Purification and in vitro antioxidant activities of tellurium-containing phycobiliproteins from tellurium-enriched *Spirulina platensis*

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Abstract: Tellurium-containing phycocyanin (Te-PC) and allophycocyanin (Te-APC), two organic tellurium (Te) species, were purified from tellurium-enriched *Spirulina platensis* by a fast protein liquid chromatographic method. It was found that the incorporation of Te into the peptides enhanced the antioxidant activities of both phycobiliproteins. With fractionation by ammonium sulfate precipitation and hydroxylapatite chromatography, Te-PC and Te-APC could be effectively separated with high purity, and Te concentrations were 611.1 and 625.3 μg g⁻¹ protein in Te-PC and Te-APC, respectively. The subunits in the proteins were identified by using MALDI-TOF-TOF mass spectrometry. Te incorporation enhanced the antioxidant activities of both phycobiliproteins, as examined by 2,2′-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) assay. Moreover, Te-PC and Te-APC showed dose-dependent protection on erythrocytes against the water-soluble free radical initiator 2,2′-azo(2-asmidinopropane)dihydrochloride-induced hemolysis. In the hepatoprotective model, apoptotic cell death and nuclear condensation induced by tert-butyl hydroperoxide in HepG2 cells was significantly attenuated by Te-PC and Te-APC. Taken together, these results suggest that Te-PC and Te-APC are promising Te-containing proteins with application potential for treatment of diseases related to oxidative stress.

Keywords: tellurium, phycocyanin, allophycocyanin, purification, antioxidant activity

Introduction

*Spirulina platensis* (*S. platensis*), a blue-green algae, is commercially available for human consumption and used as a functional food because of its nutritional and medicinal properties.¹⁻⁴ Researches have shown that *S. platensis* is a good carrier for selenium (Se) accumulation.⁵⁻⁶ By way of aquatic medium, *S. platensis* is an easily Se-supplementable vegetable.⁷ The accumulated Se is mainly incorporated into proteins of *S. platensis* cells, and the selenized protein exists mainly as seleno-methionine and seleno-cysteine.⁸⁻⁹ Moreover, it has become clear that most physiologic effects of Se are exerted by selenoproteins.⁶⁻⁹,¹⁰ Our previous studies showed the separation of high-purity Se-containing phycocyanin (Se-PC) and allophycocyanin (Se-APC) from Se-enriched *S. platensis*.⁷,¹¹ Our research also indicated that purified Se-PC was a promising organic Se species with potential applications in cancer chemoprevention, and both Se-PC and Se-APC are promising organic Se species with potential applications in antioxidant therapy on human erythrocytes and hepatoprotective activities.⁷,¹¹,¹²

Tellurium (Te) is in the same group as sulfur and Se in the periodic table and is classified as a metalloid. The similarity of Se and Te encompasses not only their names and positions on the periodic table, but to a large degree their chemical and biochemical
properties.\textsuperscript{13,14} Se was found to be an essential element in the mid-1950s after being regarded as a toxic element for a long time.\textsuperscript{13,15} Currently, we know Se plays an important role in therapies for many diseases such as cancer, human immunodeficiency virus/acquired immunodeficiency syndrome, and aging.\textsuperscript{10,13,15} In contrast, Te is a rare but non-essential trace element.\textsuperscript{16} However, studies have found that the typical amounts of Te in the human body are larger than 0.5 g, which exceeds the levels of all other trace elements in humans.\textsuperscript{17} Although thought to be toxic, the toxicity of Te depends on the chemical form and the quantity of the element consumed. Te poisoning is characterized by nausea, somnolence, and garlic odor on the breath and in urine caused by dimethyl telluride.\textsuperscript{17} Te can also interfere with cholesterol synthesis by blocking the cytochrome P450 reductase-linked terminal oxidase.\textsuperscript{18} Toxicity of Te on the nervous system is suggestive of peripheral neuropathy during a period of active myellogenesis through alteration of the transcription of the myelin proteins.\textsuperscript{19} Widy-Tyszkieiwicz et al\textsuperscript{20} found that Te induced cognitive deficits in rats, similar to neuropathological changes in the central nervous system. Santos et al\textsuperscript{21} also demonstrated the hemolytic and genotoxic evaluation of organochalcogens in human blood cells in vitro. Recently, Roy and Harde\textsuperscript{22} found that Te tetrachloride and diphenyl ditelluride caused cytotoxicity in rat hippocampal astrocytes through induction of an intrinsic apoptotic pathway. However, the toxicity of Te has received less attention than that of Se, which may be due to the less frequent contact of humans with this element and its compounds.\textsuperscript{17}

