Ex vivo decontamination of yeast-colonized dentures by iodine–thiocyanate complexes

Sarra Sebaa1,2  
Maxime Faltot1  
Sandra De Breucker3  
Zahia Boucherit-Otmani2  
Françoise Bafort4  
Jean-Paul Perraudin5  
Philippe Courtois1

1Laboratory of Physiology and Pharmacology, Université Libre de Bruxelles, Brussels, Belgium; 2Laboratory of Antibiotics and Antifungals; Physico-Chemistry, Synthesis and Biological Activity, University of Tlemcen, Tlemcen, Algeria; 3Department of Geriatrics, CUB – Hôpital Erasme, Brussels, Belgium; 4Integrated and Urban Plant Pathology Laboratory, Liège University, Gembloux, Belgium; 5Taradon Laboratory, Tubize, Belgium

Introduction: Under well-defined experimental conditions, and in the presence of hydrogen peroxide, lactoperoxidase produces stable iodine–thiocyanate complexes that have antimicrobial properties. A novel process was developed to short circuit the consumption of hydrogen peroxide by microbial catalases by producing iodine–thiocyanate complexes prior to contact with microorganisms, with the aim of being able to decontaminate the ex vivo dentures colonized by yeasts.

Materials and methods: Teabags containing lactoperoxidase adsorbed on inert clay beads were immersed for 1 minute in phosphate buffer solution (0.1 M pH 7.4) containing 5.2 mM potassium iodide, 1.2 mM potassium thiocyanate, and 5.5 mM hydrogen peroxide. After removing the adsorbed lactoperoxidase, the stability and efficacy of iodine–thiocyanate complexes for Candida-colonized denture decontamination were verified. Investigations were performed in vitro on Candida albicans ATCC 10231 and on clinical isolates from 46 dentures. A Candida plate count was performed after a 24-hour incubation at 37°C on Sabouraud–chloramphenicol or CHROMagar solid media; then, the yeast growth was evaluated in Sabouraud broth by turbidimetry and biofilm biomass by crystal violet staining.

Results: In vitro tests demonstrated the effectiveness of the oxidant solution in sterilizing a suspension of 10⁶ Candida cells per milliliter after a 5-minute incubation. A single ex vivo immersion of contaminated dentures in a solution of iodine–thiocyanate complexes led to a decrease of at least 1 log unit in the number of colony-forming units in 58.3% of the tested dentures, while immersing in water alone had no effect on denture colonization (significant \( \chi^2; p = 0.0006 \)).

Conclusion: These data suggest a promising new strategy for decontamination of dentures.

Keywords: biofilm, Candida, hygiene, lactoperoxidase, oral cavity

Introduction

Peroxidases have been known for several decades as antimicrobial factors in the oral environment.1–4 They originate from the salivary glands (sialoperoxidases) or from neutrophils (myeloperoxidase) passing into the oral cavity through the gingival sulci. In the presence of hydrogen peroxide (\( \text{H}_2\text{O}_2 \)), both enzymes convert salivary thiocyanate (\( \text{SCN}^- \)) into hypohiocyanite (\( \text{OSCN}^- \)), which is capable of inhibiting different oral bacterial species (among others Streptococcus mutans, Lactobacillus acidophilus, and black-pigmented anaerobes) and yeasts such as Candida albicans.5,6 In vitro, these peroxidases are also capable of oxidizing iodide (\( I^- \)) to hypoiiodite (\( \text{OI}^- \)); the lactoperoxidase from bovine milk also shows the ability to form OSCN− or OI−.7,8 In tissues, myeloperoxidase forms hypochlorite (\( \text{OCI}^- \)) from the chloride ion (\( \text{Cl}^- \)). Based on promising in vitro experimental data, hypohalous (\( \text{OI}^- \)) or pseudohypohalous...
(OSCNG−) compounds have been suggested as adjuvants to improve oral hygiene or to decontaminate dentures colonized by Candida. For this reason, lactoperoxidase has been incorporated into the oral hygiene products with thiocyanate and a hydrogen peroxide donor often consisting of a glucose/glucose oxidase enzyme sequence. However, the know-how transfer from in vitro to in vivo conditions has led to inconclusive data.9–11 The resulting paradox that is active in vitro, inactive or minimally active in vivo, could be explained by the competition between oral peroxidases and microbial catalase for the same substrate, hydrogen peroxide.12 In fact, the hydrogen peroxide secreted in the oral environment by the salivary glands or produced by neutrophils and some microbial species can be immediately consumed either by peroxidases or by enzymes such as catalase or glutathione peroxidase. The presence of H2O2-detoxifying enzymes helps reduce hypohalous oral concentrations, thus avoiding reaching the antimicrobial levels and further regulating the oral microflora. Ex vivo use for decontamination of dentures may short circuit the competition for H2O2 due to detoxifying enzymes in the oral environment; hence, the antimicrobial effect on the microbial flora depends on the balance between the enrichment in oxidants and the catalase activity present in the biofilms attached to the denture. Recent in vitro studies evaluated the effects of both iodine and thiocyanate when simultaneously incorporated into a peroxidase system on phytopathogens such as Penicillium expansum.13 Under well-defined experimental conditions, and in the presence of hydrogen peroxide, lactoperoxidase produces stable iodine–thiocyanate complexes such as I2(SCN)−, which have antimicrobial properties.14

