Size- and shape-dependent clinical and mycological efficacy of silver nanoparticles on dandruff

Mohammad F Anwar¹Deepak Yadav²Swati Jain³Sumeet Kapoor⁴Shweta Rastogi⁵Indu Arora⁶Mohammed Samim¹

¹Department of Chemistry, Faculty of Science, ²Faculty of Medicine, Jamia Hamdard University, New Delhi, ³Amity Institute of Nanotechnology, Amity University, Noida, Uttar Pradesh, ⁴Centre for Biomedical Engineering, Indian Institute of Technology, Delhi, ⁵Department of Chemistry, Hans Raj College, ⁶Department of Biomedical Sciences, Rajguru College of Applied Sciences for Women, University of Delhi, Delhi, India

Abstract: Dandruff is a prominent scalp problem caused by the growth of fungus *Malassezia furfur*, potentially cascading into dermal inflammation, itching, and tissue damage. The present work outlines a detailed analysis of the treatment of scalp infection using silver nanomaterials (Ag NMs), and focuses on biocidal activity owing to manipulation of size, shape, and structure. Monodisperse silver spherical nanoparticles (NPs) and nanorods (NRs) were synthesized by chemical routes that were characterized using analytical and spectroscopic techniques. Ag NMs demonstrated enhanced biocidal tendencies compared to market available drugs, itraconazole and ketoconazole, showing greater zones of inhibition. The obtained 20 nm and 50 nm spherical-shaped NPs and 50 nm NRs showed concentration-, size-, and shape-dependent antifungal activity, with 20 nm spherical-shaped NPs exhibiting excellent potency. Minimum inhibitory concentration for 20 nm was lowest at 0.2 mg/mL in comparison to 0.3 mg/mL for NRs. Primary irritation index was 0.33 and 0.16 for 20 nm and 50 nm spherical-shaped NPs, respectively, while 50 nm rod-shaped NMs exhibited negligible redness. An in vivo model for *M. furfur* infection was generated by passing fungi subcutaneously in rats' skin. Again, 20 nm particles showed best normalization of skin after 10 days on regular dosing, in comparison with bigger and rod-shaped particles. The statistical clinical score was highest for Ag nanorods, followed by 50 nm Ag NPs-treated animals. It was observed that 20 nm spherical particles exhibited the lowest score (0) compared with others as well as with antifungal drugs. Biochemical analysis performed by checking antioxidant enzymatic activities indicated tissue repair and normalization of enzymes and protein concentration by Ag NPs.

Keywords: *Malassezia furfur*, nanorods, in vivo analysis, Wistar rat model

Introduction

In recent years, skin-related issues pertaining to microbial infections garnered great importance from both scientific and commercial corners. Prominently, *Malassezia furfur* is a domineering fungus that has received much attention from clinicians and dermatologists. It is a lipo-dependent dimorphic and saprophyte fungus that is a causative organism of pityriasis versicolor, seborrheic dermatitis, and dandruff conditions constituting major skin and scalp disorders.¹ These are characterized by dermal inflammation and tissue damage resulting in dryness as well as hyperproliferation, causing flaking, itching, and redness of scalp. These conditions result from proinflammatory-free fatty acids released during lipase activity from the fungi, and *M. furfur* is considered an etiological factor for dandruff.^{2,3}

Currently available mainstay treatment options are based on compounds such as selenium sulfide, imidazoles, and hydropyridones. These antimicrobial agents demonstrate antifungal and antiproliferative effects via disruption of either cellular

Correspondence: Mohammed Samim
Department of Chemistry, Jamia
Hamdard University, Hamdard Nagar,
New Delhi 110062, India
Tel +91 11 2605 9688 ext 5557
Fax +91 11 2605 9663
Email shamim_chem@yahoo.co.in

organization integrity or transport of essential macromolecules. Most of these drugs suffer from poor compliance, clinical efficacy, and result in adverse drug effects such as oiliness.^{4,5}

These limitations have paved the way for novel nanomaterials (NMs) as potential biocidals with an added advantage of mitigating antibiotic-resistant bacteria and other microbes. Significantly essential oils are being explored as skin and hair quality enhancers as well as potential antimicrobial agents,⁶ but their efficiency is unknown and unpredictable.⁷ Among metallic nanostructures, silver has been in practice for a long period of time in the medical sector, including use as wound and burn dressing materials for controlling bacterial infections. Apparently, Ag⁺ ions are easily liberated from nanomaterial structure on contact with water medium (Equation 1).^{8,9}

