Pure paclitaxel nanoparticles: preparation, characterization, and antitumor effect for human liver cancer SMMC-7721 cells

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Introduction: Pure paclitaxel nanoparticles (PPN), consisting entirely of drug molecules, were prepared by the electrostatic spraying method as promising candidates for antitumor application. Compared with the traditional preparation method, the advantage of the electrostatic spraying method included high production rates, relatively small particle sizes, and ease of preparation. Materials and methods: Paclitaxel was used to prepare PPN by electrostatic spray. The electrostatic spray device included a constant speed pump with a syringe, a high-voltage power supply, and a metal foil receiver was used to prepare and evaluate PPN. The syringe drew off a certain amount of paclitaxel chloroform solution (150 μg/mL) and was placed on the constant speed injection pump. The dissolution behavior of PPN was evaluated by dissolution test and the presence of paclitaxel in PPN was detected by X-Ray powder diffraction and differential scanning calorimetry. Effect of PPN on SMMC-7721 cells were studied by cell uptake, cell apoptosis and antitumor study. Results: The results of X-ray powder diffraction and differential scanning calorimeter characterization showed that the PPN were in an amorphous state. A dissolution study indicated that PPN have a significantly enhanced dissolution rate of paclitaxel. Moreover, SMMC-7721 tumor cells treated with PPN exhibited a distinctly high uptake rate that promoted cell apoptosis. An in vivo antitumor study demonstrated that PPN had significant antitumor efficacy. Conclusion: All conclusions verified that electrostatic spraying is a potential technology for developing PPN, and PPN can be regarded as a promising treatment for cancer.

Keywords: electrostatic spraying, paclitaxel, SMMC-7721 cells, poorly water soluble drug, dissolution

Introduction
Paclitaxel is a broad-spectrum anticancer drug that is clinically effective against lung, liver, ovarian, breast, and other cancers. However, its solubility is less than 0.1 μg/mL in water, which seriously affects its bioavailability.¹ Paclitaxel injection uses castor oil as the solvent for enhancing its solubility. Castor oil can induce hypersensitivity, and toxicity is inevitable.²,³ Paclitaxel for injection (albumin bound) has a low drug load (10%), although it has low toxicity and good tolerance. Thus, improving its water solubility and bioavailability with effective preparation technology is a worthy scientific endeavor.

At present, research about paclitaxel dosage forms include liposomes,⁴,⁵ polymeric micelles or nanoparticles,⁶,⁷ nanocrystallines,⁸–¹¹ mesoporous materials inclusion,¹²–¹⁴ and so on. The use of paclitaxel nanoparticles as a potential dosage form has attracted the attention of researchers. The solubilization principle shows that the particle size of an insoluble crystalline drug should be in the nanometer scale. Nanoparticles are made using anti-solvent methods, high-pressure homogenization, micronization...
technology, and so on, which increases the specific surface area and, thereby, the solubility and dissolution rate of insoluble drugs according to the Ostwald–Freundlich and Noyes–Whitney equation. Increasing the specific surface area of drug nanoparticles improves the contact between drug particles and biomembranes, and promotes the absorption of insoluble drugs. There are three technological methods for the preparation of paclitaxel nanocrystals: the first is the top–down type, which includes the medium grinding method, the homogenization method, and the microjet method; the second is the bottom–up type, which includes the precipitation method and the supercritical fluid method; the third is a combination of the first two techniques. These preparation technologies still present some challenges, including the complexity of the processes, their small scale, the poor reproducibility, and the high cost.

