Comparison of two self-assembled macromolecular prodrug micelles with different conjugate positions of SN38 for enhancing antitumor activity

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Abstract: 7-Ethyl-10-hydroxycamptothecin (SN38), an active metabolite of irinotecan (CPT-11), is a remarkably potent antitumor agent. The clinical application of SN38 has been extremely restricted by its insolubility in water. In this study, we successfully synthesized two macromolecular prodrugs of SN38 with different conjugate positions (chitosan-(C₁₀-OH)SN38 and chitosan-(C₂₀-OH)SN38) to improve the water solubility and antitumor activity of SN38. These prodrugs can self-assemble into micelles in aqueous medium. The particle size, morphology, zeta potential, and in vitro drug release of SN38 and its derivatives, as well as their cytotoxicity, pharmacokinetics, and in vivo antitumor activity in a xenograft BALB/c mouse model were studied. In vitro, chitosan-(C₁₀-OH)SN38 (CS-(10s)SN38) and chitosan-(C₂₀-OH)SN38 (CS-(20s)SN38) were 13.3- and 25.9-fold more potent than CPT-11 in the murine colon adenocarcinoma cell line CT26, respectively. The area under the curve (AUC)₀–₂₄ of SN38 after intravenously administering CS-(10s)SN38 and CS-(20s)SN38 to Sprague Dawley rats was greatly improved when compared with CPT-11 (both P<0.01). A larger AUC₀–₂₄ of CS-(20s)SN38 was observed when compared to CS-(10s)SN38 (P<0.05). Both of the novel self-assembled chitosan-SN38 prodrugs demonstrated superior anticancer activity to CPT-11 in the CT26 xenograft BALB/c mouse model. We have also investigated the differences between these macromolecular prodrug micelles with regards to enhancing the antitumor activity of SN38. CS-(20s)SN38 exhibited better in vivo antitumor activity than CS-(10s)SN38 at a dose of 2.5 mg/kg (P<0.05). In conclusion, both macromolecular prodrug micelles improved the in vivo conversion rate and antitumor activity of SN38, but the prodrug in which C₂₀-OH was conjugated to macromolecular materials could be a more promising platform for SN38 delivery.

Keywords: self-assembled prodrug micelles, in vitro cytotoxicity, pharmacokinetics, in vivo antitumor activity

Introduction

7-Ethyl-10-hydroxycamptothecin (SN38), a potent inhibitor of topoisomerase I, is a semisynthetic analog of camptothecins. It is well known as a very potent anticancer agent that exhibits 100- to 1,000-fold higher in vitro cytotoxicity against various cancer cell lines than irinotecan (CPT-11).¹,² However, the clinical application of SN38 has been extremely restricted by its insolubility in water as well as in most pharmaceutically acceptable solvents.³ To overcome these disadvantages, a water-soluble prodrug of SN38, CPT-11, was developed for clinical application. However, the clinical use of CPT-11 is accompanied by severe diarrhea and myelosuppression.⁴ Moreover,
the conversion of CPT-11 to SN38 exhibits interindividual variability because of patient differences in enzyme or protein expression, and the conversion rate is only 2%-8% after administration. In order to improve the drug solubility and enhance the antitumor activity of SN38, various preparations have been developed in recent years, such as liposome-based formulations, polymeric micelles, and complex compounds. Unfortunately, these preparations are accompanied by inherent shortcomings, especially the need for complex preparation methods, low drug loading, and inefficient delivery. To overcome these problems, macromolecular prodrug systems that self-assemble in aqueous medium have been investigated in recent years. The use of macromolecular prodrugs is an attractive potential approach to improve solubility, produce sustained release, and obtain tumor localization by enhanced permeability and retention (EPR).

Like other camptothecins, SN38 is comprised of a (3,4-β) quinoline moiety, an indolizinone, and an α-hydroxy-δ-lactone ring. The lactone ring is necessary for topoisomerase inhibition. The active lactone form of SN38 is converted into the inactive carboxylate form at alkaline pH (Figure 1). However, the activity of the lactone form is much greater than the carboxylate form. Previous research has shown that modifications of hydroxycamptothecin at C10-OH or C20-OH can influence the antitumor activity and stability of the lactone ring. Although several trials have been carried out by conjugating SN38 to macromolecular materials, there are no significant reports of any differences between C10-OH and C20-OH conjugated to hydrophilic materials as the reaction group.

In order to develop new self-assembled macromolecular prodrugs for SN38 delivery and to investigate the differences between the two conjugated positions of SN38, a suitable macromolecular polymer must be chosen as the hydrophilic vehicle. Water-soluble chitosan with a high degree of deacetylation and a low molecular weight has been extensively studied due to its outstanding biocompatibility, abundant biomass, low cost, biodegradability, and low toxicity. To date, polymeric micelles based on chitosan remain an excellent choice for anticancer drug delivery. In our study, water-soluble chitosan was chosen as a hydrophilic vehicle for the delivery of SN38.

In this study, we illustrate the outstanding advantage of self-assembled chitosan-SN38 prodrug micelles. Both of the novel self-assembled chitosan-SN38 prodrug micelles exhibited better in vitro cytotoxicity and in vivo antitumor activity than CPT-11 on CT26. The in vitro and in vivo differences in CS-(10s)SN38 and CS-(20s)SN38 macromolecular prodrug micelles were investigated for the first time in this study.