In the last few decades, increasing attention has been paid to the investigation of Te compounds as therapeutic agents.\textsuperscript{23} There are two main types of Te compounds, inorganic Te and organic Te derivatives, which have been found to show antimicrobial, antihelmintic, antioxidant, immunomodulatory, and anticancer activities.\textsuperscript{17,23–30} For instance, organic Te compounds were also identified as potent inhibitors of cysteine proteases, indicating their promising application potential as antitumorigenic agents.\textsuperscript{29} RT-04, a novel organotellurium(IV) compound, was able to induce apoptosis in HL60 human promyelocytic leukemia cells by down-modulation of Bcl-2 expression.\textsuperscript{28} Moreover, an organotelluroxetane, RF-07, displayed significant protective effect on picrocarpine-induced status epilepticus, indicating its promising therapeutic potential as an antiepileptogenic agent.\textsuperscript{30} Many studies also found that small inorganic Te complexes, ammonium trichloro (dioxoethylene-O,O') tellurate (AS101) and Octa-O-bis-(R,R)-tartrate ditellurane (SAS), displayed excellent safety profiles and promising anti-cancer therapeutic potentials.\textsuperscript{17} Therefore, just like Se 50 years ago, it is possible that the toxic element Te will ultimately be found to be an essential element.\textsuperscript{14}

Compared to the numerous studies of Se-containing proteins, such as Se-PC and Se-APC,\textsuperscript{8,11,12,31–35} limited information on the chemical and biological properties of Te-containing proteins is available. Considering the biological similarity of Se and Te, some interesting and significant results might be attained. Thus, the objective of this study is to present a simple and efficient way to separate Te-PC and Te-APC from Te-enriched \textit{S. platensis}, and investigate the in vitro antioxidant activities of both phycobiliproteins. The obtained results confirmed the incorporation of Te into PC and APC, which caused no difference in identification of the subunits by MALDI-TOF-TOF. Most importantly, Te incorporation enhanced the antioxidant activities of both phycobiliproteins. Taken together, these results suggest that Te-PC and Te-APC are novel organic Te species with application potential in antioxidation, and deserve further development for treatment of diseases related to oxidative stress in the future.

**Materials and methods**

**Cultivation of high Te-enriched \textit{S. platensis}** \textit{S. platensis} was provided by the Research Center of Hydrobiology of Jinan University (Guangzhou, People’s Republic of China). All cultivations were conducted in 500 mL Erlenmeyer flasks containing 250 mL Zarrouk medium (pH 8.5) at 30°C±1°C with a light illumination of 4,000 lx and a 14:10 hour light:dark cycle. Te was added in the form of sodium tellurite (Na$_2$TeO$_3$) stepwise to obtain different final concentrations, as described in Table 1. \textit{S. platensis} was harvested and freeze dried on day 11. The dry weight of algal biomass was determined by drying the \textit{S. platensis} at 70°C in a vacuum oven until a constant weight was obtained.

**Purification of Te-PC and Te-APC**

The freeze dried Te-enriched \textit{S. platensis} cells were suspended in 1 mM Na-phosphate buffer (pH 6.8). The cell suspension was frozen (at −20°C) and thawed (at 4°C) ten

<table>
<thead>
<tr>
<th>Day 7 (mg L$^{-1}$)</th>
<th>Day 8 (mg L$^{-1}$)</th>
<th>Day 9 (mg L$^{-1}$)</th>
<th>Final Te concentration (mg L$^{-1}$)</th>
<th>Final biomass of \textit{S. platensis} (g L$^{-1}$)</th>
</tr>
</thead>
<tbody>
<tr>
<td>200</td>
<td>250</td>
<td>250</td>
<td>700</td>
<td>1.20±0.03</td>
</tr>
<tr>
<td>200</td>
<td>300</td>
<td>300</td>
<td>800</td>
<td>1.21±0.02</td>
</tr>
<tr>
<td>250</td>
<td>300</td>
<td>350</td>
<td>900</td>
<td>1.23±0.05</td>
</tr>
<tr>
<td>250</td>
<td>350</td>
<td>400</td>
<td>1,000</td>
<td>1.11±0.04</td>
</tr>
</tbody>
</table>

**Table 1 Stepwise addition time and concentration of Te in \textit{Spirulina platensis} and the final biomass of \textit{S. platensis}**

**Abbreviation**: Te, tellurium.
times and was then sonicated for 10 minutes (using a Sonics VCX 600 system, 200 W). Next, the sonicated cells were centrifuged at 10,000× g (using a Millipore microcentrifuge; EMD Millipore, Billerica, MA, USA) for 60 minutes to collect the Te-PC and Te-APC containing supernatant for further purification.