The aim of this study was to validate a new procedure for the production of iodine–thiocyanate complexes by lactoperoxidase adsorbed on a removable support that could be used to clean a Candida-contaminated denture ex vivo. In this study, the peroxidase-produced oxidants were tested in vitro, first on the reference strain C. albicans ATCC 10231 (often used for the evaluation of antifungal molecules), then on clinical strains isolated by swabbing resin dentures. Finally, an ex vivo clinical trial was performed to confirm the data obtained in vitro.

**Materials and methods**

**Microorganisms**

Yeast were aerobically grown at 37°C for 24 hours in Sabouraud broth (OXOID™ CM147; Basingstoke, UK) or on solid medium, such as Sabouraud agar with chloramphenicol and gentamicin or CHROMagar™ media (BD Diagnostics™, Erembodegem, Belgium). The investigations were carried out on C. albicans ATCC 10231 (Culti-Loops™, Thermo Scientific™, Lenexa, KS, USA), which is often chosen to test antifungals,15 and on clinical strains sampled from dentures using sterile cotton swabs (EUROTUBO®; Deltalab™, Barcelona, Spain). Clinical isolates were identified according to the appearance of their colonies on CHROMagar™ medium, by chlamydoconidia formation on Rice Extract agar with Polysorbate 80 (BD Diagnostics™) and using the API yeast identification system (bioMérieux, Marcy-l’Etoile, France).

All in vitro investigations were performed in a third subculture on Sabouraud solid medium. Candida suspensions were adjusted in phosphate buffer (0.1 M pH 7.4) to 106 (or 2×107) blastoconidia per mL by dilution after counting in a Thoma cell chamber (Marienhof™, Lauda-Königshofen, Germany).

**Oxidant production**

H2O2 (Perhydrol®), potassium iodide (KI), and potassium thiocyanate (KSCN) were purchased from Merck™ (Darmstadt, Germany). A teabag containing 50 mg lactoperoxidase (Tarahon, Tubize, Belgium) and 100 mg inert clay beads was immersed in 1 L phosphate buffer (0.1 M pH 7.4). After dissolution of KI (5.2 mM final) and KSCN (1.2 mM final), 550 µL of Perhydrol® (5.5 mM final) was added dropwise with stirring for 1 minute. The bag was then removed and the solution was stored at 4°C. Before use, the oxidative compounds produced by the aforementioned procedure were assayed using a dithiobisnitrobenzoic acid (DTNB) spectrophotometric method adapted from Aune and Thomas (1977).16 After reduction of 146 µM DTNB (Sigma-Aldrich™, St Louis, MO, USA) using 142 µM mercaptoethanol (Merck™), thionitrobenzoic (TNB) acid reagent (total volume of 2 mL) was mixed with the solution containing the oxidants (from 0 to 500 µL made up to 500 µL with 0.1 M phosphate buffer, pH 7.4). The TNB that absorbs light at 412 nm was then oxidized to the colorless DTNB. According to the Beer–Lambert law (molar extinction coefficient ε equal to 13,600 M−1·cm−1), the number of moles of oxidized TNB was calculated and transformed into oxidant concentration presuming that one mole of oxidant oxidizes two moles of TNB. Figure 1 shows the analytical range for determining the oxidant concentration by using normalized data from four independent experiments. Between 8.4 and 75.3 µM of oxidant, absorbance at 412 nm is proportional to the concentration: the relation is then linear (y = 0.014x, r = 0.9930). The inter-day precision of the method was 9.3%. The inset in Figure 1 represents the experimental absorbance after mixing different volumes of fresh oxidant solution with reagents.
Clinical, Cosmetic and Investigational Dentistry 2018:10

submit your manuscript | www.dovepress.com
Dovepress

Effect of iodine–thiocyanate complexes on Candida

Candidacidal effect

The candidacidal effect of the solution of iodine–thiocyanate complexes was evaluated by counting colony-forming units (CFUs) on Petri dishes for the reference strain ATCC 10231. The suspensions of *C. albicans* (10⁷ or 2 × 10⁸ blastoconidia in 0.1 mL of 0.1 M phosphate buffer, pH 7.4) were incubated for 5 minutes in the presence of oxidizing solution (9.9 mL). A control suspension was incubated in phosphate buffer alone. In addition, 3 different conditions were simultaneously tested on the yeast suspensions: distilled H₂O, 5.5 mM H₂O₂ alone. In addition, 3 different conditions were simultaneously tested on the yeast suspensions: distilled H₂O, 5.5 mM H₂O₂ in buffer and a solution prepared without H₂O₂ thus containing only 1.2 mM KSCN, and 5.2 mM KI substrates in buffer. After incubation, 20 μL (from undiluted test tubes and from 100-fold diluted control tubes) was spread on Sabouraud agar.

