Spermine-modified *Antheraea pernyi* silk fibroin as a gene delivery carrier

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**Abstract:** The development of a novel cationized polymer used as a gene delivery carrier that can conveniently and effectively transfect cells resulting in a stably expressed target gene remains a challenge. *Antheraea pernyi* silk fibroin (ASF) is a cytocompatible and biodegradable natural polymer, and it possesses Arg–Gly–Asp sequences but a negative charge. In order to render ASF amenable to packaging plasmid DNA (pDNA), spermine was used to modify ASF to synthesize cationized ASF (CASF), which was used as a gene delivery carrier. CASF was characterized using trinitrobenzene sulfonic acid assay, the zeta potential determination, and a Fourier transform infrared analysis, and the results of these characterizations indicated that the –NH$_2$ in spermine effectively reacts with the –COOH in the side chains of ASF. Spermine grafted to the side chains of ASF resulted in the conversion of the negative charge of ASF to a positive charge. CASF packaged pDNA and formed CASF/pDNA complexes, which exhibited spherical morphology with average particle sizes of 215–281 nm and zeta potential of approximately +3.0 mV to +3.2 mV. The results of the MTT assay, confocal laser scanning microscopy, and flow cytometry analysis in a human endothelial cell line revealed that CASF/pDNA complexes exhibited lower cytotoxicity and higher transfection efficiency compared to the pDNA complexes of polyethyleneimine. These results indicate that our synthesized CASF, a cationized polymer, is a potential gene delivery carrier with the advantages of biodegradability and low cytotoxicity.

**Keywords:** silk fibroin, spermine, cationized polymer, gene delivery carrier

**Introduction**
Gene therapy is an important approach to treat a variety of diseases. The principal step in gene therapy is the delivery of nucleic acids into cells. Gene delivery systems include both viral and nonviral delivery systems. Viral delivery systems have high gene delivery efficiency, but they also have disadvantages, such as immunogenicity, the possibility of permanent insertion into the host genome, and difficulties in handling. In contrast, nonviral delivery systems have received substantial attention because they can be structurally varied, are relatively safe, and can carry large and diverse amounts of genetic materials into cells. The effectiveness and biocompatibility of nonviral plasmid DNA (pDNA) delivery systems based on cationized liposomes or chemically synthesized polymers have developed over the past decade. Several polymeric gene delivery systems have been investigated, such as polyethyleneimine (PEI), poly-(l-lysine), chitosan, and dendrimers. Among these polymers, PEI is a typical nonviral gene delivery carrier and is often considered the gold standard of gene transfection. Branched PEI (25 kDa) has produced good results in gene therapy experiments. It has a relatively high transfection efficiency based on the proton sponge effect, which can protect DNA from endosomal or lysosomal degradation. Nevertheless, PEI with a high molecular weight is cytotoxic and is not easily...
biodegradable by cells. Therefore, a key challenge for nonviral pDNA delivery systems is the development of a new cationized polymer that not only effectively transfers therapeutic DNA into the targeted cells but also exhibits biodegradability and noncytotoxicity.

Silk fibroin produced by the silkworm is a natural protein that has been widely used in tissue engineering scaffolds and as a drug delivery carrier due to its biocompatibility, bioactivity, and biodegradability. Bombyx mori silk fibroin (BSF) and Antheraea pernyi silk fibroin (ASF) are the two most extensively used silk fibroin species. BSF has positive effects on cell adhesion, viability, growth, and differentiation and has been widely used in reconstruction of skin, vascular tissue, and cartilage tissue. Compared to BSF, the amino acid composition of ASF contains abundant tripeptide sequences of Arg–Gly–Asp (RGD), which are favorable for cell adherence. In vitro investigations have shown that the use of ASF matrices results in greater cell attachment and growth than BSF matrices. In our previous studies, we used ASF to coat PEI/pDNA complexes and transfected ASF/PEI/pDNA complexes into L929, HEK293, and HCT 116 cells. The results revealed that ASF reduces the cytotoxicity of PEI/pDNA complexes by shielding the excess positive charge of PEI and improves transfection efficiency. However, pDNA must be condensed by the positively charged PEI, as negatively charged ASF did not package and condense pDNA in previous studies. If cationized ASF (CASF) is synthesized by grafting compounds with low molecular weights and high densities of positive charge onto the side chains of ASF, then, in the presence of a suitable positive charge density in CASF, exogenous DNA can be packaged and transfected into cells through endocytosis. Such a gene delivery carrier would be completely biodegradable and exhibit low cytotoxicity.