Materials and methods
Materials
SN38 (98% pure) and CPT-11 (98% pure) were purchased from Knowshine Pharmachemicals Inc. (Shanghai, People’s Republic of China). Chitosan (average molecular weight =30,000, degree of deacetylation =97%) and coumarin-6 were supplied by Sigma-Aldrich Co. (St Louis, MO, USA). 1-Ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (EDC), N-hydroxysuccinimide (NHS), 4-dimethylaminopyridine (DMAP), N,N-diisopropylethylamine (DIPEA), succinic acid anhydrous (SA), and di-tert-butyl dicarbonate (DIBOC) were obtained from Tokyo Chemical Industry Co. (Tokyo, Japan). All other chemicals and reagents were analytical grade after dehydration.

Animals and cell line
Pathogen-free Sprague Dawley (SD) rats (male, weighing 200±20 g) and BALB/c mice (male, weighing 20±2 g) were purchased from the Laboratory Animal Center of Shenyang

Figure 1 Structures of pH-dependent lactone and carboxylate forms of SN38.
Abbreviation: SN38, 7-Ethyl-10-hydroxycamptothecin.
Pharmaceutical University. All animals were housed under standard conditions and the studies were carried out according to the regulations for animal experimentation issued by the State Committee of Science and Technology of the People’s Republic of China.

The murine colon adenocarcinoma cell line CT26 was purchased from the National Platform of Experimental Cell Resource for Sci-Tech. CT26 cells were maintained in RPMI 1640 with 10% fetal bovine serum (FBS). Ascites of the mouse liver cancer cell line H22 were supplied by the Pharmacology Laboratory of Shenyang Pharmaceutical University.

Synthesis of CS-SN38 macromolecular prodrugs with different conjugate positions

Synthesis of succinyl-(C\textsubscript{10}-OH)SN38 and succinyl-(C\textsubscript{20}-OH)SN38

The entire synthesis strategy is shown in Figure 2. These derivatives of SN38 were synthesized according to the reported references with modifications.\textsuperscript{33–35}

SN38 (1 g, 2.55 mmol) and DIBOC (0.61 g, 2.81 mmol) were stirred in anhydrous dichloromethane (DCM) with a catalytic amount of anhydrous pyridine for 14 hours at room temperature. The reaction mixture was then washed several times with 0.5% NaHCO\textsubscript{3} and 0.1 mol/L HCl. Crude compound 1 was obtained by evaporating the solvent.

Compound 1 (200 mg, 0.41 mmol), SA (162.7 mg, 1.63 mmol), DIPEA (0.032 mL, 0.30 mmol), and DMAP (24.9 mg, 0.21 mmol) were reacted in 6 mL anhydrous DCM under N\textsubscript{2} protection for 7 hours at 40°C, and then deprotected with 30% trifluoroacetic acid (TFA)/DCM at room temperature. The mixture was precipitated and purified to obtain succinyl-(C\textsubscript{10}-OH)SN38.

SN38 (1.00 g, 2.55 mmol) and SA (0.33 g, 3.32 mmol) were stirred in 10 mL of anhydrous pyridine for 48 hours at 40°C. The mixture was precipitated in water with excess HCl. After filtration and vacuum drying, the product was purified by column chromatography to obtain succinyl-(C\textsubscript{20}-OH)SN38.

Synthesis of CS-(10s)SN38- and CS-(20s)SN38

SN38-conjugated macromolecular prodrugs

To obtain CS-(10s)SN38 and CS-(20s)SN38, succinyl-(C\textsubscript{10}-OH)SN38 and succinyl-(C\textsubscript{20}-OH)SN38 were reacted with EDC and NHS in dimethyl sulfoxide (DMSO) for 4 hours, then transferred to a chitosan/H\textsubscript{2}O/DMSO solution. The temperature of the mixture was raised to 70°C for 5 hours, and the mixture was stirred at room temperature for another 24 hours. The mixture was then poured into a methanol/ammonia solution and the resulting precipitate was collected by centrifugation. After washing several times with ethanol until no SN38 or derivatives were detected, the yellow precipitate was dissolved in acidic distilled water, and then dialyzed against distilled water in regenerated cellulose dialysis bags to remove any water-soluble byproducts. The final CS-(10s)SN38 and CS-(20s)SN38 were lyophilized (freeze dryer model FD-1C-50; Boyikang Experimental Instrument Co. Ltd., Beijing, People’s Republic of China).

\textsuperscript{1}H nuclear magnetic resonance and mass spectrometry analysis

Proton nuclear magnetic resonance (\textsuperscript{1}H NMR) spectra of intermediates and CS-SN38 macromolecular prodrugs were determined using a Bruker AVANCE Spectrometer (AV-600; Bruker Corporation, Billerica, MA, USA) operated at 400 MHz and 600 MHz, respectively. Intermediates were dissolved in deuterated DMSO (DMSO\textsubscript{d6}) and the prodrugs were dissolved in a mixture of deuterated water (D\textsubscript{2}O) with 1% tetradecanol acetic acid (CD\textsubscript{3}COOD). The mass spectrometry (MS) experiments on the intermediates were performed on a mass spectrometer (micrOTOF-Q; Bruker Corporation).