Electrostatic spray technology was studied in the early twentieth century. The principle is that as the surface charge density of droplets under a high-voltage electric field is increased, the surface is stretched and deformed, the droplets form a Taylor cone, and then the electric field makes the liquid flow along the Taylor cone and gradually produces a filament. This filament – under the multiple actions of the electric field force, surface tension, Kulun repulsive forces, and gravity – breaks into small droplets that, as a result of the repeated fragmentation process, gradually produce a series of smaller droplets, and these droplets in the space charge effect are forced to produce a spray. With the extension of time and distance, these droplets will become smaller and smaller. The electrostatic spray device comprises a constant speed pump with a syringe, a high-voltage power supply, and a metal foil receiver. It is applied to electrospray ionization mass spectrometry, pesticide spraying, and pharmaceutics. In pharmaceutics, the application of the technology is mainly focused on the preparation of polymer nanoparticles (such as poly(lactic-co-glycolic), lactose, ethyl cellulose, polyvinylpyrrolidone, etc.).

Recently, we used the electrostatic spray method to prepare pure paclitaxel nanoparticles (PPN). Compared with the current nanocrystal preparation technology, the use of electrostatic spray technology for the preparation of pure insoluble drug nanoparticles has not been reported. The advantages of the method lie in the following: 1) a single-step process can produce pure drug nanoparticles with controllable and uniform particle size; 2) the equipment is simple and easy to be industrialized; and 3) pure drug nanoparticles are present in an amorphous state. In addition, the antitumor effect of PPN was evaluated by in vitro cell experiments and tumor inhibition experiments in vivo.

Materials and methods
Materials
Paclitaxel bulk drug (PBD) was provided by the Tianfeng Biotechnology Company with a purity >99%. Tween 80, methanol, acetonitrile, and trichloromethane were purchased from Tianjin Yongsheng Fine Chemical Co., Ltd. (Tianjin, People’s Republic of China). SMMC-7721 cell lines, Tris-buffered saline Tween solution (TBST), PBS, RPMI 1640 medium, Annexin V-fluorescein isothiocyanate (FITC), MTT, propidium iodide (PI), fetal bovine serum (FBS), and trypsin were purchased from Shenyang Dingguo Biotechnology Co., Ltd. Paclitaxel injections (PIJ) were purchased from Beijing Union Pharmaceutical Factory.

Preparation and evaluation of PPN
The electrostatic spray device included a constant speed pump with a syringe, a high-voltage power supply, and a metal foil receiver. The syringe drew off a certain amount of paclitaxel chloroform solution (150 μg/mL) and was placed on the constant speed injection pump. The needle type (inner diameter) was 0.16 mm, the voltage was 40 kV, the receiving distance was 20 cm, and the injection rates were 50, 100, and 200 μL/h. The AC power was on, and aluminum foil received PPN. The PPN obtained were defined as PPN50, PPN100, and PPN200, respectively.

The content of paclitaxel in PPN were determined by high-performance liquid chromatography (HPLC; L-2400, Hitachi, Japan) equipped with a UV spectroscope (UV-2000, Unico, USA) at 227 nm. A Welch C18 column (4.6×250 mm, 5 μm) was used, the mobile phase was acetonitrile and water with a volume ratio of 50:50, and the flow rate was 1 mL/min. The content (C) of paclitaxel in PPN was calculated by the equation: 

\[ C = \frac{C_1}{1 + \frac{C_1}{C_2}} \]

Characterization of PPN
The particle size analysis of PPN was measured by laser diffraction using a Nano-ZS90 instrument from Malvern. The volume-weighted particle size distribution ranged from 0.040 μm to 2,000 μm. The dispersing liquid of the suspensions was 0.02% Tween-80 aqueous solution. The morphology of PPN was characterized by scanning electron microscopy (SEM; JSM-7001F, JEOL, Japan) and transmission electron microscopy (TEM; Tecnai G2F30, FEI, USA). The crystal transition or melting point of PPN was monitored by a differential scanning calorimetry (DSC; DSC-60, Shimadzu, Japan). The heating rate was 10°C/min under nitrogen. X-ray powder diffraction (XRD; Rigaku Ultima IV, Rigaku, Japan) was carried out to validate the solid state of paclitaxel. The step size was 0.02° and the scanning rate was 4°/min in a range (2θ) from 3°C to 60°C.
In vitro dissolution

A dissolution instrument (RC806D, Tianda Tianfa, People’s Republic of China) was used to conduct the dissolution test. PBD (30 mg) and PPN, including PPN50, PPN100, and PPN200 (30 mg), were suspended, respectively, in 900 mL PBS (pH 7.4) containing 1% Tween-80. The temperature of the medium was 37°C±0.5°C, and the paddle speed was 100±1 rpm. At predetermined intervals, samples (5 mL) were withdrawn from the dissolution medium to be analyzed by HPLC after being filtered through a 0.22-µm microporous filtering film, and an equal volume of fresh medium was added. Experiments were conducted in triplicate.