Spermine is a biogenic tetraamine with two primary and two secondary amino groups. Spermine is present in all eukaryotic cells and is involved in cellular metabolism. It reversibly condenses DNA and is routinely used in DNA isolation. We hypothesized that some of the negative charges on ASF can be shielded and changed to positive charges after spermine is grafted onto the side chains of ASF, and that this process can be used to synthesize CASF. CASF can package exogenous DNA directly and exhibits advantages such as biodegradability and low cytotoxicity; therefore, CASF can be used as a gene delivery carrier. We present a novel biodegradable and bioactive CASF for use as a gene delivery vehicle through the reaction of the –NH₂ in spermine with the –COOH in ASF side chains in the presence of 1-ethyl-(3-dimethylaminopropyl)-3-carbodiimide hydrochloride (EDC). CASF was used to package pDNA to form CASF/pDNA complexes, which were used to transfect a human endothelial cell line (EA.hy926) in vitro. Cytotoxicity assays, flow cytometry analysis, and confocal microscopy observation revealed that the CASF/pDNA complexes exhibited lower toxicity and higher transfection efficiency than PEI/pDNA.

Materials and methods

Materials

A. pernyi silk fibers were purchased from Liaoning Province (People’s Republic of China). PEI (branched; molecular weight: 25 kDa), EDC, 3-(4, 5-dimethyl-2-thiazolyl)-2, 5-diphenyl-2-H-tetrazolium bromide (MTT), ethylenediamine, and trinitrobenzene sulfonic acid (TNBS) were purchased from Sigma-Aldrich (St Louis, MO, USA). Plasmid DNA encoding enhanced green fluorescent protein (EGFP; 4,733 bp; Genbank accession number: U55762) was purchased from Clontech (Mountain View, CA, USA), amplified in the DH5α Escherichia coli strain, and purified using a Qiagen Plasmid Mega Kit (Qiagen, Chatsworth, CA, USA). The concentration of plasmid DNA was determined by measuring the ultraviolet (UV) absorbance at 260 nm. The human endothelial cell line EA.hy926 (CRL-2922™) was purchased from the American Type Culture Collection (ATCC, Manassas, VA, USA). The experiments were formally reviewed and approved by the ethics committee of Soochow University. Reviewed by the ethics committee, the experimental design and implementation have fully considered the principle of security; the experimental content did not exist potential damage and risk, and followed the principles outlined in the Declaration of Helsinki.

Preparation of ASF solution

ASF solution was prepared following the procedure described previously. Briefly, the A. pernyi silk fibers were degummed by three treatments with 2.5 g/L Na₂CO₃ aqueous solution at 98°C–100°C for 30–45 minutes. After being rinsed and air-dried at 60°C, 10 g of the A. pernyi silk fibers were dissolved in 100 g of melted calcium nitrate tetrahydrate at 100°C for 5 hours. The cooled solution was dialyzed (molecular weight cutoff: 9–12 kDa) against deionized water for 3 days to obtain the regenerated ASF solution with a concentration of ~2.2 wt%. The concentration of ASF solution was adjusted to 2.0 wt% with deionized water.