In contrast to silver ions, nanoparticles (NPs) serve better owing to slow- and long-term release of ions by oxidation of zerovalent as compared to burst release from ion formulations.¹⁰ NPs can easily permeate cell membranes of living cells and have many fold increase in the cellular uptake of these NPs as compared to the bulk particles. Their cellular accumulation crumbles structural and functional integrity of cells, interjecting and obstructing natural physiological processes. They disrupt DNA via phosphorylation process of prokaryotic cell and bind with thiol groups of cell membrane proteins causing cell rupture and death of organisms.¹¹ Reports also suggest NPs are less cytotoxic to mammalian cells than extremely small ions.¹² Recently, NPs have also been explored as antifungal agents against clinically relevant fungi. Ag NPs have primarily been tested to kill *Candida* spp. in a research established by Lee et al.^{13,14} They have reported the effect of biological synthesis of Ag NPs using different fungi and have investigated their effect against antibiotic resistant bacteria and *M. furfur*. Recently, in vitro antimicrobial assessment has also been accomplished by Pant et al,¹⁵ indicating possible use of Ag spherical NPs in antidandruff shampoos and treatment of dermal diseases.

We inspect silver nanomaterials (Ag NMs) as potential fungistatic material for containing and eradicating scalp-related diseases such as dandruff. Dandruff disease was created on rat models by optimizing conditions for passing fungal infection on their skin. We report for the first time a detailed analysis of uptake of silver nanostructures on

diseased rat skin and their treatment with clinical scores as an indication for treatment regime.

Experimental

Materials and methods

Sodium, bis (2-ethylhexyl) sulfosuccinate (AOT [aerosol-OT]), hexane, silver nitrate (AgNO₃), trisodium citrate, sodium borohydride (NaBH₄), and aqueous solution of hydrazine chemicals were purchased from Sigma-Aldrich, St Louis, MO, USA. Antifungal drugs such as itraconazole and ketoconazole were obtained from HiMedia Laboratories Pvt. Ltd., India. NMs syntheses were carried out in ultrapure Millipore water. All solvents and reagents were purchased from commercial sources and were used as received unless noted otherwise.

The inoculum suspensions of *M. furfur* (MTCC-1765) obtained from MTCC (Chandigarh, India) were prepared uniformly by the standard procedure. Male albino Wistar rats weighing 250–350 g used in the present study were procured from the institute's animal house, Jamia Hamdard, New Delhi, under specific pathogen-free conditions and facilities for all the in vivo experiments of clinical efficacy and histopathology. All animals were individually housed in standard size polycarbonate cages with controlled conditions of temperature (23°C±1°C), humidity (55%±10%) in 12 hours:12 hours of light and dark cycle, fed with animal feed (Hindustan Pvt. Ltd., India), and water supplied ad libitum. All animal experiments were performed in compliance with regulations and standards of the ethical committee of the institute (CPCSEA/173).

Synthesis of spherical-shaped silver nanoparticles by microemulsion method

Different sizes and shapes of Ag NPs were obtained by following the chemical processes optimized previously.^{16,17} Reverse micelles of AOT were prepared by dissolving 0.444 g AOT in 10 mL hexane, and precise volume of water containing metal component/hydrazine was added to these solutions. The amount of solubilized water is usually expressed as a molar ratio of water to AOT, that is, $W_0 = [\text{H}_2\text{O}]/[\text{AOT}]$. For the preparation of Ag NPs, 75 µL 0.1 M and 75 µL 1.5 M aqueous solutions of AgNO₃ were added to 5 mL 0.1 M AOT solution. The resulting solution was stirred for 6 hours on magnetic stirrer to form microemulsion (A). In another set of reactions, 75 µL 0.3 M and 75 µL 0.45 M of hydrazine hydrate solutions were added dropwise to 5 mL of 0.1 M AOT solution. The molar ratio of hydrazine hydrate to AgNO₃ was held constant for all the experiments at a value

of three. Similarly, this solution was also stirred for 6 hours by magnetic stirrer to form microemulsion, referred to as B. In order to reduce the silver ions, microemulsion (B) containing hydrazine hydrate was added slowly into microemulsion (A) containing AgNO_3 drop by drop. After the addition of hydrazine hydrate microemulsion, the resultant solution was subjected to continuous stirring for 6 hours. Subsequently, the color of the microemulsion changed from colorless to a yellowish brown color, indicating the formation of Ag NPs.

Synthesis of rod-shaped silver nanoparticles by aqueous-phase method

Aqueous solution of 1×10^{-4} M AgNO_3 and anhydrous solution of 1×10^{-3} M AOT in hexane were prepared. In 100 mL of round bottom flask, 50 mL of AgNO_3 solution and 35 μL of anhydrous AOT solution were mixed by stirring for 1 hour. After mixing 600 μL aqueous solution of 1×10^{-2} M trisodium citrate, it was introduced to above solution and refluxed at 100°C with constant stirring for 70 minutes. After refluxing, the reaction mixture became turbid and turned yellowish green. This solution contained both spherical- as well as rod-shaped Ag NPs that were separated by centrifuging the sample at 2,000 rpm for 30 minutes at 25°C . The Ag nanorods (NRs) settled at the bottom of their container after centrifugation, while AOT and spherical NPs were removed using a pipette. This separation procedure was repeated several times until supernatant become colorless. NRs were washed by centrifugation process and stored in a colored glass bottle at room temperature till further use.

