Antimicrobial potential of green synthesized CeO$_2$ nanoparticles from *Olea europaea* leaf extract

**Abstract:** This article reports the green fabrication of cerium oxide nanoparticles (CeO$_2$ NPs) using *Olea europaea* leaf extract and their applications as effective antimicrobial agents. *O. europaea* leaf extract functions as a chelating agent for reduction of cerium nitrate. The resulting CeO$_2$ NPs exhibit pure single-face cubic structure, which is examined by X-ray diffraction, with a uniform spherical shape and a mean size 24 nm observed through scanning electron microscopy and transmission electron microscopy. Ultraviolet-visible spectroscopy confirms the characteristic absorption peak of CeO$_2$ NPs at 315 nm. Fourier transform infrared spectroscopy reflects stretching frequencies at 459 cm$^{-1}$, showing utilization of natural components for the production of NPs. Thermal gravimetric analysis predicts the successful capping of CeO$_2$ NPs by bioactive molecules present in the plant extract. The antimicrobial studies show significant zone of inhibition against bacterial and fungal strains. Thehigher activities shown by the green synthesized NPs than the plant extract lead to the conclusion that they can be effectively used in biomedical application. Furthermore, reduction of cerium salt by plant extract will reduce environmental impact over chemical synthesis.

**Keywords:** XRD, SEM, TEM, FTIR, antibacterial, antifungal, nanomedicine, green fabrication

**Introduction**

Green synthesis of metallic oxide nanoparticles (NPs) is an emerging field of nanoscience and technology. Synthesis methodology and surface modification play a critical role in the physical, chemical, electrical, and optical properties of nanomaterials. Cerium oxide (CeO$_2$) is a semiconductor with a band gap energy of 3.0–3.9 eV and large excitation energy. In bulk form, CeO$_2$ is a singular rare earth oxide and does not show any crystallographic modifications by temperature increase up to its melting point.$^1$–$^5$ Its various applications such as acting as a catalyst; use in sensor, solid oxide fuel cells, and sunscreen cosmetics; and its O$_2$ storage capacity, role in CO$_2$ solar conversion, and antibacterial applications are due to its idiosyncratic ability to alter the oxidation states.$^6$–$^{10}$

Ordinarily, CeO$_2$ NPs are fabricated by physical and chemical methods including hydrothermal,$^{11}$ flame spray pyrolysis,$^{12}$ thermal decomposition,$^{13}$ aqueous precipitation,$^{14}$ and coprecipitation technique.$^{15}$ In contrast, green synthesis of metallic oxide NPs offers several advantages such as cost-effectiveness, large-scale commercial production, and pharmaceutical applications. The CeO$_2$ NPs are considered less toxic toward cell lines, when compared to TiO$_2$ and ZnO.$^{16}$ Biosynthesis of CeO$_2$ NPs using plant extracts such as *Gloriosa superba* L. and *Hibiscus sabdariffa* has been reported$^1$ with promising results.

*Olea europaea* belongs to the family Oleaceae and is native to tropical and warm temperate regions of the world.$^{17}$ Mediterranean region serve as the major cultivation area,
accounting for about 98% of the world’s olive production.\textsuperscript{18} \textit{O. europaea} exhibits antioxidant, anti-inflammatory, antiatherogenic, anticancer, antimicrobial, and antiviral activities, along with hypolipidemic and hypoglycemic potential.\textsuperscript{19} Bioactive compounds principally present in olive leaves are oleuropein, followed by hydroxytyrosol, the flavone-7-glucosides of luteolin and apigenin, and verbascoside.\textsuperscript{20,21}

This study reports green synthesis of \textit{CeO\textsubscript{2}} NPs utilizing \textit{O. europaea} leaf extract as an effective chelating agent without any acid or base treatment. The synthesized \textit{CeO\textsubscript{2}} NPs are further characterized for their antibacterial and antifungal potential.

**Experimental procedures**

**Collection and processing of plant material**

The authenticated \textit{O. europaea} leaves were procured from Barani Agricultural Research Institute, Chakwal, Pakistan. Fresh leaves were washed with distilled water and air-dried under shade at room temperature to avoid photodissociation of bioactive compounds. For extract preparation, 20 g of dried leaves was ground and suspended in 200 mL of distilled water. The mixture was placed in a shaking incubator at 50\textdegree C for 2 hours at 50 rpm. Thereafter, it was filtered using Whatman No 1 filter paper and the filtrate was stored at room temperature for further usage.