Preparation of CASF

First, 50 mL of ASF solution (2.0 wt%) was stirred in an ice bath. Spermine was then added to the solution at 0.5 wt%,
1.0 wt%, 1.5 wt%, 2.0 wt%, 4.0 wt%, 6.0 wt%, 8.0 wt%, and 10.0 wt% of the ASF weight in solution. The pH value of the mixture was adjusted to 7.0 by the addition of hydrochloric acid solution. EDC was added to the mixture to account for 20 wt% of the weight of ASF in solution. The mixture was then stirred for 10–12 hours in an ice bath. Finally, the CASF solution was obtained by dialysis against deionized water at 4°C for 72 hours to remove excess reagents.

**Determination of free amino groups**

Free amino group content was determined using the TNBS assay.\(^{35}\) Ethylenediamine was used as a standard. Standard solutions (0.5–7.5 \(\mu\)mol, 1 mg/mL) and test samples (10 mg/mL, 5 mL) were prepared. To each solution, 200 \(\mu\)L of 15 mg/mL TNBS was added and diluted to 3 mL with double-distilled water; then, 1 mL of 0.8 M NaHCO\(_3\) was added to the mixture. The reaction continued for 2 hours at 37°C, followed by the addition of 3 mL of 1 M HCl. UV absorbance was measured at 360 nm using a spectrophotometer (UV-3600; Shimadzu, Tokyo, Japan). The sample absorbance readings were changed to reflect free amino group content using a standard curve obtained with ethylenediamine.

**Measurement of the zeta potential**

The zeta potential of CASF was measured in triplicate using a zeta potential analyzer (Malvern Zetasizer Nano ZS90; Malvern Instruments, Malvern, UK) at 25°C.

**Fourier transform infrared measurement**

ASF, CASF (modified with 4 wt% spermine against ASF), ASF + spermine (a mixture of spermine and ASF at a spermine/ASF ratio of 4/100), and spermine solution were dried at 25°C and cut into particles \(\leq 40 \mu\)m wide. Approximately 1 mg of these particles was pressed into a pellet with 200 mg of potassium bromide. Fourier transform infrared (FTIR) analysis was performed with a Nicolet 5700 spectrometer. The measurements were taken in the range of 400–4,000 cm\(^{-1}\).\(^{36}\)

**Preparation of CASF/pDNA complexes**

CASF at a concentration of 10 mg/mL was filtered (0.22 \(\mu\)m, EMD Millipore, Billerica, MA, USA). Suspensions of CASF/pDNA complexes with weight ratios (CASF to pDNA) of 8/1, 16/1, 32/1, 48/1, and 64/1 were prepared by mixing CASF solutions with pDNA, followed by gentle vortexing for 30 seconds and incubation for 30 minutes at 4°C. PEI/pDNA (w/w, 3/2)\(^{30}\) was chosen in this study as a control. The preparation method of complex suspension of PEI/pDNA (3/2) used was similar to that of CASF/pDNA.

**Gel retardation assay**

The binding ability between CASF and pDNA was investigated using agarose gel electrophoresis. Complex suspensions of CASF/pDNA (weight ratios 8/1, 16/1, 32/1, 48/1, and 64/1) and ASF/pDNA (64/1) were prepared according to the procedure described earlier. These complex suspensions were loaded onto a 1.0 wt% agarose gel with ethidium bromide (0.5 \(\mu\)g/mL) and run with Tris-acetate buffer at 100 V for 30 minutes. Finally, the results were observed by irradiation under UV light.

**Characterization of CASF/pDNA complexes**

The suspensions of CASF/pDNA complexes (weight ratios 32/1, 48/1, and 64/1) were checked for size and surface charge using a zeta potential analyzer at 25°C. The morphology of the CASF/pDNA (weight ratio 48/1) complexes was observed with scanning electron microscopy (SEM; Hitachi S-4800; Hitachi Ltd., Tokyo, Japan).

**Cell culture**

EA.hy926 cells were cultured in Dulbecco’s Modified Eagle’s Medium (DMEM; Gibco, Thermo Fisher Scientific, Waltham, MA, USA) supplemented with 10% fetal bovine serum (FBS, Gibco) and 1% penicillin/streptomycin (Gibco). The cells were cultured in a 37°C incubator with 5% CO\(_2\).