Colonies were counted after 24 hours of incubation at 37°C using an e-Count™ colony counter (Heathrow Scientific, Vernon Hills, IL, USA). Oxidant concentration was determined by spectrophotometry, and *C. albicans* ATCC 10231 sensitivity to the oxidant was measured in each solution to check in vitro efficiency before further investigations and clinical trials.

Biofilm production and quantification

*Candida* biofilms were produced in flat-bottom 96-well (12 columns of 8 wells) polystyrene plates (GreinerBio-one®, CellStar™, Neuburg, Germany) for reference and 6 clinical strains randomly taken from denture swabs. Each column of the microplate corresponded to an experimental condition repeated 8 times. The minimal inhibitory concentration of iodine–thiocyanate complexes was determined by evaluation of turbidimetric growth in broth and by quantification of attached biomass in the reactive wells for the reference strain and for 6 clinical isolates from dentures. Yeast growth and their capacity to form biofilms were evaluated after 24 hours at 37°C in flat-bottom 96-well plates in liquid Sabouraud medium with increasing concentrations of oxidants ranging from 31 to 1000 μM. Using Combitips pipettes or a MultiChannel pipette (Eppendorf®, Germany), each well

Figure 1 DTNB oxidation by iodine–thiocyanate complexes: relation between absorbance at 412 nm and oxidant concentration in the reaction medium (mean ± SEM). Inset: relation between absorbance and oxidant solution volume transferred in the reaction medium. The various volumes were made up to 0.5 ml with phosphate buffer and then added to the same reagent volume (2 ml) to obtain a 2.5 ml final volume.

Figure 2 Evaluation of iodine–thiocyanate complexes concentration in solutions stored at 4°C (average of 4 replicates for each of the solutions tested). Oxidant production was assayed by the DTNB method in 47 different oxidant solutions after variable storage times (from 1 day to 132 weeks). Initial concentrations were 1929 ± 46 μM (mean ± SEM, n = 8). Dotted line indicates the value corresponding to the mean – 3 SD of the initial concentrations.

Abbreviations: DTNB, dithiobisnitrobenzoic acid; SEM, standard error of the mean.

Figure 2 shows the concentration of oxidants found after different storage times at 4°C (from 1 day to 132 weeks). Initial concentrations were 1929 ± 46 μM (mean ± standard error of the mean [SEM], n = 8): the coefficient of variation (CV) was 6.7%. These levels appear to slowly decrease over 4 weeks and then decayed exponentially (r = 0.9751) to reach zero after 50 weeks. After 5 weeks of storage at 4°C, the concentration of iodine–thiocyanate complexes had decreased by 20.9% of the initial value; at 50 weeks of storage at 4°C, the concentration of iodine–thiocyanate complexes had decreased by 88.8%. In the context of this investigation, the oxidant concentration was always determined on the day of use.
Initially contained 0.5 × 10⁶ blastoconidia suspended in 25 µL of 0.1 M phosphate buffer at pH 7.4. After adding the same volume (25 µL) of oxidizing solution, the plate was mixed for 5 minutes at room temperature using a magnetic stirrer (IkaLab, Biotecom, Belgium). Sabouraud broth was added (final volume of each well: 250 µL). After a 24-hour incubation at 37°C, the growth of the yeast was evaluated photometrically at 600 nm using a microplate reader (Packard SpectraCount™; PerkinElmer, Wellesley, MA, USA). This turbidimetric measurement indifferently evaluates the planktonic phase and the phase attached to the bottom of the well plus sedimented cells. The data were compared with a control without the oxidants and a sterility control without the well plus sedimented cells. The toxicity of the oxidant solution was evaluated using the Packard SpectraCount™ Microplate Reader is acceptable up to an optical density of 2,500 (CV and coefficient of accuracy <5%). However, all samples with an absorbance value higher than 2,000 were diluted 10-fold. Measurement of crystal violet absorbance in series of 8 ATCC-control replicates showed a CV between 10.9% and 29.5%.

