Nanotechnology-based drug delivery systems for treatment of oral cancer: a review

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Abstract: Oral cancer (oral cavity and oropharynx) is a common and aggressive cancer that invades local tissue, can cause metastasis, and has a high mortality rate. Conventional treatment strategies, such as surgery and chemoradiotherapy, have improved over the past few decades; however, they remain far from optimal. Currently, cancer research is focused on improving cancer diagnosis and treatment methods (oral cavity and oropharynx) nanotechnology, which involves the design, characterization, production, and application of nanoscale drug delivery systems. In medicine, nanotechnologies, such as polymeric nanoparticles, solid lipid nanoparticles, nanostructured lipid carriers, gold nanoparticles, hydrogels, cyclodextrin complexes, and liquid crystals, are promising tools for diagnostic probes and therapeutic devices. The objective of this study is to present a systematic review of nanotechnology-based drug delivery systems for oral cancers.

Keywords: targeted delivery, oral squamous cell carcinoma, oral cancer treatment

Introduction

Oral squamous cell carcinoma (OSCC) is the sixth most common cancer for both sexes worldwide, and the 5-year survival rate for this disease is approximately 50%.1 In 2011, OSCC accounted for nearly 3% of all cancer cases worldwide; its estimated incidence is approximately 275,000 cases per year, with two-thirds of these cases occurring in developing countries.1 In 2013, there were 41,380 estimated new cases of oral cavity and pharyngeal cancer in the US for both sexes, with 32.8% associated with the tongue, 27.5% with the mouth, 33.7% with the pharynx, and 5.9% in other parts of the oral cavity. There were also 7,890 estimated deaths in the US, of which an estimated 19.1% were new cases, with 26.2% associated with the tongue, 22.8% with the mouth, 30.4% with the pharynx, and 20.8% in other parts of the oral cavity.2 OSCC is a malignant tumor of the squamous epithelium lining the oral mucosa. These tumors are malignant and tend to spread rapidly. The main causes of oral cancer include excessive alcohol intake and tobacco use.3–5 Exposure to sunlight is a causative factor for cancer of the lips, which is similar to that for skin cancer.6–9 Human papilloma virus is also a risk factor for causing oral cancer.10–13 Immunosuppressed patients (eg, human immunodeficiency virus [HIV] and renal transplant patients) have the highest risk factor for developing oral cancer.14,15

The prevalence has shown a 5.3-fold increase for men and a twofold increase for women over the past 2 decades.16 In addition, the annual death toll for oral cancer in males has been rapidly increasing.17 The 5-year mortality rate for oral cancer is approximately 50% worldwide,18–20 which signifies a poor prognosis for developing countries.1 The rates for OSCC recurrence vary from 18% to 76% for patients who undergo standard treatment, and a delay in starting treatment is considered the major cause for no relevant improvement in the survival rate.21
Pathophysiology of oral cancer

The genomic pathway plays a role in OSCC, whereby alterations in the genome result in the altered expression of proteins, chemical mediators, and enzymes.32 Carcinogenesis is a process with multiple steps, each characterized by the sequential stimulation of additional genetic defects followed by clonal expansion. Due to oncogene activation and tumor suppressor gene inactivation, OSCC leads to the deregulation of cell proliferation and death. The genetic alterations include gene amplification, oncogene overexpression, mutations, deletions, and hypermethylation, leading to the inactivation of particular genes such as the p53 tumor suppressor gene.33

Oncogenes do not play an important role in the cancer process, although they do play a role in initiation. Initiation transforms a normal cell into a premalignant cell, and it requires the inactivation of negative regulators in the cell (eg, tumor suppressor genes), which is considered an important event that leads to the development of malignancy. Tumor suppressor genes are most often inactivated by point mutations, deletions, and rearrangements in both gene copies.34,35

Mutations in p53 and p16 are involved in the carcinogenesis process. The p53 gene plays a role in maintaining genomic stability, cell cycle progression, cellular differentiation, DNA repair, and apoptosis, and p16 is involved in cell cycle regulation, including cell cycle arrest and apoptosis.36

The tumor suppressor gene p53 is known to be mutated in approximately 70% of all adult solid tumors.37 These p53 gene mutations have been associated with smoking and the use of tobacco in squamous cell carcinomas of the head and neck.38 An in vivo study of functionally inactivated p53 in...
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<th>Abbreviation</th>
<th>Drug</th>
<th>Carrier</th>
<th>Composition</th>
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<tr>
<td>NLC</td>
<td>Docetaxel</td>
<td>Stearic acid, glyceryl monostearate, soya lecithin, oleic acid</td>
<td>To prolong the drug release and increase the drug inhibitory effect.</td>
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<td>Etoposide</td>
<td>Monostearin, soybean oil, soya lecithin, PEG, DSPE</td>
<td>To enhance bioavailability.</td>
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<td>PNP</td>
<td>Anti-epidermal growth factor receptor antibody</td>
<td>Gold</td>
<td>To provide a potential technique for oral cancer diagnosis.</td>
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<td>PNP</td>
<td>Anti-epidermal growth factor receptor antibody</td>
<td>Gold</td>
<td>To establish early diagnosis of oral cancer.</td>
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<td>Polycationic liposome (Metafectene®; Biontex Laboratories GmbH, München, Germany)</td>
<td>To study the effect of plasmonic photothermal therapy.</td>
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<td>PHE, NIPAAm, EBA</td>
<td>To investigate the FOXO3a activity-modulating and antitumor effects of rapamycin and cisplatin in OSCC cells.</td>
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<td>To improve cytotoxic activity against various tumor cells.</td>
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**Abbreviations:** CD, cyclodextrin; DSPE, distearoylphosphatidylethanolamine; EBA, N,N'-ethylenebis-acrylamide; LC, liquid crystal; LCNP, liquid crystalline nanoparticle; NIPAAm, N-isopropylacrylamide; NLC, nanostructured lipid carrier; PNP, polymeric nanoparticle; OSCC, oral squamous cell carcinoma; PBLG, poly[(γ-benzyl L-glutamate)]; PEG, polyethylene glycol; PHE, N-acryloyl-l-phenylalanine; SLN, solid lipid nanoparticle.
oral tumors and the restoration of p53 in oral cancer lines and tumors induced in animal models demonstrated reversal of the malignant phenotype.39