**Cytotoxicity assay**

Evaluation of cytotoxicity was performed using the MTT assay. The EA.hy926 cells were seeded in 96-well plates at a density of \(5 \times 10^4\) cells per well and allowed to attach overnight. Then, the cells were incubated with the complexes in serum-free DMEM. After 4 hours, the medium was replaced with 200 \(\mu\)L of fresh DMEM medium containing 10% FBS for an additional 1 day or 2 days. Thereafter, the medium was removed, and 20 \(\mu\)L of MTT solution (5 mg/mL in DMEM) was added to each well; the cells were incubated at 37°C for 4 hours. Formazan crystals that formed in the living cells were dissolved in 100 \(\mu\)L of isopropanol containing 4\(\mu\)L HCl per well for 30 minutes with slight shaking. Absorbance was determined using a microplate reader (Synergy HT, Bio-Tek, Winooski, VT, USA) at 490 nm. Cell viability was evaluated by calculating the percentage of sample absorbance compared to that of the control after subtracting the blank absorbance. All experiments were repeated three times.

**In vitro gene transfection**

The EA.hy926 cells were seeded in 20 mm cover glass bottom cell culture dishes (NEST, Lot No 2013008) at a density...
of $10^7$ cells per dish and allowed to reach 80% confluence. Complex suspensions of CASF/pDNA (weight ratios 32/1 and 48/1) and PEI/pDNA (3/2) were freshly prepared before transfection, and the medium in each well was exchanged for fresh serum-free DMEM. Complexes were added to the dishes with serum-free DMEM, and the cells were incubated with the complexes for 4 hours. The medium was then removed and replaced with complete medium. After 48 hours, the cells were observed using confocal laser scanning microscopy (FV1000; Olympus, Tokyo, Japan). Levels of GFP expression were determined by flow cytometry analysis. The cells were seeded in six-well plates and transfected using the method described earlier. After 48-hour post-transfection, cells were trypsinized, washed, and resuspended in phosphate-buffered saline. Cell-associated fluorescence was quantitatively determined by green fluorescence emissions measured using a FACSCalibur flow cytometer (BD Biosciences, San Jose, CA, USA). Untransfected EA.hy926 cells were used as background controls.

### Statistical analysis

Data were expressed as the mean ± standard deviation (SD). Significant differences between groups were tested by one-way analysis of variance. Statistical significance was set in advance to a probability level of 0.05.

### Results

#### Characterization of CASF

As shown in Figure 1, the content of free amino groups in unmodified ASF was $\sim 3.7 \mu mol/mL$. After modification with 2 wt% and 4 wt% spermine against ASF weight in solution, the concentration of free amino groups in ASF increased to $\sim 4.4 \mu mol/mL$ and $\sim 5.4 \mu mol/mL$, respectively. Compared to ASF modified with 4 wt% spermine, the concentrations of free amino groups in 6 wt%, 8 wt%, and 10 wt% spermine-modified ASF were not significantly different. These results illustrated that using 4 wt% spermine against ASF to modify ASF can efficiently increase the concentration of free amino groups in ASF side chains.

The zeta potential values of different ratios of spermine against ASF were measured at various weight ratios ranging from 0% to 10% (Figure 2). The zeta potential of ASF before modification was $\sim 9.2$ mV. Using 0.5 wt% of spermine against ASF changed the negative surface charge of ASF to a positive charge and converted ASF into CASF, at which point the zeta potential of CASF sharply increased to $\sim 1.7$ mV. When the weight of spermine used against ASF was 2% or 4%, the zeta potential of CASF continued to increase significantly (to $\sim 5.5$ mV and $\sim 6.8$ mV, respectively). As the weight of spermine against ASF increased, there was no significant effect on the increase in the zeta potential of CASF. The results indicated that using 4 wt% of spermine against ASF to modify ASF not only increases the concentration of free amino groups in the side chains of ASF but also changes the surface charge of ASF from negative to positive and increases the zeta potential to approximately $\sim 6.8$ mV. Subsequent experiments used CASF modified with 4 wt% of spermine against ASF.