**Fabrication of \textit{CeO\textsubscript{2}} NPs by \textit{O. europaea} leaf extract**

For green synthesis of \textit{CeO\textsubscript{2}} NPs, 8.68 g of Ce(NO\textsubscript{3})\textsubscript{3} \cdot 6H\textsubscript{2}O was added to 200 mL of \textit{O. europaea} leaf extract and stirred on a magnetic hotplate at 50\textdegree C, 1,500 rpm for 2 hours. The blackish brown \textit{CeO\textsubscript{2}} NPs were collected by centrifugation at 10,000 rpm (GR BioTek, Orpington, England) for 10 minutes. The NPs were washed repeatedly with deionized water and finally dried in hot air oven at 60\textdegree C for 6 hours and further annealed in Gallenkamp furnace (Apeldoorn, the Netherlands) at 500\textdegree C for 2 hours. The yellow-colored NPs obtained were stored in an air-tight jar at room temperature. A schematic diagram for the fabrication of \textit{CeO\textsubscript{2}} NPs using \textit{O. europaea} leaf extract is shown in Figure 1.

**Characterization of \textit{CeO\textsubscript{2}} NPs**

**Scanning electron microscopy (SEM)**

The size and shape of synthesized \textit{CeO\textsubscript{2}} NPs were studied using JEOL-JSM-6490LA SEM (JEOL, Tokyo, Japan) operating at 20 kV with a counting rate of 2,838 cps.

**Transmission electron microscopy (TEM)**

To further investigate the internal morphology, TEM analysis of green fabricated \textit{CeO\textsubscript{2}} NPs was performed by TEM (model no JEOL-1010) operating at 80 kV.

**X-ray diffraction (XRD; crystallographic structure)**

To examine the crystallographic structure of green synthesized \textit{CeO\textsubscript{2}} NPs, XRD analysis was carried out using X’Pert\textsuperscript{3} Powder (PANalytical) with nickel monochromator in the range of 20 from 20\textdegree to 80\textdegree using Cu K\textalpha radiation of wavelength 1.5406 Å. Operating voltage of 40 kV with 30 mA current was provided at room temperature. To calculate the theoretical size of \textit{CeO\textsubscript{2}} NPs, Scherrer’s equation \textit{(D=0.9λ/β cosθ)} was employed, where \textit{D} is the average crystalline domain size perpendicular to the reflecting planes, \textit{λ} the X-ray wavelength (1.5406 Å), \textit{β} the angular full width at half maximum in radians, and \textit{θ} is the diffraction angle (20 is the measured angle of diffraction in degrees) or Bragg’s angle.

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**Figure 1** Schematic diagram of fabrication of \textit{CeO\textsubscript{2}} NPs using \textit{Olea europaea} leaf extract.

**Abbreviations:** BARI, Barani Agricultural Research Institute; NPs, nanoparticles; SEM, scanning electron microscopy; \textit{O. europaea}, \textit{Olea europaea}. 

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Furthermore, the lattice constant was measured using the following equation:

\[ d = \frac{a}{\sqrt{h^2 + k^2 + l^2}} \]

where \(a=b=c\) are lattice constants, \(d\) represents the interplanar spacing which was calculated by using \(2d \sin \theta = n\lambda\), and \(h, k, l\) are Miller indices. The unit cell volume was calculated using \(V = a^3\) formula unit.

**Fourier transform infrared (FTIR) spectroscopy**

For the identification of unknown bio-active compounds in plant extract, utilization of capping agents from plant extract and confirmation of production materials (\(\text{CeO}_2\) NPs), FTIR spectroscopic analysis was performed using KBr pellet methodology (model SHIMADZU FTIR, Kyoto, Japan) in the wavenumber ranges 400–4,000 cm\(^{-1}\). FTIR spectra for \(\text{CeO}_2\) NPs, residue (as synthesized), and \(O.\ europa\) a leaf extract were recorded separately and explained comparatively.

**Ultraviolet (UV)-visible spectroscopy**

The optical characterization of green fabricated \(\text{CeO}_2\) NPs was made through UV-visible spectroscopy in the range of 290–400 nm with a Shimadzu spectrophotometer (model UV-1800, Kyoto, Japan) operating at a resolution of 1 nm. Three milligrams of synthesized \(\text{CeO}_2\) NPs was dissolved in 10 mL of deionized \(\text{H}_2\text{O}\) and this solution was sonicated for 20 minutes. Subsequently, the liquid sample was subjected to UV-visible spectrophotometry.

