacterial colony.⁵⁶ Chitosan conjugated nanocomposites have profound antibacterial activity, due to the presence of an ammonia group in chitosan. Ammonia adsorbs on to the cell wall electrostatically and potentiates the destruction of cell wall by causing leaking of macromolecules from the bacterial cell.^{12,57} It is reported that synergistic antibacterial activity of chitosan with AgNPs may be resulted due to blistering (blebs), blockage of the electron transport chain, and clumping of membranes.⁵⁴ In addition, chemical interaction as the possible cause of the synergistic rise in antibacterial activity of the synthesized AgNPs with antibiotic.^{31,58} However, it was also postulated that the combined effect of antimicrobials drives synergy by membrane alterations generated by AgNPs, and no chemical interactions were detected between AgNPs and antibiotics.⁵⁹ Moreover, it was also reported that amino and hydroxyl groups present in tested

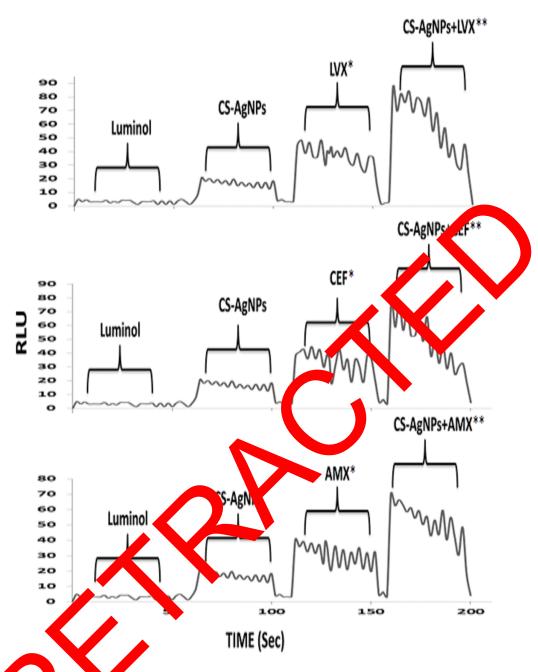


Figure 5 CS-AgNP associate peneration. He roxyl radicals in the presence of different antibacterial agent. *The chemiluminescence of luminol (stage 1) in the presence of CS-AgNPs (spee 2) antibic (stage 3) CS-agNPs combined with different antibiotics (stage 4). **RLU, relative luminescence units; AMX, amoxicillin; CEF, cefixime; LVX, levofloxacin. All control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol), ** $p \le 0.005$ highly significant as compared to control (luminol) (lumino

antibiotics bind with CS-AgNPs by chelation, which may maximize agglomeration and increase antibacterial activity.¹⁹ Kohanski et al (2007) suggested that production of hydroxyl radicals is an intriguing bacterial killing mechanism of several antibiotics.⁶⁰ Consistent with reported studies, we tested whether or not bacterial cells treated with AgNPs generated hydroxyl radicals and how this affected the synergistic effects. In the luminol model, synthesized CS-AgNPs generated hydroxyl radicals as shown in Figure 5. All studied bactericidal antibiotics, caused the formation of hydroxyl radicals. However, treatment with combinations of CS-AgNPs and antibiotics showed increased hydroxyl radical formation compared with each antibacterial agent alone. Results indicated a significant increase in the generation of hydroxyl radicals, which might be an important cause

Table 2 Minimum Inhibitory Concentrations (MICs) of Different Antibacterial Agents Against Clinical Isolates

Antibacterial Agents	Minimum Inhibi (Mean ± S.D)	Minimum Inhibitory Concentrations (µg/mL) (Mean ± S.D)							
	E. coli	K. pneumoniae	S. aureus	S. typhi	P. aeruginosa				
SAEE	7500 ± 245.3	7500 ± 211.4	7500 ± 254.2	7500 ± 287.2	7500 ± 241.2				
CS	256 ± 14.7	256 ± 15.5	256 ± 26.3	256 ± 17.5	256 ± 18.4				
AgNPs	64 ± 9.4	64 ± 14.5	64 ± 8.4	64 ± 5.1	64 ± 9.3				
CS-AgNPs	32 ± 3.4	32 ± 11.4	32 ± 6.5	32 ± 4.1	32 ± 7.3				
AMX	1024 ± 141.2	1024 ± 102.4	256 ± 22.5	256 ± 29.4	512 ± 64.7				
CEF	256 ± 12.4	128 ± 26.4	256 ± 22.4	32 ± 3.0	256 ± 23.4				
LVX	16 ± 2.3	64 ± 7.9	128 ± 24.8	I ± 0.12	1024 ± 126.4				

Note: All experiments were performed in triplicates and reported as mean ± SD.