FTIR spectra provided further evidence for the successful modification of ASF by spermine. Figure 3 shows the FTIR spectra of CASF, ASF, ASF + spermine, and spermine. The characteristic absorption bands of ASF are at 1,655 cm$^{-1}$ (amide I), 1,540 cm$^{-1}$ (amide II), 1,236 cm$^{-1}$ (amide III), 965 cm$^{-1}$ (amide IV), and 655 cm$^{-1}$ (amide V).

![Figure 1](https://www.dovepress.com/)

**Figure 1** TNBS assay for the determination of free amino groups in spermine-modified ASF.

**Notes:** *P* < 0.05 compared with unmodified ASF. The error bars represent the standard deviation of the free amino groups data.

**Abbreviations:** ASF, *Antheraea pernyi* silk fibroin; TNBS, trinitrobenzene sulfonic acid.

![Figure 2](https://www.dovepress.com/)

**Figure 2** Zeta potential of spermine-modified ASF at different weight ratios of spermine against ASF.

**Note:** The error bars represent the standard deviation of the zeta potential data.

**Abbreviation:** ASF, *Antheraea pernyi* silk fibroin.
Similar absorption bands have been reported previously. Compared with the spectrum of ASF (Figure 3b), there were no significant changes in the spectrum of the mixture of ASF and spermine (Figure 3c), indicating that spermine mixing with ASF cannot change the structure of side chains of ASF. Several changes were found in the spectrum of CASF (Figure 3a) in contrast with uncationized ASF (Figure 3b). The significant absorption at 965 cm$^{-1}$ in Figure 3a compared with Figure 3b could be attributed to the O=C–N in-plane bending vibration, which was caused by the reaction between –COOH in the side chains of ASF and –NH$_2$ in spermine. The peak at 3,293 cm$^{-1}$ in Figure 3b shifted to 3,409 cm$^{-1}$ in Figure 3a, attributable to the new N–H or =N–H symmetrical and asymmetrical stretching vibrations, which appeared near 3,373 cm$^{-1}$ in the spectrum of spermine (Figure 3d). Compared with unmodified ASF, the amide I and amide II absorption bands of modified ASF at 1,655 cm$^{-1}$ and 1,540 cm$^{-1}$ shifted to lower wavenumbers (1,635 cm$^{-1}$ and 1,520 cm$^{-1}$, respectively). The amide V absorption band at 655 cm$^{-1}$ shifted to a higher wavenumber, 698 cm$^{-1}$, suggesting that the chemical environment of the amide bonds that were formed by the reaction of –COOH in the side chains of ASF with the –NH$_2$ in spermine was significantly different from that of the peptide bonds in the main ASF chains. These results demonstrated that ASF was successfully modified by spermine.

**Agarose gel retardation assay and morphology of CASF/pDNA complexes**

Negatively charged pDNA was packaged by positively charged CASF via electrostatic interaction to form CASF/pDNA complexes (Figure 4A). The morphology of the CASF/pDNA (w/w, 48/1) complexes was observed by SEM (Figure 4B). The micrographs show that most complexes were uniform nanospheres ~130–260 nm in diameter. To investigate the binding capability of CASF to pDNA, an agarose gel retardation electrophoresis assay was performed at different weight ratios of CASF/pDNA (Figure 4C). The DNA band in Lane 2 was similar to that of naked pDNA (Lane 1), which indicated that unmodified ASF cannot package pDNA. However, with an increasing CASF/pDNA weight ratio, a corresponding reduction in migration was observed. When the weight ratio of CASF/pDNA increased to 16/1 (Lane 4), most of the pDNA was retained within the well, with only faint signs of migration; when the weight ratio increased to 32/1, 48/1, or 64/1 (Lanes 5, 6, and 7), there was absolutely no sign of DNA migration. These results suggest that, when the CASF/pDNA weight ratio was equal to or...
greater than 32/1, CASF could completely package pDNA. Complexes with CASF/pDNA weight ratios of 32/1, 48/1, and 64/1 were chosen for subsequent experiments.