Acrylic resin (Palapress®; Kulzer, Hanau, Germany) pieces (3 × 18 × 5 mm) were produced as previously described and stored at 4°C in 0.1% (w/v) sodium azide. On the day of the experiment, resin pieces were washed for 5 minutes at room temperature, 3 times with 60 mL sterile distilled water, and once with 60 mL Sabouraud liquid medium. For biofilm production on this support, two pieces were transferred to 15 mL conical Falcon tubes made of polypropylene (BD Biosciences, Erembodegem, Belgium) and immersed in 4 mL Sabouraud medium containing a C. albicans ATCC 10231 suspension (10⁶ blastoconidia/mL). After a 24-hour incubation at 37°C, the liquid phase was carefully aspirated and its turbidity measured at 600 nm. A control without yeast attested to the absence of contamination during the process. Resin pieces were then transferred into new sterile Falcon tubes by sliding and were washed 3 times in sterile phosphate buffer (0.1 M, pH 7.4). Finally, they were immersed for 30 minutes in the solution of iodine-thiocyanate complexes or in phosphate buffer as a control, both supplemented with glucose (2 g/100 mL). Swabs from each surface were seeded onto Sabouraud (with 0.4 g/L chloramphenicol and 0.04 g/L gentamycin) solid medium for a CFU count performed after 48 hours at 37°C.

**Dentures**

Denture wearers were recruited at the Erasme Hospital (ULB, Brussels) from the dentistry or geriatrics departments between January and March 2016. The protocol was approved on October 22, 2015, by the Erasme-ULB Ethics Committee under number P2015/413; the participants provided written informed consent for their data to be anonymously published. The recruited denture wearers had a healthy oral mucosa free of any clinical oral infection, namely of prosthetic stomatitis or other mucosal candidiasis. Enrolled patients (N = 46, between 41 and 98 years old) were alternately allocated to a test or control group of 23 patients each. All participants completed a questionnaire on the denture characteristics and their hygiene habits. The ex vivo treatment of dentures was carried out in 3 steps in a single session: 1) swabbing (time zero) of the denture resin (palatin side) to detect the presence of *Candida*, 2) immersing the denture for 5 minutes in the iodine–thiocyanate complexes solution (test group) or sterile water (control group) followed by rinsing with sterile water, and 3) reswabbing of denture to detect the residual presence of *Candida*. Moreover, the dentures in the control group were immersed in the solution of iodine–thiocyanate complexes, and then swabbed 3 times. At this point, the clinician was not aware of the dentures that were contaminated with *Candida*.

The swabs were immediately seeded on ChromAgar® agar for the *Candida* plate count and species diagnostics. Subcultures were stored at 4°C and all in vitro investigations were performed on a third subculture that was started the day before each experiment.

**Toxicity of oxidants**

The toxicity of the oxidant solution was evaluated using the Trypan blue method on oral epithelial cells present in saliva after desquamation. Trypan blue that enters living cells is immediately expelled: the living cells do not have a colored
appearance whereas the dead cells appear colored in blue. The saliva of 6 adults (5 women and 1 man aged 25–65 years) was collected in sterile containers (DeltaLab™). A volume of 50 µL of saliva diluted twice (in 0.9% NaCl) was mixed with 50 µL of a Trypan blue solution (Merck™) at a concentration of 0.4 g/100 mL. The percentage of dead cells (on a total of at least 120 cells in each preparation) was observed between a slide (Thermo Scientific™, Braunschweig, Germany) and coverslip (Marienfeld, Lauda-Königshofen, Germany) on a Leica DM2000 optical microscope (Leica™, Wetzlar, Germany). A 30-minute incubation (room temperature) of 100 µL of saliva with 100 µL of iodine–thiocyanate complexes solution or 0.4% (w/v in phosphate buffer) chlorhexidine digluconate (Sigma-Aldrich™) preceded the microscopic examination to check the toxicity of these compounds.

Statistics
Experimental data were normalized in percentage by dividing each experimental value by the arithmetic mean of the control values obtained in the same experiment. Mean values are expressed with their related SEM, unless indicated otherwise. Statistical tests (average, standard deviation, SEM, Kolmogorov–Smirnov test, unpaired Student’s t-test, analysis of variance supplemented by a Dunnett’s multiple comparison post-test, chi-square test) were performed using GraphPadPrism software, version 7.01, for Windows (GraphPad Software, San Diego California USA; www.graphpad.com). This same software was used to carry out the mathematical analysis of the data and to draw graphs.

Results
Candida susceptibility to iodine–thiocyanate complexes
Figure 3 shows the effect of the various solution components, combined or not, on the survival rate of C. albicans ATCC 10231. Only the complete system had a candidacidal effect after a 5-minute incubation in the presence of 10⁶ Candida cells/mL. Neither thiocyanate/iodide substrates after incubation with the peroxidase enzyme nor H₂O₂ alone reduced the Candida CFU count. The survival rate of yeast in water did not differ from that observed in the buffer.

When tested on higher blastoconidia concentrations, the solution of iodine–thiocyanate complexes reduced the viability of Candida cells at least up to 2×10² blastoconidia/mL in a 5-minute incubation (data not shown).