**Thermal gravimetric analysis (TGA)**

To examine the thermal properties and capping action of bioactive compounds for tailoring \(\text{CeO}_2\) NPs, we performed TGA (model Diamond TGA; PerkinElmer, Waltham, USA) under nitrogen environment from 25°C to 800°C at 10°C/minute.

**Antibacterial activity of \(\text{CeO}_2\) NPs**

The antibacterial activities of green synthesized \(\text{CeO}_2\) NPs were studied against Gram-positive (G+ve) (\(\text{Staphylococcus aureus}\) ATCC 6538) and Gram-negative (G–ve) (\(\text{Escherichia coli}\) ATCC 15224, \(\text{Pseudomonas aeruginosa}\) ATCC 15442, \(\text{Klebsiella pneumoniae}\) ATCC-BAA 1706) strains by disk diffusion method with slight modification. In brief, the bacterial strains were cultured in nutrient broth (Sigma-Aldrich Co., St Louis, MO, USA) at 37°C until the culture reached 1.5×10\(^6\) colony forming units (CFU) per milliliter. About 20 mL of autoclaved molten nutrient agar was poured into the Petri dishes and allowed to cool. All of the four bacterial cultures were swapped over solidified agar medium. Disks were loaded with 20 \(\mu\text{g/5}\mu\text{L}\) \(\text{CeO}_2\) NPs solution. A solution of cefixime or roxithromycin (4 mg/mL each) was used as positive control and deionized \(\text{H}_2\text{O}\) as negative control. The plates were incubated at 37°C for 24 hours and the zones of inhibition (ZOIs) around the disks were measured thereafter.

**Antifungal assay**

Antifungal activity was determined against \(\text{Mucor}\) species (FCBP-0300), \(\text{Aspergillus flavus}\) (FCBP-0064), \(\text{Fusarium solani}\) (FCBP-434), and \(\text{Aspergillus niger}\) (FCBP-0198) by disk diffusion method. Sabouraud dextrose agar (pH 5.7) (Sigma-Aldrich Co.) was autoclaved and poured in Petri plates ensuring sterile conditions. Fungal lawns were prepared by inoculating spores on the surface of the growth media. Thereafter, disks were loaded with \(\text{CeO}_2\) NPs solution 20 \(\mu\text{g/5}\mu\text{L}\) through micropipette. Clotrimazole 10 \(\mu\text{g/disk}\) was used as positive control, while deionized water served as negative control. The plates were incubated at 25°C for 24–48 hours and ZOI was measured.

**CFU counting assay**

To examine the percentage reduction in bacterial count with time, we carried out CFU counting assay. The experimentation was performed in sample tubes, each having 2 mL of sterilized (autoclaved) nutrient broth. The first sample tube containing 2 mL of nutrient broth acted as sterile negative control and the second tube was impregnated with cefixime (as positive control), while the remaining four samples were inoculated with \(S.\ aureus\) (G+ve), and \(E.\ coli, P.\ aeruginosa,\) and \(K.\ pneumoniae\) (G–ve), respectively, having a concentration of 10\(^7\) and 10\(^8\) CFU/mL each, and then placed in a shaking incubator at 37°C. Thereafter, the bacterial culture tubes were loaded with \(\text{CeO}_2\) NPs solution (20 \(\mu\text{g/5}\mu\text{L}\)). After an interval of 30 minutes, 10 \(\mu\text{L}\) from each sample was collected and spread over the already prepared agar plate surface. To estimate the bacterial colony in liquid, McFarland turbidity standards were applied, and their values were calibrated before and after incubation. Positive and negative control tests were performed on \(E.\ coli\) only. Later on, the plates were incubated at 37°C and the percentage reduction with time in bacterial count was calculated by the following approximation:22

\[
\%\,\text{reduction} = \frac{\text{Viable count at 0 minutes} - \text{Viable count at 150 minutes}}{\text{Viable count at 0 minutes}} \times 100
\]
Statistical analysis
Results are calculated as the mean of at least three individual experiments presented with standard deviation (SD). The statistical analysis of the results was carried out with Student’s t-tests by using Statistical Package for the Social Sciences software package version 19.0 (IBM Corporation, Armonk, NY, USA) and considered significant at P-values <0.05.