Fourier transform infrared analysis

In order to investigate the chemical interaction between SN38 and chitosan of the two different prodrugs, Fourier transform infrared (FT-IR) spectra were recorded on an FT-IR spectrometer (Bruker Corporation). The samples were ground with KBr crystals, and then pressed into pellets for infrared (IR) measurement over the range of 400–4,000 cm\textsuperscript{-1}.

Differential scanning calorimetry analysis

Five samples, including SN38, chitosan, the physical mixture of SN38 and chitosan, CS-(10s)SN38, and CS-(20s)SN38, were weighed and sealed in aluminum pans. Differential scanning calorimetry (DSC) studies were then carried out using a differential scanning calorimeter (STAR® DSC System; Mettler Toledo, Greifensee, Switzerland) under dynamic nitrogen flow and a heating–cooling rate of 10°C/minute from 30°C to 240°C.

Drug-loading measurement

To permit the complete release of SN38 from the macromolecular prodrug, a certain amount of prodrug was dissolved in a 0.1 mol/L NaOH solution and incubated for 30 minutes. An excess amount of HCl was then added to transform the carboxylate form of SN38 to the lactone form. The samples
Figure 2 Synthesis of succinyl-SN38 derivatives, CS-(10s)SN38 and CS-(20s)SN38.

Abbreviations: CS-(10s)SN38, chitosan-(C₁₀-OH)SN38; CS-(20s)SN38, chitosan-(C₂₀-OH)SN38; DCM, dichloromethane; DIPEA, N,N-diisopropylethylamine; DMAP, 4-dimethylamino pyridine; DMSO, dimethyl sulfoxide; EDC, 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride; NHS, N-hydroxysuccinimide; SA, succinic acid; TFA, trifluoroacetic acid; SN38, 7-Ethyl-10-hydroxycamptothecin.
were diluted and analyzed. Samples were tested in triplicate, and the drug loading (%) was determined by dividing the weight of SN38 by the weight of the prodrug. Information about the recovery rate of this method can be found in Figure S1.

Particle size and zeta potential measurement of CS-SN38 micelles
In aqueous medium, both macromolecular prodrugs could self-assemble into micelles. The size and the zeta potential of the micelles were determined by laser light scattering with a Malvern Zetasizer (Zetasizer Nano S90; Malvern Instruments, Malvern, UK) at 25°C.

Transmission electron microscopy micrographs
A transmission electron microscope (TEM; JEM-2100; JEOL, Tokyo, Japan) was used for the morphological analysis of the self-assembled prodrug micelles following negative staining with a phosphotungstic acid solution (2% weight per volume) at an accelerating voltage of 200 kV. The samples were air-dried before the observations were carried out.

Measurement of the critical micelle concentration
To determine the critical micelle concentration (CMC) of CS-(20s)SN38 and CS-(10s)SN38, a known amount of coumarin-6 in chloroform was added to a series of vials. After the chloroform was evaporated, a series of concentrations of the macromolecular prodrug solution was added to each vial, then ultrasonicated for 30 minutes and stirred overnight in the dark. The fluorescence emission intensity was measured using a spectrofluorophotometer (F-2500 FL Spectrophotometer; Hitachi Ltd., Tokyo, Japan) at 466 nm (emission wavelength) and 504 nm (lasing wavelength). The CMC of the self-assembled macromolecular prodrug micelle was determined at the intersection of the tangents to the two linear portions of the fluorescence intensity curve.

In vitro drug release studies
The behavior of the SN38 in vitro release from both of the self-assembled macromolecular prodrugs was studied by dialysis at 37°C±1°C in different release media for 48 hours. Briefly, CS-(10s)SN38 and CS-(20s)SN38 were dissolved in 1 mL pH 6.5 phosphate-buffered saline (PBS), pH 7.4 PBS, 25% liver homogenate, and 50% plasma, where the concentration of the prodrugs was 250 µg/mL (equivalent to SN38), respectively. The solutions were then put into dialysis tubes (molecular weight cut-off: 10,000–12,000 kDa) and the tubes were immersed in 20 mL of PBS (pH 6.5 for the first tube and pH 7.4 for the others). Experiments were performed at 37°C in a shaking water bath at 100 rpm. Samples were collected at 1, 3, 5, 8, 12, 24, 36, and 48 hours and the release media were replaced at certain intervals, so that the media volume could meet the sink condition. The concentration of SN38 was determined by high-performance liquid chromatography (HPLC) (LC-10AT; Shimadzu Corporation, Kyoto, Japan) at 385 nm with an ultraviolet (UV) detector (SPD-10A; Shimadzu Corporation) using a C18 reversed-phase column (Dikma Technologies Inc., Lake Forest, CA, USA). The mobile phase consisted of acetonitrile and acetate buffer (48:52 volume per volume).

In vitro cytotoxicity assays
CT26 was used to determine the in vitro cytotoxicity. Briefly, adherent cells (about 5,000 cells per well) were plated in 96-well plates and incubated for 24 hours at 37°C with 5% CO2. After incubation, the cells were treated with serial dilutions of CS-(10s)SN38, CS-(20s)SN38, succinyl-(C10-OH)SN38, succinyl-(C20-OH)SN38, CPT-11, and SN38, and then further incubated for 48 hours. At the end of this time, 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2-H-tetrazolium bromide (MTT) was added and the cells were incubated for another 2 hours. After removing surplus MTT, the colored formazan product was assayed using an enzyme-linked immunosorbent assay (ELISA) analyzer (Tecan, Männedorf, Switzerland) at a test wavelength of 492 nm.

