Rapid efficient synthesis and characterization of silver, gold, and bimetallic nanoparticles from the medicinal plant *Plumbago zeylanica* and their application in biofilm control

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**Background:** Nanoparticles (NPs) have gained significance in medical fields due to their high surface-area-to-volume ratio. In this study, we synthesized NPs from a medicinally important plant – *Plumbago zeylanica*.

**Materials and methods:** Aqueous root extract of *P. zeylanica* (PZRE) was analyzed for the presence of flavonoids, sugars, and organic acids using high-performance thin-layer chromatography (HPTLC), gas chromatography-time of flight-mass spectrometry (GC-TOF-MS), and biochemical methods. The silver NPs (AgNPs), gold NPs (AuNPs), and bimetallic NPs (AgAuNPs) were synthesized from root extract and characterized using ultraviolet-visible spectra, X-ray diffraction (XRD), energy-dispersive spectrometry (EDS), transmission electron microscopy (TEM), and dynamic light scattering (DLS). The effects of these NPs on Acinetobacter baumannii, Staphylococcus aureus, and Escherichia coli biofilms were studied using quantitative biofilm inhibition and disruption assays, as well as using fluorescence, scanning electron microscopy, and atomic force microscopy.

**Results:** PZRE showed the presence of phenolics, such as plumbagin, and flavonoids, in addition to citric acid, sucrose, glucose, fructose, and starch, using HPTLC, GC-TOF-MS, and quantitative analysis. Bioreduction of silver nitrate (AgNO₃) and chloroauric acid (HAuCl₄) were confirmed at absorbances of 440 nm (AgNPs), 570 nm (AuNPs), and 540 nm (AgAuNPs), respectively. The maximum rate of synthesis at 50°C was achieved with 5 mM AgNO₃ within 4.5 hours for AgNPs, and with 0.7 mM HAuCl₄ within 5 hours for AuNPs. The synthesis of AgAuNPs, which completed within 90 minutes with 0.7 mM AgNO₃ and HAuCl₄, was found to be the fastest. Fourier-transform infrared spectroscopy confirmed bioreduction, while EDS and XRD patterns confirmed purity and the crystalline nature of the NPs, respectively. TEM micrographs and DLS showed about 60 nm monodispersed Ag nanospheres, 20–30 nm Au nanospheres adhering to form Au nanotriangles, and about 90 nm hexagonal blunt-ended AgAuNPs. These NPs also showed antimicrobial and antibiofilm activity against *E. coli*, *A. baumannii*, *S. aureus*, and a mixed culture of *A. baumannii* and *S. aureus*. AgNPs inhibited biofilm in the range of 96%–99% and AgAuNPs from 93% to 98% in single-culture biofilms. AuNPs also showed biofilm inhibition, with the highest of 98% in *S. aureus*. AgNPs also showed good biofilm disruption, with the highest of 88% in *A. baumannii*.

**Conclusion:** This is the first report on rapid and efficient synthesis of AgNPs, AuNPs and AgAuNPs from *P. zeylanica* and their effect on quantitative inhibition and disruption of bacterial biofilms.

**Keywords:** *P. zeylanica*, AgNPs, AuNPs, AgAuNPs, biofilm inhibition and disruption, GC-TOF-MS
Introduction

Nanomaterial synthesis and characterization have been an extensive area of research since the last decade. Biological synthesis of nanoparticles (NPs) is considered environmentally friendly, nontoxic, and cost-effective. Therefore, biological synthesis of NPs from various plant extracts and microorganisms like bacteria, yeasts, fungi, and algae have proven to be promising sources. However, plant extracts are preferred over microbial sources, as they do not require sterile conditions. Also, bioreduction of NPs using plant extracts is rapid, ecofriendly, biocompatible, and nontoxic.

Silver has been exploited as antimicrobials from ancient period. With the evolution of nanomedicine as a study for treating infections, metallic silver in the form of NPs has regained its significance. Several bacteria have developed resistance against antibiotics, which has challenged the treatment of human infections. Therefore, silver NPs (AgNPs) as an antimicrobial agent seem to be beneficial compared to antibiotics. Moreover, few reports show synergistic enhancement of activity of antibiotics with AgNPs. There are a number of reports on green synthesis of gold NPs (AuNPs) by plants and microorganisms, and they are being exploited for research on drug delivery and cancer studies. However, as gold is not known to have an antimicrobial property, very few reports are available on the antimicrobial nature of AuNPs. It is of interest to some researchers whether AuNPs can inhibit biofilm formation and kill bacteria as they show surface bioconjugation. So far, syntheses of bimetallic NPs (AgAuNPs) from Diospyros kaki and Azadirachta indica extracts have been reported by Song and Kim, respectively, with an Au core and Ag shell observed in case of A. indica. However, research on their application as antimicrobial agents is still in its infancy.

Plumbago zeylanica, commonly known as agni (meaning fire) in Sanskrit, is an important medicinal plant used in ayurveda, an Indian system of traditional medicine. Ahmad and Beg reported on the antimicrobial activity of aqueous and alcoholic extracts from roots of P. zeylanica against Staphylococcus aureus, Pseudomonas aeruginosa, Proteus vulgaris, Bacillus subtilis, and Candida albicans. Plumbagin (5-hydroxy-2-methyl-1,4-naphthoquinone) is a major flavonoid present in P. zeylanica, especially in roots, showing antifungal and antimicrobial properties. Plumbagin is also well known for its plasmid-curing property. Plumbagin has cured plasmid-encoding resistance to silver, cadmium, antimony, streptomycin, and ampicillin in Acinetobacter baumannii species.

The phytochemical analysis of an aqueous root extract of P. zeylanica (PZRE) showed the presence of alkaloids, glycoside, reducing sugars, phenolics, tannins, lignin, saponins, and flavonoids. The majority of these compounds are known to act as reducing as well as stabilizing agents in the bioreduction of silver and gold salt. Moreover, there have been no reports on the synthesis of metal NPs by P. zeylanica to date. Therefore, we propose that P. zeylanica may synthesize novel metal NPs exhibiting antimicrobial and biofilm inhibition/disruption properties.