Abbreviations: S.D, standard deviation; SAEE, Syzygium aromaticum ethanolic extract; CS, chitosan; AMX, amoxicillin; CEF, cettere; LVX, levoflo

 Table 3 Fractional Inhibitory Concentration (FIC) Index and Minimum Inhibitory Concentrations (MICs) Applotics-Conjugated CS-AgNPs Against Clinical Isolates

Clinical Isolates	Fractional Inhibitory Concentration (FIC) Index							
	CS-A	gNPs + AMX	CS-A	gNPs + CEF		gNPs + LVX		
	FIC	MIC of CS-AgNPs + AMX	FIC	MIC of CS-AgNPs	FIC	MIC of CS-AgNPs + LVX		
E. coli	0.12	4 + 64	0.18	+ 32	0.12	4 + 1		
K. pneumoniae	0.15	8 + 32	0.12	+ 8	0.18	8 + 4		
S. aureus	0.18	4 + 16	0.18	16	0.15	4 + 4		
S. typhi	0.18	4 + 16	0.18	4 -	0.12	2 + 0.0625		
P. aeruginosa	0.18	4 + 32	0	4 + 32	0.25	4 + 128		

Abbreviations: AMX, amoxicillin; CEF, cefixime; LVX, levofloxacin.

 Table 4 Biocompatibility Index (BI) and Mimum Borericidal Concentration (MBC) Values of Antibiotics of Different Antibacterial Agents Against Clinical Isolates

Antibacterial Agents	Biocompute ility Index (E										
	E. coli		K. pneun	noniae	S. aureus	5	S. typhi		P. aerugi	nosa	IC ₅₀
		мвс		мвс	BI	мвс	ві	мвс	BI	мвс	(µg/mL)
AgNPs	1.6	64	1.6	64	1.6	64	1.6	32	1.6	64	102.7*
CS-AgNPs		32	3.9	32	3.9	32	3.9	32	3.9	32	125.7*
AMX	0.	J24	0.1	1024	0.7	256	0.3	512	0.1	1024	194.6
CS-AgNP AMX	5.2+2	16+64	5.2+1.4	16+128	5.2+1.4	16+128	10.4+2.8	8+64	5.2+2.8	16+64	84.4+180.1
CEF	03	512	0.7	256	0.3	512	2.8	64	0.3	512	179.3
CS-AgNPs	4.9+2.4	16+64	4.9+9.9	16+16	4.9+2.4	16+64	9.9+39.6	8+4	4.9+2.4	16+64	79.8+158.4
LVX	5.2	32	1.3	128	0.6	256	83.4	2	0.16	1024	166.8
CS-AgNPs + LVX	8.1+37.4	8+4	2.0+2.3	32+64	8.1+4.6	8+32	16.2+149.9	4+1	4.0+1.1	16+128	65.0+149.9

Note: *These IC_{50} values were taken from our previous study [Muhammad Arif et al 2020].

Abbreviations: MBC, minimum bactericidal concentration; IC₅₀, inhibitory concentration; AMX, amoxicillin; CEF, cefixime; LVX, levofloxacin.

of the synergism seen. It is suggested that oxidative stress caused by AgNPs through increase in radicals production leads to damage of the nucleic acids and proteins, and consequently inhibition of proliferative processes in bacterial cells.⁵⁵ Researchers confirmed that AgNPs

produced different toxic radicals including hydrogen peroxide (H₂O₂) and hydroxyl radicals (OH).^{61,62} Moreover, results of physical interaction investigation also pointed out that the bacterial cells were killed more effectively by "CS-AgNPs-antibiotic complexes", which is in consistent

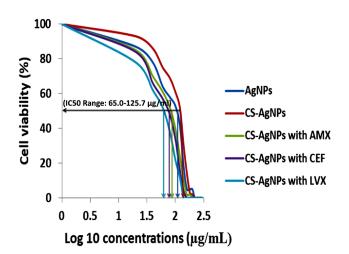


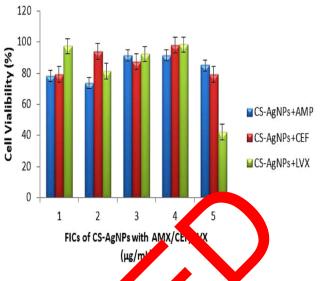
Figure 6 % viability of HeLa cells against AgNPs, CS-AgNPs alone, and combination with different antibiotics while IC_{50} of each agent are given in Table 4. All experiments were performed in triplicates and reported as mean.