Another OSCC characteristic is telomerase activity. Several oral tumors have been confirmed to have the expression of telomerase, which is strongly associated with malignancy in oral tissues. Telomerase activity has been identified in OSCC, with 80% of patients with head and neck squamous cell carcinoma40 having telomerase activity, and it has been reported that most immortal OSCC cell lines have high levels of telomerase and have tumor radioresistance.41,42

Other chemical mediators are involved in oral cancer pain, such as endothelin-1 (ET-1), proteases, and nerve growth factor.43 ET-1 is a potent vasoactive peptide that produces nociception. In oral cancer, ET-1 binds to the endothelin-B receptor and is expressed on nonmyelinating Schwann and dorsal root ganglion satellite cells.44 In patients with OSCC, the ET-1 levels are higher in the tumor microenvironment, and nociception was reported with mechanical stimuli parallel to the mechanical allodynia.45,46 The role of ET-1 in oral cancer pain was confirmed and characterized in a mouse model by Pickering et al47 and the ET-1 concentration was a more important factor than tumor volume in establishing cancer pain.

Protease-activated receptor type 2 (PAR2) is involved in oral cancer.48,49 This receptor is activated by serine proteases, trypsin, and tryptase.50 PAR2 activates dual messenger pathways in a second step that sensitizes transient receptor potential vanilloid type-1 (TRPV1) and transient receptor potential vanilloid type-4 (TRPV4) receptors on nociceptive afferents where there is resulting TRPV1-dependent thermal hyperalgesia and TRPV4-dependent mechanical allodynia.51 In OSCC, the fibroblasts in the stroma produce trypsin, and this serine protease is capable of activating PAR2 on sensory neurons. This continual release of serine proteases in the microenvironment could produce an ongoing excitation of primary nociceptive afferents, leading to mechanical allodynia in oral cancer patients.49

In the microenvironment of many cancers, sensory neurons are chronically exposed to nerve growth factor (NGF).1,52 The acute peripheral administration of this chemical mediator leads to thermal hyperalgesia, whereas chronic administration produces mechanical allodynia.53 The activity of NGF is mediated via a receptor tyrosine kinase;54 thus, NGF can also facilitate the proliferation and invasion of multiple cancers,55,56 including oral cancer;56 a process related to pain. The pain mechanism in oral cancer can be established by association with perineural involvement, with invasion and proliferation of a cancer occurring within a nerve associated with pain.56,57 Higher NGF levels were found in cancer tissues from oral cancer patients.57

Angiogenesis is a crucial step in the processes of uncontrolled tumor proliferation and metastasis, and inhibiting angiogenesis is considered to be effective for treating oral cancer. Vascular endothelial growth factor (VEGF) is thought to be an important angiogenic factor,58 and studies have shown that OSCC is associated with an elevated VEGF concentration in the serum. These higher levels of VEGF are correlated with lymph node metastasis, clinical stage, and the prognosis and treatment of OSCC.59–62

Cancer cells induce the development of an exaggerated inflammatory state in the stroma, which in turn promotes cancer growth, invasion, and metastasis. Inflammatory cells in the microenvironment, such as myeloid dendritic cells, macrophage subtypes (M1 and M2), mast cells, neutrophils, and T and B lymphocytes, secrete chemokines, prostaglandins, proteases, and complement components that collectively adopt an exaggerated inflammatory state that promotes cancer growth, tissue invasion, and metastasis.63–65

The chemical mediators produced by an upregulation in inflammation include transforming growth factor beta (TGF-β), tumor necrosis factor alpha (TNF-α), interleukin (IL)-6, cyclooxygenase 2 (COX-2), and matrix metalloproteinase-7 (MMP-7).66,67 TNF-α and IL-6 are produced by malignant keratinocytes, stromal fibroblasts, and macrophages, and these cytokines promote tumor growth by modifying the expression of cell-adhesion molecules and extracellular matrix proteins and stimulate angiogenesis.68

A high level of COX-2 expression exists in stromal cells and cancerous cells at the invasive front in OSCC.69,70 Thus, COX-2 plays a role in the process of local invasion and metastasis.62 Increased COX-2 expression in OSCC is associated with a high rate of recurrence after treatment with a poor response to radiotherapy and poor prognosis.71 Matrix metalloproteinases are involved in the cell migration, angiogenesis, and proteolytic activation of growth factors, events necessary for invasion into surrounding connective tissue for neoplastic cells.72,73 MMP-7 plays a pivotal role in inflammatory diseases and malignant invasion by tissue remodeling72,74 and destroying the extracellular matrix, including the basement membrane, and this process is necessary for invasion and metastasis.75 Increased MMP-7 expression has been found to be related to oral cancer.72,76

Nuclear factor-kappa B (NF-κB) participates in the expression of genes involved in inflammatory and immune responses, cell proliferation, and survival.77 NF-κB protein levels gradually increase from the premalignant lesion stage to invasive cancer, indicating an important role for NF-κB
in the early stages of carcinogenesis. In OSCC, reduction in NF-κB activity results in low IL levels, including those for IL-2, IL-6, and IL-8. In addition, IL-8 plays a role in the induction of the angiogenesis process.