Materials and methods

Plant material and preparation of extract

P. zeylanica roots were collected from the natural landscape of Western Ghats, on Dr Homi Bhabha Road, Pune, Maharashtra, India. They were thoroughly washed with distilled water, shade-dried, chopped into thin slices, and ground to make fine powder. The aqueous extract was prepared by boiling 5 g of the powder in 100 mL sterile distilled water, followed by centrifugation at 8,000 rpm for 10 minutes. The supernatant of PZRE was filtered through a 0.45 μm membrane filter and stored at 4°C until further use.

Preparation of the PZRE samples for chromatographic analysis

PZRE was sequentially extracted along a polarity gradient using ethyl acetate and methanol for high-performance thin-layer chromatography (HPTLC; semiautomated Linomat 5, Camag, Muttenz, Switzerland) fingerprinting and gas chromatography-time of flight-mass spectrometry (GC-TOF-MS). The sequential extraction was done by suspending 5 mL of root extract in 5 mL of high-performance liquid chromatography-grade ethyl acetate, followed by stirring for 3 hours and centrifugation at 10,000 rpm at room temperature (RT). The supernatant was recovered, and 5 mL methanol was added to the pellet and redispersed. The solution was vigorously mixed by vortexing for 15 minutes, and then methanol extract was recovered. The ethyl acetate and methanol extracts were concentrated.

Phytochemical analysis of P. zeylanica

HPTLC analysis of PZRE was performed according to the protocol given in earlier studies. Plumbagin, quercetin, diosgenin, glucose, and fructose (Sigma-Aldrich, St Louis, MO, USA) were taken as standard for the analyses. Similarly, 5 μL of the ethyl acetate extract of PZRE was subjected to GC-TOF-MS (Pegasus 4D GCxGC-TOFMS system with Agilent 6890 gas chromatograph equipped with
a dual-jet thermal modulator between primary and secondary columns and a Pegasus IV TOFMS as a detector; LECO, St Joseph, MI, USA). The program used was as follows: primary column, 5% phenyl polysilphenylene-siloxane HP-5MS capillary, 30 m ×0.32 mm, 0.25 μm; secondary column, 1.00 m ×0.10 mm ID ×0.10 μm of Rxi®-17 ms (Restek, Bellefonte, PA, USA) in GC oven (primary oven program, 100°C for 0.5 minute, followed by an increase up to 215°C at an increment of 20°C/minute, held for 0.5 minute, further to 270°C at 25°C/minute and held for 10 min; secondary oven program, rate and duration were identical to primary oven, target temperature was set at 30°C above primary oven). Sample injection: Gerstel PTV (Gerstel, Mülheim an der Ruhr, Germany) using solvent vent mode; modulator temperature +20°C. Carrier gas: helium at ramped pressure mode; transfer line 240°C. MS parameters: electron impact ionization 70 eV; ion source temperature 250°C; detector voltage 1,700 V; data-acquisition mass range m/z 50–500 at an acquisition rate of ten spectra; software ChromaTOF® 3.34 (LECO). Identification of components was based on comparison of their mass spectra with those of National Institute of Standards and Technology library spectra (v. 2.0).\textsuperscript{35} Citric acid, plumbagin, starch, and total reducing sugar content of PZRE were determined according to the methodology mentioned in the following sections.

**Total reducing sugar**
To 50 μL of PZRE, we added the same volume of 80% phenol (volume/volume) and vortexed. To this mixture, 2 mL of sulfuric acid was added and incubated for 10 minutes at RT, followed by recording absorbance at 490 nm. Total reducing sugar content was estimated using a standard glucose curve.\textsuperscript{36}

**Total phenolic content**
PZRE (0.125 mL) was added to 0.5 mL of deionized water. To this mixture, 0.125 mL Folin–Ciocalteu reagent was added and incubated for 5 minutes at RT. After incubation, 1.25 mL of 7% Na₂CO₃ solution was added, and the volume was made up to 3 mL with distilled water. This reaction mixture was then incubated for 90 minutes at RT, and absorbance was recorded at 760 nm. Total phenolic content was estimated using a standard gallic acid curve.\textsuperscript{37}

**Total flavonoid content**
Half a milliliter of 2% AlCl₃ in methanol was mixed with a 0.5 mL sample of PZRE and incubated for 10 minutes at RT. After 10 minutes, absorbance was recorded at 368 nm. The total flavonoid content was determined using a standard quercetin curve.\textsuperscript{37}

**Citric acid content**
One milliliter of PZRE was evaporated in order to dry it completely, and reconstituted in 1 mL 5% trichloroacetic acid. Dropwise addition of 8 mL of anhydrous acetic anhydride was done, followed by incubation at 60°C for 10 minutes. The tubes were then stoppered. After 10 minutes, 1 mL of pyridine was added to each tube and restopped. The tube was then incubated at 60°C for 40 minutes, followed by incubation in an ice-water bath for 5 minutes. Absorbance was recorded at 420 nm. The total citric acid content was determined from a standard citric acid curve.\textsuperscript{37}

**Starch content**
P. zeylanica root powder was weighed, and 5 g of the powder was washed with 70% ethanol. The washes were repeated to ensure the removal of sugars until it did not develop color when treated with anthrone reagent. The residue was dried and boiled in 100 mL distilled water. Then 1 mL of PZRE was evaporated completely and reconstituted in 60% perchloric acid. To this, 4 mL of anthrone reagent was added and was boiled in a water bath for 8 minutes. The intensity of the dark green color was recorded at 630 nm, and the starch content was estimated by comparing the absorbance with the standard glucose curve.\textsuperscript{37}

**Plumbagin content**
Plumbagin was extracted using acetone, chloroform, and heptane according to the method described by Israni et al.\textsuperscript{38} In a mortar and pestle, 5 g root powder was macerated using acetone (3×50 mL). The acetone extract was then evaporated to dryness at RT and 25 mL chloroform was added, followed by an equal volume of distilled water (3×) to remove watersoluble impurities. The chloroform extract was evaporated till a dark brown oily residue was obtained. This residue was treated with 10 mL of phosphoric acid (10%) for 30 minutes followed by extraction with 5 mL n-heptane. To 0.2 mL of this yellow solution, 10% alcoholic KOH was added, the volume was made up to 5 mL using absolute ethanol, and absorbance was recorded at 520 nm. The concentration of plumbagin was determined using a standard plumbagin curve.