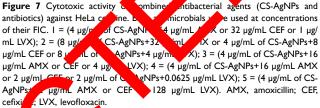
with previous study.²⁶ However, the exact molecular mechanism by which increase in production of radical species, still requires further studies.

Toxicity Studies

In vitro Cytotoxic Evaluation of Antibiotics-Loaded CS-AgNPs

The antibiotics-loaded CS-AgNPs were subjected to evaluate the cytotoxicity on the HeLa cell lines. The dosedependent effects of all tested solutions were observed against eukaryotic cells. The IC₅₀ values f AMV CFF and LVX was found to be 194, 17, and the μ g/mL, respectively (Table 4). The percentine viability of HeLa





cells t differer concentrations of AgNPs, CS-AgNPs alone and combination with different antibiotics is als presented in Figure 6. In our previous work, IC_{50} alue of CS-AgNPs was reported as 125 µg/mL.²³ In the surrent study, IC_{50} values of CS-AgNPs were slightly decreased after conjugation with different antibiotics. However, a considerable reduction in MICs and FICs (Tables 2 and 3, Figure 7), suggested the dose reduction

 Table 5 Body Weight and Positive Major Organ
 Veights of Both Male and Female Rats After 28th Day's Exposure of CS-AgNPs at Different Concentrations

Groups	کی Weight Gain (ارتباع ± ۲۰۰۰)	Relative Major Organ Weights (% ± S.D) (After 28th Days)							
	(After oth Days)	Brain	Heart	Kidneys	Liver	Lungs			
Male									
Control	24.4 ± 4.2	0.61 ± 0.04	0.36 ± 0.01	1.02 ± 0.09	4.33 ± 0.23	0.47 ± 0.07			
CS-AgNPs (30 mg/kg	22.8 ± 3.0	0.58 ± 0.02	0.34 ± 0.03	0.98 ± 0.12	4.57 ± 0.25	0.43 ± 0.09			
CS-AgNPs (60 mg/kg)	19.7 ± 8.1	0.60 ± 0.04	0.33 ± 0.03	0.93 ± 0.06	4.66 ± 0.37	0.52 ± 0.10			
CS-AgNPs (90 mg/kg)	18.1 ± 7.1	0.63 ± 0.01	0.35 ± 0.02	1.01 ± 0.10	4.72 ± 0.20	0.45 ± 0.05			
Female									
Control	18.9 ± 5.7	0.53 ± 0.03	0.38 ± 0.02	1.04 ± 0.05	4.12 ± 0.18	0.52 ± 0.04			
CS-AgNPs (30 mg/kg)	19.3 ± 5.3	0.51 ± 0.05	0.36 ± 0.01	1.07 ± 0.04	4.41 ± 0.13	0.48 ± 0.10			
CS-AgNPs (60 mg/kg)	20.4 ± 6.9	0.58 ± 0.07	0.30 ± 0.04	1.02 ± 0.11	4.37 ± 0.40	0.43 ± 0.02			
CS-AgNPs (90 mg/kg)	19.0 ± 5.4	0.55 ± 0.02	0.36 ± 0.03	0.99 ± 0.07	4.48 ± 0.20	0.51 ± 0.07			

Note: All experiments were performed in triplicates and reported as percentage \pm SD. **Abbreviation:** S.D, standard deviation (n=10).