Figure 4 shows the effect of different oxidant concentrations (from 31 to 1000 µM) on the growth of Candida (reference strain and clinical isolates) in the wells of a 96-well polystyrene plate. The data from each independent experiment were expressed as the percentage of the paired controls conducted in the same plate, allowing us to estimate the variability of the results within each condition with respect to the day-control and to avoid the bias of variability between experiments. After a 24-hour incubation in the presence of C. albicans ATCC 10231, the mean turbidity

Figure 3 Candida inhibition after a 5-minute incubation at room temperature by various combinations of reagents used to form iodine–thiocyanate complexes. Candida albicans ATCC 10231 survival rate was evaluated by counting colony-forming units (CFUs) after cultivation on Sabouraud agar. Lp was clay-adsorbed inside a teabag. Data are expressed as the percentage of CFUs counted after incubation in 0.1 M phosphate buffer (pH 7.4). Error bars indicate the standard error of the mean (N = 4).

Abbreviation: Lp, lactoperoxidase; H₂O₂, hydrogen peroxide; KSCN, potassium thiocyanate; KI, potassium iodide.

Figure 4 Effect of different concentrations of iodine–thiocyanate complexes on Candida albicans growth (ATCC 10231 and 6 clinical isolates) after a 5-minute incubation at room temperature in the wells of a 96-well polystyrene plate. Yeast growth was then evaluated after addition of sabouraud broth and incubation at 37°C for 24 hours. Data are expressed as the percentage of turbidity observed in the same conditions after incubation in 0.1 M phosphate buffer (pH 7.4). The different oxidant concentrations were compared with the paired control by ANOVA complemented by a Dunnett’s multiple comparison test (**p < 0.001). Error bars indicate the standard error of the mean (N = 6).

Abbreviation: ANOVA, analysis of variance.
of the culture medium in the control wells corresponded to an absorbance at 600 nm of 0.783 ± 0.023 (N = 6). In 6 independent experiments, concentrations of iodine–thiocyanate complexes ≤250 µM did not modify the turbidity observed in the wells when compared to the average of the paired controls. The absorbance at 600 nm after 24 hours of incubation at 37°C then varied between 99.8% ± 1.3% and 102.1% ± 1.7% of that observed for the paired controls (Dunnett, NS). A iodine–thiocyanate complex concentration ≥500 µM significantly suppressed the growth evaluated by turbidimetry (Dunnett: p < 0.001). The results obtained on clinical strains isolated from the dentures were similar. The mean turbidity of the culture medium in the control wells corresponded to an absorbance at 600 nm of 0.885 ± 0.024 (N = 6). In the presence of iodine–thiocyanate complexes at a concentration ≤250 µM, the 600 nm absorbance after a 24-hour incubation at 37°C ranged from 93.6% ± 4.4% to 100.1% ± 0.4% of that observed for paired controls (Dunnett, NS). Higher oxidant concentrations (500 and 1,000 µM) significantly reduced growth by 78.1% ± 8.8% and 99.4% ± 0.6% of control growth, respectively (Dunnett, NS). Higher oxidant concentrations (500 and 1,000 µM) significantly inhibited the biomass by 78.1 ± 8.8% and 99.4% ± 0.4% of the control biomass as assessed using the crystal violet method (Dunnett: p < 0.001).

On the other hand, after 30 minutes of contact at room temperature, iodine–thiocyanate complexes completely abolished the presence of Candida biofilms on a resin piece contaminated with C. albicans ATCC 10231 as shown in Figure 6.

**Denture decontamination using iodine–thiocyanate complexes**

Table 1 details the number of patients recruited (N = 46 divided into 2 groups of 23 individuals) and enrolled after demonstration of Candida contamination (N = 27; 12 in the test group and 15 in the control group). The mean ages were 74.3 years for the control group (N = 23, median: 68.0 years) and 79.5 years for the test group (N = 23, median: 82.0 years). The mean ages of Candida-positive patients were 74.2 years for the control group (N = 15) and 77.5 years for the test group (N = 12). None of these patients suffered from oral mucosal disease, including prosthetic stomatitis. Hygiene techniques

---

**Figure 5** Effect of different concentrations of iodine–thiocyanate complexes on biofilm formation by Candida albicans strains (ATCC 10231 and 6 clinical isolates) after a 5-minute incubation at room temperature in the wells of 96-well polystyrene plate. Biofilm was then evaluated by staining the attached biomass after incubation at 37°C for 24 hours in Sabouraud broth. Data are expressed as the percentage of attached biomass observed in the same conditions after incubation in 0.1 M phosphate buffer (pH 7.4). The different oxidant concentrations were compared with the paired control by ANOVA complemented by a Dunnett’s multiple comparison test (**p < 0.001). Error bars indicate the standard error of the mean (N = 6).