The above supernatant was fractionated by precipitation with ammonium sulfate at 30% saturation, followed by 65% saturation. The precipitate obtained from 30% saturation of ammonium sulfate was discarded, and the supernatant was further brought to 65% saturation of ammonium sulfate. The collected precipitate from 65% saturation of ammonium sulfate was redissolved in 5 mL Na-phosphate buffer (1 mM, pH 6.8) and dialyzed overnight at 4°C against the same buffer. Then, the supernatant proteins containing mainly Te-PC and Te-APC were collected by centrifugation at 10,000× g (using an EMD Millipore microcentrifuge) for 60 minutes at 4°C.

Next, the above supernatant was applied to a Bio-Logic DuoFlow chromatography system equipped with a Bio-Rad Econo-column (1.5×30 cm) packed with hydroxyapatite (HA). The column was developed with six different concentration gradients of Na-phosphate buffer with 0.2 M NaCl (pH 6.8): namely, 0.001, 0.005, 0.01, 0.02, 0.05, and 0.10 M. The flow rate was 0.5 mL/min, and the effluent was collected in 2 mL fractions. All fractions having a purity ratio of absorbance (A\text{280}/A\text{650}) >4.5 and A\text{450}/A\text{280} >2.5 were pooled, and they possessed a high purity of Te-PC and Te-APC, respectively.

Te-APC was purified for the second time to obtain higher purity with the same HA chromatography system. This time, the column was developed with four different concentration gradients of Na-phosphate buffer with 0.2 M NaCl (pH 6.8): that is, 0.005, 0.02, 0.05, and 0.10 M. All fractions having a purity ratio of A\text{650}/A\text{280} >5.0 were pooled, which contributed to the high purity of Te-APC. The purification procedures for PC and APC from S. platensis were the same as that for Te-PC and Te-APC from Te-enriched S. platensis, as described previously in this section.

Spectroscopic measurements

All spectra were recorded at room temperature. Ultraviolet–visible (UV–VIS) absorption spectra were measured on a CARY 500 spectrophotometer between 250 and 700 nm, with a 1 cm path length. Purity ratio (A\text{450}/A\text{280} and A\text{650}/A\text{280}) was calculated based on the UV–VIS spectra. The fluorescence emission spectra were recorded on a F4500 Fluorescence spectrometer (Hitachi Ltd., Tokyo, Japan).

Sodium dodecyl sulfate polyacrylamide gel electrophoresis analysis

Sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS–PAGE) on 15% polyacrylamide gel was carried out using the Laemmli buffer system. Each protein sample was mixed with an equal volume of sample buffer containing 2% (w/v) SDS, 10% (v/v) glycerol, 5% (v/v) 2-mercaptoethanol, 0.002% (w/v) bromophenol blue, and 60 mM Tris (pH 6.8), and boiled for 10 minutes. At room temperature, electrophoresis was carried out, and the gel was stained with colloidal coomassie blue.

In-gel digestion

Protein gel bands, carefully excised from colloidal coomassie blue stained gels, were excised on extremely clean surfaces and cut into small pieces with a surgical needle. Then, the in-gel digestion was done according to the following protocol. Gel pieces were destained with 100 mM NH\text{4HCO\text{3}} in 50% methanol (v/v) three times. Then, the gel pieces were dehydrated with acetonitrile (ACN) and dried by speed vacuum (SAVANT™ refrigerated condensation trap system). The dried gel samples were incubated in a trypsin solution containing 50 mM NH\text{4HCO\text{3}} and 40 μg/μL trypsin for 30 minutes on ice. After addition of a buffer solution (25 mM NH\text{4HCO\text{3}}) to cover all gel pieces, the gel samples were incubated at 32°C overnight. Next, the digested peptides were extracted from the gel pieces with buffer solution and washing solution (ACN containing 5% trifluoroacetic acid [TFA]) by sonication for 10 minutes (Branson 5210 system). The extraction was repeated twice. There were three extractions in total. Finally, the supernatant containing the digested peptides was dried down using a speed vacuum.