Pharmacokinetic and statistical analysis
Male SD rats were used for the pharmacokinetic study. All animals were fasted overnight before starting the experiment. CS-(10s)SN38 (10 mg/kg), CS-(20s)SN38 (10 mg/kg), and CPT-11 (10 mg/kg) were administrated to rats intravenously. All preparations were equivalent to SN38. Blood samples were collected into heparinized tubes at the following times: 5, 10, 20, and 40 minutes, and 1, 1.5, 2, 3, 4, 6, 8, 12, and 24 hours after administration. The plasma was then obtained from the blood samples by centrifugation at 12,000× g for 10 minutes and then stored at −60°C until required for analysis.

Analysis of the free SN38 (chitosan-unbound SN38) and CPT-11 was performed as follows: 100 µL of plasma, 10 µL of internal standard solution (hydroxycamptothecin, 5 µg/mL), and 5 µL 0.2 mol/L HCl were mixed for 30 seconds; the
samples were then extracted with 2 mL of ethyl acetate for 5 minutes at room temperature. After centrifugation at 3,000× g for 10 minutes, the supernatants were transferred into glass tubes and then dried under nitrogen at 40°C. The samples were redissolved until the HPLC analysis was carried out.

The concentration of chitosan-bound SN38 (SN38 remaining bound to chitosan) was calculated as follows: the concentration of whole SN38 minus the concentration of free SN38 is equal to the concentration of chitosan-bound SN38. The concentration of whole SN38 was detected by the following method: 50 µL of plasma was incubated with 5 µL 1 mol/L NaOH for 30 minutes at 37°C. After incubation, 2 mol/L HCl (5 µL), 10 µL of internal standard (hydroxycamptothecin, 5 µg/mL), and 200 µL of methanol were added. The samples were centrifuged at 12,000 rpm for 10 minutes before being analyzed by HPLC at 385 nm with a UV detector.

At day 14, tumors from the mice that received different treatments were excised. The tumors were fixed in 10% formalin and embedded in paraffin. Sections with a thickness of 4 µm were then prepared. The sections were stained with hematoxylin and eosin (H&E) for microscopic observation.

**Results and discussion**

**Characterization of succinyl-(C\textsubscript{10} -OH) SN38 and succinyl-(C\textsubscript{20} -OH)SN38**

Characterization of succinyl-(C\textsubscript{10}-OH)SN38: 1H NMR (400 MHz, DMSO\textsubscript{d6}, δ): 0.95 (t, 3H, CH\textsubscript{3}2), 1.27 (t, 3H, CH\textsubscript{3}2), 2.14 (q, 2H, CH\textsubscript{2}), 2.77 (m, 2H, CH\textsubscript{2}), 3.00 (q, 2H, CH\textsubscript{2}), 5.18 (d, 2H, J=18.8, CH\textsubscript{2}), 5.48 (s, 2H, CH\textsubscript{2}), 5.75 (s, 1H, OH), 7.00 (s, 1H, ArH), 7.32 (d, 1H, J=2.4, ArH), 7.38 (dd, 1H, J=8.8, J=2.4, ArH), 7.97 (d, 1H, J=8.8, ArH); MS (ESI-TOF) m/z: [M + H]+ calculated for C\textsubscript{22}H\textsubscript{19}N\textsubscript{2}O\textsubscript{5}, 493.1611; found, 493.1605.

Characterization of succinyl-(C\textsubscript{20}-OH)SN38: 1H NMR (400 MHz, DMSO\textsubscript{d6}, δ): 0.92 (t, 3H, CH\textsubscript{3}2), 1.26 (t, 3H, CH\textsubscript{3}2), 2.14 (q, 2H, CH\textsubscript{2}), 2.47 (m, 2H, CH\textsubscript{2}), 2.73 (m, 2H, CH\textsubscript{2}), 3.07 (q, 2H, CH\textsubscript{2}), 5.27 (d, 2H, CH\textsubscript{2}), 5.47 (s, 2H, CH\textsubscript{2}), 7.01 (s, 1H, ArH), 7.39 (d, 1H, J=2.4Hz, ArH), 7.42 (dd, 1H, J=9.2Hz, J=2.4Hz, ArH), 8.03 (d, 1H, J=9.2Hz, ArH), 10.33 (s, 1H, OH); MS (ESI-TOF) m/z: [M + H]+ calculated for C\textsubscript{22}H\textsubscript{19}N\textsubscript{2}O\textsubscript{5}, 493.1611; found, 493.1605.

Succinyl-(C\textsubscript{10}-OH)SN38 and succinyl-(C\textsubscript{20}-OH)SN38 were characterized by 1H NMR and MS, as shown earlier; the original spectrum can be found in Figure S2. The absorption peaks of succinyl-linker H atoms were observed in both of the derivatives, and the chemical shift was 2.4–2.8 ppm. For succinyl-(C\textsubscript{10}-OH)SN38, the absorption peak of C\textsubscript{10}-OH disappeared due to the esterification of SA with the phenolic hydroxyl of SN38. The chemical shift of C\textsubscript{20}-OH was 5.75 ppm. In the case of succinyl-(C\textsubscript{20}-OH)SN38, the chemical shift of C\textsubscript{10}-OH was seen at 10.33 ppm. The disappearance of the absorption peak of C\textsubscript{20}-OH indicated that the succinyl-linker was conjugated to SN38 at C\textsubscript{20} -OH. Since these two derivatives are isomers, the same molecular weight was found in the MS study.