Characterization of nanomaterials

Double beam UV–visual spectrophotometer (UV-1601, Shimadzu, Japan) was used to measure the absorbance of synthesized Ag NMs and record optical density. Hydrodynamic radii and stability of synthesized NPs were investigated using a dynamic light scattering (DLS) instrument (Zetasizer Version, 6.1, Malvern, equipped with vertically polarized He–Ne laser [632.8 nm] and autocorrelator) at ambient temperatures and constant angle of 90° . Shape, size, and morphology of synthesized NPs and NRs were visually assessed by transmission electron microscope (FE-TEM; JEOL, Tokyo, Japan; JEM-2100) at 200 kV acceleration voltage. Samples were prepared by drop coating 10 μL Ag NMs onto carbon-coated copper grids followed by air drying, forming thin layer on surface. X-ray diffraction (XRD) study of completely dried powder of Ag NMs was performed on Philips PW 1830 using a monochromatized X-ray beam with nickel-filtered $\text{CuK}\alpha$ radiation ($\lambda=1.5418 \text{ \AA}$) at 40 kV. A continuous scan mode

was used to collect 2θ data between 30° and 40° , with a 0.02 sample pitch and $4^\circ/\text{min}$ scan rate.

Evaluation of antifungal properties

In vitro assessment of antifungal properties of Ag NMs was carried out via plating methodology. Briefly, *M. furfur*, maintained in glycerol stock 15% (vol/vol) at -80°C , was revived in pityrosporum medium and olive oil broth in an incubator cum shaker (200 rpm) at 28°C to obtain 10^5 CFU/mL of fungi. A total of 0.1 mL of *M. furfur* was plated on specialized potato dextrose agar (PDA) plates and sterile paper disks impregnated with predetermined concentrations, ranging from 0.1 mg/mL to 0.5 mg/mL of synthesized Ag NMs were kept on them. Autoclaved and sterilized specialized PDA agar plates were inoculated with 0.1 mL of *M. furfur* containing 10^5 CFU/mL. Sterile disks of Whatman filter paper (6 mm size) impregnated with 100 μL predetermined concentrations, ranging from 0.1 mg/mL to 0.5 mg/mL, of synthesized Ag NMs were kept on them. These plates were incubated for 28 hours at 35°C and the resulting zones of inhibition (ZOI) were measured in millimeters. A negative control without Ag NMs, constituted by disk impregnated with fungal filtrate, was also included in the experiment. The efficacy of Ag NMs was compared with commercially available drugs ketoconazole and itraconazole at 0.5 mg/mL concentration level. The diameter of fungal colony developing on Ag NM plates was compared with the diameter of the colony obtained on negative control and drugs, and an average of six experiments was used to represent numerical data. minimum inhibitory concentration (MIC)s were noted as the concentration at which no growth was observed.

Skin irritation test and scratching behavior assessment

Five healthy Wistar male rats, weighing 250–350 g, were selected for the study. An area of 2 cm^2 was shaved for each rat to expose sufficient test area, and assay was followed according to Kaur et al.¹⁸ The rats were divided in five groups as group I (control), group II (standard), group III (20 nm spherical-shaped Ag NPs), group IV (50 nm spherical-shaped Ag NPs), and group V (50 nm rod-shaped Ag NPs). Aqueous solution of formalin (0.7%) was applied to standard irritant, and the control was untreated. The test sites were visually observed for erythema and edema daily up to 7 days as scored in Table 1.¹⁹ Scratching behavior of rat models was recorded on a digital camera (Sony W55) by observing their itching tendency for 15 minutes. Rats of groups III, IV, and V were visually observed for developing

Table I Standards for skin irritation study

Skin responses	Score
Erythema and eschar formation	
No erythema	0
Very slight erythema (barely perceptible)	1
Well-defined erythema	2
Moderate to severe erythema	3
Severe erythema (beet redness) to slight eschar formation (injuries in depth)	0
Edema formation	
No edema	0
Very slight edema (barely perceptible)	1
Slight edema (edges of area well-defined by definite raising)	2
Moderate edema (raised ~1.0 mm)	3
Severe edema (raised >1.0 mm and extending beyond exposure area)	4
Total possible score for irritation	8

itching or scratching tendency and compared with group I control untreated rats.