Peptide mass fingerprint analysis by MALDI-TOF-TOF mass spectrometry and database searching

The digested peptide samples were dissolved in 0.1% TFA and applied to a Zip Tip u-18 column (EMD Millipore) for desalting. Then, the desalted peptide samples were redissolved in 50% ACN containing 0.1% TFA and were added onto a 192-well MALDI plate and air-dried prior to analysis in the MALDI-TOF-TOF system. The α and β subunit fractions of Te-PC and Te-APC collected from SDS–PAGE were dried by speed vacuum prior to trypsin digestion. The digested subunit samples were directly added onto the 192-well MALDI plates for analysis. MALDI-TOF-TOF analysis was performed using an ABI Applied Biosystems
4700 Proteomics Analyzer (GE Healthcare UK Ltd., Little Chalfont, UK). Mass spectra were obtained using a laser (337 nm, 200 Hz) as a desorption ionization source. Data were acquired in the reflection positive mode using delayed extraction. Spectra were calibrated using trypsin autolysis products (m/z 842.51 and 2,211.10) as internal standards. After mass spectrometry (MS) acquisition, the ten strongest peptides per spot were selected automatically for tandem MS analysis. Identification of proteins was performed by searching against the National Center for Biotechnology Information (NCBI, Bethesda, MD, USA) non-redundant protein sequence database. The peptide mass tolerance was set as 0.5 Da (50 ppm), and variable modifications of oxidation and carbamidomethylation were considered. Automatic data analysis and database searching were carried out by the GPS Explore software (Thermo Fisher Scientific, Waltham, MA, USA). The probability score generated was used as criterion for identification. Proteins with total score >59 or best ion score >29 were considered to be credible.

**Determination of total Te, inorganic Te, and organic Te concentrations**

Te concentration was determined by inductively coupled plasma atomic emission spectroscopy (ICP-AES; PerkinElmer Inc., Waltham, MA, USA) method, as previously described. Briefly, the sample was digested with 3 mL concentrated nitric acid and 1 mL H2O2 in a digestive stove (Shanghai Qian Jian Instruments Co, Ltd., Shanghai, People’s Republic of China) at 180°C for 3 hours. Then, the digested product was reconstituted to 10 mL with Milli-Q H2O and used for total Te determination. To determine inorganic Te concentration, the dried algal samples were extracted with 15% HCl; then, the extract was analyzed by ICP-AES directly. Organic Te concentration was calculated as the difference between the total Te concentration and the inorganic Te concentration.

**Trolox equivalent antioxidant capacity assay**

ABTS⁺ free radical scavenging activities of antioxidants were measured as follows. The 2,2’-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid (ABTS) reagent was prepared by passing the ABTS stock solution (5 mM) through manganese dioxide to allow the completion of radical generation; then, the solution was passed through a 0.2 μM polyvinyl difluoride (PVDF) syringe filter. After that, the filtrate was diluted with 5 mM phosphate-buffered saline (PBS) buffer (pH 7.4) to an absorbance of 0.70±0.02 at 734 nm, as recorded on a spectrophotometer (Genesys 5; Spectronic Instruments, Inc., Rochester, NY, USA). To determine the antioxidant activity, 1 mL of ABTS reagent was mixed with 50 μL of sample or positive control. The absorbance was then measured at 734 nm, 6 minutes after the initiation of mixing.

**Erythrocyte hemolysis assay**

The AAPH (2,2’-azobis[2-amidinopropane]dihydrochloride) free radical scavenging activities of antioxidants were measured as the inhibition of erythrocyte hemolysis, as per our previous study.

**Cell culture and determination of cell viability**

HepG2 human hepatocarcinoma cell line was obtained from the American Type Culture Collection (ATCC, Manassas, VA, USA). The cells were grown in Dulbecco’s Modified Eagle’s Medium (DMEM) supplemented with fetal bovine serum (10%), penicillin (100 IU/mL), and streptomycin (50 units/mL) at 37°C in a humidified incubator with 5% CO2 atmosphere. Cell viability was determined by measuring the ability of the cells to metabolize 3-[4,5-dimethyl-thiazol-2-yl]-2,5-diphenyltetrazolium bromide (MTT) to a purple formazan dye, as previously described.

**Flow cytometric analysis**

Flow cytometric analysis was carried out according to our previous study. Briefly, HepG2 cells after treatments were harvested by centrifugation and washed with PBS. Cells were stained with propidium iodide (Sigma-Aldrich Co., St Louis, MO, USA) after fixation with 70% ethanol at −20°C overnight. Labeled cells were washed with PBS and then analyzed by flow cytometer. The cell cycle distribution was analyzed using MultiCycle software (Phoenix Flow Systems, San Diego, CA, USA). The proportions of cells in gap 0/gap 1 (G1), synthesis, and gap 2/mitosis phases were represented as deoxyribonucleic acid (DNA) histograms. Apoptotic cells with hypodiploid DNA contents were measured by quantifying the sub-G1 peak. For each experiment, 10,000 events per sample were recorded.