**Characterization of CS-(10s)SN38 and CS-(20s)SN38**

After purification, CS-(10s)SN38, CS-(20s)SN38, and chitosan were characterized by their 1H NMR spectra. A typical 1H NMR spectrum of chitosan is presented in Figure 3A. As shown in Figures 3B and C, the proton peaks at 0.8–1.2 ppm, which are attributed to the methyl protons of
SN38, were observed in the spectra of CS-(10s)SN38 and CS-(20s)SN38. Furthermore, no such peaks were present at the same chemical shifts for chitosan. The protons corresponding to the methylene of SN38 appeared between chemical shifts 2.1–2.8 ppm, which is in agreement with a previous study.\(^{15,18}\) These results indicated that CS-(10s)SN38 and CS-(20s)SN38 had been synthesized successfully.

The FT-IR spectra of chitosan, SN38, the mixture of chitosan and SN38, CS-(10s)SN38, and CS-(20s)SN38 are shown in Figure 4. The spectrum of CS indicated that the peaks at 1,633 cm\(^{-1}\) and 1,520 cm\(^{-1}\) were due to the C–O stretching vibration of the carbonyl group and the N–H bending vibration of the secondary amine (amide I and II), respectively (original spectra of chitosan could be found in Figure S3). Compared with CS, the new signals at 1,656 cm\(^{-1}\) and 1,566 cm\(^{-1}\) were assigned to the new amide link for amide I and II. In addition, peaks at 1,598 cm\(^{-1}\) and 1,600 cm\(^{-1}\) were observed in the spectra of CS-(10s)SN38 and CS-(20s)SN38, representing the characteristic vibration of the benzene skeleton. All of the results showed that the grafting of SN38 onto the amino groups of CS had been successful.

To further confirm that SN38 had been covalently linked to chitosan, a DSC experiment was performed. As illustrated in Figure 5, the endothermic peak that was related to the melting point of SN38 in its crystalline form appeared at 208°C–210°C. For the physical mixtures, all the characteristic peaks of every component showed only a slight shift. Both macromolecular prodrugs exhibited no sharp peaks, which suggested that SN38 was present in an amorphous state in the prodrugs.

The drug loading was found to be 16.5%±0.7% for CS-(10s)SN38 and 17.4%±0.8% for CS-(20s)SN38 as found using HPLC with a UV detector, and this was much higher than the amphiphilic block copolymer micelles loaded with SN38.\(^ {13,39}\) In our study, the water-insoluble drug was directly linked to chitosan by a covalent bond. The force of the
covalent bond was stronger than the hydrophobic interactions between SN38 and the hydrophobic group of the amphiphilic block copolymers, and this could be an important reason for the high drug-loading efficiency. The solubility of both macromolecular prodrugs was more than 40 mg/mL (corresponding to 7.2 mg/mL for SN38) while SN38 had a very poor solubility of 11–38 µg/mL in water. These observations further confirm that SN38 was covalently bound to chitosan.

Characterization and morphology of self-assembled micelles
Both CS-(10s)SN38 and CS-(20s)SN38 were SN38-loaded macromolecular micelles produced in aqueous medium by the self-assembly of amphiphilic graft copolymers and the CMC values were calculated as 15 µg/mL and 12 µg/mL by the method described earlier (Figure 6). The CMC value reflects the threshold concentration of the self-aggregation formation by the hydrophilic and hydrophobic groups of the macromolecular prodrugs. Such low CMC values for CS-(10s)SN38 and CS-(20s)SN38 indicate their excellent self-aggregation behavior in aqueous medium and stabilization following dilution. The TEM image and particle size are shown in Figure 7. The shape of both macromolecular prodrugs was almost spherical and the mean particle size was 128.0±1.8 nm for CS-(10s)SN38 micelles (Figure 7A) and 122.4±1.1 nm for CS-(20s)SN38 micelles (Figure 7B) (n=3). The particle sizes were similar when the drug loading of the two produg micelles was similar. This indicated that particle size could mainly be determined by the degree of grafting of the hydrophobic groups and the characteristics of the hydrophilic groups, rather than the conjugate position of SN38. The mean zeta potential of CS-(10s)SN38 was 21.1±0.5 mV (n=3), whereas the mean zeta potential of CS-(20s)SN38 was 22.3±0.4 mV (n=3). These relatively high zeta potentials of the micelles play an important role in the mutual repulsion and stability against aggregation, which could be due to the ionized amino groups of chitosan.

