PDLA a potential new potent topical analgesic: a case report

Joel S Goldberg1,2
1Durham Veterans Affairs Medical Center, 2Duke University School of Medicine, Durham, NC, USA

Abstract: Polymer D-lactic acid (PDLA) is a hydrogel that has been shown to sequester L-lactate (lactate). This reaction is rapid, spontaneous, and non-enzymatic. Lactate has been shown to have many functions within the nervous system including its use as a secondary fuel to sustain neural activity and as a neuromodulator. In the central nervous system, lactate is produced in glial cells and shuttled to neurons to be used mostly as a fuel. Lactate dehydrogenase (LDH)1 is the predominant LDH isoform within neurons and unlike LDH5, it preferentially converts lactate to pyruvate which can be used to produce adenosine triphosphate (ATP). Considering that lactate is intimately involved in the sustenance of neural activity, PDLA was applied to an open wound and its effects were examined. The results showed that the application of PDLA induced topical analgesia. This may be the first report to demonstrate that sequestering lactate, a source of energy required to sustain the firing of action potentials in neurons, may produce analgesia.

Keywords: polymer D-lactic acid, topical analgesia, hydrogel

Introduction

In biological systems, L-lactate (lactate) is usually considered as a waste product of glycolysis. Recent evidence supports that lactate has an independent function in modulating neurotransmission and it serves as an important energy source for neurons.1,2 The brain contains considerable amounts of lactate that can be used as the substrate for glucose required for the production of adenosine triphosphate (ATP) when glucose is in short supply.3 Lactate is produced in glial cells and can be shuttled to neurons presumably for fuel. Lactate can be converted to pyruvate within neurons by lactate dehydrogenase (LDH)1. LDH1 is the predominant LDH isoform within neurons and it preferentially converts lactate to pyruvate unlike LDH5.4 Pyruvate can then be converted to acetyl-coA, which is utilized by neurons to produce ATP. It was postulated that if such a system of lactate utilization occurs in the central nervous system, a similar system may also exist in the peripheral nervous system.

In 1932, Feng used a sciatic nerve preparation to show that the addition of lactate improves a peripheral nerve’s capacity to maintain sustained neural activity as measured by a galvanometer muscle deflection.5 He also showed that this sustained activity did not occur in nitrogenous environment confirming that the effects of lactate were through oxidation.5 Immke and McCleskey further verified the role of lactate as a neuromodulator by showing that lactic acid could enhance neural transmission of visceral afferents.1
In 2012, it was discovered that lactate binds to polymer D-lactic acid (PDLA) forming a presumed stereocomplex. This reaction has been described as sequestering or “trapping” lactate. The evidence that this reaction occurs was demonstrated by two separate enzymatic methods and high-performance liquid chromatography. The reaction is rapid, spontaneous, and non-enzymatic at 37°C. Sequestering lactate with PDLA is an extraordinarily versatile chemical reaction, which has many potential medical and non-medical uses. This case report shows that topical application of PDLA in the form of a hydrogel could produce analgesia by sequestering lactate, which is the energy source required by nociceptors to sustain the propagation of action potentials.

Methods
Pandey and Aswath first described the synthesis of poly-L-lactic acid (PLLA) from L-lactic acid in a home microwave. The method to produce PDLA from D-lactic acid follows the same procedures as those used to produce PLLA from L-lactic acid. Beginning with low power and short time intervals, D-lactic acid was dehydrated and polymerized to form PDLA. D-lactic acid was then microwaved until there was approximately 50% loss of weight secondary to dehydration and esterification. In addition, the PDLA dimer (n=2) and tetramer (n=4) were synthesized at the Duke University Small Molecule Synthesis Facility by Dr David M Gooden and tested for enzymatic activity using Accuvin test strips.

Figure 1 Enzymatic testing for lactic acid using Accuvin test strips.
Notes: Accuvin test strips from Accuvin LLC, St Napa, CA, USA. 180 mg/L lactic acid tested positive for lactic acid. 180 mg/L lactic acid and PDLA dimer (n=2) tested negative for lactic acid. 180 mg/L lactic acid and PDLA tetramer (n=4) tested positive for lactic acid. 180 mg/L lactic acid and PDLA oligomer mixture tested negative for lactic acid.
Abbreviation: PDLA, polymer D-lactic acid.

Results
The test strips were activated by the control (180 mg/L of lactic acid), and the solution of 180 mg/L lactic acid containing the PDLA tetramer (n=4). The test strips were not activated by 180 mg/L solution of lactic acid containing the PDLA dimer (n=2) nor the 180 mg/L of lactic acid containing PDLA, which consisted of a mixture of oligomers formed during the synthesis in the microwave (Figure 1).

Case report
The case patient had a 0.5×1.5 cm skin lesion in the left index finger with an approximate depth of 1.5 mm. The lesion produced vigorous bleeding and pain. Application of approximately 10 mg of PDLA in the form of a hydrogel to the lesion produced an intense burning pain that lasted for approximately 30 seconds until there was complete analgesia without complete loss of sensation (numbness). The analgesia lasted for 24 hours until PDLA was removed by water. After washing the wound with water, the pain returned and additional PDLA was applied, which produced continuous analgesia until the lesion was completely healed after 10 days.

Discussion
The test strip experiment confirms that the PDLA dimer (n=2) sequesters lactate and is a likely component of the PDLA microwave mixture. It also confirms that the PDLA tetramer (n=4) does not sequester lactate. The case report demonstrates that the PDLA mixture prepared in a microwave has analgesic properties when applied to a patient’s open wound.

Topical analgesics and local anesthetics employ a variety of mechanisms to produce analgesia. Topical analgesics can produce analgesia either by stimulation of A-beta fibers (eg, through application of counter irritants or transcutaneous...
electrical stimulation) or by some forms of massage involving presumed inhibition of nociception at the dorsal horn. Topical local anesthetics in current clinical use interfere with conductance of sodium by blocking the propagation of action potentials. Topical capsaicin deletes substance P, which is a known excitatory neurotransmitter. This case report shows that application of PDLA hydrogel produces analgesia presumably by sequestration of lactate depleting the energy source required by nociceptors to sustain neural activity. This mechanism is supported by prior work demonstrating the importance of lactate in the sustenance of action potentials. 1, 5 Since the analgesic effects could be reversed by washing the wound with water, it suggests that such effects are not secondary to neurolysis. Also, PDLA used in this case report exists as a hydrogel and hence, it may serve as a protective barrier for healing.

Depleting the energy source required by nerves to propagate action potentials may be a new mechanism to produce analgesia. If the active oligomers of PDLA could be introduced into the central nervous system, the ability of PDLA to sequester lactate may be used as a potential mechanism for the treatment of posttraumatic stress disorder (PTSD) and mania. PTSD develops in susceptible individuals who undergo infusions of sodium lactate, and patients with mania are shown to have elevated lactate levels. 8, 9 Although the toxicity of PDLA is not known, PDLA is a known component in cardiac stents, biodelivery systems, and wound coverings. 10, 11 It also has been used as part of drug delivery systems.

Conclusion

Topical application of a PDLA hydrogel induced analgesia in the patient with an open wound. PDLA may have two beneficial effects for wound healing: 1) providing a protective barrier as a hydrogel; and 2) serving as an analgesic by sequestering lactate, a primary fuel source required for the generation of nociceptive action potentials.

Acknowledgment

The author would like to thank Dr David M Gooden and staff of the Small Molecule Synthesis Facility at Duke University, who were commissioned to synthesize the PDLA dimer and tetramer.

Disclosure

The author reports no conflicts of interest in this work.

References