**4,6-diamidino-2-phenylindole staining**

Nuclear staining was examined by fluorescence staining using DAPI (4,6-diamidino-2-phenylindole; Sigma-Aldrich) as a molecular probe. Briefly, cells cultured in chamber slides were fixed with 3.7% formaldehyde and permeabilized with 0.1% Triton X-100 in PBS. Cells were incubated with 1 μg/mL DAPI for 15 minutes at 37°C. After that, cells were washed with PBS and examined with a fluorescence
microscope (Nikon Eclipse 80i, magnification 200×; Nikon Corporation, Tokyo, Japan).

Statistical analysis
Data were expressed as mean ± standard deviation (SD) from at least three independent experiments. The difference between two groups was analyzed by two-tailed Student’s t-test, and the significance among three or more groups was analyzed by one-way analysis of variance (ANOVA) multiple comparisons. In all analyses, a value of P<0.05 was considered statistically significant.

Results and discussion
Accumulation and transformation of tellurite in S. platensis, and the effects on the biomass concentration
Our previous studies have showed that S. platensis is a good carrier for Se.3 In the preliminary screening of the current study, we have tested the tolerance of S. platensis to tellurite for a wide range of concentrations from 0 to 1,000 mg/L. The results showed that S. platensis was resistant to tellurite. This study aimed to purify Te-containing proteins from the microalgae. Therefore, we chose the higher concentrations of tellurite for Te accumulation with the purpose of obtaining higher Te-containing proteins. As shown in Figure 1, we demonstrated that tellurite could be efficiently accumulated in S. platensis during cultivation, and inside the cells tellurite was transformed to organic Te at a high ratio. The percentage of organic Te in S. platensis was 81.3, 76.1, 82.6, and 76.8%, under different Te concentrations in the culture medium. Even at a high tellurite concentration of 1,000 mg L⁻¹, S. platensis could still transform tellurite to organic Te. In order to investigate whether the tellurite accumulation and biotransformation would significantly (P<0.05) affect the growth of S. platensis, we monitored the changes in the final biomass after Te addition (Table 1). The results showed that the final biomass of S. platensis increased by 3.46%, 4.76%, and 6.06% under the final tellurite concentrations of 700, 800, and 900 mg L⁻¹, respectively. The highest final tellurite concentration (1,000 mg L⁻¹) resulted in the lowest biomass concentration (3.90% less than that of the control), possibly due to the toxic effect of high Te stress. Therefore, just like Se, tellurite also had either stimulating or toxic effects on S. platensis, depending on the tellurite level. Interestingly, during the experiments, strong garlic-like odors could be smelled. This information further confirmed the transformation of tellurite into volatile Te products by S. platensis cells, which might mitigate the toxicity of tellurite.11 As a result, S. platensis possessed tolerance to high levels of tellurite, and it was a good carrier for tellurite accumulation.

Purification of Te-PC and Te-APC, and determination of Te concentration
Te-enriched S. platensis has been obtained by adding tellurite into S. platensis. Considering the percentage of organic Te in S. platensis and the toxicity of tellurite, we used the final tellurite concentration of 700 mg L⁻¹ to acquire Te-enriched S. platensis, from which Te-PC and Te-APC were purified (Figure 2, Table 2 and 3).

Phycocyanin (PC), a natural blue dye with many commercial uses, is the major soluble protein extracted from cyanobacteria such as S. platensis.37 Allophycocyanin (APC) is the minor soluble protein component of S. platensis cell extract. To separate Te-PC and Te-APC, we applied a two-step precipitation procedure with ammonium sulfate at 30% and 65% saturation. The initial 30% saturation of ammonium sulfate removed contaminating proteins, and the next 65% saturation of ammonium sulfate precipitated both the Te-PC and Te-APC crude fractions, which were then subjected to HA chromatography. The elution involved six different concentration gradients of 0.001, 0.005, 0.01, 0.02, 0.05, and 0.10 M Na-phosphate buffer with 0.2 M NaCl (pH 6.8). From this step, a good separation of Te-PC from Te-APC was achieved (Figure 2A). Te-PC and Te-APC were eluted in the fourth and sixth concentration gradients of 0.02 and 0.10 M, respectively. The purity ratio (A₆52/A₂₈₀) of Te-PC was up to 5.59, and the purity ratio (A₆52/A₂₈₀) of Te-APC was 3.11. Therefore, both the two-step precipitation and the six different concentration gradients elution procedure contributed to the efficient separation of contaminating protein and Te-PC from Te-APC.
Further purification of Te-APC was achieved by the second application of HA chromatography (Figure 2B). The purity ratio was enhanced from 3.11 to 5.16. As compared with other published purification methods, our method was more convenient and efficient, with good purity performance.