**Synthesis and characterization of nanoparticles**
**UV-visible spectroscopy and FTIR analysis**
Synthesis and optimization studies of AgNPs and AuNPs were carried out according to the methods given earlier.\textsuperscript{3,37} AgAuNPs were prepared using 1 mM silver
nitrate (AgNO₃) and 1 mM chloroaucric acid (HAuCl₄) salt solutions in a 1:1 ratio. To 5 mL PZRE, 47.5 mL of HAuCl₄ solution was added, followed by addition of 47.5 mL AgNO₃. Synthesis of NPs was monitored by recording the spectra on an ultraviolet (UV)-visible spectrophotometer (SpectraMax M5; Molecular Devices, Sunnyvale, CA, USA) operating at a resolution of 1 nm. Optimization studies for rate of NP synthesis were carried out at different temperatures ranging from 4°C to 50°C, and at varying concentrations from 0.1 mM to 5 mM. For AgAuNPs, the salt mixture of AgNO₃ and HAuCl₄ was mixed in the ratio of 1:1, with concentrations differing from 0.1 mM to 5 mM. AgNPs, AuNPs, and AgAuNPs were also chemically synthesized and characterized according to the protocol reported earlier. Dried PZRE before and after bioreduction was subjected to Fourier-transform infrared spectroscopy (FTIR; Affinity-1; Shimadzu, Tokyo, Japan) using the potassium bromide pellet technique, and was exposed to an IR source of 500–4,000 cm⁻¹. For application purposes, the NPs were synthesized, followed by washing twice with double-distilled water in order to use for antimicrobial and biofilm assays.

EDS and XRD analysis

NPs were studied using energy-dispersive spectrometry (EDS; JSM 6360A analytical scanning electron microscope [SEM]; JEOL, Tokyo, Japan) at an energy range 0–20 keV for elemental confirmation. Phase formation of the bioreduced NPs was studied using X-ray diffraction (XRD; D8 Advanced; Bruker Optik, Ettlingen, Germany) with a CuKα (1.54 Å) source.

DLS measurement and TEM analysis

Freshly prepared 3 mL AgNPs, AuNPs, and AgAuNPs were taken in polystyrene cuvettes and subjected to particle-size analysis employing dynamic light scattering (DLS; Zetasizer Nano-2590; Malvern Instruments, Malvern, UK). Zeta-potential values of each sample were also recorded on the same instrument. The surface morphology and size of the NPs were determined using transmission electron microscopy (TEM; Tecnai™ G2 20 U-Twin; FEI, Hillsboro, OR, USA).

Antimicrobial assay

The effects of NPs synthesized using PZRE were tested against *A. baumannii* AIIMS 7, *Escherichia coli* NCIM (National Collection of Industrial Microorganisms) 2931 and *S. aureus* MTCC (Microbial Type Culture Collection) 3160 on Mueller–Hinton (MH) agar plates using a disk-diffusion assay. Different concentrations of freshly prepared AgNPs, AuNPs, and AgAuNPs were prepared from 2,048 μg to 0.25 μg/disk by the double-dilution method. After incubation at 37°C for 18 hours, the zones of inhibition were measured. Chemically synthesized NPs were used as controls. All assays were performed in triplicate.

Determination of biofilm inhibition and disruption

Biofilm-inhibition assays were performed by coinoculation of 5 μL of culture (optical density adjusted to 0.05 at 600 nm) with AgNPs, AuNPs, and AgAuNPs (1,024 μg/well) in 200 μL MH broth for 24 hours at 37°C. In the case of biofilm-disruption assays, bacterial biofilm was allowed to form for 24 hours, followed by washing with sterile phosphate-buffered saline (PBS; NaCl 8 g/L, KCl 0.2 g/L, Na₂HPO₄ 1.44 g/L, KH₂PO₄ 0.24 g/L). Biofilm disruption was initiated by the addition of 1,024 μg/well of freshly synthesized NPs. Additionally, the effect of NPs was checked against a mixed-biofilm formation by *A. baumannii* and *S. aureus*. Biofilm formation in the absence of NPs served as a positive control. The extent of biofilm inhibition and disruption was determined using the crystal violet staining method as reported earlier. Chemically synthesized NPs were used as controls. Biofilm inhibition and disruption assays were carried out in triplicate.

Visualization of biofilm employing fluorescence microscopy, SEM, and AFM

The biofilm inhibition and disruption assays were carried out on grease-free, sterile glass slides with tenfold-higher volumes to make the final volume 2 mL in 12-well plates and kept at 37°C for incubation. The concentration of the NPs was 1,024 μg/well. After 24 hours, the glass slides were washed with sterile PBS. The glass slides were stained with acridine orange and observed under a Zeiss fluorescence microscope using 465–495 nm excitation filters (AxioScope A1; Carl Zeiss Meditec, Jena, Germany). Glass slides fixed with glutaraldehyde were dehydrated sequentially using ethyl alcohol and analyzed using SEM (JSM 6360A; JEOL). Slides washed with PBS and air-dried were used for image analysis at a resolution of 2×2 μm under semi-contact atomic force microscopy (AFM; NTEGRA, NT-MDT, Moscow, Russia).

Results

Phytochemical analysis of *P. zeylanica*

HPTLC fingerprints for PZRE showed the bands corresponding to the standard plumbagin and diosgenin in the methanolic and ethyl acetate fractions of PZRE.
Characterization of nanoparticles

Visual observation, UV-visible spectroscopy, and FTIR analysis

AgNPs, AuNPs, and AgAuNPs were bioreduced by PZRE. The yellowish-brown color of AgNPs appeared due to excitation of surface plasmon vibrations with absorbance maxima at 440 nm and a violet color of the AuNPs at 570 nm. A peak appeared at t=1 hour, with a steady increase till t=4.5 hours in AgNPs and t=5 hours in AuNPs (Figure 1A and B). AgAuNPs developed a dark-violet brownish color with absorbance maxima shifting towards 540 nm. A small peak appeared at t=0 minutes with the rate of synthesis increasing steadily till 90 minutes (Figure 1C).