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Parameters	Groups* (Mean ± S.D)	± s.D)						
	Male				Female			
•	Life	CS-AgNPs (3 ⁰ mg/kg)	CS-AgNPs (60 mg/kg)	CS-AgNPs (90 mg/kg)	Control	CS-AgNPs (30 mg/kg)	CS-AgNPs (60 mg/kg)	CS-AgNPs (90 mg/kg)
Serum electrolytes								
Na ⁺ (mEq/L)	135.4 ± .54	2	I 34.6 ± 8.46	142.2 ± 11.57	137.5 ± 6.91	134.8 ± 8.43	I38.7 ± 6.9I	138.2 ± 7.73
K ⁺ (mEq/L)	5.64 ± 0.78	6.0. ± 0.61	5.89 ± 0.56	4.94 ± 1.08	5.24 ± 0.59	5.83 ± 0.73	6.09 ± 0.48	5.63 ± 0.77
Ca' (meq/L)	0./U ± 82.č	5.4I ± 0.82	1.32 ± 0.48	6.08 ± 1.12	5.06 ± 0.43	5.3/ ± 0.60	11.0 ± 7.5	5.30 ± 0.35
Blood profile								
RBCs (10 ¹² /L)	8.42 ± 0.63	8.17 ± 0.5	8.35 - 12	8.09 ± 0.72	7.97 ± 0.51	7.99 ± 0.70	7.83 ± 0.62	7.70 ± 0.48
WBCs (10 ⁹ /L)	13,485 ± 3482	14,834 4870	15 / ± / 4	15,873 ± 8340*	11,540 ± 4017	12,604 ± 5137	13,034 ± 5379	13,974 ± 7842*
Basophils (%)	0.00 ± 0.00	0.00 ± 0.00	± 0.00	$0.60 \pm 0.04^{*}$	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00	0.60 ± 0.06*
Eosinophils (%)	I.87 ± 0.20	1.44 ± 0.17	10	$2.87 \pm 0.38^{*}$	2.49 ± 0.14	1.72 ± 0.28	2.92 ± 0.36	3.47 ± 0.24*
Lymphocytes (%)	75.89 ± 6.87	78.07 ± 8.16	81.56 11.36	89.94 ± 12.57*	78.15 ± 7.46	80.17 ± 5.42	79.05 ± 8.47	91.53 ± 9.91*
Monocytes (%)	I.45 ± 0.27	I.48 ± 0.14	1.51 0.23		1.37 ± 0.34	I.53 ± 0.39	1.46 ± 0.16	2.59 ± 0.41
Neutrophils (%)	22.41 ± 5.98	24.73 ± 4.21	29.4 ± 5 PC	3/ 9 ± 9.40*	19.86 ± 4.76	22.33 ± 5.91	26.74 ± 4.37	37.08 ± 5.10*
Hb levels (g/dL)	14.98 ± 0.94	I4.56 ± 0.85	15.20 ± 0.98	12 ± 1.12	13.47 ± 0.57	13.82 ± 0.97	13.27 ± 0.62	14.01 ± 0.83
ESR (mm/h)	1.42 ± 0.27	I.48 ± 0.43	1.43 ± 0.69	.54 ± ∩	1.53 ± 0.18	I.46 ± 0.37	I.53 ± 0.56	I.58 ± 0.39
Liver profile			-					
ALP (U/L)	218.47 ± 28.7	285.41 ± 87.6	312.56 ± 74.0*	44 38 ± 102.5**	27 60 ± 25.3	267.38 ± 69.3	298.19 ± 37.6	430.41 ± 58.3**
AST (U/L)	235.72 ± 39.4	312.40 ± 28.1	$378.37 \pm 56.8^{*}$	439. * ± 67.5**	203 + 24.3	240.37 ± 37.6	311.98 ± 34.2*	381.09 ± 43.1**
ALT (U/L)	43.21 ± 18.27	59.79 ± 14.08	62.13 ± 17.25*	81.19 ±	52.1 - 31	61.08 ± 12.40	$68.97 \pm 20.73^*$	75.37 ± 19.33**
Bilirubin (mg/dL)	0.68 ± 0.14	0.52 ± 0.17	0.81 ± 0.13*	0.99 ± 0.10**	6 ± 0.24	0.64 ± 0.42	0.73 ± 0.37*	$0.92 \pm 0.29^{**}$
Kidney profile								
Creatinine (mg/dL)	1.87 ± 0.24	1.82 ± 0.14	1.85 ± 0.21	1.86 ± 0.17	1.13 ±	n ± 0.24	1.21 ± 0.11	1.18 ± 0.29
BUN (mg/dL)	28.70 ± 2.41	27.21 ± 3.87	29.05 ± 3.28	25.96 ± 4.84	26.2 ± 3.19	26.97 ± 2.72	28.33 ± 3.94	27.94 ± 3.48
Uric acid (mg/dL)	1.71 ± 0.14	I.68 ± 0.08	1.73 ± 0.10	I.74 ± 0.19	1.67 9.18	1.70	I.68 ± 0.32	1.66 ± 0.24
Cardiac Profile								
CK-MB (IU/mL)	12.84 ± 0.26	12.06 ± 0.19	11.97 ± 0.51	12.15 ± 0.18	13.10 ± 0.62	.2.42 ± 0.36	13.18 ± 0.47	12.52 ± 0.29
Cholesterol (mg/dL)	72.84 ± 12.74	71.68 ± 14.87	72.62 ± 13.87	70.02 ± 17.45	77.67 ± 10.67	79.34 ± 12.1	78.04 ± 14.93	78.90 ± 9.82
Inflammatory biomarkers								
CRP (mg/dL)	60.45 ± 4.58	66.91 ± 7.48	69.73 ± 5.81	86.13 ± 7.34*	57.33 ± 5.39	60.90 ± 6.53	64.74 ± 5.52	83.62 ± 7.67*
IL-I (pg/mL)	38.78 ± 3.34	41.37 ± 4.67	46.85 ± 4.65	59.62 ± 5.32	37.99 ± 4.25	43.75 ± 4.76	48.92 ± 6.78	57.67 ± 4.34*
IL-2 (pg/mL)	22.14 ± 3.45	26.22 ± 4.87	31.58 ± 5.82	39.32 ± 3.66*	26.85 ± 5.24	31.48 ± 4.10	35.34 ± 6.08	45.08 ± 6.84*