**Figure 6** Sabouraud agar plates inoculated by swabbing contaminated acrylic resin pieces and incubated for 48 hours at 37°C. The resin pieces were previously subjected to different processes. (A) Uncontaminated acrylic resin piece processed to attest the sterility of handling. (B) Contaminated (C. albicans ATCC 10231) acrylic resin piece immersed for 30 minutes in phosphate buffer (0.1 M, pH 7.4) at room temperature. (C) Contaminated (C. albicans ATCC 10231) acrylic resin piece immersed for 30 minutes in the solution of iodine–thiocyanate complexes at room temperature. The data are representative of 4 independent experiments.

---

Sebaa et al

Clinical, Cosmetic and Investigational Dentistry 2018:10

submit your manuscript | www.dovepress.com

Dovepress

Clinical, Cosmetic and Investigational Dentistry downloaded from https://www.dovepress.com/ by 54.70.40.11 on 29-Dec-2018
For personal use only.
Powered by TCPDF (www.tcpdf.org)
Table 1 Status of Candida contamination for investigated dentures

<table>
<thead>
<tr>
<th>Number of cases</th>
<th>Test</th>
<th>Control</th>
</tr>
</thead>
<tbody>
<tr>
<td>Number of recruited denture wearers</td>
<td>23</td>
<td>23</td>
</tr>
<tr>
<td>Number of Candida-colonized dentures</td>
<td>12</td>
<td>15</td>
</tr>
<tr>
<td>Plurispecies colonization</td>
<td>3</td>
<td>1</td>
</tr>
<tr>
<td>Monospecies colonization</td>
<td>9</td>
<td>14</td>
</tr>
<tr>
<td>Number of enrolled Candida-contaminated dentures</td>
<td>12</td>
<td>15</td>
</tr>
<tr>
<td>Number of recruited patients reporting using the procedure for cleaning denture</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Brushing with water</td>
<td>3</td>
<td>4</td>
</tr>
<tr>
<td>Brushing with toothpaste</td>
<td>18</td>
<td>11</td>
</tr>
<tr>
<td>Soaking in a commercial cleansing tablet solution</td>
<td>14</td>
<td>7</td>
</tr>
<tr>
<td>Soaking in a commercial mouthwash solution</td>
<td>9</td>
<td>9</td>
</tr>
<tr>
<td>Soaking in bleach</td>
<td>0</td>
<td>1</td>
</tr>
</tbody>
</table>

Note: Status of Candida contamination (prevalence, number of different species, and hygiene procedure used) in both investigated groups (test and control).

had no effect on the denture colonization by Candida. Indeed, the denture wearers from both groups claimed to practice the different procedures in similar proportions (\(\chi^2\) test, \(p = 0.5003, \text{NS}\)). The rate of denture colonization was 58.7% in the patients examined in this study (N = 46). In 55.6% of the wearers with Candida contamination, denture swabbing returned more than 50 CFUs per Petri dish. More than half of the dentures (85.2%, N = 23 for a total of 27 wearers with a Candida-contaminated denture) showed a “monospecies” contamination (a single species of yeast based on the color of macroscopic colonies observed on CHROMagar® medium). The remaining 14.8% (N = 4 for a total of 27 edentulous patients) were colonized by several “species” (2 or 3 different yeast species). The albicans species was found in 74.1% of the positive cultures (N = 27). The second most commonly identified species was C. glabrata (18.5%). The association C. albicans plus C. glabrata was found in 11.1% of Candida positive patients.

Table 2 reports the results of the clinical trial. A single 5-minute ex vivo immersion in the iodine–thiocyanate complexes resulted in a decrease of at least 1 log unit in the number of CFUs in 7 out of 12 cases (58.3%), whereas immersing in water had no effect on denture colonization. This effect was statistically significant (\(\chi^2\), \(p = 0.0006\) and Fisher’s test, \(p = 0.0009\)). A decrease of at least 1 log unit corresponds to a decrease of more than 90% of the CFUs initially counted. The dental prostheses of the control group were also submitted to a final immersion of 5 minutes in the iodine–thiocyanate complexes and then to reswabbing: 60% of the control dentures then showed a decrease of at least 1 logarithmic unit. This confirmed the data observed in the test group and gave these patients the same chance of having a decontaminated prosthesis in the context of this study. Moreover, in the test group, a general downward trend in the number of colonies present after soaking was apparent: a reduction in the Candida (whatever the species) count by half was observed in 83.3% of cases after a 5-minute incubation in the oxidant solution. In the same manner, the trend was similar in the secondarily treated control group: 73.3% showing at least a halving of Candida plate count. The control solution (water) led to a decrease in the number of CFUs of at least half in 26.7% of the cases but never a decrease of the order of 1 log unit. It should be noted too that the Candida CFU count showed an increase in the number of CFUs in 13.3% of cases after soaking in water.