The effects of temperature on the rate of AgNPs and AgAuNPs synthesis showed a gradual increase in synthesis up to 50°C (Figure 2A and B). The study of the effect of salt concentration revealed an increase in the rate of AgNPs synthesis with an increase in concentration and maximum synthesis at 5 mM AgNO₃ (Figure 2C). The rate of synthesis was highest in 0.7 mM and decreased further from 1 mM to 5 mM in AgAuNPs (Figure 2D). Similar results were observed in the case of AuNPs (Figure S2).

Table 1: High-performance thin-layer chromatography of the root extract of Plumbago zeylanica

<table>
<thead>
<tr>
<th>Compounds</th>
<th>Type of extract</th>
<th>Amount (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Plumbagin</td>
<td>Hydromethanolic</td>
<td>0.002</td>
</tr>
<tr>
<td></td>
<td>Ethyl acetate</td>
<td>1.6</td>
</tr>
<tr>
<td>Diosgenin</td>
<td>Hydromethanolic</td>
<td>–</td>
</tr>
<tr>
<td></td>
<td>Ethyl acetate</td>
<td>2.8</td>
</tr>
<tr>
<td>Quercetin</td>
<td>Hydromethanolic</td>
<td>0.0005</td>
</tr>
<tr>
<td></td>
<td>Methanolic</td>
<td>0.21</td>
</tr>
<tr>
<td>Glucose</td>
<td>Hydromethanolic</td>
<td>14.3</td>
</tr>
<tr>
<td>Fructose</td>
<td>Hydromethanolic</td>
<td>0.5</td>
</tr>
</tbody>
</table>

Note: All experiments were performed in triplicate, and standard deviations were negligible.

Abbreviation: –, not detected.

(Table 1 and Figure S1). Similarly, GC-TOF-MS analysis was done. The compounds detected in GC-TOF-MS analysis of ethyl acetate extract of PZRE are enlisted in Table 2. Sucrose was observed to be a major component of the PZRE, at 22%. Plumbagin content was found to be 2% in the plant extract. Total reducing sugars, phenolic content, flavonoid content, starch, citric acid and plumbagin content are quantified and given in Table 3.

Table 2: Gas chromatography-time of flight-mass spectrometry analysis of Plumbago zeylanica root extract (PZRE)

<table>
<thead>
<tr>
<th>Name</th>
<th>Formula</th>
<th>Quantity</th>
<th>Area %</th>
<th>RT (s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sucrose</td>
<td>C₆H₁₂O₁₁</td>
<td>57</td>
<td>21.912</td>
<td>558.3</td>
</tr>
<tr>
<td>2-Furancarboxaldehyde, 5-(hydroxymethyl)-</td>
<td>C₅H₆O₂</td>
<td>126</td>
<td>19.378</td>
<td>444.9</td>
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<tr>
<td>D-Allose</td>
<td>C₈H₁₂O₄</td>
<td>60</td>
<td>6.2624</td>
<td>587.4</td>
</tr>
<tr>
<td>Triethylene glycol</td>
<td>C₂H₆O₄</td>
<td>45</td>
<td>5.0681</td>
<td>440.1</td>
</tr>
<tr>
<td>Spiro[2,3]hexan-5-one, 4,4-diethyl-</td>
<td>C₁₀H₁₆O</td>
<td>110</td>
<td>3.9047</td>
<td>446.0</td>
</tr>
<tr>
<td>Ethanol, 2-(2-aminoethoxy)-</td>
<td>C₇H₁₄N₂</td>
<td>45</td>
<td>3.8949</td>
<td>559.7</td>
</tr>
<tr>
<td>4-Chromanol</td>
<td>C₉H₄O₂</td>
<td>121</td>
<td>2.9764</td>
<td>690.5</td>
</tr>
<tr>
<td>Ethanol, 2,2′-[oxybis(2,1-ethanediyloxy)]bis-</td>
<td>C₃H₈O₅</td>
<td>45</td>
<td>2.8263</td>
<td>568.9</td>
</tr>
<tr>
<td>Benzothiazole, 2-(2-hydroxyethylthio)-</td>
<td>C₅H₈N₂O₅</td>
<td>167</td>
<td>2.8117</td>
<td>832.1</td>
</tr>
<tr>
<td>N-Butyl-tert-butylamine</td>
<td>C₇H₁₄N</td>
<td>114</td>
<td>2.7494</td>
<td>432.3</td>
</tr>
<tr>
<td>Plumbagin</td>
<td>C₃H₆O₃</td>
<td>188</td>
<td>2.4452</td>
<td>639.9</td>
</tr>
<tr>
<td>Pentane, 2,3,3,4-tetramethyl-</td>
<td>C₅H₁₀</td>
<td>85</td>
<td>2.0557</td>
<td>430.0</td>
</tr>
<tr>
<td>Phenol, 2,4-bis(1,1-dimethyl)-</td>
<td>C₅H₈O</td>
<td>191</td>
<td>1.214</td>
<td>569.4</td>
</tr>
<tr>
<td>Acetaldehyde, tetramer</td>
<td>C₅H₈O₄</td>
<td>45</td>
<td>1.2032</td>
<td>877.2</td>
</tr>
<tr>
<td>Ethanol, 2-[2-(2-ethoxyethoxy)ethoxy]-</td>
<td>C₅H₈O₄</td>
<td>45</td>
<td>1.1105</td>
<td>456.6</td>
</tr>
<tr>
<td>Benzoic acid</td>
<td>C₅H₄O₂</td>
<td>122</td>
<td>1.003</td>
<td>417.1</td>
</tr>
<tr>
<td>Ethanol, 2,2′-[oxybis(2,1-ethanediyloxy)]bis-</td>
<td>C₃H₈O₅</td>
<td>45</td>
<td>0.84417</td>
<td>750.8</td>
</tr>
<tr>
<td>1,4-Benzenediol, 2-methoxy-</td>
<td>C₅H₈O₂</td>
<td>125</td>
<td>0.78487</td>
<td>530.8</td>
</tr>
<tr>
<td>Semioxamidine</td>
<td>C₇H₈N₂O₄</td>
<td>103</td>
<td>0.73226</td>
<td>454.8</td>
</tr>
<tr>
<td>n-Hexadecanoic acid</td>
<td>C₁₄H₂₇O₂</td>
<td>60</td>
<td>0.72788</td>
<td>814.7</td>
</tr>
<tr>
<td>2-Octanamine</td>
<td>C₈H₁₇N</td>
<td>55</td>
<td>0.72463</td>
<td>438.4</td>
</tr>
<tr>
<td>Tridecanoic acid, methyl ester</td>
<td>C₁₃H₂₇O₂</td>
<td>74</td>
<td>0.71858</td>
<td>792.4</td>
</tr>
<tr>
<td>Benzoic acid, 4-ethoxy-, ethyl ester</td>
<td>C₁₃H₂₇O₂</td>
<td>121</td>
<td>0.65169</td>
<td>576.1</td>
</tr>
<tr>
<td>Ethane, 1,1′-oxybis, 2-methoxy-</td>
<td>C₁₀H₁₆O₂</td>
<td>45</td>
<td>0.59053</td>
<td>428.9</td>
</tr>
<tr>
<td>Ethane, iodo-</td>
<td>C₁₁H₁₀I</td>
<td>156</td>
<td>0.58063</td>
<td>640.3</td>
</tr>
<tr>
<td>N3-Diethyl-3-heptanamine</td>
<td>C₁₇H₁₄N</td>
<td>142</td>
<td>0.5475</td>
<td>633.5</td>
</tr>
</tbody>
</table>