In vitro cytotoxicity

PPN cytotoxicity assays were carried out in a liver cancer cell line (SMMC-7721). SMMC-7721 cells, at 10^6 cells/well, were seeded in 96-well plates and cultured in a CO_2 atmosphere at 37°C for 24 h. RPMI 1640 medium with 10% FBS was removed and replaced with equal amounts of PBD suspension, PIJ, and PPN50 suspension. The medium of the suspension was RPMI 1640 medium containing 10% FBS and 2% hyromellose. The treated cells were incubated for 24, 48, and 72 h. Each well was added 20 µL MTT solution (5 mg/mL), and then the plate was incubated for 4 h. After discarding the medium, 200 µL DMSO was added to each well, and the plate was shaken for 10 min. The absorbance was obtained using a microplate reader at a wavelength of 492 nm.

Flow cytometry assays

Cell apoptosis was recorded by flow cytometry (Becton Dickinson, Zürich, Switzerland). SMMC-7721 cells (1×10^4 cells/well) were seeded in six-well plates and cultured in a CO_2 atmosphere at 37°C for 48 h. PBD suspensions, PIJ, and PPN50 suspensions (corresponding to 10 ng/mL paclitaxel concentration) were added to the six-well plates and incubated for 48 h. The medium was discarded, and trypsin was added for digesting the cells. Then, 500 µL binding buffer dispersed the collected cells after centrifugation. Annexin V-FITC (5 µL) and PI (5 µL) were sequentially added to the system and, after 20 min of incubation, the apoptosis rate of the SMMC-7721 cells was determined in regard to time.

Western blotting assay

After the PBD suspension, PIJ, and PPN50 suspension treatments of the SMMC-7721 cells, the ultrasonic grinding method was used to extract the total protein from the SMMC-7721 cells. The total protein concentration was adjusted and determined using the bicinchoninic acid (BCA) method. The protein samples and the buffer solution with a volume ratio of 4:1 were mixed and boiled for 5 min. A 10-µL sample of each solution and the standard protein were analyzed by SDS-PAGE electrophoresis. TBST containing 5% skim milk blocked the polyvinylidene fluoride (PVDF) membrane containing the separated proteins. The PVDF membrane was incubated with the primary antibodies (Bax, Bcl-2, and Caspase-3) overnight; thereafter, secondary antibodies were incubated with the membranes for 2 h. After enhanced chemiluminescence reagents (ECL) staining, the protein expression levels were investigated by a UVP gel analysis system (UVP, LLC., Upland, CA, USA).

Cellular uptake study

The counted SMMC-7721 cells were seeded in culture bottles and cultured in a CO_2 atmosphere at 37°C for 24 h. PPN50 suspensions (corresponding to 50, 100, and 150 µg/mL of paclitaxel concentrations) were added to the culture bottles and incubated for 0.5 h. The PBD suspension was used as a comparator. In addition, a PPN50 suspension (corresponding to 50 µg/mL paclitaxel concentration) was added to the culture bottles and incubated for 0.5, 1, and 1.5 h. The medium was discarded and the cells washed with PBS to remove free drugs. The cells were collected and broken by an ultrasonicator. The drug was extracted from the cell lysis solution. The extraction solvent was methyl tert-butyl ether. The organic layer was evaporated by a high-speed centrifuge (Hualida LNG-T120, People’s Republic of China). The dried samples were dissolved in 50 µL of the mobile phase again and vortexed for 5 min. After centrifugation, 20 µL of the supernatant was analyzed by HPLC with a UV–vis detector (HITACHI L-2400, Japan) at 227 nm. The mobile phase was acetonitrile and water (50:50, V/V), and the flow rate was 1 mL/min.