with minimal potential toxicity, which is corresponded with a previous study.²¹ Researchers claimed that the combination of antibiotics with silver nanoparticles restored the antibacterial potential of these agents against resistance acquired bacteria along with the reduction in toxicity.^{18,22} These observations are also in agreement with the present study, where the BI values after conjugation with antibiotics were found more than 1, expressing more bactericidal activity with low toxicity (Table 4).

In vivo Acute and Sub-Acute Toxicity Studies of CS-AgNPs

Initially, the acute oral toxicity exynthesized S-AgNPs was determined for 14 days to sign cant sign f toxicity was observed in studied animals after 14 days of oral dosing of CS-AgNPs. he LD, value of CS-AgNPs was found to be >200 g/kg, h mbled p vious findings.⁶³ In no signing of manges were observed the sub-acute at any dose of CS-. NPs in both male and female body weights relative on weights compared to control s (P<0.05) Table 5. All studied groups animals surgrov viv until the enhanasia was performed for further stu-Turthermo , no sign of adverse effects or any dies. oticed during the study period of 28 days. infection Le also reported similar observations after the treatent of rats with AgNPs at concentrations of 500 mg/kg.³⁷ In the present work, the effect of gender was not oserved in the hematological and biochemical analysis of Sprague Dawley rats. At the dose of 90 mg/kg, synthesized CS-AgNPs significantly increased the counts of WBCs, basophils, eosinophils, lymphocytes, and neutrophils compared to control at the significance level of P<0.05 (Table 6). There were no significant changes observed in other hematological parameters of studied groups after 28 days of oral dosing. Moreover, serum levels of sodium, potassium, and calcium were found in normal ranges during sub-acute toxicity studies compared to the control group.

The serum biochemical analyses are a widely used tool to assess responses in animals induced by different exogenous chemicals and toxins. These biochemical tests are also used to diagnose various diseases of the liver, kidney, heart, and other organs. The elevated levels of any serum biochemical parameter demonstrate any damaged or dysfunction of organs. After 28 days exposure of CS-AgNPs, significant (P<0.05) and highly significant (P<0.005) increased were observed in alkaline phosphatase (ALP), aspartate transaminase (AST), alanine transaminase

			CS-AgNPs (90 mø/kø)	89.97 ± 10.85* 23.53 ± 3.54*
			CS-AgNPs (60 mg/kg)	(3.89 ± 5.56 63.85 ± 4.36
	± S.D)		s <mark>9, _aA-2.</mark> S0 mølk	61.2 ± 5.98 61.2 ± 5.98 12.68 ± 3.53
	G s* (Mea	Male	Contro	59.44 ± 4.3 10.26 ± 3.21
Table 6 (Continued).	Parameters			IL-6 (pg/mL) IL-10 (pg/mL)
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erythrocyte sedimentation rate; ALP, alkaline phosphatase; AST, alkaline phosphatase; ALT, alanine transaminase; BUN,