Abbreviations: RT, reaction time; s, seconds.
The plant extracts before and after bioreduction were subjected to FTIR analysis. The sample of plant extract before bioreduction showed a peak at 3,300 cm$^{-1}$, signifying the presence of hydroxyl groups (Figure 3). In samples after reduction, a peak at 3,300 cm$^{-1}$ was not observed and the free hydroxyls were observed at 3,390 cm$^{-1}$, indicating that hydroxyl groups present in polyphenols from *P. zeylanica* were responsible for the bioreduction of the AgNO$_3$ and HAuCl$_4$ solutions to form respective NPs. A significant peak at 1,741 cm$^{-1}$ of the symmetric stretch of C=O indicated the presence of aldehydes, ketones, esters, or carboxylic acids.

EDS, XRD, and DLS analysis

In EDS analysis, the characteristic peak at 3 keV confirmed the presence of AgNPs, whereas particle-size analysis employing DLS showed that the bioreduced AgNPs had an average size of 63.92 nm (Figure 4A and B). Stability of the particles was confirmed from the zeta-potential value, which was found to be $-31$, whereas phase formation was observed using XRD patterns. Phase formation was confirmed from characteristic peaks, such as (111), (200), (220), and (311). The data matched with the standard Joint Committee for Powder Diffraction Set, card 040783, confirming a face-centered cubic structure for the AgNPs (Figure 4C and D).

EDS of AgAuNPs showed that peaks were obtained at around 2 and 3 keV, confirming the presence of silver and gold (Figure 5A). The silver content was around 66%, and that of gold was 14%. The average size of the AgAuNPs was recorded to be 93.04 nm, with a zeta-potential value of $-21$ (Figure 5B and C). In XRD analysis, the position of the

### Table 3 Phytochemical analysis of *Plumbago zeylanica* root extract (PZRE)

<table>
<thead>
<tr>
<th>Sample</th>
<th>Phytochemicals (mg/mL)</th>
</tr>
</thead>
<tbody>
<tr>
<td>PZRE</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Plumbagin</td>
</tr>
<tr>
<td></td>
<td>0.12±0.04</td>
</tr>
</tbody>
</table>

**Note:** The data is indicated as the means ± standard deviation, n=3.

![Figure 1A–C](https://www.dovepress.com/)

**Figure 1A–C** Ultraviolet-visible spectra of nanoparticles synthesized from PZRE at 37°C as a function of time. (A) AgNPs from PZRE in 1 mM AgNO$_3$; (B) AuNPs from PZRE in 1 mM HAuCl$_4$; (C) AgAuNPs from PZRE in 1 mM AgNO$_3$ and HAuCl$_4$ solution in the ratio 1:1.

**Abbreviations:** HAuCl$_4$, chloroaucic acid; AgNO$_3$, silver nitrate; PZRE, *Plumbago zeylanica* root extract; AgNPs, silver nanoparticles; AuNPs, gold NPs; AgAuNPs, bimetallic NPs; OD, optical density.
peak (38.9°C), located between the (111) reflections of Ag and Au, as seen in Figure 5D, represents the formation of alloy-like AgAuNPs. Similarly, characterization of AuNPs was carried out using the aforementioned techniques, and results are given in Figure S3.

TEM analysis
The TEM micrograph showed spherically shaped AgNPs synthesized by PZRE (Figure 6A). The TEM micrographs revealed some unique features of AuNPs (Figure 6B and C). They were found to be anisotropic in nature, with spheres, triangles, and hexagons exhibiting an interesting and rare phenomenon of shape evolution. Figure 6B demonstrates an arrangement of gold nanospheres forming a gold nanotriangle. Furthermore, the shape evolution continued, with nanospheres adhering to the triangle, leading to the growth of nanotriangles with attached nanospheres (Figure 6C). In Figure 6D, the arrows indicate nanohexagons and nanospheres of AgAuNPs, but the majority of them were nanohexagons. A unique feature of AgAuNPs was their shape: blunt ended polygonal NPs. Chemically synthesized NPs were also characterized using EDS, XRD, TEM, and DLS (Figures S4–S6).