In vivo antitumor efficacy

In vivo antitumor efficacy was evaluated using solid SMMC-7721 cell tumor-bearing nude mice (Beijing Weitong Lihua Experimental Animal Technology Co. Ltd.). The animal experimental study conformed to the guidelines of the Jin Zhou Medical University Laboratory Animal Ethics Committee (no 11400700217882) and the committee approved the experiment. The right front legs of nude mice were implanted by with 5×10^4 SMMC-7721 cells dispersed in 0.1 mL of physiological saline through hypodermic injection. When the tumor volume reached 50–70 mm³, the tumor-bearing mice were randomly divided into four groups. Physiological saline, PBD, PIJ, and PPN50 suspension, at a dose of 20 mg/kg, were administered via intraperitoneal injection at 3-day intervals. The length and width of each tumor was measured by a caliper before each administration. The tumor volume was calculated using the equation: V=0.5×length×
After dosing five times, the solid tumors were excised from the animals and fixed with paraformaldehyde for further study. Tumor sections were obtained by paraffin embedding and slicing, and were stained by H&E. Then, fluorescence microscopy (Leica DMI 4000B, Germany) was used to observe cell apoptosis in the tumors.

Statistical analysis
Results are shown as the mean±SD, and all data after statistical analysis testing (analysis of variance) were considered statistically significant.

Results and discussion
Preparation and morphological characterization of PPN
The PPN were prepared in a single-step process using an electrostatic spraying method. The process is shown in Scheme 1. The droplets generated under high voltage formed a Taylor cone and produced a filament. Due to the combined action of the electric field force, surface tension, and gravity in the solvent evaporation process, the filaments turned into a string of beads; then, PPN were obtained by redispersion. Parameters such as drug concentration, injection velocity, spinning voltage, needle-collector distance, and needle diameter affected the particle sizes of PPN. The drug concentration and injection velocity were two key factors. A paclitaxel concentration of 150 mg/mL could produce uniform particles, but the high paclitaxel concentration caused needle clogging. SEM and TEM images in Figure 1A–C indicated that PPN with good dispersibility have a particle size of 90 nm. Particle size determination results (Figure 1D) of PPN were in agreement with the images of SEM and TEM. When other parameters were constant, the particle size of PPN significantly increased with injection velocity (50, 100, and 200 µL/h). As shown in Figure 2A–C, SEM images displayed that the particle sizes of PPN (50, 100, and 200) were 90 nm, 200 nm, and 3 µm. Moreover, the smaller the needle diameter, the smaller the PPN particle size. The content of paclitaxel in PPN reached 100%, compared with 10% of the marketed Abraxane® preparation.27

XRD and DSC analysis
XRD was undertaken to check the crystalline nature of paclitaxel in the PPN. The diffractograms of PBD and PPN are shown in Figure 3A. The pattern indicated that, compared with the crystal characteristic diffraction peaks of PBD at 5.6°, 8.4°, and 12.5°, there were no crystal characteristic diffractions in PPN50, PPN100, and PPN200. This showed that paclitaxel in the PPN was presented in the amorphous state after the electrostatic spray. This may have been caused by the rapid volatilization of the solvent during the movement of the droplets in the electric field. Paclitaxel molecules do not form ordered crystals. DSC thermograms of PBD and PPN are shown in Figure 3B. PBD, as a crystal, presented a single endothermic peak at 223°C. By comparison, PPN (50, 100, and 200) show an endothermic peak from 35°C to 85°C, which showed PPN had a low glass-phase transition temperature.28 Paclitaxel began to melt and decompose, and became exothermic at ~150°C. These data indicated paclitaxel in the PPN existed in an amorphous state, which was identical to the XRD conclusion.