Toxicity of oxidants
The percentage of Trypan blue–stained epithelial cells was determined in saliva from 6 adults, before and after addition of the oxidative solution. The effect of iodine–thiocyanate complexes on Candida growth and its ability to form biofilms revealed lower oral epithelial cell toxicity than that of chlorhexidine, an antiseptic that is widely used in stomatology. Indeed, Table 3 shows an increase in the number of epithelial cells stained by Trypan blue after 30 minutes of incubation at room temperature, lower than that found in the presence of 0.2% (w/v) chlorhexidine under the same experimental conditions. An ANOVA supplemented by a Dunnett test showed that this increase was significant for the 1000 \(\mu\)M oxidant solution (\(p < 0.001\)) and 0.2% (w/v) chlorhexidine (\(p < 0.001\)).

Discussion
This study aimed to evaluate the effect of oxidants produced by peroxidase on an oral microorganism by avoiding interference from the components of the peroxidase system.
that generate them. Indeed, the new investigated process bypassed the presence, in the final solution, of enzymatic proteins (potentially immunogenic for in vivo applications) and the presence of hydrogen peroxide, one of the peroxidase substrates that is unstable and which is itself oxidizing. This study demonstrated the effectiveness of iodine–thiocyanate complexes produced by lactoperoxidase adsorbed on clay at inhibiting *C. albicans* (both from a reference source and clinical strains) in vitro. The prior preparation of iodine–thiocyanate complexes in the absence of (oral) microorganisms bypassed the consumption of hydrogen peroxide by microbial catalases. In addition, the presence of lactoperoxidase – without its substrates – could become a microbial nutrient and thus stimulate the growth of certain bacterial species in the oral cavity. The method already implemented in agronomy makes it possible to obtain lethal oxidant concentrations for *C. albicans* close to the concentrations of hypohiocyanite observed in whole saliva, namely up to 300 µM.

The oxidant concentrations produced were less toxic to the epithelial cells of the mouth than chlorhexidine. Indeed, 1,000 µM oxidant solution increased the percentage of epithelial cell mortality in saliva by 20.0% on average (from 34.8% to 54.8%) while chlorhexidine increased it by 65.2% (from 34.8% to 100.0%). When diluted 3 times, the oxidant solution retained its anti-

<table>
<thead>
<tr>
<th>Table 3 Toxicity evaluation of oxidant solution</th>
</tr>
</thead>
<tbody>
<tr>
<td>Assay condition</td>
</tr>
<tr>
<td>Control</td>
</tr>
<tr>
<td>Iodine–thiocyanate complex solution, 300 µM</td>
</tr>
<tr>
<td>Iodine–thiocyanate complex solution, 1000 µM</td>
</tr>
<tr>
<td>Chlorhexidine 0.2%</td>
</tr>
</tbody>
</table>

Notes: Percentage of saliva epithelial cells stained by Trypan blue after a 30-minute incubation in the presence of iodine–thiocyanate complexes (300 vs 1,000 µM) or chlorhexidine (0.2%). The different conditions were compared with the paired control by ANOVA complemented by a Dunnett’s multiple comparison test (**p < 0.01). Hypoiodite decreases the survival rate at a lower concentration than hypohiocyanite. However, the hypohalosial ion present in the oral cavity is hypohiocyanite rather than hypoiodite because the substrate SCN− at the origin of the former is 1,000 times more abundant than iodide (I−) at the origin of the latter: in fact, the salivary thiocyanate results in salivary concentrations in the mM while iodide scarcely reaches µM concentrations. Mixtures of the two substrates, thiocyanate and iodide, have been less often investigated but seem to result in the formation of a more stable oxidating complex over time. Few of the papers claiming that lactoperoxidase activity produces a OSCN−/OI− mixture as products in the presence of an SCN−/I− substrate mixture actually analyzed the products generated under their experimental conditions. The relative concentration of the substrates, and the pH and ionic strength of the media modified the nature of active ions produced by lactoperoxidase and the biological activity of the enzyme. Under well-defined experimental conditions – ie, 1) high ionic strength, 2) a ratio of KI:KSCN around 4.5, and 3) in vitro enzymatic reaction in a neutral or acid buffer – an iodine–thiocyanate complex is produced. Indeed, using 13C nuclear magnetic resonance spectroscopy, OSCN− is observed only when using H2O2 and SCN− as substrates, while in the presence of a mixture of SCN−/I−, the peak of OSCN− is replaced by another corresponding to I2SCN−. In particular, the use of a highly concentrated neutral buffer such as 0.1 M phosphate buffer pH 7.4 optimized the oxidoreduction of iodide and thiocyanate by lactoperoxidase and yielded more concentrated ion solutions in the form of I2SCN−, which showed in vitro inhibition of *P. expansum*, a fungus responsible for various fruit pathologies. The formation of I2(SCN)− iodine–thiocyanate complex has already been incriminated in older studies when mixing I2 or I− with SCN−.