A number of complex mechanisms are involved in the genesis and progression of oral cancer. OSSC is a multistep process in which multiple genetic events occur that alter the normal function of oncogenes and tumor suppressor genes. These events can result in the increased production of growth factors. Recent advances in the understanding of the molecular control of these various pathways will allow for more accurate diagnosis and assessment of prognosis and might lead the way for more novel approaches for treatment and prevention.

**Oral cancer treatment**

Treatment protocols for oral cavity cancers are generalized therapies based on stage, chemoradiation therapy, and induction chemotherapy for locally advanced disease. In current therapies, some anticancer drugs have been used alone or in combination for the treatment of oral cancer, such as cisplatin, cetuximab, fluorouracil, paclitaxel, docetaxel (DTX), and methotrexate.

The oral administration of anticancer agents is preferred by patients for its convenience and potential for outpatient treatment. In addition, oral administration facilitates prolonged exposure to a cytotoxic agent. However, low solubility in aqueous fluids, low apparent permeability, and poor bioavailability are noted as limitations for oral chemotherapy. Intravenous administration is the most direct, and it overcomes the variable absorption patterns of the gastrointestinal tract. Intravenous administration leads to immediate and complete bioavailability; thus, this route has the potential to be hazardous because high concentrations of drugs are delivered to normal tissues, causing greater damage to healthy tissues and increased adverse reactions.

To overcome the disadvantages of current cancer treatment techniques, the scientific community has turned toward nanotechnology to develop new and more effective nanotechnology-based drug carrier systems to optimize oral, buccal, and intravenous treatment routes.

**Nanotechnology-based drug delivery systems**

**Nanoparticles**

Nanoparticles can be defined as ultradispersed solid supramolecular structures with a submicrometer size ranging from 10 to 1,000 μm. The drugs can be dissolved, entrapped, encapsulated, or attached to a nanoparticle matrix, which acts as a reservoir for particulate systems and therefore plays an important role as a drug delivery system for clinical applications, particularly in oncology.

Nanoparticles fabricated from polysaccharides, proteins, and biocompatible/biodegradable polymers, such as polyethylene glycol (PEG), poly(γ-benzyl l-glutamate) (PBLG), poly(D,L-lactide), poly(lactic acid) (PLA), poly(D,L-glycolide), poly(lactide-co-glycolide), polycyanacrylate, chitosan, gelatin, and sodium alginate are called PNP.

The nanoparticles (NPs) are mainly prepared via the dispersion of preformed polymers, the polymerization of monomers, ionic gelation, or the coacervation of hydrophilic polymers, but other methods for their generation have also been reported, such as supercritical fluid technology and particle replication in non-wetting templates (PRINT®; DeSimone Lab, Chapel Hill, NC, USA).

NPs can improve the stability of drugs and control their targeted delivery, allowing for a constant and uniform concentration at the site of a lesion and facilitating drug extravasation into the tumor system, thus reducing side effects.

Damascelli et al evaluated the effectiveness of the intra-arterial infusion of paclitaxel incorporated in NPs based on human albumin (albumin NPs) for use as induction chemotherapy before definitive advanced tongue cancer treatment. Paclitaxel is a lipophilic drug; therefore, surface-active agents must be added for dissolution in organic fluids. In addition, paclitaxel causes severe allergic reactions with intravenous use. Albumin NPs are attractive formulations because they can incorporate a significant amount of drugs into a particle matrix due to the different drug-binding sites present in albumin molecules. Damascelli et al reported that the intra-arterial infusion of paclitaxel in albumin nanoparticles is reproducible and effective.

Sulíkkarali et al investigated the anti-buccal tumor effects of naringenin (NAR)-loaded nanoparticles (NARNPs) prepared in a NAR:aminoalkyl methacrylate copolymers Eudragit® (Röhm GmbH & Co. KG, Darmstadt, Germany) E:polyvinyl alcohol (1:10:10; weight (w)/w) ratio by a nanoprecipitation method. NAR has promising pharmacological activity; however, it has low oral bioavailability, which is a crucial obstacle. The results of the study revealed that NARNPs have more potent antitumor effects than free NAR, preventing the formation of OSSC. In addition, NARNPs improved the biochemical status to a normal range in 7,12-dimethylbenz(a)anthracene-induced oral carcinogenesis. This result may be attributed to the fact that NAR nanoparticles can arrive at tumor sites via a process called “enhanced permeation and retention” due to the fact that the tumor tissue vasculature is porous with leaky endothelium.
which increases and sustains the drug concentration inside tumor cells over time, leading to higher antitumor efficacy compared with free NAR.\textsuperscript{119}