Drug release studies
The release of SN38 from the self-assembled macromolecular prodrugs was performed at 37°C. We chose releasing media to simulate body fluids (pH 7.4 PBS), the tumor extracellular environment (pH 6.5 PBS), blood circulation (50% plasma), and liver circulation (25% liver homogenate).
When SN38 is bound to chitosan by a succinyl linker, the macromolecular prodrug may have two different pathways to release the active metabolite, as illustrated in Figure 8: 1) rupture at the amide bond, followed by hydrolysis to release SN38; or 2) direct hydrolysis of the ester bond to obtain SN38. Thus, we investigated the cumulative profiles of both SN38 and succinyl-SN38 derivatives of the two prodrug micelles. SN38 was released from CS-(10s)SN38 and CS-(20s)SN38 in different media, as shown in Figures 9A and B. For both prodrugs, the SN38 release rates were relatively higher in the 50% plasma and 25% liver homogenate than in PBS. The ester bond between the drug and the spacer could be cleaved by esterases, but in PBS, the release was only due to nonspecific hydrolysis. The release of succinyl-(C_{10}-OH) SN38 is shown in Figure 9C, and that of succinyl-(C_{20}-OH) SN38 is shown in Figure 9D. The low cumulative curves of the two succinyl-SN38 derivatives in 50% plasma and 25% liver homogenate reflected fast in vivo elimination. The succinyl-SN38s had a fluctuating release profile in different media, which could be due to a dynamic transformation from succinyl-SN38 to free SN38. Comparing the cumulative profiles of the two macromolecular prodrugs, CS-(20s) SN38 showed slower drug release by 48 hours in different media (P<0.05), which indicates better stability of CS-(20s) SN38 in vitro for the controlled release of SN38. Esterification at C_{20}-OH could enhance steric hindrance, prevent the ring-opening of SN38, and protect the prodrug micelles from hydrolysis, which could be an explanation for the slower release of CS-(20s)SN38.

Cytotoxicity study

In the toxicity study, both the CS-(10s)SN38 and CS-(20s) SN38 prodrugs, succinyl-SN38 derivatives, CPT-11, and SN38 were incubated with CT26 cells for 48 hours in order to compare their in vitro cytotoxicity (Table 1). SN38 showed the highest activity against the CT26 cell line with a half maximal inhibitory concentration (IC_{50}) value of 20.4 nmol/L. CPT-11 is the main clinical form of SN38, and its IC_{50} value was (46.7±8.4)×10^3 nmol/L. The IC_{50} values of CS-(10s)SN38 and CS-(20s)SN38 were (3.5±0.2)×10^3 nmol/L and (1.8±0.4)×10^3 nmol/L, respectively, which were 13.3- and 25.9-fold more sensitive than CPT-11. The in vitro cytotoxicity of all SN38 derivatives and prodrugs was highly potent when compared with that of CPT-11. Moreover, it was observed that CS-(20s)SN38 and succinyl-(C_{20}-OH)SN38 were more cytotoxic than CS-(10s) SN38 and succinyl-(C_{10}-OH)SN38 (both P<0.05). This suggests that the C_{20}-OH position derivatives might have superior anticancer activity to C_{10}-OH position derivatives, which could perhaps be due to the stabilization of the lactone ring of SN38. Only the lactone form is active against topoisomerase I, and the activity is reduced to 1/10 when the lactone form is transformed into the carboxylate form under basic conditions. Also, modification at C_{20}-OH could...
reduce the hydrogen-bonding interaction between C_{20}-OH and the carbonyl groups.\textsuperscript{43}

**Pharmacokinetic analysis of CS-(20s)SN38, CS-(10s)SN38, and CPT-11**

To investigate the in vivo pharmacokinetic behaviors of CS-(10s)SN38, CS-(20s)SN38, and CPT-11, free SN38 and the prodrugs (equivalent to SN38) were examined by HPLC. The mean plasma concentration–time curves of SN38 after intravenous administration of the macromolecular prodrug micelles and CPT-11 are shown in Figure 10. The pharmacokinetic parameters of SN38 were calculated using noncompartmental analysis with DAS 2.0 software (Table 2). The area under the curve (AUC)\textsubscript{0–24} values of free SN38 after intravenous administration of CS-(10s)SN38 and CS-(20s)SN38 were 7,011.18±969.76 µg⋅h/L and 8,925.92±1,907.46 µg⋅h/L.
respectively, which were statistically insignificant ($P>$0.05). However, when compared with CPT-11, both AUC$_{0-24}$ values were much higher than that of CPT-11 ($P<$0.05). For CS-(10s)SN38 and CS-(20s)SN38, the half-life ($T_{1/2}$) and mean residence time (MRT) of plasma SN38 were prolonged, while the renal clearance (CL) was reduced when compared with CPT-11 ($P<$0.05). The nonspecificity of hydrolysis could be an explanation for the higher conversion rate observed for SN38 of the prodrugs when compared with CPT-11. The AUC$_{0-24}$ for free SN38 of CS-(10s)SN38 was slightly higher than that of CS-(20s)SN38, which might be due to the faster release of SN38 from CS-(10s)SN38 in plasma. The lactone form of SN38 is necessary for topoisomerase inhibition, and the determination of lactone form of SN38 is valuable for macromolecular prodrug study. We would like to discuss the percent of lactone form of SN38 in vivo in our future work.