Characterization of CASF/pDNA complexes

As shown in Table 1, the zeta potentials of three weight ratios of CASF/pDNA complexes were +3.0 ± 0.2 mV, +3.2 ± 0.5 mV, and +3.2 ± 0.2 mV. Compared to CASF/pDNA (weight ratio 48/1), the zeta potentials of CASF/pDNA (weight ratio 64/1) complexes were not increased. The average diameters of CASF/pDNA complexes at weight ratios of 32/1, 48/1, and 64/1 were 215.4 ± 4.9 nm, 234.7 ± 5.0 nm, and 280.9 ± 7.7 nm, respectively. The particle size of CASF/pDNA complexes at 64/1 weight ratio was the largest of the three weight ratios because excess CASF was absorbed at the surface of the complexes. Compared with CASF/pDNA complexes of weight ratios 32/1 and 48/1, the zeta potential of CASF/pDNA complexes at weight ratio 64/1 was not increased, but the particle size significantly increased. Small particle sizes lead to higher levels of cellular uptake, and therefore these might be favorable for transfection. In subsequent experiments, the highest weight ratio of modified ASF against pDNA was used (weight ratio 48/1).

Cytotoxicity of CASF/pDNA complexes

The effects of CASF on the viability of EA.hy926 cells were evaluated by an MTT assay in comparison with PEI (Figure 5). After 1 day of culture, cell viability significantly improved after transfection with CASF/pDNA complexes compared with transfection with PEI/pDNA (P < 0.05). The results on the second day were similar to those on the first day, and the cell viability of the CASF/pDNA-transfected EA.hy926 cells was still significantly higher than that of the PEI/pDNA-transfected cells (P < 0.05). This result implies the great potential of CASF as a low-toxicity gene delivery carrier.

In vitro transfection of CASF/pDNA complexes

In this study, 2 µg of pDNA was delivered to achieve visible expression after 48 hours. Green fluorescent images

Table 1 The zeta potential (mV) and particle sizes of CASF/pDNA complexes at different CASF/pDNA weight ratios

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<tr>
<th>CASF/pDNA weight ratio</th>
<th>Zeta potential (mV)</th>
<th>Particle size (nm)</th>
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<tr>
<td>32/1</td>
<td>+3.0 ± 0.2</td>
<td>215.4 ± 4.9</td>
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<tr>
<td>48/1</td>
<td>+3.2 ± 0.5</td>
<td>234.7 ± 5.0</td>
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<tr>
<td>64/1</td>
<td>+3.2 ± 0.2</td>
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Abbreviations: CASF, cationized Antheraea pernyi silk fibroin; pDNA, plasmid DNA.
of transfected EA.hy926 cells were obtained by inversion fluorescence microscopy. As shown in Figure 6, there were greater number of fluorescent cells after transfection with CASF/pDNA complexes (Figure 6A and C) than with PEI/pDNA complexes (Figure 6E). The morphology of the transfected cells was normal, with polygonal characteristics, after transfection with CASF/pDNA complexes (Figure 6B and D), but cells in the PEI/pDNA complex-transfected group appeared to have shrunk and decreased in number (Figure 6F). This result further indicated that CASF possesses lower cytotoxicity than PEI. The transfection efficiency was confirmed by flow cytometry (Figure 7). The transfection efficiencies of the complexes of CASF/pDNA at weight ratio 32/1 and CASF/pDNA at weight ratio 48/1 were 15.8%±2.0% and 13.8%±2.3%, respectively, higher than the values of 12.0%±1.4% for the PEI/pDNA complexes. The untreated cells showed the lowest levels of transfection efficiency, namely, only 1.6%±0.2%.