Scheme 1 Flow chart of pure paclitaxel nanoparticles (PPN) preparation process.
In vitro dissolution

The dissolution profile of PBD and PPN is shown in Figure 4. The dissolution amount of PPN (50, 100, and 200) at 5 min was above 80%, whereas PBD could dissolve only to 13.1%±1.24%. At 1 h, the dissolution amount of PPN (50, 100, and 200) reached ~100%, whereas that of PBD had only risen to 38%±3.53%. The results indicated that the dissolution rate of PPN was significantly improved and showed a fast-release
effect. PPN50 had the best dissolving effect and the reason for the increase in the dissolution rate could be explained using the Noyes–Whitney equation.\textsuperscript{29} The increase in surface area due to the reduction in particle size led to the enhancement of the dissolution rate. Reduction of particle size decreased the diffusion distance, which was also a reason for the significantly increased dissolution rate. However, the most important factor was the amorphous state of the paclitaxel.

\textbf{In vitro cytotoxicity and apoptosis analysis}

In view of PPN50’s good dissolution effect and smaller particle size, in vitro inhibition efficacy studies on the liver cancer cell line SMMC-7721 were carried out to assure that paclitaxel retained its activity after being processing through the electrostatic spray. To test cell viability in SMMC-7721 cells treated with PBD suspensions, PIJ and PPN50 suspensions, at different concentrations ranging from 1 ng/mL to 1,000 ng/mL, were determined by an MTT assay.\textsuperscript{1} As shown in Figure 5A–C, after 24 h of incubation, the PBD suspension (1,000 ng/mL) caused up to 28.6\%±4.29\% cell death, whereas the cell death rate of the PPN50 suspension was 44.5\%±1.26\%. This was close to that of PIJ (41.4\%±0.78\%). When the concentration of the PPN50 suspension was less than 10 ng/mL, the inhibitory effect was similar to that of the PBD suspension, but the inhibition rate increased significantly when the concentration was increased. At 48 h and 72 h, the inhibitory effect of PPN50 (72.2\%±8.65\% and 82.4\%±3.53\%) at 1,000 ng/mL concentration was obviously better than that of the PBD suspension (55.1\%±1.79\% and 58.5\%±3.86\%) and PIJ (56.8\%±1.20\% and 68.4\%±5.19\%); moreover, the inhibition rate was closely related to drug concentration. The IC\textsubscript{50} value after 24 h on SMMC-7721 of PBD suspension was found to be 48.9\%±4.68 µg/mL, whereas those of the PPN50 suspension and PIJ were 0.521\%±0.072 and 3.09\%±0.491 µg/mL, respectively. The IC\textsubscript{50} value of the PBD suspension, the PPN50 suspension, and PIJ at 48 h were 622.24\%±27.9, 68.49\%±5.64, and 211\%±23.2 ng/mL, respectively. At 72 h, the IC\textsubscript{50} value of PPN50 reached 10.03\%±1.04 ng/mL, compared with 189.65\%±17.5 ng/mL for the PBD suspension and 40.1\%±1.62 ng/mL for PIJ. This demonstrated that PPN could significantly inhibit cell proliferation, which is attributed to PPN maintaining a high drug concentration in cells due to the direct cell uptake of PPN at the nanoscale as well as its increased drug dissolution.\textsuperscript{30} Apoptosis is a programmed process of self-destruction under certain physiological or pathological environments. It was measured by flow cytometry.\textsuperscript{31} As shown in Figure 6, PPN (10 ng/mL) induced 46.4\%±5.94\% of SMMC-7721 cell apoptosis compared with 10.5\%±3.46\% of PBD and, correspondingly, the apoptosis
rate of PIJ was 38.1±4.27%. This indicated that PPN had an excellent apoptosis-promoting effect on SMMC-7721 cells. The expression levels of the proapoptotic protein Bax, the anti-apoptotic protein Bcl-2, and Caspase-3 further verified the conclusions of the flow cytometry studies. As shown in Figure 7, compared with the PBD group, in the PPN group the protein expression of Bax was induced prominently, and the protein expression level of Bcl-2 was suppressed, which led to an increase of the Bax/Bcl2 ratio. The proapoptotic tendency dominated in the progress of cells. Moreover, Caspase-3, as one of the main apoptosis proteins, was distinctly enhanced. The relative gray values further proved the above results. These results make it clear that the intrinsic apoptotic pathway might be involved in PPN-induced apoptosis. It obviously means that PPN has good antitumor potential.