The plasma concentration of the two different macromolecular prodrugs and CPT-11 are shown in Figure 11. The calculated parameters of the plasma prodrugs are given in Table 3. CS-(20s)SN38 has a higher AUC$_{0-24}$ than CS-(10s) SN38 ($P<$0.05). The $C_{max}$ values of the two prodrugs were

Table 1: Cytotoxicity of different forms of SN38 in murine colon adenocarcinoma cell line CT26

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<th>Compound</th>
<th>IC$_{50}$ (nmol/L)</th>
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<tr>
<td>SN38</td>
<td>20.4±1.8</td>
</tr>
<tr>
<td>Succinyl-(C$_{10}$-OH)SN38</td>
<td>565.9±51.8</td>
</tr>
<tr>
<td>Succinyl-(C$_{20}$-OH)SN38</td>
<td>375.7±73.6</td>
</tr>
<tr>
<td>CS-(10s)SN38</td>
<td>(3.5±0.2)×10$^3$</td>
</tr>
<tr>
<td>CS-(20s)SN38</td>
<td>(1.8±0.4)×10$^3$</td>
</tr>
<tr>
<td>CPT-11</td>
<td>(46.7±8.4)×10$^3$</td>
</tr>
</tbody>
</table>

Notes: All preparations were equivalent to SN38; data are shown as mean ± standard deviation; n=3.
Abbreviations: CPT-11, irinotecan; CS-(10s)SN38, chitosan-(C$_{10}$-OH)SN38; CS-(20s)SN38, chitosan-(C$_{20}$-OH)SN38; IC$_{50}$, half maximal inhibitory concentration; SN38, 7-Ethyl-10-hydroxycamptothecin.

Figure 9: Drug release profiles of CS-(10s)SN38 compared with CS-(20s)SN38 in different media at 37°C: (A) SN38 released from CS-(10s)SN38, (B) SN38 released from CS-(20s)SN38, (C) succinyl-(C$_{10}$-OH)SN38 released from CS-(10s)SN38, and (D) succinyl-(C$_{20}$-OH)SN38 released from CS-(20s)SN38.
Notes: Data are shown as mean ± standard deviation; n=3.
Abbreviations: CS-(10s)SN38, chitosan-(C$_{10}$-OH)SN38; CS-(20s)SN38, chitosan-(C$_{20}$-OH)SN38; PBS, phosphate-buffered saline; SN38, 7-Ethyl-10-hydroxycamptothecin.
not significantly different. The existence of SN38 bound to the macromolecular prodrug might take advantage of the EPR effect and a better in vivo antitumor effect was thus expected.

**In vivo antitumor activity and pathological examination**

In order to further evaluate the in vivo antitumor activity of the two macromolecular prodrug micelles, an experiment was performed using a CT26 xenograft BALB/c mouse model. The results obtained are shown in Table 4 and Figure 12A. The growth in TV was very fast in the saline group (group A), while the TV growth of mice injected with a low dose of CS-(10s)SN38 and CS-(20s)SN38 (groups C and D) showed no statistically significant difference. The tumor growth was significantly inhibited at high doses of CS-(10s)SN38 and CS-(20s)SN38 (groups E and F), especially for CS-(20s) SN38. The average tumor weight of the saline group was 3.27±0.28 g. At the low dose, the IRT (%) of the CS-(20s) SN38 group (31%) was not significantly different from that of the CS-(10s)SN38 group (27.7%) (P>0.05). At the high dose of the CS-(20s)SN38 group, the IRT (%) (64.7%) was much higher than that of CPT-11 or CS-(10s)SN38 at the same dose (both P<0.05). Compared with CPT-11, both prodrug micelles could increase in vivo antitumor activity. The reason for this could be due to the fact that both prodrug micelles increased the solubility of SN38 and promoted the tumor accumulation of SN38. In addition, another reason for the better in vivo antitumor activity of CS-(20s)SN38 could be the following: firstly, the CS-(20s)SN38 micelles had a better circulation effect in vivo than did CS-(10s)SN38, which could enhance the EPR effect; secondly, occupation at the C_{20}-OH of SN38 might increase the lactone ring stability of SN38 and lead to antitumor activity. Additionally, body weight was used as a toxicity marker (Figure 13). In the saline group, body weight significantly increased by 24% by the end of the experiment. Similar results were shown for the low-dose group of the macromolecular prodrugs. In the high-dose macromolecular prodrug groups, the body weights were slightly increased with...
Table 3 Pharmacokinetic parameters of prodrugs after intravenous administration of 10 mg/kg CS-(10s)SN38, CS-(20s)SN38, and CPT-11

<table>
<thead>
<tr>
<th>Formulation</th>
<th>AUC&lt;sub&gt;0-24&lt;/sub&gt; (×10&lt;sup&gt;3&lt;/sup&gt; µg h/L)</th>
<th>C&lt;sub&gt;max&lt;/sub&gt; (×10&lt;sup&gt;3&lt;/sup&gt; µg/L)</th>
<th>T&lt;sub&gt;1/2&lt;/sub&gt; (h)</th>
<th>MRT&lt;sub&gt;0-24&lt;/sub&gt; (h)</th>
<th>CL (L/h/kg)</th>
</tr>
</thead>
<tbody>
<tr>
<td>CPT-11</td>
<td>19.71±2.26</td>
<td>8.71±0.99</td>
<td>2.89±0.51</td>
<td>2.61±0.27</td>
<td>0.46±0.02</td>
</tr>
<tr>
<td>CS-(20s)SN38</td>
<td>133.07±11.10</td>
<td>48.43±4.27</td>
<td>7.05±2.04</td>
<td>4.94±1.58</td>
<td>0.07±0.01</td>
</tr>
<tr>
<td>CS-(10s)SN38</td>
<td>89.84±8.01</td>
<td>46.89±6.56</td>
<td>5.40±1.29</td>
<td>4.90±0.66</td>
<td>0.11±0.01</td>
</tr>
</tbody>
</table>

Notes: Data are shown as mean ± standard deviation; n=6; All formulations were equivalent to SN38.