Our previous research indicated that under photoautotrophic culture conditions, the organic Se in *S. platensis* accounted for about 80% of the total Se accumulated in the algal cells and was comprised of about 60% of the water-soluble protein-bound form. Organic Te demonstrated similar characterization in *S. platensis*, accounting for 50% of the water-soluble protein-bound form. We obtained the Te concentration of 611.1 μg g⁻¹ in the Te-PC after HA chromatography once, and the Te concentration of 625.3 μg g⁻¹ in the Te-APC after HA chromatography twice. Further confirmation for Te-PC and Te-APC was then performed, as described in the “Characterization of Te-PC and Te-APC by UV–VIS and fluorescence spectroscopy” section.

### Characterization of Te-PC and Te-APC by UV–VIS and fluorescence spectroscopy

The standard analytical methods for characterization of PC and APC are UV–VIS spectroscopy, fluorescence spectroscopy, and SDS–PAGE of the denatured proteins.

As shown in Figure 3A, the characteristic absorption peak at 618 nm and strong absorbance below 300 nm in the UV–VIS spectrum of Te-PC verified the presence of the Te-PC. The fluorescence spectrum of Te-PC showed maximum emission at 646 nm (Figure 3C), which was in accordance with the spectral property of PC reported in the literature.

Te-APC was corroborated by the characteristic absorption peak at 652 nm, a shoulder at 620 nm, and strong absorbance at 280 nm in the UV–VIS spectra (Figure 3B). The decrease in relative intensity of peaks at 280 nm and 620 nm indicated the successive improvement in the purity of Te-APC in terms of purity ratio ($A_{652}/A_{280}$). The fluorescence spectrum of purified Te-APC showed maximum emission at 660 nm (Figure 3D), which was in agreement with the spectral property of APC reported in the literature. Moreover, the lack of fluorescence emission around 646 nm indicated the efficient separation of Te-APC from Te-PC. From the above results, we found that Te-PC and PC exhibited similar spectra, the same as Te-APC and APC. It could be concluded that the incorporation of Te into PC and APC had no effect on the spectral properties of these proteins.

### Table 2: Purity ratio and Te concentration of Te-PC obtained in different purification steps

<table>
<thead>
<tr>
<th>Purification step</th>
<th>Purity ratio ($A_{652}/A_{280}$)</th>
<th>Te concentration (μg g⁻¹ protein)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Crude extract</td>
<td>0.31</td>
<td>372.0</td>
</tr>
<tr>
<td>Precipitation with 30% saturation of (NH₄)₂SO₄</td>
<td>0.41</td>
<td>405.5</td>
</tr>
<tr>
<td>HA chromatography</td>
<td>5.59</td>
<td>611.1</td>
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</tbody>
</table>

**Abbreviations:** Te-PC, tellurium-containing phycocyanin; HA, hydroxyapatite; Te, tellurium.

### Table 3: Purity ratio and Te concentration of Te-APC obtained in different purification steps

<table>
<thead>
<tr>
<th>Purification step</th>
<th>Purity ratio ($A_{652}/A_{280}$)</th>
<th>Te concentration (μg g⁻¹ protein)</th>
</tr>
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<tbody>
<tr>
<td>Crude extract</td>
<td>0.22</td>
<td>372.0</td>
</tr>
<tr>
<td>Precipitation with 30% saturation of (NH₄)₂SO₄</td>
<td>0.23</td>
<td>405.5</td>
</tr>
<tr>
<td>The first HA chromatography</td>
<td>3.11</td>
<td>603.2</td>
</tr>
<tr>
<td>The second HA chromatography</td>
<td>5.16</td>
<td>625.3</td>
</tr>
</tbody>
</table>

**Abbreviations:** Te-APC, tellurium-containing allophycocyanin; HA, hydroxyapatite; Te, tellurium.
SDS–PAGE and identification of α and β subunits by peptide mass fingerprint

SDS–PAGE analysis indicated two major bands corresponding to α and β subunits of both Te-PC and Te-APC (Figure 4A). For Te-PC, the molecular masses of α and β subunits were estimated to be 17.6 kDa and 18.1 kDa, respectively (Figure 4B); thus, the molecular mass of the Te-PC monomer (α, β) was calculated to be 35.7 kDa. For Te-APC, molecular masses of α and β subunits were 17.1 kDa and 18.4 kDa, respectively (Figure 4C), and the molecular mass of the Te-APC monomer (α, β) was 35.5 kDa.