In vivo efficacy of Ag Nano

Animal model for dandruff

An animal infection model of dandruff was performed according to protocol of Tatsumi et al²⁰ and Ghannoum et al²¹ with little modification. Before inoculation, the hairs of both flanks at upper left side and lower right side skin (3 cm × 3 cm) of each rat were removed by electric hair clipper (MOSER, Germany) followed by the application of hair remover cream for complete removal of hairs. The skin was slightly abraded with sandpaper, which makes it more susceptible to infection. A 200 µL of inoculum of fungi containing 2×10^7 CFU/mL was applied on the marked area of skin by using a sterile pipette tip and rubbed thoroughly with a sterile cotton-tipped swab (HiMedia Laboratories Pvt. Ltd.) and was passaged thrice on Albino Wistar rat's skin. This was continuously applied for 5 days on the marked area of skin followed by scrubbing and evaluated for fungal infection.

Animal grouping and dosing

The experimental animals were distributed into seven groups comprising six Wistar rats in each group. Group I rats were designated as control and were treated with saline, and group II was designated as dandruff model and the animals were treated with *M. furfur*. Groups III and IV were treated with commonly available market drugs, itraconazole and ketoconazole, and were used as standards. Groups V, VI, and VII were treated with 200 µL of various sizes and shapes of Ag NPs (20 nm and 50 nm Ag NPs and 50 nm rod-shaped Ag

NRs). The treatment comprising 200 µL of drugs or NMs started on day 1 postinfection and was applied topically daily for 10 consecutive days.

Clinical and mycological evaluation of treatment efficacy of silver nanoparticles

Clinical and mycological end points were used to determine the efficacies of the antifungal agents tested, and the final assessments were performed on the 14th day postchallenge. Changes (mild, moderate, or severe) in redness, ulceration, and scaling or hair loss at the site of inoculation were visually examined and recorded daily. The infected area marked on the back of each animal was divided into four equal quadrants and were scored accordingly. The redness score was graduated as 0, normal; 1, pink; 2, red; and 3, violet and the lesion score was graduated: 0, normal; 1, papule; 2, skin scales; 3, single layer of skin scales and ulcers; and 4, multiple layers of skin scales and ulcers. Areas that had been used for prior specimen sampling were excluded for subsequent clinical evaluation to avoid bias by redness and ulcers due to prior scraping. Scores from the quadrants were summed for each animal (maximum possible score per animal was 20) and used to determine the clinical efficacy of different treatment groups that are expressed as a percentage relative to the infected–untreated control group using the following formula:

$$\text{Percentage efficacy} = 100 - \left(T \times \frac{100}{C} \right) \quad (2)$$

where T is the total score of the test group and C is the total score of the infected–untreated control. The total score for any treatment group signifies the average clinical score from animals in the same group.²¹

Biochemical assays

These assays aimed to estimate potential effect of Ag NMs in rats and the method is described as supplementary data. The measured biochemical parameters comprised total proteins,²² catalase (CAT),²³ superoxide dismutase (SOD),²⁴ glutathione peroxidase (GPx),²⁵ malondialdehyde (MDA),²⁶ and total hydroperoxides (THP).²⁷

Histopathological examination

Ag nano-treated groups were compared with control group to observe the histopathology of skin tissues. On the 14th day of experiment, the rats were anesthetized with ether and perfused transcardially with saline. Skin samples were removed and postfixed in 10% neutral buffered formalin for

24 hours and processed for histopathological examinations.²⁸ After fixation, 3–4 mm slices of tissues were dehydrated and embedded in paraffin. Cross sections of 5 μm thickness were cut and followed by washing with xylene, and then the tissues were mounted with DPX (dibutyl phthalate and xylene). Hematoxylin and eosin stains were used to visualize and differentiate between normal tissue, standard drugs-treated tissue, and Ag nano-treated tissues. Slides were studied for histological changes, and microphotographs were taken using Olympus BX50 microscope (Olympus Corporation, Tokyo, Japan) at Jamia Hamdard, New Delhi, India.

Statistical analysis

The results are expressed in terms of mean $\pm$ standard deviation. Statistical analysis of all the data between groups receiving no infection vs *M. furfur* infected alone along with those receiving *M. furfur* vs Ag NPs-treated infected rats was performed by unpaired two-tailed Student's *t*-test with consideration of *P*-value <0.05 as statistically significant parameter.

Results and discussion

Synthesis and characterization of Ag nanoparticles and nanorods

Successful synthesis of spherical- and rod-shaped Ag NPs was easily facilitated by AOT-based microemulsion process by varying the concentrations of precursor and reducing agent (Figure S1). Route 1 systematically yielded high concentrations of spherical NPs, whereas NRs of native silver were obtained by route 2, which were validated primarily by UV–vis spectroscopy. Synthesized NMs were found to stable for 4 months when kept under appropriate conditions of temperature and humidity.

An increase in hydrodynamic size was recorded for NPs in correlation with different AgNO_3 concentrations in microemulsion. Ag NPs showed an increase of 20.25 ± 1 nm and 50.27 ± 1 nm at 75 μL and 100 μL of 0.1 M AgNO_3 , respectively. This is attributed to enhanced layering of formed seed particles with more native silver atoms gradually released in microemulsion, thereby maintaining polydispersity and stability. Spherical-shaped Ag NPs of 20 nm and 50 nm showed absorption peaks at 406 nm and 425 nm, respectively.