Antibacterial activity
The minimum inhibitory concentration (MIC) values of the NPs against Gram-negative and Gram-positive bacteria are

Figure 2A–D Bioreduction of AgNPs and AgAuNPs from PZRE at varying reaction temperatures and concentrations of the salts as a function of time. (A) Time course of AgNPs synthesis with varying reaction temperatures; (B) time course of AgAuNPs synthesis against reaction temperature; (C) time course of AgNPs formation obtained at different AgNO₃ concentrations; (D) time course of AgAuNPs formation obtained at 1:1 AgNO₃ and HAuCl₄. Abbreviations: HAuCl₄, chloroauric acid; AgNO₃, silver nitrate; PZRE, Plumbago zeylanica root extract; AgNPs, silver nanoparticles; AgAuNPs, bimetallic NPs.

Figure 3A–D Fourier-transform infrared absorptive spectra before and after bioreduction of dried PZRE. (A) PZRE; (B) AgNPs; (C) AuNPs; (D) AgAuNPs. Abbreviations: PZRE, Plumbago zeylanica root extract; AgNPs, silver nanoparticles; AuNPs, gold NPs; AgAuNPs, bimetallic NPs.
Figure 4A–D Determination of purity, size, and nature of AgNPs synthesized by PZRE at 37°C. (A) Representative spot energy-dispersive spectrum; (B) histogram of size distribution; (C) zeta-potential analysis; (D) representative X-ray diffraction profile of thin film.

Abbreviations: PZRE, Plumbago zeylanica root extract; AgNPs, silver nanoparticles.

Figure 5A–D Determination of purity, size, and nature of AgAuNPs synthesized by PZRE at 37°C. (A) Representative spot energy-dispersive spectrum; (B) histogram of size distribution; (C) zeta-potential analysis; (D) representative X-ray diffraction profile of thin film.

Abbreviations: PZRE, Plumbago zeylanica root extract; AgAuNPs, bimetallic NPs.
given in Table 4. PZRE-mediated AgNPs showed effective antimicrobial activity compared to PZRE-mediated AuNPs and AgAuNPs, as well as the chemically synthesized AgNPs when used as a control. AgNPs exhibited a lower MIC value against *E. coli* (2 μg/disk) in comparison with *A. baumannii* and *S. aureus* strains (8 μg/disk). AuNPs showed the highest antimicrobial activity against *S. aureus*, with an MIC of 8 μg/disk, whereas chemically synthesized AuNPs did not show antimicrobial activity. Moreover, bioreduced AgAuNPs were found to show more effective antimicrobial activity compared to AgAuNPs synthesized chemically.

**Table 4** Minimum inhibitory concentration of silver nanoparticles (AgNPs), gold NPs (AuNPs) and bimetallic NPs (AgAuNPs) against bacterial cultures

<table>
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<tr>
<th>Name of the microorganism</th>
<th>Biologically synthesized</th>
<th>Chemically synthesized</th>
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<tr>
<td></td>
<td>AgNPs</td>
<td>AuNPs</td>
</tr>
<tr>
<td><em>A. baumannii</em> AIIMS 7</td>
<td>8</td>
<td>128</td>
</tr>
<tr>
<td><em>E. coli</em> NCIM 2931</td>
<td>2</td>
<td>256</td>
</tr>
<tr>
<td><em>S. aureus aureus</em> MTCC 3160</td>
<td>8</td>
<td>8</td>
</tr>
</tbody>
</table>

**Note:** All experiments were performed in triplicate and standard deviations were negligible.

**Abbreviations:** AIIMS, All India Institute of Medical Sciences; NCIM, National Collection of Industrial Microorganisms; MTCC, Microbial Type Culture Collection; –, No antimicrobial activity observed.
Biofilm inhibition and disruption using nanoparticles

As all three NP types – AgNPs, AuNPs, and AgAuNPs – showed superior antibacterial activity, these NPs were further studied for biofilm inhibition and biofilm disruption with chemically synthesized NPs as a control. Biofilm inhibition with NP concentration of 1,024 μg/well after 24 hours of coinoculation is represented by the graph in Figure 7A. It was observed that AgNPs and AgAuNPs inhibited the biofilm in all the test-microorganism single cultures around 96%–99% and 93%–98%, respectively. This was interesting compared to the control AgNPs and AgAuNPs, which showed inhibition of 67%–91% and 0%–82%, respectively. In the case of mixed culture, chemically synthesized AgNPs did not inhibit biofilm formation, whereas AgNPs from PZRE showed 91% biofilm inhibition. AgAuNPs showed A. baumannii biofilm inhibition up to 94% and E. coli biofilm inhibition up to 98%, with the highest inhibition of S. aureus biofilm of 99%. Similar to control AgNPs, control AgAuNPs showed poor biofilm inhibition compared to biological AgAuNPs, whereas AuNPs from both biological and chemical sources showed poor biofilm inhibition and disruption, except AuNPs from PZRE, which showed 97% S. aureus biofilm inhibition.

To elaborate more on the control of bacterial biofilms, a biofilm-disruption assay was carried out to check the effect of NPs on biofilm grown for 24 hours (Figure 7B). Bacterial
biofilms were seen to be disrupted in presence of AgNPs up to 88% in *A. baumannii*, 67% in *E. coli*, 78% in *S. aureus*, and 64% in mixed biofilm, whereas chemically synthesized AgNPs showed poor disruption activity of about 19%–60%. AuNPs from PZRE showed lower biofilm disruption of around 40%, with a maximum of 95% in *S. aureus*, whereas chemically synthesized AuNPs did not show any biofilm-disruption activity. AgAuNPs also showed poor disruption of about 61%–77%, but this was superior to that of chemical AgAuNPs (40%–70%).