Yu et al also investigated the action of NPs against oral cancer. These authors assessed the anticancer effects of herpes simplex virus thymidine kinase (HSV-TK)-loaded PEG–PBLG nanoparticles and PEG–PBLG nanoparticle-mediated HSV-TK/ganciclovir nanoparticles toward OSCC.\textsuperscript{120} HSV-TK is a good apoptosis-inducing gene; however, its transference into the tumor is crucial. However, the results demonstrated that HSV-TK-loaded PEG–PBLG nanoparticles had a core-shell structure, DNA protection, and higher gene-transfer efficiency and released DNA gradually; thus, they can be used as gene carriers in future clinical applications. Furthermore, PEG–PBLG nanoparticle-mediated HSV-TK/ganciclovir had a strong anticancer effect on buccal carcinoma induced in golden hamsters.\textsuperscript{120}

In another study,\textsuperscript{121} the potential antitumor activity of cisplatin-loaded nanoparticles based on PEG-poly(glutamic acid) block copolymers was assayed in four OSCCs. The results showed that the growth inhibitory effects of cisplatin-loaded nanoparticles were significantly less than that for free cisplatin. However, the caspase-3 and -7 cascades, which are activated by a cisplatin stimulus, induced the release of cytochrome c from mitochondria and led to an irreversible commitment to apoptotic cell death in both cisplatin- and NC-6004-treated OSC-19 cells. Other interesting data obtained from this study revealed that nephrotoxicity, a crucial side effect of cisplatin-loaded nanoparticles, is much lower than that for free cisplatin. Therefore, it can be interpreted that these NPs are as efficient against OSCC as free cisplatin but with much less renal toxicity.\textsuperscript{121}

Li et al prepared NPs based on biocompatible and biodegradable hyperbranched poly(ether ester) polymers that possess many hydroxyl and carboxyl functional groups available for functionalization, including the covalent attachment of drug molecules.\textsuperscript{122} These hyperbranched poly(ether ester) NPs were attached to the photosensitizer chlorin(e6) (ce6), and they demonstrated an improvement in the in vitro photodynamic therapy activity over free ce6 in CAL 27 human oral cancer cells, which may be due to factors including increased cellular uptake of the photosensitizer and the disaggregating effect of covalently binding ce6 to a hydrophilic polymer that improve the quantum yield of the reactive oxygen species produced during photodynamic therapy. In addition, photosensitizer-loaded nanoparticles can reach the most sensitive subcellular sites, demonstrating a capability for treating superficial oral cancer or precancerous lesions.\textsuperscript{122}

Nevertheless, some studies revealed that some of the aforementioned polymers may lead to cytotoxicity after internalization into cells, restricting the use of NPs as a drug delivery system. In addition, the large-scale production of PNPs is also problematic and is not relevant for the pharmaceutical market.\textsuperscript{123–125}

Therefore, SLNs were developed to overcome the disadvantages of PNPs because they demonstrate physical stability, protection of incorporated labile drugs from degradation, controlled release, and excellent tolerability; thus, they can be used for different routes of administration, such as parenteral, oral, dermal, ocular, pulmonary, and rectal.\textsuperscript{126–129}

SLNs are made from solid lipids at room temperature and are stabilized by surfactant. SLNs can be obtained by a high-pressure homogenization (HPH) process that forms an average particle size of <500 nm and a low microparticle content, other production procedures that use organic solvents (HPH/solvent evaporation), or the dilution of microemulsions.\textsuperscript{130–132} The schematic structure of SLNs is shown in Figure 1.

Holpuuch et al tested a SLN formulation as a local oral cancer chemoprevention strategy.\textsuperscript{133} These authors demonstrated that SLNs composed of idarubicin hydrochloride (0.2 mg idarubicin/mL), sodium tetradecyl sulfate (0.159 mg/mL), emulsifying wax (2 mg/mL), polyoxyyl 20 stearyl ether ([Brij® 78; Uniqema, Wilmington, DE, USA] 2.3 mg/mL), and D-alpha-tocopheryl PEG 1,000 succinate ([vitamin-E TPGS] 3 mg/mL) underwent internalization by OSCC cells and could provide higher final intracellular levels relative to bolus administration. Furthermore, the penetration and subsequent internalization of nanoparticles within proliferating basal layer cells demonstrates the feasibility of nanoparticle formulations for local delivery and the stabilization of oral chemopreventive compounds.\textsuperscript{133}

However, SLNs have some limitations because the HPH process leads to drug degradation, the coexistence of different lipid modifications and colloidal species, and a low drug-loading capacity, and because of the kinetics of the distribution processes.\textsuperscript{134}

To overcome these difficulties, a new generation of SLNs has emerged, ie, NLCs, which consist of solid matrix entrapping variable liquid lipid nanocompartments, as shown in Figure 1. The presence of liquid lipid nanocompartments avoids solid lipid crystallization and improves the drug payload and release because these are still controlled by a surrounding solid lipid barrier.\textsuperscript{135–137}
Aditya et al made curcumin and genistein co-loaded NLCs based on oleic acid, lecithin, glycerol monostearate, and Tween® 80 (Meryer (Shanghai) Chemical Technology Co., Ltd, Shanghai, People’s Republic of China). These NLCs were found to be promising vehicles for the oral delivery of poorly bioaccessible molecules such as curcumin and genistein. In addition, NLCs had great effects against prostate cancer due to the enhanced intracellular uptake of NLCs by cells. Curcumin has also shown encouraging results in in vitro and in vivo models of OSCC. Therefore, future extensive research can determine the beneficial effects of curcumin-loaded NLCs for oral cancer treatment.