Route 2 of aqueous-phase synthesis proceeds with the reverse micelle formation of water droplets of citrate ions in hexane with self-seeding template-less mechanism stabilized by AOT. The small water droplets act as reactor site for conversion of salt to native metal on reduction in organic solvent.²⁹ The obtained NMs showed two plasma bands in

UV–vis absorption spectrum at 531 nm and 432 nm corresponding to longitudinal and transverse plasmonic bands, respectively, confirming the formation and isolation of Ag NRs (Figure S2). The time scale analysis clearly revealed transformation of particulate structure to rod at 70°C – 80°C , where greenish yellow color indicated reduction of salt by citrate. The exact mechanism of the synthesis is yet to be established, but it is predicted that concentration of mild reducing agent plays a pivotal role in anisotropic reduction of AgNO_3 , usually aspect ratio decreases with decreasing reducing agent.³⁰

The number of water molecules added per molecule of AOT W_0 , present in micelles, is representative of their core size, and it has been observed in AOT microemulsion systems that the water pool radius increases with the water content.³¹ The UV–vis spectra of the Ag NPs are shown in Figure S2 at $W_0=10$ and 15. It was observed that with the increase in water contents in microemulsion, core size increased and formed bigger NPs than with lower water contents, which favorably produce small reverse micelles. The red shift in wave length from 406 to 425 nm (Figure S2) was also observed as the particles size increased from 20 to 50 nm, when W_0 was increased from 10 to 15 at constant silver concentration. At low water content, the water solubilized in the polar core is bound by surfactant molecules, which increases the boundary strength and decreases the intermicellar exchange rate. This process induces formation of monodisperse Ag NPs with small particle size. However, as water content increases, this bounding loosens up, creating macro zones and facilitating water pool exchange via collisions, resulting in coprecipitation of compounds solubilized in two different reverse micelles to complete more quickly. Reactants would be rapidly transferred from one water core to another, and thus the resultant particle size is large and the size distribution becomes relatively wide.

The well-defined shape that can be visualized under TEM (Figure 1) suggested that route 1 results in spherical NPs having narrow size distribution, and aqueous-phase reaction yield rod-shaped particles. TEM micrographic analysis also confirmed the relationship between the size and precursor amount concentration. These analyses established that process 2 – aqueous-phase reaction–yielded rod structures of Ag NMs of 50 nm diameter. Figure 1D illustrates the XRD pattern of powdered Ag NRs that depicted four different peaks at 2θ 39° , 45° , 64° , and 73° corresponding to [1 1 1], [2 0 0], [2 2 0], and [3 1 1] miller indices, respectively;³² therefore, the synthesized Ag NRs are found to be having regular face-centered cubic organization.

Figure 1 TEM micrograph.

Notes: (A) 20 nm spherical-shaped Ag NPs; (B) 50 nm with spherical-shaped Ag NPs; (C) TEM image showing 50 nm of rod-shaped Ag NPs. The insets show the diffraction pattern recorded by aligning the electron beam perpendicular to one of the square faces of an individual nanoparticle. (D) XRD pattern of Ag NR.

Abbreviations: TEM, transmission electron microscopy; NPs, nanoparticles; XRD, X-ray diffraction; NR, nanorod.

In vitro biocidal efficiency

Antifungal activity of the prepared Ag NMs was investigated against *M. furfur*, which is responsible for grave dermatological disorders like pityriasis versicolor and seborrheic dermatitis along with dandruff. Ag NMs and known antifungal drugs such as ketoconazole as well as itraconazole were challenged with an inoculum of 10^5 CFU/mL of fungi, and killing capacity was quantifiably measured by denoting ZOI on pityrosporum PDA medium by agar diffusion method. Figure S3 shows the photographic image of antimicrobial test denoting ZOI for different agents. It was observed that 20 nm NPs created maximum ZOI than 50 nm spherical NPs as well as 50 nm NRs at different concentrations of Ag NMs, ranging from 0.2 mg/mL to 0.5 mg/mL. Pyrrole-based market drugs had no response with negligible ZOI even at highest tested concentration of 0.5 mg/mL as presented in the inset of Figure 2, indicating superior characteristics of Ag NMs.

Spherical-shaped Ag NPs of 20 nm (0.5 mg/mL) exhibited maximum antifungal activity in contrast to 50 nm rod-shaped Ag NPs, demonstrating minimum killing. These results indicate size and shape-dependent antifungal activity of Ag NMs. Smaller-sized Ag NPs are more efficient for antifungal application due to higher breaching of fungal cell wall and interfering with the cellular respiration, leading to

cell death. This is the first report to demonstrate that *M. furfur* mitigation activity depends on structural differences of NMs, which will become important in further analyzing shampoos as well as drug delivery formulations for scalp treatment of fungal infection (work in progress). Antifungal activity linearly increases with increase in concentration of NMs tested against 10^5 CFU/mL of fungi. The MICs against 10^5 CFU/mL of *M. furfur* were 0.2 mg/mL for 20 nm and 50 nm Ag NPs and 0.3 mg/mL of rod-shaped materials. These results show that Ag NMs have potent antimicrobial activity even at low concentrations.