The oxidant concentrations produced were less toxic to the epithelial cells of the mouth than chlorhexidine. Indeed, 1,000 µM oxidant solution increased the percentage of epithelial cell mortality in saliva by 20.0% on average (from 34.8% to 54.8%) while chlorhexidine increased it by 65.2% (from 34.8% to 100.0%). When diluted 3 times, the oxidant solution retained its anti- activity while reducing the mortality of the oral cells by ~13.0%. The resistance of epithelial cells can be explained on the one hand by the presence of bacteria covering their surface and forming a protective biofilm, and on the other hand by detoxifying enzymatic activities in some commensal species present in the oral environment. Rinsing dentures after ex vivo incubation in the oxidant solution will further limit eventual effect...
on epithelial cells. A recent paper described on HeLa cells a beneficial balance between the active concentration and the toxic one for iodine–thiocyanate complexes produced in a non-enzymatic manner, by comparison with other biocide molecules such as hydrogen peroxide. 27 Further studies will have to analyze the interference between this solution of iodine–thiocyanate complexes and the oral microbial ecology. The efficacy of the iodo-thiocyanate complexes prepared without peroxidase activity has already been demonstrated in vitro on Pseudomonas aeruginosa and on Staphylococcus aureus both in suspension and in biofilms.28

Ex vivo denture decontamination

Denture hygiene is difficult for patients to achieve; Candida-contaminated dentures are a gateway for local and systemic infections. Nevertheless, yeast carriage seemed poorly influenced by the hygiene techniques currently available. According to the information gathered by the anamnesis of all the patients and detailed in Table 1, it appears that patients with a contaminated denture show similar levels of hygiene as other patients, and that regular use of toothpaste and mouthwashes (sometimes with antiseptics) does not prevent the colonization of dentures by Candida. A previous paper showed that commercial denture cleaners have effective in vitro candidacidal activity both in terms of biofilm removal and disinfection, but the persistence of a residual biofilm could lead to continuous recolonization of the prostheses.29 Alternative cleaning methods are needed to improve biofilm removal but these cannot be based on the use of antifungal therapy agents due to the development of resistance. The results reported in this ex vivo clinical study clearly show that a solution of iodine–thiocyanate complexes affected Candida carriage in 60% of the tested dentures after only a 5-minute incubation. Failures can be explained by a Candida carriage at the initial stage above the upper analytical range, thus precluding documentation of a reduction in the number of colonies. They could also be attributed to the presence of resistant strains. The candidacidal effect was active against the two Candida species observed on the denture studied: C. albicans and C. glabrata. A subsequent clinical investigation should follow a cohort of patients who have decontaminated their dentures every day for several days.

Conclusion

The poor susceptibility to antifungals of microorganisms incorporated in biofilms and the emergence of strains that are resistant to conventional antifungals should encourage their restriction to the treatment of proven infections. This investigation suggests a promising new strategy that could be used to remove Candida from dentures with the aim of preventing mucosal invasion and oropharyngeal candidiasis, especially in medically compromised patients. The new method proposed herein affects the growth of Candida in liquid medium and limits its presence in biofilms. Other preclinical studies should investigate the influence of iodine–thiocyanate complexes on the whole oral microbiome. The preliminary preparation of the solution and the ex vivo treatment of dentures reduced the catalase competition for hydrogen peroxide in the oral environment and allowed sufficient concentrations of oxidants produced by lactoperoxidase to be reached for yeast inhibition. The experimental conditions of the described procedure promote the production of iodine–thiocyanate complexes by lactoperoxidase that is characterized by a stability compatible with hygiene applications achievable in patients’ homes.

Acknowledgments

This study was supported by a grant from the Xenophilia Funds (Université Libre de Bruxelles, Brussels, Belgium). The authors thank Ch. Hauet and L. Malashkina from the Haute Ecole Francisco Ferrer (Medical Biology section) for their enthusiastic participation in the investigations in the context of their final year dissertation.

Disclosure

Françoise Bafort and Jean-Paul Perraudin are two of the patented owners of the biological applications of iodine–thiocyanate complexes (Bafort F, Perraudin J-P, Jijakli H Composition comprenant des ions I$_2$SCN$^-$ et/ou des ions I(SCN)$^-$, WO2017046407A1, 2015). The authors report no other conflicts of interest in this work.

References

Clinical, Cosmetic and Investigational Dentistry

Publish your work in this journal

Clinical, Cosmetic and Investigational Dentistry is an international, peer-reviewed, open access, online journal focusing on the latest clinical and experimental research in dentistry with specific emphasis on cosmetic interventions. Innovative developments in dental materials, techniques and devices that improve outcomes and patient satisfaction will be highlighted. The manuscript management system is completely online and includes a very quick and fair peer-review system, which is all easy to use. Visit http://www.dovepress.com/testimonials.php to read real quotes from published authors.

158