Chinsriwongkul et al researched NLCs based on a blend of cetyl palmitate and different liquid lipids, including soybean oil, medium-chain triglyceride, soybean oil/oleic acid (3:1) and medium-chain triglyceride/oleic acid (3:1), at a 1:1 weight ratio for the parenteral delivery of the anti-cancer drug all-trans retinoic acid (ATRA). NLCs based on oleic acid enhanced the ATRA loading capacity in the NLCs; however, all ATRA-loaded NLCs had prolonged release of ATRA in addition to being more cytotoxic than the free drug in an in vitro and in vivo models of OSCC. Therefore, future extensive research can determine the beneficial effects of curcumin-loaded NLCs for oral cancer treatment.

Liu et al designed DTX-loaded NLCs (DTX–NLCs) based on stearic acid, glyceryl monostearate, soya lecithin, and oleic acid prepared by the modified film ultrasonication–dispersion method. DTX was held in the lipid core of NLCs, which results in a prolonged release that could reduce the frequency of administration. Furthermore, DTX–NLCs had more cytotoxicity than free DTX, which is likely because DTX–NLCs carry drugs into cancer cells by endocytosis and enhance intracellular drug accumulation by nanoparticle uptake. These results are promising for cancer therapy, including that for oral cancer, because DTX provides an alternative for the management of OSCC.

Zhang et al aimed to develop three NLC formulations (NLC, PEG–NLC, distearoylphosphatidylethanolamine (DSPE)–PEG–NLC) for etoposide (VP16) and evaluate potential NLCs as an oral delivery system. The NLCs were based on VP16 (15 mg), monostearin (100 mg), soybean oil (30 mg), and soya lecithin (70 mg); PEG–NLCs were based on VP16 (15 mg), monostearin (100 mg), soybean oil (30 mg), soya lecithin (70 mg), and PEG-40 (140 mg); and DSPE–PEG–NLCs were based on VP16 (15 mg), monostearin (100 mg), soybean oil (30 mg), soya lecithin (70 mg), PEG-40 (140 mg), and DSPE–PEG (12 mg). All NLCs were prepared by an emulsification and low-temperature solidification method. A pharmacokinetic study conducted in rats revealed that the relative bioavailability of VP16–NLCs, VP16–PEG–NLCs, and VP16–DSPE–NLCs was enhanced approximately 1.8-, 3.0-, and 3.5-fold, respectively, compared with a VP16 suspension. Moreover, VP16–DSPE–NLCs showed the highest cytotoxicity against human epithelial-like lung carcinoma cells, which is likely due to NLC absorption at the cell surface and the release of VP16 close to the membrane, or NLC was internalized in cells and then released from the nanoparticles. These formulations may also provide an alternative for the treatment of oral cancer because VP16 also appears to have action in OSCC.

Liu et al designed querectin (QR)-loaded cationic NLCs that were based on a glycerol monostearate:medium-chain triglycerides ratio of 4:1, lecithin concentration of 3%,...
didodecyldimethylammonium bromide concentration of 1%, and QR concentration of 5%. Liu et al reported that the QR-loaded cationic NLCs released QR slower than QR in solution released QR in vitro, mainly due to the slow erosion or degradation of the lipid matrix, which could prolong the residence time of the drug at the tumor site, eg, an oral cancer tumor site.

Nanoparticles based on noble metals, particularly gold, have an immense potential for cancer diagnosis and therapy based on their surface-plasmon resonance-enhanced light scattering and absorption.

El-Sayed et al prepared gold nanoparticles (AuNPs) by the citrate reduction of chloroauric acid. This group used a simple and inexpensive conventional microscope with proper rearrangement of the illumination system and a light collection system to image cells incubated with AuNPs or anti-epidermal growth factor receptor (EGFR) antibody-loaded AuNPs. Both types of AuNPs were then incubated with a single nonmalignant epithelial cell line, HaCaT (human keratinocytes), and two malignant epithelial cell lines, HOC 313 clone 8 and HSC 3 (human OSCC cell lines). The results showed that the scattering images and absorption spectra recorded from anti-EGFR antibody-conjugated AuNPs incubated with cancerous and noncancerous cells were different and provided a potential technique for oral cancer diagnostics.

Kah et al also investigated AuNPs for the early diagnosis of oral cancer based on surface plasmon resonance. These authors prepared AuNPs via the reduction of 0.259 mM hydrogen tetrachloroaurate by 34 mM trisodium citrate (Sigma-Aldrich Co., St Louis, MO, USA) at a temperature of 90°C, and the AuNPs were conjugated to a monoclonal anti-EGFR antibody as a cancer biomarker for imaging via established protocols for the passive absorption of anti-EGFR on the surface of AuNPs. It was demonstrated that the use of EGFR-loaded AuNPs improved optical contrast under reflectance-mode imaging in vitro. Furthermore, the use of gold nanoparticles in surface-enhanced Raman scattering enhanced Raman spectroscopy signals for the analysis of cancer-related chemical changes in saliva.