**Cellular uptake analysis**

PPN could be absorbed directly and accumulated at the tumor site by the enhanced permeability and retention effect. HPLC was conducted to measure if PPN could be effectively taken up by SMMC-7721 cells. Figure 8A shows the concentration change of paclitaxel in SMMC-7721 cells treated with PPN at various time periods. At 0.5 h, the concentration of paclitaxel was 2.59±0.786 µg/mL, compared with 1.37±0.213 µg/mL of PBD. With increasing duration from 0.5 h to 1.5 h, the concentration of paclitaxel in the PPN group increased from 2.59±0.786 µg/mL to 7.35±0.289 µg/mL. By comparison, the concentration of paclitaxel in the PIJ group was only up to 6.37±0.516 µg/mL at 1.5 h. The increase in the uptake amount indicated that the cellular uptake of PPN was time dependent and, over time, the intake of PPN was more than that of PIJ. Figure 8B shows that the amount of PPN uptake also gradually increased and was close to that of PIJ at 0.5 h when the concentration of PPN increased from 50 µg/mL to 150 µg/mL. The PBD group had the least amount of cell uptake. Thus, the uptake of PPN was also concentration dependent. These results show that PPN could be taken up directly and maintained at high drug concentrations in cells, which promotes cell death.

**Evaluation of therapeutic effect in solid tumor-bearing mice**

The greatly enhanced tumor cell apoptosis caused by PPN heightened our expectation of the eventual efficacy of PPN in...
anticancer therapy. Therefore, a comparative evaluation of the antitumor performance of PBD, PIJ, and PPN was carried out on a tumor-bearing nude mouse model. As shown in Figure 9A and B, the tumor progression results indicate that both PPN suspensions yielded definite tumor inhibition in comparison with the control group. The treatment group with PPN was significantly better at inhibiting tumor volume compared with the PIJ group. Calculated tumor inhibition rates, based on the averaged volumes of tumors on the animals, were 22.9%±1.07% and 61.7%±5.36% for the PBD and PPN suspension treatment groups, respectively. The tumor inhibition rates between these two groups were significantly different from each other. Moreover, the PPN group obtained a slightly better effect than the PIJ group (56.4%±1.78%). The results can be attributed to the increased paclitaxel dissolution from PPN and accumulation of PPN with smaller particle sizes in the tumors, which led to a high drug concentration atmosphere and accelerated the death of cancer cells. Histopathological examination of the excised tumor tissues was undertaken, and the representative histological photomicrographs are shown in Figure 9C. As suggested by pathologists, the tumor tissue in the PPN group had larger necrotic areas compared with the PIJ group, and the nuclei showed more marked shrinkage and fragmentation. After H&E staining, the purple in the sections represented the viable tumor cells. The saline group observed many viable tumor cells. The PPN and PIJ groups evidently presented less viable tumor cells, and the PPN group had a better anti-proliferation effect.

Conclusion
In summary, PPN obtained by an electrostatic spraying method in a simple manner had adjustable particle sizes. In vitro antitumor experiments showed that PPN could be ingested by cells in significant amounts and inhibit cell proliferation. Compared with PBD and PIJ, PPN had good antitumor effects in vivo. Overall, this study suggested that the electrostatic spraying method in preparing PPN in comparison with traditional nanocrystal preparation methods holds promise in application. These advantages such as single-step process, the simple equipment, amorphous state, and so on are beneficial to industrial production.

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Disclosure
The authors report no conflicts of interest in this work.

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