Skin irritation study

Once in vitro antifungal efficiency in relation to size was established, we sought to evaluate in vivo antimicrobial effects of Ag NPs to understand their effects on rat's skin surface. Skin of all rat groups showed visual observation for irritation marked using redness scoring (Figure S4) on exposure to NMs at different time periods. Skin irritation studies on Rat gave 0 scale for erythema as well as 0 level scale for edema as compared to the standard formalin solution. The results give clear indication of pronounced topical irritation with small spherical NPs owing to their enhanced reactivity as compared with bigger dimensions as well as shape.

Figure 2 Comparison of different antifungal agents and correlation of their concentration with biocidal efficiency.

Notes: (A) Represents the concentration dependent antifungal activity of Ag NPs on *M. furfur*. (B) Represents the comparative antifungal activity of Ag NPs with standard drugs.

Abbreviations: NPs, nanoparticles; *M. furfur*, *Malassezia furfur*.

The results of skin irritation test (Table 2) indicate that Ag NPs do not produce any dermatological reaction and were well tolerated by rats showing no drastic visual flakes formation and inflammation. Primary irritation index (PII) was calculated as 0.33 and 0.16 for 20 nm and 50 nm spherical NPs, respectively, while the rats in direct contact with 50 nm rod-shaped NMs exhibited negligible redness and were marked 0 scores. These results confirmed slightly more irritation tendency of spherical-shaped NMs, but they were chosen for further studies on rats due to their higher antimicrobial efficiency than Ag NR. This also serves as an indirect proof elucidating more penetration capabilities as well as reactivity of small spherical particles than rod-shaped Ag NPs.

Itching is an unpleasant sensation acting as physiological mechanism to support the defense against external agents, particularly aggravated in seborrheic dermatitis and other

dermatological complications. Rats of III, IV, and V groups were recorded by camera for 15 minutes after the treatment of different Ag NMs and their scratching pattern was compared with normal untreated rats. No adverse dermatological effects of NMs occlusion in rat's skin were noted as measured in irritation studies, indicating no specific reaction of Ag NMs on the surface. The numbers of scratch sequences were lower for 50 nm NRs as compared to spherical NPs, similar to the results obtained for other irritation studies. This is attributed to pronounced reactivity and inclusion of 20 nm and 50 nm spherical NPs (Figure S5).

Clinical and mycological treatment analysis

After creating dandruff model, an initial investigation was made to determine the *in vivo* antifungal effects of Ag NMs. The manifestation and progression of infection were visually evident

Table 2 Rat skin irritation test with all nanomaterials and drugs

	Time (h)	Score				
		Rat 1 (control)	Rat 2 (formalin)	Rat 3 (20 nm Ag NPs)	Rat 4 (50 nm Ag NPs)	Rat 5 (50 nm Ag NR)
Erythema and scar	6	0	1	0	0	0
	12	0	2	0	0	0
	48	0	3	2	1	0
Edema	6	0	0	0	0	0
	12	0	1	0	0	0
	48	0	2	0	0	0
Primary irritation index (PII)		PII = 0/6 = 0.0	PII = 9/6 = 1.5	PII = 2/6 = 0.33	PII = 1/6 = 0.16	PII = 0/6 = 0.0

Abbreviations: h, hour; NPs, nanoparticles; NR, nanorod.

on the skin of rats that developed red patches, flakes, and scales, increasing gradually with time. Different groups of rats were subcutaneously delivered with single dose of 200 μ L of antimicrobial agent on day 1 postinfection for 10 consecutive days. They were marked by clinical scores clearly indicating the extent of redness in comparison to normal skin of control set (Table 3). On continued application of Ag NMs and other drugs, regression in redness was visible (Figure 3) and marked on the final assessment on tenth day of clinical and mycological investigation.

Furthermore, a contingency statistical analysis corroborated these findings demonstrating significant variation in percent efficacy for each treatment group. The mean clinical score for Ag NRs was highest at any given time followed by a mean score for 50 nm Ag NPs-treated animals comprised in group 2 (aqueous solution of formalin). It was observed that 20 nm spherical particles exhibited the lowest score 0 compared to others as well as with antifungal drugs. The same trend was denoted for the increase in lesion formation on infection and their eventual decrease on treatment with all materials under investigation. These results confirmed substantial treatment of spherical shape and small size NPs that penetrate easily in different layers of skin and have higher capabilities for interacting with fungal cells. Large-sized particles, including those with rod-like morphology, hamper their deep penetration and killing of fungus via active DNA damage or cell rupture mechanisms.³³ Nanostructures release silver ions in control manner owing to slow oxidation on rat skin surface, prolonging the treatment effects and efficacy of the fungistatic agent. No mortality or acute intoxication or adverse skin damages characterized the treatment methodology during the observation period of 15 days. Behavior of rats during this time frame was normal. On final culmination of studies, the animals were sacrificed, and no pathological side effects or deformities were observed in any organ.