Afifi et al used hamster buccal pouch carcinoma as a model for OSCC to study the effects of plasmonic photothermal therapy using AuNPs combined with visible laser irradiation. AuNPs were synthesized using the citrate reduction method. The results demonstrated an amplified decrease in proliferation rates for cancer cells upon plasmonic photothermal therapy using AuNPs in addition to maintaining no adverse effects on normal cells, which can be explained by the enhanced permeability and retention effect. These findings indicate that AuNPs directly injected into hamster buccal pouch carcinomas can be used as a treatment for human OSCC in the future.

Liposomes

Liposomes are unilamellar or multilamellar microscopic particles composed of membrane-like lipid layers, often phospholipids and cholesterol, surrounding aqueous compartments, as depicted in Figure 2. Liposomes are the most widely used drug delivery systems for the systemic administration of many drugs for decreasing drug toxicity and increasing their accumulation at target sites. Therefore, liposomes have been intensively studied for the delivery of chemotherapeutic drugs to improve therapeutic efficacy and decrease the toxicity to normal cells.

Furthermore, liposome-based formulations for gene therapy, such as synthetic cationic liposomal-DNA called

Figure 2 Schematic representation of unilamellar (A) and multilamellar (B) liposomes.

Note: The arrow indicates an enlarged view of the outer layers of multilamellar liposomes.
lipoplexes, have clear potential, particularly for oral cancer treatment. In this context, Konopka et al investigated the effects of high concentrations of fetal bovine serum on the transfection efficiency of a polycationic liposome (Metafectene™; Biontex Laboratories GmbH, München, Germany) and a polyamine reagent (GeneJammer; Agilent Technologies, Santa Clara, CA, USA) in HSC-3 and H357 human OSCC cells. The results showed that both polycationic liposomes could mediate gene delivery, which is not excessively inhibited even in the presence of 60% fetal bovine serum; therefore, they can be used in the delivery of genes in biological environments.

Figueirô Longo et al studied the effects of photodynamic therapy mediated by a liposomal formulation prepared by dimyristoyl phosphatidylcholine in the presence and absence of additives such as cholesterol or cardiolipin to release aluminum phthalocyanine chloride, a photosensitizer, in tongue tumors induced in Swiss mice. This treatment produced intense necrosis in the tumor tissue accompanied by the infiltration of polymorphonuclear cells and thrombi formation on tumor-associated blood vessels. Thus, these results showed that photodynamic therapy mediated by a liposomal formulation of aluminum phthalocyanine chloride can be effective against chemically induced oral cancer.

Velloso et al showed that liposomal aluminum phthalocyanine chloride-based photodynamic therapy inhibits the phosphatidylinositol 3-kinase/Akt/mammalian target of rapamycin (PI3K/Akt/mTOR) signaling pathway in cultured human OSCC cells. mTOR, a 289 kDa serine/threonine kinase located downstream of the PI3K/Akt pathway, has been shown to be a major regulator of cell growth, proliferation, migration, differentiation, and survival. In OSCC, activation of PI3K is a frequent event, and mTOR can be involved in the pathophysiology of oral cancer. Thus, these results are promising for oral cancer treatment.

Hydrogels

A hydrogel is a mesh of hydrophilic polymeric chains dispersed in water that is swellable and can release drugs for dissolution and disintegration through the spaces in their mesh, as shown in Figure 3. In addition to swelling, physical properties include permeability, mechanical resistance, and surface aspects that can be modulated through structural modification.

Hydrogels are attractive for oral administration because their polymeric chains can closely interact with saliva glycoproteins, causing a mucoadhesion phenomenon. There has been a great deal of interest in the use of hydrogels as chemotherapeutic drug delivery systems for drugs including paclitaxel, doxorubicin, DTX, tamoxifen, and cisplatin. Furthermore, studies have revealed that strategies are required to overcome the disadvantages of chemotherapeutic drugs such as cisplatin, which is usually intravenously administered, whereby 90% becomes linked to hemoproteins and 10% is free to enter into the cells.

In this context, a research group investigated the incorporation of cisplatin loaded-hydrogels called P9, CP2, MH2, and CMH2. A stock solution of cisplatin was added stepwise to each polymer solution. The acrylic hydrogels P9 and CP2, which contain a carboxyl group, were obtained by free radical polymerization of the monomers N-acryloyl-l-phenylalanine and N-isopropylacrylamide (NIPAAm), and they were cross-linked with N,N'-ethylene-bis-acrylamide.

![Figure 3](https://example.com/image.png) Hydrophilic polymeric chains network and release the drug for dissolution through the spaces of their mesh.
MH2 and CMH2 hydrogels were obtained by free radical polymerization of the methacrylate monomer N-methacryloyl-l-histidine and NIPAAm, and the authors assessed the in vitro cytotoxicity of cisplatin-loaded hydrogels. They reported that P9 hydrogels could modulate the rate of cisplatin release.\textsuperscript{172} P9 hydrogels have also been described as a promising platform for chemotherapeutic treatment, including that for oral cancers.\textsuperscript{173}