Results and discussion
Biosynthesis process using *O. europaea* leaf extract (visual observations)
Synthesis of NPs using plant extracts has several advantages over other conventional biosynthesized routes (based on fungi, bacteria, etc). It is because the green extract is more ecofriendly and environment friendly, and fewer cytotoxic reducing agents are required for tailoring NPs. It is also commonly observed that maintaining microbial cultures under optimum conditions is time consuming and costly. In bulk (at an industrial level), safe mode synthesis of NPs, microbial-based NPs production has its own limitations.23,24

*O. europaea* aqueous extract contains a variety of molecules that might be effectively utilized as chelating agents in the green synthesis of CeO$_2$ NPs. Upon addition of (Ce(NO$_3$)$_3$)$_6$H$_2$O, the biologically active compounds of the plant extract act as reducing agents for the nanofabrication process. Oleuropein is the most abundant bioactive compound in *O. europaea* aqueous extract, which breaks down into highly polarizable and reactive hydroxytyrosol.21 It may take part in the redox mechanism in tuning CeO$_2$ NPs, as shown in Figure 2. Figure 3 shows the change in color from dark blackish green to dirty green, indicating the formation of CeO$_2$ NPs. The blackish brown CeO$_2$ NPs were collected by centrifugation at 10,000 rpm for about 10 minutes. Thereafter, the NPs were washed repeatedly with deionized water to eliminate the uncoordinated biomolecules from the extract, and finally dried in hot air oven at 60°C for ~6 hours and further annealed at 500°C for 2 hours. The yellow-colored NPs obtained were stored at room temperature. The entire recovery process is presented in Figures 1 and 2.

SEM analysis
Figure 4 depicts the SEM images of green synthesized CeO$_2$ NPs. It is obvious from the SEM micrograph that the CeO$_2$ NPs are highly homogeneous and symmetrical in morphology with spherical shape, and the findings clearly represent

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**Figure 2** Hypothesized redox process of CeO$_2$ NPs synthesis.
Abbreviation: NPs, nanoparticles.
a mean size of 24 nm. Similar morphological shape with uniform average grain size of less than 30 nm was also reported previously.\textsuperscript{10,25,26}

\section*{TEM analysis}

It is clear from the TEM micrograph that the green synthesized CeO\textsubscript{2} NPs exhibit spherical morphology with highly homogeneous symmetry and an average particle size of 24 nm (Figure 5).

\section*{XRD (crystallographic structure) analysis}

Figure 6 shows the XRD profile of green fabricated CeO\textsubscript{2} NPs. It shows that the CeO\textsubscript{2} NPs possess single-phase cubic fluorite structure without any additional peaks. Presence of several broad Bragg peaks corresponds to (111), (200), (220), (311), (222), (400), (331), and (420) orientations and they are precisely well indexed to Joint Committee on Powder Diffraction Standards (JCPDS) (card no 34-0394). This later crystallographic phase reflects that each cerium site...
is encountered by eight oxygen sites in face-centered cubic arrangement, while each oxygen has the geometry of tetrahedron cerium site. Using Debye–Scherrer approximation and from the broadening of peaks, the approximate crystal size of CeO$_2$ NPs was found to be 6 nm, which shows that each grain of size 24 nm was composed of a cluster of 6 nm crystallites. The observed value was found to be close to that reported previously. The calculated value of lattice constant “a” is 0.585 nm, while the unit cell volume “V” is measured to be 0.20 nm.

**FTIR spectroscopy analysis**

Comparative FTIR results illustrate significant absorption peaks in the wave number range 400–4,000 cm$^{-1}$ (Figure 7). The range between 2,500 and near 4,000 cm$^{-1}$ is assigned to O–H and C–H stretch. The plant extract exhibiting N–O symmetric stretch at 1,355 cm$^{-1}$ was found to be less in the residue after CeO$_2$ NPs were isolated, reflecting that Ce is actively oxidized to CeO$_2$ by the nitro compounds in extract chemistry. The onset of a broad intense band that can be viewed as one tends toward higher vibrations (2,800 cm$^{-1}$) can be ascribed to O–H stretching range due to the presence of water.