Biochemical assay

The activities of the antioxidant enzymes – CAT, SOD, GPx, MDA, and THP of all groups were measured as an important

indication of treatment process, and the values are reported in Table 4. These antioxidative enzymes work as defense systems that manage reactive oxygen species, resulting in oxidative stress under high concentration or inadequate removal conditions. This severely affects metabolic functioning while damaging cellular biomolecules irreversibly.³⁴ Enzyme activities and protein concentrations were calculated using kits operated using skin of all the six rats in each group taken as an average and including standard deviation. These concentration values for control group without fungal infection and medication were considered the desired levels for comparison with the rest of the sets. Diseased condition of group II animals deleteriously reduced the levels of CAT, SOD, GPx, and THP while increased the levels of MDA from that of control group because of fungal attack. The reduction amounted to two- to fourfolds for tested parameters, which is significant in assessing the grave situation of skin infection.

On introduction of Ag NPs and drugs, these levels further changed indicating tissue repair and normalization of enzymes and proteins concentration. Significant increase in the CAT, SOD, GPx, THP and decrease in MDA concentration were observed prominently for groups V, VI, and VII as compared to groups III and IV, verifying enhanced mitigation of fungal growth by Ag NMs. It is clear that Ag NMs, particularly 20 nm Ag NPs, restore activities and concentrations of CAT and GPx that are considered to be primary antioxidants accountable for direct removal of reactive oxygen species. A total of 25%–50% improvement in concentration levels was observed for Ag NMs than both drugs.

Further, treatment with Ag NMs doubled CAT and total proteins concentrations, increased SOD, GPx, and THP by ~3.5 times, whereas MDA was reduced to half values than diseased condition (group II). The maximum effect was seen for 20 nm NPs that attained the desired values as that of control. The order of correctness of diseased condition can be defined in ascending configuration of their effectiveness as $It < Ket < Ag\ NR < 50\ nm\ NP < 20\ nm\ spherical\ NPs$. This is an important analysis in understanding minimal

Table 3 Clinical and mycological evaluation of treatment efficacy of silver nanoparticles in rat models

S no	Group factor	Control	Infected	Treated (itraconazole)	Treated (ketocanazole)	Treated with Ag NPs		
						20 nm	50 nm	50 nm
1	Redness	0	3	2	1	1	1	1
2	Ulceration	0	3	2	2	0	0	1
3	Scaling	0	3	2	2	0	1	1
4	Total score	0	9	6	5	1	2	3
5	Efficacy (%)	NA	NA	33.34	44.45	88.89	77.78	66.67

Abbreviations: NPs, nanoparticles; NA, not applicable.

Figure 3 Skin photographs of the dorsal surface of rats.

Notes: (A) Control group I showing no redness, no inflammation; (B) *M. furfur* diseased group, scales on the skin; (C) and (D) Groups III and IV treated with itraconazole and ketoconazole, respectively, showing skin with decreased scales as compared to Group II with pinkish color; (E) Group V (20 nm spherical Ag NPs) showing complete treatment of *M. furfur* as compared to Group II; (F) and (G) Groups VI and VII (50 nm spherical- and rod-shaped Ag NPs) also showed normalization of the skin but less as compared to group V.

Abbreviations: *M. furfur*, *Malassezia furfur*; NPs, nanoparticles.

Table 4 Results of biochemical activity assays in different groups of studied rats

Groups of animals	CAT (KU/g protein)	SOD (KU/g protein)	GPx (UI/g protein)	MDA (mmol/g tissue)	THP (mmol/g protein)	Total protein (mg/dL)
Group I (control)	74.27±4.64	70.92±11.34	41.22±2.11	18.03±1.97	8.40±1.5	7.91±0.49
Group II (diseased model)	34.98±3.53	17.80±2.60	9.00±1.37	47.47±0.64	2.63±0.20	4.69±1.2
Group III (itraconazole)	49.43±4.65	32.53±3.21	17.02±2.63	31.21±2.81	4.24±1.41	5.53±0.88
Group IV (ketoconazole)	48.28±5.73	34.21±6.11	19.87±3.72	34.72±3.32	4.92±2.1	5.21±1.5
Group V (20 nm spherical Ag NPs)	72.32±3.21	68.87±3.87	38.67±4.11	19.93±4.72	7.87±0.89	7.74±1.56
Group VI (50 nm spherical Ag NPs)	69.54±2.73	64.67±3.63	35.83±3.39	23.21±3.21	7.11±1.1	6.77±3.2
Group VII (50 nm rod Ag NPs)	62.73±3.72	59.32±2.32	32.64±2.79	26.54±5.11	6.93±0.68	6.13±2.22

Abbreviations: CAT, catalase; SOD, superoxide dismutase; GPx, glutathione peroxidase; MDA, malondialdehyde; THP, total hydroperoxides; NPs, nanoparticles.