Moura et al investigated in vitro cisplatin release from chitosan hydrogels cross-linked with glycerol phosphate disodium salt and chitosan hydrogels that were ionic/covalently co-cross-linked.\textsuperscript{174} Their results demonstrated that the rate of release of cisplatin from ionic cross-linked chitosan hydrogels was significantly lower than that for chitosan hydrogels ionic/covalently co-cross-linked, and the amount of drug released was also quite different (60%–70% for hydrogels containing genipin against 20% for ionic hydrogels). Despite these differences, the release profiles were similar for both types of hydrogels, with an initial burst reaching a maximum concentration at approximately 2 to 3 hours. The researchers concluded that hydrogels containing both cross-linking agents can improve the chemical and mechanical properties presented when compared with hydrogels obtained with only one of the reticulating agents,\textsuperscript{175} making it attractive for the treatment of oral cancers because the release profile of the system occurs quickly, thus releasing the drug formulation before it is removed from the oral cavity by the salivary flow.\textsuperscript{175}

Emoto et al studied hydrogels obtained with cross-linkable hyaluronic acid for the intraperitoneal administration of cisplatin for extended retention and consequent action against peritoneal carcinomatosis.\textsuperscript{176} Hyaluronic acid was dissolved in water and sodium periodate was added and stirred for 2 hours. Afterward, ethylene glycol was added to stop the reaction, and the mixture was immediately dialyzed against water. The formation and swelling kinetics of hydrogels and the in vitro release kinetics of cisplatin from hydrogels were studied. The tests showed that there was sustained cisplatin release within 4 days. The researchers also evaluated the antitumor effects of the intraperitoneal administration of cisplatin-loaded acid hyaluronic hydrogels using a mouse model of gastric cancer. They observed a significant reduction in the weight of the peritoneal nodules in the gel-cisplatin group, whereas no significant reduction was detected in a phosphate-buffered saline-cisplatin group. It was concluded that this hydrogel is desirable for retention and modulates the release of cisplatin, thus increasing its antitumor effects.\textsuperscript{176}

Researchers have tested a system composed of a heat-sensitive copolymer formed by PEG-poly(ε-caprolactone)-PEG (PECE) for the incorporation of suberoylanilide hydroxamic acid (SAHA) with cisplatin and subsequently evaluated the in vitro release profile of these drugs against oral carcinoma.\textsuperscript{175} For the formation of hydrogels, the PECE copolymer was first completely dissolved in water and cooled to 4°C to form a colloidal solution. SAHA and cisplatin solutions were then mixed into the PECE colloidal solution to form a homogeneous solution, and the PECE concentration was maintained at 30% (w/w). The authors concluded that the SAHA-cisplatin/PECE hydrogel system with direct intratumoral injections may be a useful method for the treatment of oral cancer and other solid tumors.\textsuperscript{177}

Liquid crystals

LCs are materials in a differential state, demonstrating a property between a solid and a liquid. This state is called mesophase: the prefix “meso-” means “intermediate”.\textsuperscript{178} LCs are divided into two categories: thermotropics, which are structured by means of temperature, and lyotropics, which occur by association with amphipathic compounds and solvents. The mesophase lyotropics are mostly lamellar, hexagonal, or cubic,\textsuperscript{179} as shown in Figure 4.

LCs are usually based on water as a solvent, surfactant (may contain cosurfactants), and an oily phase. One of the advantages of LCs is that they can be stored for long periods because they are thermodynamically stable.\textsuperscript{180}

Polarized light microscopy is one of the characterization techniques that is used for the preliminary identification of mesophase LCs.\textsuperscript{181} In this analysis, a sample undergoes the incidence of polarized light, which is enough to deflect light and is called anisotropic (can be mesophase lamellar or hexagonal). However, if the latter does not bend light, it is isotropic (cubic arrangements); therefore, other techniques are needed for confirming this structure,\textsuperscript{179} including small-angle X-ray scattering, small-angle neutron scattering, neutron diffraction, nuclear magnetic resonance, and cryo-fracture electron microscopy.\textsuperscript{179,182,183}

The LC systems significantly change the drug release profile and reduce the toxicity of drugs, improving clinical efficiency.\textsuperscript{178} Hosmer et al in 2012, studied mesophase lamellar LCs formed with glycerides for the incorporation of the anticancer drug paclitaxel.\textsuperscript{184} Paclitaxel is highly effective against various types of cancer, including oral cancer.\textsuperscript{185–187} However, it has severe adverse effects associated with systemic drug administration, including hypersensitivity reactions, thrombocytopenia, and neutropenia.\textsuperscript{188,189} Hosmer et al\textsuperscript{184} found that, among the formulations studied, the Brij-based lamellar phase containing 20% medium-chain
mono-/diglycerides maximized the delivery of paclitaxel and showed good efficacy against paclitaxel-sensitive fibroblasts; therefore, LCs may be a promising strategy for the treatment of cancers, including oral cancer.

Zeng et al developed liquid crystalline nanoparticles consisting of soy phosphatidyl choline and glycerol dioleate for the incorporation of paclitaxel using a solvent precursor method described by Rizwan et al in which a 50:50 (w/w) mixture of soy phosphatidyl choline and glycerol dioleate was agitated for 3 hours to form a uniform oily phase. 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid (HEPES) was then carefully added followed by stirring at room temperature for 24 hours. The precursor of LC was added under magnetic stirring at 60°C, forming a coarse dispersion. This dispersion was subsequently homogenized using a Microfluidizer® (Microfluidics, Newton, MA, USA) at a pressure of 10,000 psi for three cycles and 30,000 psi for two cycles. The systems were characterized by polarized light microscopy, indicating the coexistence of reversed cubic and hexagonal phases in the optimized LC matrix. Transmission electron microscopy and cryo-field emission scanning electron microscopy revealed the internal water channel and “twig-like” surface morphology of the LC matrix. Tests were performed, including pharmacokinetics in vivo, and particle size distribution, phase behavior characterization, transmission electron microscopy, and cryo-field emission scanning electron microscopy in vitro. Zeng et al concluded that these systems demonstrated potential as nanocarriers for water-insoluble drugs such as paclitaxel, improving intravenous bioavailability.