In order to identify α and β subunits separated by SDS–PAGE, peptide mass fingerprinting (PMF) by using MALDI-TOF-TOF mass spectrometry was employed in the present study. The introduction of the Zip Tip column for desalting the digested protein samples, and the application of sonication to facilitate extraction of the peptides resulted in superior performance of the subsequent PMF analysis. Table 4 shows the accession numbers, protein molecular masses, the number of peptides identified in each subunit, and the protein score of each PMF analysis. The protein bands excised from the SDS–PAGE gel of purified PC, Te-PC, APC, and Te-APC were well identified by MALDI-TOF-TOF, which confirmed the effectiveness of this fast protein liquid chromatographic system to purify Te-PC and Te-APC from *Synechococcus platensis*. Taken together, the results from UV–VIS, fluorescence spectroscopy, SDS–PAGE, and PMF analyses fully supported the successful incorporation of Te into PC and APC, and there was no change in the structures of PC and APC. Furthermore, it was observed that the incorporation of Te into PC and APC caused no difference in identification of the subunits by MALDI-TOF-TOF.
Antioxidant activities of Te-PC and Te-APC

In a free radical scavenging assay, the relatively long-lived ABTS•⁺, generated by the direct oxidation of ABTS with manganese dioxide, is decolorized during the reaction with hydrogen-donating antioxidants. This assay provides a direct comparison of the antioxidant activities of tested samples as expressed in terms of percentage inhibition. Te-PC, PC, Te-APC, and APC (Figure 5) all inhibited the formation of ABTS free radicals in a concentration-dependent effect. Moreover, Te-PC and Te-APC demonstrated stronger inhibition than PC and APC. For example, Te-PC inhibited ABTS oxidation by 28.0% at 0.0125 mg mL⁻¹ and up to 97.2% at 0.1 mg mL⁻¹, which were higher percentages than those of PC (23.5% and 92.1%, respectively). In previous papers, we found that both Se-PC and Se-APC performed stronger antioxidant activities than PC and APC.⁷,¹¹,¹² So, it was indicated in this paper that Te maintained similar antioxidant activity as Se, and the incorporation of Te into the peptides enhanced the antioxidant activities of both phycobiliproteins. The results also demonstrated that PC and Te-PC had stronger inhibition than APC and Te-APC. At the concentration of 0.1 mg mL⁻¹, the scavenging activity of Te-PC (97.2%) was 2.4 times higher than that of Te-APC (39.9%). Moreover, Te-PC and PC also showed significantly (P<0.05) high inhibition on trolox (6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid), which was also a well-known antioxidant (data not shown).

The antioxidant activities of Te-PC and Te-APC were further confirmed by erythrocyte hemolysis assay, in which AAPH radicals could attack the erythrocyte cell membrane components, including proteins and lipids, and cause changes in the structure and function of the membranes. As shown in Figure 5C, Te-PC and Te-APC showed dose-dependent protection on erythrocytes against the AAPH-induced hemolysis. For instance, Te-PC inhibited the hemolysis by 45.6% at 0.025 mg mL⁻¹ and by up to 78.6% at 0.2 mg mL⁻¹, which were higher percentages than those of Te-APC (32.2% and

Table 4 Identification of α and β subunits of PC, Te-PC, APC, and Te-APC by MALDI-TOF-TOF

<table>
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<tr>
<th>Protein</th>
<th>Subunit</th>
<th>Accession no</th>
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<th>No of peptide identified</th>
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<td>gi</td>
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<td>291568829</td>
<td>18,429.9</td>
<td>11</td>
</tr>
</tbody>
</table>

Notes: Results searched from NCBI non-redundant protein sequence database. Te-PC and Te-APC were identified as PC and APC, respectively, since the NCBI database contains no sequence information on Te-containing proteins.

Abbreviations: PC, phycocyanin; Te-PC, tellurium-containing phycocyanin; APC, allophycocyanin; Te-APC, tellurium-containing allophycocyanin; MALDI-TOF-TOF, matrix assisted laser desorption/ionization-time of flight/time of flight; No, number; MW, molecular weight, NCBI, National Center for Biotechnology Information.
Figure 5 (A) Antioxidant activities of Te-PC and (B) Te-APC as determined by ABTS assay and (C) hemolysis assay.

Notes: Values expressed are means ± SD of triplicates. *The significant difference between the two tested groups at the same concentration is indicated at $P<0.05$.

Abbreviations: PC, phycocyanin; Te-PC, tellurium-containing phycocyanin; APC, allophycocyanin; Te-APC, tellurium-containing allophycocyanin; ABTS, 2,2′-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid; SD, standard deviation.

Figure 6 Te-PC, PC, Te-APC, and APC inhibit t-BOOH-induced apoptosis in HepG2 cells.