 $**_{\mathcal{P}} \leq 0.005$ highly significant as compared to control

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÷ h (n=10). cells; ł

Notes: All experiments were performed in triplicates and reported as mea

blood urea nitrogen; CK-MB, creatine kinase-MB; CRP, C-reactive protein; Abbreviations: S.D, standard deviation; RBCs, red blood cells; WBCs,

34.95 ± 4.10

± 4.8I

9.45 ± 3.67

17.85 ± 2.78

TNF-a (pg/mL)

27.62 ± 5.95*

33.68 ± 5.87*

25.16 ± 4.88

87.85 ± 9.01*

69.12 ± 7.93 19.95 ± 3.27

± 6.53 I6.56 ± 3.62 21.98 ± 4.86

63.79 ±

13.27 ± 4.34 18.03 ± 2.87

58.54 ± 4.30

CS-AgNPs (90 mg/kg)

CS-AgNPs (60 mg/kg)

CS-AgNPs (30 mg/kg)

Control

Female

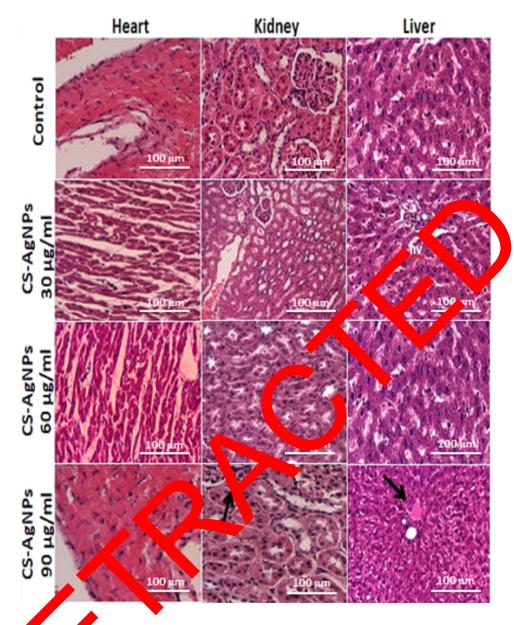


Figure 8 Histopathologics, traminations of heart, kidney and liver of male rats after administration of CS-AgNPs at different doses using the 400× magnification. Arrows in figures shows the minor tissue of mation at their doses of CS-AgNPs.

(ALT) and biling in levels, at the doses of 60 mg/kg and 90 mg/kg are rectivery. More than 50% increase in bilirubin levels can be demonstrated to hepatocellular injury.⁶⁴ The highly sign, cant increase in bilirubin levels at higher doses, implying that liver injury may have occurred. Increased levels of ALP, AST, and ALT also endorsed the possibility of hepatocellular injury. These enzymatic elevations, coupled with hyperbilirubinemia, may also be observed in cholestatic drug reactions.⁶⁵ This implies that orally administered CS-AgNPs at higher doses may cause hepatocellular toxic reactions similar to those caused by other drugs. The significant increase in levels of CRP, IL-

1, IL-2, IL-6, IL-10 and TNF- α at high doses also indicated the minor tissue inflammation in treated rats. The present observations are also consistent with previously reported results related to the sub-acute oral toxicity of AgNPs.^{37,63} In contrast, the normal levels of creatinine, blood urea nitrogen (BUN), uric acid, creatinine kinase (CK-MB), and cholesterol at all doses indicated that CS-AgNPs did not produce any toxic effects on renal and cardiac systems.

Histopathological examinations were performed to evaluate the effects of CS-AgNPs on body organs of male and female rats, ie, heart, kidney, and liver (Figures