Complexes of cyclodextrins

Cyclodextrins are cyclical and composed of at least six units of glucose, resulting in a truncated cone form with a hollow cavity, as shown in Figure 5.

Although natural cyclodextrins (the best known cyclodextrins include α-, β-, and γ-cyclodextrin) are of interest for the development of pharmaceutical formulations by presenting excellent biocompatibility, the ability to mask undesirable organoleptic properties of drugs, and the ability to increase solubility and permeability, these compounds demonstrate limitations for the transport of drugs, enabling the loading of only lipophilic drugs by virtue of the cyclodextrin hydrophilic exterior and interior hydrophobic cavity. Soon, natural cyclodextrins may be produced with chemical modifications in accordance with the interest in this field, enabling the attainment of cyclodextrins with both lipophilic and conjugated polar groups, making them amphiphilic.

Cyclodextrins represent a group of excipients with great potential for use in pharmaceutical formulations. Once the bioavailability and multifunctional features of cyclodextrins are able to reduce the undesirable properties of drugs that are included in complexes, application via several routes of administration will be enabled. It is also important to highlight the ability to include drugs with solid or liquid characteristics.

Ramineni et al studied cyclodextrins for the inclusion of imiquimod to treat precancerous dysplastic lesions of the oral cavity. This group developed a mucoadhesive film for the conveyance of a highly hydrophobic drug. To prepare these films, a polymer aequous solution of polyvinylpyrrolidone mixed with ethanol following the addition of propylene...
glycol as a plasticizer was used. At the same time, a solution of carboxymethylcellulose was prepared and added to polymer aqueous solution at high speed. The polymer mixture was dried, forming films, and the antineoplastic agent was incorporated by four different methods: sonication, solubilization in linoleic acid, complex formation by co-evaporation, and solubilization in methanol and acetate buffer. The films were translucent and flexible, with the exception of those prepared with linoleic acid; therefore, the films are a promising platform for the delivery of drugs with mucoadhesion properties, which are able to be administered at the desired location in addition to sustaining the delivery of imiquimod.200

Researchers have also developed a new type of hollow complex based on the combination of cyclodextrin with hyaluronic acid, which can be added to paclitaxel.201 The system was innovatively performed and was associated with the pursuit of controlled drug release by complexation with cyclodextrins that are recognized by cancer cells and sensitized to enzymatic hydrolysis caused by the natural biological properties of hyaluronic acid. Under physiological conditions, paclitaxel was released slowly, demonstrating that the cyclodextrins were stable.201

A group of researchers202 developed α-cyclodextrins with pH-sensitive nanoparticles. To reach the specific location of a tumor that is around pH 5.7–7.8, this group prepared a system for drug targeting. These authors incorporated paclitaxel and analyzed the activity of various tumor cells in addition to conducting tests in vivo in mice with melanoma. These mice were given a single intravenous dose of paclitaxel (10 mg/kg) and were compared with a negative control that was administered saline solution, and a reduction in tumor cells was observed. Thus, it is noted that formulated nanomedicines can effectively reverse the multidrug resistance of cancer cells resistant to paclitaxel. In summary, pH-sensitive α-cyclodextrin materials can be conveniently produced by a facile acetonation process, which may be further processed into NPs with controllable size and size distribution. The results demonstrated that the systems have biocompatibility and lead to a reduction in adverse effects and improved antitumor activity.202

New approaches and challenges
The ultimate goal of cancer treatment is to kill as many cancer cells as possible without affecting healthy cells. However, the ability of a drug to target specific sites in the body to achieve defined therapeutic effects needs improvement. In this context, nanodelivery systems emerge as a potential alternative for overcoming some previously encountered obstacles to efficiently target several cancer cell types because they have shown several promising characteristics, including optimal anti-oral tumor effects, which are not available with traditional chemotherapy.

Thus, the US Food and Drug Administration (FDA) recently approved a clinical trial of a nanoparticle-based system to use in humans for treatment of solid tumors.203 Furthermore, Yang et al in 2003, evaluated targeted delivery to cervical lymph nodes by perioral cancer submucosal injection of cucurbitacin BE polylactic acid nanoparticles (CuBE-PLA-NPs) and their clinical therapy efficacy. The results showed that the drug concentrations in cervical lymph nodes after CuBE-PLA-NP injection were far higher than those in the control group. Furthermore, the drug concentrations in the blood in the CuBE-PLA-NP group were far lower than those in the control group.204

Hence, in the near future, oncologists and patients will benefit from suitable nanotechnology-based drug delivery systems that could lead to improved therapeutic outcomes with reduced costs. There are few clinical studies on oral cancer in the field of nanotechnology, but nanotechnology is also predicted to alter health care in dentistry, with novel methods of identifying the cancer as well as customization of a patient’s therapeutic profile.205

Further studies are needed to turn concepts of nanotechnology into practical applications and to elucidate correct drug doses and ideal release from these systems for the treatment of several cancers with different molecular and cellular mechanisms.

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