**Figure 5** TEM micrographs of green synthesized CeO$_2$ NPs.

**Figure 6** Typical XRD analysis of CeO$_2$ NPs at room temperature.

**Figure 7** FTIR spectra of biosynthesized CeO$_2$ NPs from Olea europaea leaf extract.

Notes: (a) Plant extract; (b) residue (as synthesized); (c) CeO$_2$ NPs.

Abbreviations: FTIR, Fourier transform infrared; NPs, nanoparticles.
of surface-adsorbed water molecules from the surroundings. The spectra due to stretching frequency of Ce–O can be seen at 459 cm\(^{-1}\), which indicates the fabrication of CeO\(_2\) NPs. Similarly, Ce–O stretching band at 451 cm\(^{-1}\) and at 450 cm\(^{-1}\) was reported by Arumugam et al.\(^{10}\) and Goharshadi et al.,\(^{28}\) respectively.

UV-visible spectroscopy studies

Figure 8 illustrates the UV-visible absorption spectra of CeO\(_2\) NPs, which demonstrates the presence of a single intensified peak at 315 nm that was also closely observed in previous studies.\(^{25}\) The position of the absorption peak confirms that green synthesized CeO\(_2\) NPs possess fluorite cubic morphology,\(^{26}\) as justified in Figures 4 and 5.

TGA studies

TGA of biosynthesized CeO\(_2\) NPs was performed to study the capping action of biomolecules and the thermal behavior of the synthesized CeO\(_2\) NPs, which is shown in Figure 9. TGA represents different stages of decomposition. Weight loss occurring in the first step at temperatures from 25°C to 115°C was ascribed to the loss of surface water molecules.\(^{27}\) The minor weight loss that occurred in the second step at temperatures from 50°C to 200°C was due to the combustion of bioactive surface organic components acting as capping agents. The slight weight loss occurring at around 600°C was because of oxygen loss at high temperature.\(^{30}\) Our experimental observations show total weight loss of 0.52 mg (8.8%) of the sample.

Antibacterial analysis of CeO\(_2\) NPs

The antibacterial assay was performed against G\(^+\)ve and G\(^−\)ve bacterial entities using CeO\(_2\) NPs sample loaded at a concentration of 20 µg/0.5 µL on disks. Figure 10 reflects the measurements in size of ZOI around CeO\(_2\) NPs poured disks. In the case of K. pneumoniae, S. aureus, and P. aeruginosa, a ZOI of 9, 10, and 8 mm was observed, respectively, which shows mild to moderate antibacterial behavior. Promising ZOI of 19 mm was recorded against E. coli; the prominent difference in ZOI might be due to the unique cell membrane structure that helps the bacterium to resist antimicrobial agents. Besides all these, other features such as the rate at which the NPs diffuse also play an important role against various bacterial strains.\(^{31}\) Figure 10 shows that G\(^+\)ve bacteria are more susceptible to CeO\(_2\) NPs than G\(^−\)ve bacteria;
similar trends were reported before. Green synthesized NPs have proven efficiency and comparatively low genotoxic and cytotoxic behavior toward healthy cells, when compared to NPs synthesized by various chemical methods. CeO₂ NPs fabricated by way of chemical methods were also found to be cytotoxic to somatic cells. O. europaea leaf extract also takes part in inhibiting K. pneumoniae, S. aureus, and P. aeruginosa, with ZOI of 8, 7.6, and 7.8 mm, respectively; its potential activity has also been reported in previous studies. Bioactive compounds from the leaf extract, such as oleuropein, are also found to be magically active in inhibiting several bacterial pathogens.

From the results of XRD, SEM, and TEM, the average CeO₂ NP size was found to be 24 nm, with the smaller crystallite size being 6 nm. Higher surface area due to smaller crystal size results in greater antibacterial activity. The antibacterial activity of CeO₂ NPs primarily depends upon the electrostatic attraction between positively charged NPs and negatively charged bacterial cell surface and is crucial for the activity of NPs as a bactericidal agent. The interaction is also responsible for reactive oxygen species (ROS) generation, which ultimately leads to bactericidal effect. ROS include the highly reactive hydroxyl radical (OH), singlet oxygen (O₂), and the least toxic superoxide anion radical (•O₂⁻), contributing to the major oxidative stress in biological systems. ROS production is closely related to the efficiency of a photocatalyst, depending on the generation rate, rate of migration, and energy levels of the photoexcited electron–hole pairs. In the present study, such an increase in ROS levels is perhaps linked to the electronic and microstructure properties like grain size, specific surface area, pore size, and so on. The mechanism of photoexcited generation of ROS can be given as follows:

\[
\text{CeO}_2 + h\nu \rightarrow \text{e}^- + \text{h}^+ + \text{H}_2\text{O} \rightarrow \text{OH} + \text{H} + \text{e}^- + \text{O}_2 \\
\rightarrow \text{O}_2^- + \text{H}^+ \rightarrow \text{HO}_2^- + \text{H}_2\text{O} + \text{H}^+ + \text{e}^- \rightarrow \text{H}_2\text{O}_2
\]