Visualization of biofilm using fluorescence microscopy, SEM, and AFM
The cells were seen in aggregates in the absence of NPs (Figure 8A) in images obtained under fluorescence microscopy at a resolution of 40×. In the presence of AgNPs, the biofilm as well as cells were absent (Figure 8B), whereas in Figure 8C, although biofilm was absent in the presence of AgAuNPs, the cells were seen to be isolated. In SEM images, *A. baumannii* cells and the biofilm formed, seen in Figure 8D, were intact. Due to the presence of intact biofilm, the cells

Figure 8A–I Images of *Acinetobacter baumannii* biofilm formation and inhibition using fluorescence microscopy, scanning electron microscopy (SEM), and atomic force microscopy (AFM). Upper panel, fluorescence microscopy; middle panel, SEM; lower panel, AFM. (A,D,G) Biofilm formation; (B,E,H) biofilm inhibition with AgNPs; (C,F,I) biofilm inhibition with AgAuNPs.
Abbreviations: AgNPs, silver nanoparticles; AgAuNPs, bimetallic NPs.
were present in colonies, and no change in the morphology was observed. Coincubation of the bacterial cells with AgNPs led to a change in the cell morphology and disrupted the biofilm of *A. baumannii* (Figure 8E), whereas cells were seen to be lysed in the presence of AgAuNPs (Figure 8F).

AFM images in 3-D represent *A. baumannii* biofilm formation (Figure 8G) and inhibition in the presence of AgNPs and AgAuNPs (Figure 8H and I). The roughness of the glass slide containing the biofilm and NPs was calculated, which varied for different bacterial cultures. Table 5 presents the relative reduction in the roughness values of the biofilms when treated with AgNPs, AuNPs, and AgAuNPs. The roughness of the untreated biofilms was observed to be 16 nm, 5 nm, and 11 nm in *A. baumannii, S. aureus,* and *E. coli,* respectively. The percentage of relative roughness observed when the bacterial cultures were coincubated with biologically synthesized NPs was 12%–36% in *A. baumannii* and *S. aureus* and 40%–60% in *E. coli.* Chemically synthesized AgNPs (18%–45%) showed a reduction in roughness values, but were poor compared to biological AgNPs (12%–18%) except *E. coli* (20%). Chemically synthesized AuNPs and AgAuNPs showed lower reductions in roughness values of 80%–90% and 40%–60%, respectively, compared to biologically synthesized AuNPs (18%–60%) and AgAuNPs (25%–40%). Furthermore, it was also observed when fully formed biofilm was treated with AgNPs, roughness was reduced to 12% in *A. baumannii* and 50% in *E. coli,* which indicated that AgNPs also played a role in disrupting bacterial biofilms (Figure S7).

**Discussion**

*P. zeylanica* is reported to be a rich source of phytochemicals and flavonoids, as observed in HPTLC fingerprinting and GC-TOF-MS profile. Higher concentrations of sucrose, glucose, and fructose determined by chromatographic techniques and quantitation assays may have stabilized the NPs synthesized from the root extract and may have controlled NP shapes, such as spherical and hexagonal, during the course of the synthesis. Starch concentration in PZRE was the same, whereas citric acid was found to be three times that of *Dioscorea bulbifera.* Starch stabilizes the NPs by capping them. Lee et al demonstrated that citrate can adsorb on NPs during their shaping and stabilize them. Therefore, owing to higher concentrations of citric acid in PZRE, NPs were better stabilized and evolved.

The 90 minutes required for the synthesis of AgAuNPs is the shortest reported for the biological synthesis of bimetallic AgAuNPs, in contrast to those reported from neem extract, which took 296 minutes for complete synthesis. The rate of synthesis was observed to be significantly dependent on the concentration of the salt solutions and temperature, which is well in agreement with the previous reports. The hydroxyl groups of polyphenols might be playing a significant role in bioreduction, as revealed by FTIR analysis.

UV-visible spectroscopy indicating a shift from 570 nm to 540 nm in synthesis of AgAuNPs, as well as EDS analysis indicating a lower percentage of gold compared to silver, may be attributed to the Au core and Ag shell. The size of AgNPs synthesized by PZRE was comparable to earlier reports on *Aloe vera* and *Cardiospermum halicacabum,* as well as commercially available AgNPs. TEM micrographs of AuNPs showed a unique feature of shape evolution in the form of nanotriangles, in which adherence of gold nanospheres covering the whole surface area of gold nanotriangles was observed. However, Shankar et al reported on the Turkevich protocol for synthesis of gold nanotriangles and nanospheres, while Ghosh et al reported on anisotropic AuNPs from *D. bulbifera.* AgAuNPs were comprised of nanopolygons that were blunt-ended.

We observed that chemically synthesized AgNPs showed poor antibacterial activity compared to biological AgNPs. It is important to note that AuNPs synthesized by PZRE showed antimicrobial activity. However, chemically synthesized AuNPs failed to show any activity. The reason for this may be the protein coating of NPs obtained from the biological extract.
Biofilm formation by human pathogenic microbes causing severe chronic lung, urinary tract, and various nosocomial infections is attributed to lower penetration and activity of drugs, altered genotype of bacteria exposed to drugs, lower oxygen, and acidic pH. Therefore, research on biogenic NPs for control of bacterial biofilms is very promising. Various attempts have been made to control biofilms by biosurfactants, rhamnolipids, and enzyme-based detergents. Recently, Thawal et al demonstrated 80% disruption of biofilm by a phage from the family Podoviridae isolated from *A. baumannii*.

Control of bacterial biofilms using NPs like chitosan, zinc oxide, and silver are interesting, because of their high surface-to-volume ratio. Our results are in agreement with the report of Kalishwaralal et al where 50 nM of AgNPs significantly arrested *P. aeruginosa* biofilm formation without affecting cell viability, whereas 100 nM inhibited the growth of the organism itself and resulted in a 95%–98% decrease in biofilm. Also, AgNPs and AuNPs showed better biofilm inhibition than disruption; however, in the case of mixed cultures, AgAuNPs showed better biofilm disruption than biofilm inhibition. This shows the potential of AgAuNPs to disrupt mixed biofilms, which could be because of the synergistic effect of silver and gold components. Mixed biofilms, however, were inhibited and disrupted by AgNPs and AuNPs to a lesser extent compared to single bacterial biofilm, which might be due to more resistance to the penetrability of NPs challenging their potential for biofilm disruption.