Notes: The cells were pre-incubated with different proteins for 12 hours and then co-incubated with 100 μM t-BOOH for 10 hours. Cells after treatment were collected and stained with PI after fixation by 70% ethanol. Apoptotic cells with hypodiploid DNA content were measured by quantifying the sub-G1 peak. Bars with different characters (a–d) are statistically different at $P<0.05$.

Abbreviations: PC, phycocyanin; Te-PC, tellurium-containing phycocyanin; APC, allophycocyanin; Te-APC, tellurium-containing allophycocyanin; t-BOOH, tert-butyl hydroperoxide; PI, propidium iodide; DNA, deoxyribonucleic acid; sub-G1, sub-gap 1.
58.7% at the same concentrations). The difference in antioxidant activities of Te-PC and Te-APC obtained by this assay was quite consistent with those obtained by ABTS, DPPH (2,2-diphenyl-1-picryl-hydrazyl), and superoxide anion scavenging assays. Taken together, and in comparison with PC and APC, the incorporation of Te into proteins significantly enhances their antioxidant activities. This observation is consistent with those of Se-PC and Se-APC, in which the increased antioxidant activities were due to the incorporation of Se into the protein as seleno-amino acids.7 Possibly, Te was incorporated into Te-PC and Te-APC in organic forms like Te-amino acids. Previous studies have demonstrated the antioxidative properties of organotellurium compounds in cell systems.24,25 Therefore, considering the biological similarity between Se and Te and the comparison of the antioxidant activities, we concluded that the incorporation of organic Te into PC and APC accounts for the enhanced antioxidant activities.

Te-PC and Te-APC attenuate tert-butyl hydroperoxide-induced apoptosis in HepG2 cells

The livers in humans and animals are particularly susceptible to toxic and oxidative insults, because the portal vein brings blood to this organ after intestinal absorption. The absorbed drugs and xenobiotics in a concentrated form can cause over-production of reactive oxygen species (ROS) and other free radicals, which could result in inflammatory and fibrotic processes.11 Our previous studies11 have established a hepatoprotective model by using the HepG2 cell line as a cell model and with tert-butyl hydroperoxide (t-BOOH) as an oxidation agent. The cytotoxicity of t-BOOH toward HepG2 cells was mainly attributed to the induction of ROS-mediated apoptosis.11 Therefore, in the present study, this model was used to examine the in vitro antioxidant activities and hepatoprotective effects against free radicals. The cytotoxic effects of t-BOOH and the protective effects of Te-PC, PC, Te-APC, and APC on HepG2 cells were first examined by DNA flow cytometric analysis. As shown in Figure 6, exposure of cells to 100 μM t-BOOH resulted in a marked increase in the proportion of apoptotic cells, as reflected by the increase in sub-G1 peaks from 4.3% (control) to 36.9%. However, pretreatment of the cells with 2.5 and 5 μM Te-PC and Te-APC, respectively, significantly blocked the t-BOOH-induced apoptosis. Their protective effects were significantly higher than those of PC and APC at 5 μM. These results demonstrated that Te incorporation significantly enhanced the hepatoprotective effects against free radicals. Moreover, we showed that 5 μM Te-PC, PC, Te-APC, and APC alone did not cause an increase in apoptotic sub-G1 peak, indicating that these proteins are non-toxic to the cells under the effective protection concentrations.

The protective effects of Te-PC and Te-APC on HepG2 cells against t-BOOH-induced apoptosis were further characterized by DAPI staining assay. After the exposure to 100 μM t-BOOH, HepG2 cells exhibited typical morphological alterations, while cells exposed to Te-PC and Te-APC alone showed normal nuclei of oval shape, which clearly demonstrates that Te-PC and Te-APC block t-BOOH-induced apoptosis in HepG2 cells.

Conclusion

In summary, we purified Te-PC and Te-APC from Te-enriched S. platensis by a fast, simple, efficient liquid
chromatographic method. Te was incorporated into PC and APC, as identified by MALDI-TOF-TOF analysis. Most importantly, Te incorporation enhanced the antioxidant and hepatoprotective activities of both phycobiliproteins. For future applications, the Te-containing proteins inside human bodies could be digested by different enzymes, such as pepsin, trypsin, and α-chymotrypsin. The enzymatic digestion could cause the active peptides and amino acids to become exposed to the environment, and thus enhance their biological activities. Therefore, Te-PC and Te-APC are novel organic Te species with application potential in antioxidation, and deserve further development for treatment of diseases related to oxidative stress in the future.

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Disclosure

The authors declare that there are no conflicts of interest in this work.

References