Mostly, it is believed that NPs release ions which interact with the thiol (–SH) group of the protein residing at the bacterial cell surface, resulting in denaturation of surface protein along with loss of cell membrane permeability, ultimately causing cell death. The rough surface for metallic NPs also intensifies their mechanical bactericidal capabilities of E. coli. In our experimentation, UV analysis showed strong, intensified CeO₂ NPs absorption peaks at 315 nm. It is a well-known fact that Ce, as compared to its bulk, forms low band gap energy which results in its photoexcitation at room temperature. This phenomenon will also assist in mass production of ROS inside bacterial cellular environment, and intoleration of ROS due to CeO₂ NPs results in malfunctioning of microbial biosynthetic machinery, causing lethality. In Figure 11 is presented a possible schematic diagram summarizing the above fact.

**Antifungal assay analysis of CeO₂ NPs**

The antifungal properties of green synthesized CeO₂ NPs have been shown in Figure 12. A. flavus and A. niger show greater CeO₂ NPs susceptibility with ZOI of 19 mm than F. solani for which the ZOI is 10 mm. The fungicidal activity of CeO₂ NPs was found to be maximum in the case of Mucor species with ZOI of 22 mm, showing great competency as an antifungal agent, like other antifungal candidates.
exact mechanism of the antifungal activity of CeO₂ NPs is not understood, but the antimicrobial potential is probably due to the electromagnetic interaction and ROS generation, like the production of lethal hydroxyl radical (OH) when present in the immediate vicinity of the lipid membrane. ROS cause oxidative deterioration of cell membrane lipids, denaturing the cell membrane permeability with leakage of potassium ions, ultimately causing cell death.

Analysis of CFU counting assay
The results of progression in bacterial growth with time are documented in Figure 13. It is obvious that for all samples, the bacterial population was in lag phase during the first 20 minutes of the experiment, but with the addition of CeO₂ NPs (20 μg/5 μL), persistent decline in growth was observed. Reduction in viability of E. coli was markedly maximum, while for all other bacterial strains, the growth inhibition rate with time was almost similar. So, it is very clear that due to extremely small particle size and pure phase crystallinity with high surface area (Figures 4 and 5), CeO₂ NPs exhibit prominent bacteriostatic and bactericidal activity. This might be due to intolerance of metallic CeO₂ NPs by bacterial metabolic machinery, lethal ROS generation, and greater cytotoxicity toward microbial pathogens. Future work regarding the molecular level interaction of CeO₂ NPs in bacterial growth inhibition needs more exploration.

Conclusion
The leaf extract of O. europaea is proved to be an effective chelating agent for the green synthesis of highly homogeneous, ecofriendly, and extremely small sized CeO₂ NPs. Both XRD and SEM findings confirm the cubic fluorite structure of CeO₂ NPs without any impurities and with a mean crystallite size of 6 nm and an average grain size of 24 nm. FTIR results reveal discharge of bioactive compounds from the leaf extract and Ce–O stretching mode at 459 cm⁻¹. This study shows that CeO₂ NPs exhibit mild to moderate antibacterial activity against both G+ve and G–ve strains, especially high susceptibility was shown by E. coli. CeO₂ NPs exhibited promising inhibitory effect against Mucor species, ensuring their antifungal properties. It can be deduced from this study that CeO₂ NPs synthesized through green chemistry have great potential for future antimicrobial therapies. The follow-up study will focus on differential cytotoxic behavior of photosynthesized CeO₂ NPs to healthy and cancer cells; more specifically, the size-dependent anticancer and antimicrobial properties will be examined.

Acknowledgments
The authors acknowledge Sidra Ifikhar (Department of Mathematics, University of Wah, Pakistan) for providing technical assistance regarding SEM and XRD operations.

Disclosure
The authors report no conflicts of interest in this work.

References


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