The novelty of the study lies in the fact that PZRE-synthesized NPs were found to be antimicrobial against *A. baumannii* and able to control its biofilm. Although *A. baumannii* is found in the human skin microbiome, it is also considered to be a dreadful nosocomial pathogen, due to its high resistance to antibiotics and metal salts leading to severe mortality (17%–62%) among immunocompromised patients because of bacteremia. Deformity in *A. baumannii* cells treated with AgNPs, as observed in fluorescence and SEM micrographs, might be due to enhanced penetrability of AgNPs within the biofilm, unlike metallic silver. This can be a powerful strategy for sensitizing *A. baumannii* with high metal resistance.

This is the first report of its kind where inhibition as well as disruption of bacterial biofilms have been demonstrated employing AFM after treatment with NPs synthesized by PZRE. Significant reduction of roughness confirmed the efficacy of NPs in biofilm inhibition and disruption. This proved to be advantageous, owing to semi-contact mode, better resolution, biofilms observed in their own physiological condition with no addition of chemicals, direct contact with the surface with minimum use of lenses and electron beams, and deduction of roughness values based upon variation in height and texture.

**Conclusion**

This is the first report of rapid and efficient synthesis of AgNPs, AuNPs, and AgAuNPs by *P. zeylanica*. All the three types of NPs had unique features: being spherical in case of AgNPs, revealing shape evolution in case of AuNPs, and showing blunt-ended hexagonal nanostructures in AgAuNPs. These diverse properties were obtained due to phenolics, flavonoids, sugars, starch, and citric acid present in PZRE. Bioreduced NPs were able to inhibit to significant levels, particularly in the case of *E. coli* and *S. aureus*. *S. aureus* biofilm was inhibited and disrupted significantly by AuNPs, which shows that AuNPs can be more effective against Gram-positive bacteria than Gram-negative bacteria. Therefore, the NPs synthesized, particularly AgNPs from *P. zeylanica*, proved to be very efficient in the control of bacterial biofilms.

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**Disclosure**

The authors report no conflicts of interest in this work.

**References**


Supplementary materials

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Abbreviations: HPTLC, high-performance thin-layer chromatography; PZRE, Plumbago zeylanica root extract; AuNPs, gold nanoparticles; AuAuNPs, bimetallic NPs; AgNPs, silver nanoparticles; HAuCl₄, chloroauric acid; AFM, atomic force microscopy.

Figure S1 HPTLC fingerprint of PZRE.
Notes: (A) PZRE ethyl acetate extract, and (B) standard mixture of plumbagin (P), diosgenin (D), quercetin (Q) and catechin (C).
Abbreviations: HPTLC, high-performance thin-layer chromatography; PZRE, Plumbago zeylanica root extract.

Figure S2 Bioreduction of AuNPs from PZRE at varying reaction temperatures and concentrations of the salts as a function of time. (A) Time course of AuNPs synthesis against reaction temperature, and (B) time course of AuNPs formation at different HAuCl₄ concentration.
Notes: (A) Time course of AuNPs synthesis against reaction temperature, and (B) time course of AuNPs formation at different HAuCl₄ concentration.
Abbreviations: PZRE, Plumbago zeylanica root extract; AuNPs, gold nanoparticles; HAuCl₄, chloroauric acid; h, hours.
Figure S3 Determination of purity, size and nature of the AuNPs synthesized by PZRE at 37°C.
Notes: (A) Representative spot energy-dispersive spectrum, (B) histogram of size distribution, (C) zeta potential analysis, and (D) representative X-ray diffraction profile of thin film.
Abbreviations: PZRE, Plumbago zeylanica root extract; AuNPs, gold nanoparticles.

Figure S4 Determination of purity, size and nature of the AgNPs synthesized using 1% trisodium citrate.
Notes: (A) Representative X-ray diffraction profile of thin film, (B) representative spot energy-dispersive spectrum, (C) transmission electron micrograph, and (D) histogram of size distribution.
Abbreviation: AgNPs, silver nanoparticles.
Figure S5 Determination of purity, size and nature of the AuNPs synthesized using 1% trisodium citrate.
Notes: (A) Representative X-ray diffraction profile of thin film, (B) representative spot energy-dispersive spectrum, (C) transmission electron micrograph, and (D) histogram of size distribution.
Abbreviation: AuNPs, gold nanoparticles.

Figure S6 Determination of purity, size and nature of the AgAuNPs synthesized using 1% trisodium citrate.
Notes: (A) Representative X-ray diffraction profile of thin film, (B) representative spot energy-dispersive spectrum, (C) transmission electron micrograph, and (D) histogram of size distribution.
Abbreviation: AgAuNPs, bimetallic nanoparticles.
Materials and methods for cytotoxicity assay

MCF7 (human breast adenocarcinoma), was obtained from National Center for Cell Sciences, India and cultured in a humidified atmosphere (37°C, 5% CO₂) in Roswell Park Memorial Institute1640 medium (Life Technologies, Waltham, MA, USA) supplemented with 10% fetal bovine serum.1 Nanoparticles were evaluated for cytotoxicity against MCF7 cancer cell lines. The compounds were evaluated in vitro at a various concentration. MTT colorimetric assay was used to determine growth inhibition. 100 µL of cell suspension (2×10⁴ cells) were plated in 96-well plates and allowed to attach for 24 hours. Cells were exposed in triplicate wells to AgNPs, AuNPs and AgAuNPs both synthesized chemically as well as by PZRE for 48 hours. After 48 hours, 20 µL MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide) solution (5 mg/mL) was added to each well. After 4 hours of incubation the formazan precipitate was dissolved in 100 µL of solubilization buffer (10% SDS in 0.1 N HCl) and kept overnight for complete dissolution. The absorbance at 576 nm was determined on a 96 well microplate reader (SpectraMax M5®, Molecular Devices Corporation, Sunnyvale, CA). The absorbance values were used to calculate % inhibition.

Reference