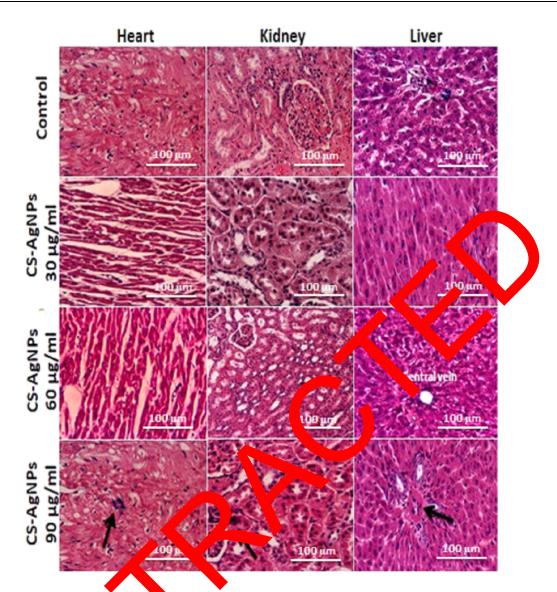


Figure 9 Histopathological examinations of heart, kidney Uiver of female rats after administration of CS-AgNPs at different doses using the 400× magnification. Arrows in figures shows the minor tissue influentation at higher dose of CS-AgNPs.

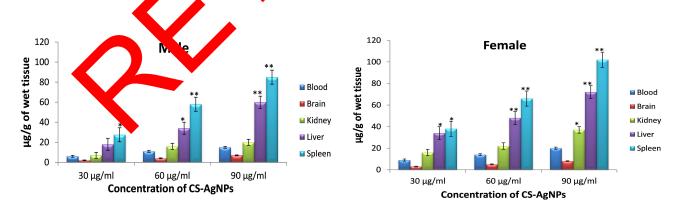


Figure 10 Silver concentrations (ng/g tissue wet weight) in different organs of both male and female after administration of CS-AgNPs. All experiments were performed in triplicates and reported as mean. * $p \le 0.05$ significant as compared to control (considered as zero concentration of silver ion), ** $p \le 0.005$ highly significant as compared to control.

8 and 9). Only mild histopathological changes, including inflammation and interstitial hyperemia, were observed only at high dose (90 mg/kg) in the heart, kidney, and liver when exposed to CS-AgNPs, compared to control in both gender groups. Moreover, there were no abnormal changes found at low (30 mg/kg) and medium (60 mg/kg) doses of CS-AgNPs in studied organs. Different researchers also reported the relevant histopathological examinations of AgNPs on different organs and animals at numerous dosing scheduled.^{34,63}

Determination of Silver lons in Tissues

The concentrations of Ag ion in blood, brain, kidney, liver, and spleen were determined by atomic absorption techniques following the oral administration of CS-AgNPs in rats at the above-defined three doses for 28 days. The concentrations of Ag ion compared to males with female rats at different concentrations of CS-AgNPs are presented in Figure 10. After CS-AgNPs administration, the Ag ion concentrations were highest in the spleen followed by liver, kidney, blood, and brain at all doses in both males and females. At 90 mg/ kg, the average levels of Ag ion in the spleen of male and female after 28 days were $85.2 \pm 7.8 \,\mu\text{g/g}$ and $102.5 \pm 9.3 \,\mu\text{g/}$ g, respectively, revealing gender difference. These f are in agreement with previous studies on AgNPs distribution to various tissues at different concentrations.^{37,63} Moreo all organs showed a gender-dependent c centra n of ion, with significantly high levels in male g apared to male rats following the 28 days a minist f of all three doses of CS-AgNPs. Kim et 2008) report two folds higher accumulation of Ag in heale compared to male rats.⁶⁶ Different researchers also reported similar findings while evaluated the ssue distribution of Ag ion given by different routes, which is suggested that tissue distribution of Ag ion was independent of the administration route.^{63,67} ares related to be specific mechanisms of the Further st gender lated d' cronce in the distribution of in vivo CS-AgNPs an ranted.

Conclusion

In the present study, the green synthesized and characterized CS-AgNPs were conjugated and physically interacted with different antibiotics by the simple centrifugation method. The stable functionalization of antibiotics on CS-AgNPs surfaces converses the resistance and increased antibacterial potential many folds against tested pathogenic isolates by radical scavenging potential. Combined therapy also decreases the therapeutic concentration and toxicity. In vivo toxicity studies in rats also confirmed the low toxicity of synthesized CS-AgNPs at a higher dose. Moreover, gender-related differences are found in tissue distribution of Ag ion that is an extensive distribution to the spleen and liver following oral administration. This particular study has pointed out that antibiotic resistance in clinical isolates can be reduced if combined with chitosan functionalized green synthesized silver nanoparticles.

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Disclosure

e authors report it conflicts of interest in this work.

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