**Discussion**

This is the first report of the synthesis of CASF and its use as a gene delivery carrier. In order to synthesis CASF, ASF whose molecular weight was distributed from ~30 kDa to ~300 kDa was modified with spermine in the presence of EDC. Carbodiimide coupling is a common method used for the reaction of primary amines with carboxylic acids, resulting in the formation of an amide bond. ASF contains ~8.66 mol% of acidic amino acid residues. Spermine...
Figure 7 (A) Flow cytometric analysis in EA.hy926 cells after being transfected with CASF/pDNA complexes at weight ratios of (a) 32/1 and (b) 48/1, as well as (c) PEI/pDNA complexes at weight ratio of 3/2. (d) Control group (untreated cells). (B) Transfection efficiency of the two complexes at various weight ratios.

Note: The values given are mean ± SD (n = 3).

Abbreviations: CASF, cationized Antheraea pernyi silk fibroin; pDNA, plasmid DNA; PEI, polyethyleneimine; SD, standard deviation.

is a polyamine that has two primary and two secondary amino groups. As illustrated in Figure 8, EDC was coupled with the carboxyl groups in the aspartic and glutamic side chains of ASF and formed unstable intermediate urea derivatives (O-acylisourea). Then, O-acylisourea was able to react with the primary amine groups in spermine, resulting in the formation of an amide linkage between ASF and spermine to achieve CASF. In this modification process, the self-crosslinking of –COOH and –NH₂ of ASF is a possible reaction. However, the spermine was excessive in the reaction system, and the –NH₂ of spermine was dominant around the –COOH in ASF side chains; therefore, the –COOH of ASF would preferentially react with the –NH₂ of spermine to form CASF. CASF was purified by dialysis and characterized based on the TNBS assay, zeta potential, and FTIR. These measurements confirmed the introduction of spermine into the side chains of ASF. The TNBS assay provided quantitative evidence that free amino groups had been introduced into the ASF (Figure 1). Zeta potential measurements (Figure 2) revealed that the grafting of spermine brings the zeta potential of ASF from negative to positive. When the weight ratio of spermine against ASF was >4%, the zeta potential and free
amino groups were not significantly increased in spermine-modified ASF. The free amino groups and zeta potential of CASF increased significantly as the amount of spermine increased in the initial reaction period. However, with further increase of spermine input amount, on the one hand, the primary amines of spermine reacted with the −COOH groups in ASF, resulting in the decrease of remaining −COOH groups. On the other hand, one spermine could react with two −COOH groups in ASF to form ASF–spermine–ASF because spermine has two primary amines, which accelerated the consumption of −NH₂ and −COOH groups. Consequently, the zeta potential of CASF did not continue to increase. The concentration of free amino groups depended on the content of spermine that was grafted onto the side chains of ASF, but zeta potential was attributed to the ratio of the amines to the carboxyl groups. After modification by spermine, although the increase in amines by spermine grafting was ~35%, the content of carboxyl groups on the ASF surface was obviously decreased. The ratio of the amines to the carboxyl groups significantly increased, which led to the significant increase of zeta potential. FTIR spectra (Figure 3) provided further evidence for the successful synthesis of CASF. The significant absorption at 965 cm⁻¹ in the spectrum of CASF was attributed to the O=C−N in-plane bending vibration, which was caused by the reaction between the −COOH in the side chains of ASF and the −NH₂ in spermine. These results indicated that spermine has been grafted onto the side chains of ASF.