Figure 4 Histopathological microphotographs of different groups of studied rats.

Notes: (A) Control I; (B) *M. furfur* diseased model group II; antifungal agent-treated groups; (C) III (itraconazole); (D) IV (ketoconazole); (E) V (20 nm spherical-shaped Ag NPs); (F) VI (50 nm spherical), and (G) VII (50 nm rod-shaped Ag NPs).

Abbreviations: *M. furfur*, *Malassezia furfur*; NPs, nanoparticles.

cytotoxicity of Ag NMs as oxidative stress is less in rats treated with them. These results signify that Ag NMs exert cytoprotective effects on infected mice with antioxidative and peroxidant properties, restoring the activities of major biochemical enzymes in the process at the used concentration parameters.³⁵ The complete toxicity evaluation is currently under progress.

Histopathological evaluation

Histopathological results of skin are shown in Figure 4. Control group I presented normal histology of epidermis and dermis. Group II (diseased model) exhibited proliferation of epidermis with hyperkeratosis, in which the stratum corneum thickens. The dermis presents discrete chronic inflammatory

infiltrates. The experimental groups III and IV (itraconazole and ketoconazole) also exhibited thick proliferation of the epidermis with a low inflammatory response in dermis. With the introduction of 20 nm Ag NPs in group V rats, no significant proliferation of the epidermis and stratum corneum hyperkeratosis was observed, while groups VI and VII (50 nm spherical and 50 nm rod-shaped Ag NPs, respectively) exhibited mild thickening of epidermis and mild stratum corneum hyperkeratosis.

Conclusion

Spherical- and rod-shaped Ag NMs have been successfully prepared of 20 nm, 50 nm, and 50 nm sizes, respectively. These NMs exhibit excellent antifungal activity eradicating

10^5 CFU/mL of *M. furfur* with an effective MIC of 0.2, 0.3, and 0.5 mg/mL for 20 and 50 nm spherical NPs and 50 nm NRs, respectively. In addition, these NMs have been found to be superior to tested drugs. Fungal infection passed on rat model induced with dandruff state was successfully treated with synthesized NPs, demonstrating differential mitigation efficacy. PII was calculated as 0.33 and 0.16 for 20 nm and 50 nm spherical NPs, while NRs, respectively, exhibited negligible redness and were marked 0 scores. The mean clinical scores marked for redness and flakes on treatment with Ag NMs indicate working of these antifungal agents, which is topically explained by their easy entry into cellular organization by breaching cell wall and membranes. Importantly, 20 nm NPs yielded best results at the end of treatment period followed by 50 nm spheres and rods. These particles responded in the most efficient manner against induced dandruff and dermatophytes as validated by contingency statistical analysis indicated by mean clinical score and descending order of NPs, 50 nm NPs, 20 nm NPs (0). These results were also expressed through biochemical and histopathological studies. Small-sized spherical Ag NPs can be used as an important cost-effective fungistatic agent in formulations for treating scalp problems, since they can be produced in larger quantities and very small amount is required for the desired effect.

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Disclosure

The authors report no conflicts of interest in this work.

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Supplementary materials

Figure S1 Schematic representation of synthesis of Ag nanoparticles and nanorods.
Abbreviation: AOT, aerosol-OT.

Figure S2 UV-vis spectra showing absorption at 406 nm and 425 nm for 20 nm and 50 nm spherical-shaped Ag NPs, respectively, and silver nanorods showed two plasma bands at 531 nm and 432 nm.
Abbreviation: NPs, nanoparticles.

Figure S3 Comparison of ZOI for different antifungal agents in groups of studied rats (A–E).

Notes: (A) Control; (B) aqueous solution of formalin (7%) (irritant); (C) 50 nm rod shaped Ag NPs; (D) 50 nm spherical shaped Ag NPs; (E) 20 nm spherical shaped Ag NPs.

Abbreviation: ZOI, zones of inhibition.

Figure S4 Skin photographs showing irritation studies.

Notes: (A) Comparative ZOI study of Ag NPs with itraconazole; (B) ZOI study of 20 nm spherical Ag NPs; (C) ZOI study of 50 nm spherical Ag NPs; (D) ZOI study of 50 nm rod Ag NPs.

Abbreviations: NPs, nanoparticles; ZOI, zones of inhibition.

Figure S5 The graph showing scratching behavior assessment of rat for scratching tendency and comparison with control untreated rats.

Abbreviations: NP, nanoparticle; NR, nanorod.

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