Abbreviations: AUC<sub>0-24</sub>, area under concentration-time curve from 0 hours to 24 hours; CL, clearance; C<sub>max</sub>, maximum concentration; CPT-11, irinotecan; CS-(10s)SN38, chitosan-(C<sub>10</sub>-OH)SN38; CS-(20s)SN38, chitosan-(C<sub>20</sub>-OH)SN38; MRT, mean residence time; T<sub>1/2</sub>, half-life; SN38, 7-Ethyl-10-hydroxycamptothecin.

Figure 12 In vivo antitumor results of CT26 xenograft Balb/c mice; (A) the tumor growth curves of CT26 xenograft Balb/c mice after intravenous administration of saline, CPT-11, CS-(10s)SN38, and CS-(20s)SN38, (B) pathological section images of cancer tumor of Balb/c mice treated by CPT-11, CS-(10s)SN38, and CS-(20s)SN38.

Notes: Data are shown as mean ± standard deviation; n=8; magnification 400×. The white arrows in the saline group represent tumor cells and muscle infiltration; the white arrows in the other groups represent tumor tissue necrosis.

Abbreviations: CPT-11, irinotecan; CS-(10s)SN38, chitosan-(C<sub>10</sub>-OH)SN38; CS-(20s)SN38, chitosan-(C<sub>20</sub>-OH)SN38; SN38, 7-Ethyl-10-hydroxycamptothecin.
excellent antitumor activities (Figure 12A). However, there was no statistically significant difference between the prodrug groups and the saline group (P>0.05). This might be due to the good biocompatibility and low toxicity of the prodrug micelles. Therefore, these prodrug micelles demonstrated excellent antitumor activity and good safety.

To further confirm therapeutic efficacy, a histological section examination was performed. Nuclei were stained violet and the extracellular matrix and cytoplasm were stained pink using H&E staining. As shown in Figure 12B, a large number of tumor cells and muscle infiltration were observed in the saline group. Partial tumor tissue necrosis was found in the group treated with 2.5 mg/kg of CPT-11, 1.25 mg/kg of CS-(10s)SN38, and 1.25 mg/kg of CS-(20s)SN38. By contrast, extensive tissue necrosis was found in the group treated with 2.5 mg/kg of CS-(10s)SN38 and 2.5 mg/kg CS-(20s)SN38. The most severe damage in the tumor tissue was observed in the high-dose CS-(20s)SN38 group. These findings revealed that the delivery of SN38 by both macromolecular prodrugs produced superior anticancer efficacy at the high dose, especially for CS-(20s)SN38, due to the destruction and reduction of tumor cell proliferation.

Conclusion

In this study, we successfully synthesized two SN38 produgs in the form of macromolecular prodrug micelles. These produgs can self-assemble into micelles in aqueous medium and release SN38 in vitro. Both produgs increased the AUC$_{0.24}$ of SN38 and antitumor activity against CPT-11. The in vitro cytotoxicity and in vivo antitumor activity of CS-(20s)SN38 were better than those of CS-(10s)SN38. Macromolecular biomaterial involving conjugated SN38 with C$_{20}$-OH could be a more promising strategy to increase the antitumor activity of SN38, and this should be a subject for further study.

Acknowledgments

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Disclosure

The authors report no conflicts of interest in this work.

References


Supplementary materials

**Figure S1** The recovery rate of the lactone form of SN38 at 37°C.

**Notes:** The recovery rate experiment was carried out to confirm that the treatment of NaOH only caused hydrolysis of SN38 and reversible transformation from the lactone form to the carboxylate form. SN38 was not degraded during the treatment. Data are shown as mean ± standard deviation; n=3.

**Abbreviation:** SN38, 7-Ethyl-10-hydroxycamptothecin.

**Figure S2** 

**A** 

1H NMR spectra of (A) succinyl-(C$_{10}$-OH)SN38 and (B) succinyl-(C$_{20}$-OH)SN38.

**Notes:** For succinyl-(C$_{10}$-OH)SN38 (A), the absorption peak of C$_{10}$-OH disappeared (10.33 ppm) due to the esterification of succinic acid with the phenolic hydroxyl group (C$_{10}$-OH) of SN38. The chemical shift of C$_{20}$-OH was 5.75 ppm (marked with the blue circle). In the case of succinyl-(C$_{20}$-OH)SN38, the chemical shift of C$_{20}$-OH can be seen at 10.33 ppm (marked with the red circle in B). The disappearance of the absorption peak of C$_{20}$-OH, indicating that the succinyl-linker was conjugated to SN38 at C$_{20}$-OH.

**Abbreviations:** NMR, nuclear magnetic resonance; SN38, 7-Ethyl-10-hydroxycamptothecin.
Two chitosan-SN38 prodrugs with different conjugate positions

Figure S3 Scanned images of FT-IR spectra of chitosan: (A) old FT-IR spectrum and (B) new FT-IR spectrum.

Abbreviation: FT-IR, Fourier transform infrared.