Gene delivery systems based on cationized polymers are significantly affected by the efficiency of DNA condensation. In this study, positively charged CASF could form complexes with negatively charged pDNA via electrostatic interaction (Figure 4A). In the preparation process of CASF/pDNA nanoparticles, a mixture of the ASF/spermine/ASF–spermine conjugate/pDNA was probably formed; however, the resulting nanoparticles were mainly CASF/pDNA in our experimental system. If the nanoparticles were ASF/spermine/ASF–spermine conjugate/pDNA, their zeta potential would have been negative. In a previous report, spermine was shown to bind poorly to DNA, resulting in the loss of DNA protection against nuclease degradation, and could not mediate gene transfer at all. Thus, the spermine/pDNA complex was not taken as a control in the gel electrophoresis and transfection studies. A gel retardation assay (Figure 4C) showed that CASF was able to package pDNA completely at a weight ratio of 32/1. The CASF displayed a positive charge. Thus, when pDNA combined with the different weight ratios of CASF, the zeta potential would change from a negative charge to a positive charge. The surface zeta potentials of CASF/pDNA complexes in this study were approximately +3.0 mV to +3.2 mV, which is conducive for the complexes to bind to anionic cell surfaces to facilitate cellular uptake. Previous studies have indicated that cells typically take up particles with sizes in the range of ~50 nm to several hundred nanometers. In this study, the average size of the formed CASF/pDNA complexes is 215–281 nm, which is suitable for transfection. The morphology of the CASF/pDNA weight ratio 48/1 complexes observed by SEM was spherical, with diameters in the range of 130–260 nm. But the average particle size of the CASF/pDNA weight ratio 48/1 complexes determined by Zetasizer was ~234.7±5.0 nm. Differences in size measurement between SEM imaging and Zetasizer measurements were attributed to the dried state of complexes for SEM measurement versus hydrated state of particles for Zetasizer measurements. Cytotoxicity is a major concern in the application of gene carriers. ASF has shown negligible level of toxicity in many cell lines. Moreover, no significant cytotoxicity of pullulan-spermine/DNA in sensory neurons was detected even at 20 mg/mL DNA, which is tenfold higher than the concentration conventionally used for transfection. Hence, in this study, we only investigated the cytotoxicity of CASF/pDNA nanoparticles in Figure 5. Compared with PEI/pDNA 3/2 ratio, the viability of cells transfected with CASF/pDNA weight ratio 32/1 and CASF/pDNA weight ratio 48/1 was significantly improved. High molecular weight PEI is relatively cytotoxic because of the high density of its cationic groups and its

Figure 8 Schematic presentation of modification reaction of ASF with spermine in the presence of EDC.

Abbreviations: ASF, Antheraea pernyi silk fibroin; CASF, cationized A. pernyi silk fibroin; EDC, 1-ethyl-3-(dimethylaminopropyl)-3-carboxydimide hydrochloride.
nondegradability.57–59 In this study, CASF/pDNA complexes had a low density of positive charges. The zeta potential of the CASF/pDNA complexes was only approximately +3.0 mV to +3.2 mV, much lower than the +18.0 mV zeta potential of the PEI/pDNA complexes.30 The positive charge of the CASF/pDNA complexes was beneficial to enhance the complexes’ contact with the cell membrane, and the low density of positive charges avoided destroying the cell membrane permeability when the CASF/pDNA complexes interacted with cells. The biodegradability of CASF allows CASF/pDNA complexes to be metabolized and eliminated, thereby showing low cytotoxicity.

The gene transfer ability of the CASF/pDNA complexes was evaluated in EA.hy926 cells. Both qualitative (Figure 6) and quantitative (Figure 7) data revealed that the transfection ability of CASF/pDNA was higher than that of PEI/pDNA. CASF formed by grafting spermine onto ASF not only possessed the ability to package pDNA but also exhibited low cytotoxicity and, therefore, effectively transfected the exogenous gene into cells.

Conclusion

In this study, spermine was used to modify ASF. The obtained CASF was used to package pDNA, and the resulting CASF/pDNA complexes were transferred into EA.hy926 cells. Characterizations using the TNBS assay, zeta potential, and FTIR analysis indicated that the –NH in spermine effectively reacted with the –COOH in the side chains of ASF in the presence of EDC. The concentration of free amino groups in CASF was significantly increased via the formation of amide bonds, converting the negative charge of ASF to a positive charge. CASF effectively packaged and condensed pDNA, and the CASF/pDNA complexes exhibited lower cytotoxicity and higher transfection efficiency than PEI/pDNA because CASF possessed excellent cytocompatibility and biodegradability. CASF may be applicable as a novel delivery carrier for clinical gene therapy.

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Disclosure

The authors report no conflicts of interest in this work.

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