Poloxamer 407/TPGS mixed micelles for delivery of gambogic acid to breast and multidrug-resistant cancer

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Background: Delivery of a high concentration of anticancer drugs specifically to cancer cells remains the biggest challenge for the treatment of multidrug-resistant cancer. Poloxamers and D-α-Tocopheryl polyethylene glycol 1000 succinate (TPGS) are known inhibitors of P-glycoprotein (P-gp). Mixed micelles prepared from Poloxamer 407 and TPGS may increase the therapeutic efficacy of the drug by delivering high concentrations inside the cells and inhibiting P-gp. Gambogic acid (GA) is a naturally derived novel anticancer agent, but poor solubility and toxic side effects limit its use. In this study, we have developed Poloxamer 407 and TPGS mixed micelle-encapsulating GA for the treatment of breast and multidrug-resistant cancer.

Methods: GA-loaded Poloxamer 407/TPGS mixed micelles were prepared using a thin film hydration method, and their physicochemical properties were characterized. Cellular accumulation and cytotoxicity of the GA-loaded Poloxamer 407/TPGS mixed micelles were studied in breast cancer cells, MCF-7 cells, and multidrug-resistant NCI/ADR-RES cells.

Results: The diameter of GA-loaded Poloxamer 407/TPGS mixed micelles was about 17.4 ± 0.5 nm and the zeta potential was -13.57 mV. The entrapment efficiency of GA was 93.1% ± 0.5% and drug loading was about 9.38% ± 0.29%. Differential scanning calorimetry and X-ray powder diffraction studies confirmed that GA is encapsulated by the polymers. The in vitro release studies showed that mixed micelles sustained the release of GA for more than 4 days. Results from cellular uptake studies indicated that GA-loaded Poloxamer 407/TPGS mixed micelles had increased cellular uptake of GA in NCI/ADR-RES cells. Cytotoxicity of GA-loaded Poloxamer 407/TPGS mixed micelles was found to be 2.9 times higher in multidrug-resistant NCI/ADR-RES cells, and 1.6 times higher in MCF-7 cells, as compared with unencapsulated GA.

Conclusion: This study suggests that Poloxamer 407/TPGS mixed micelles can be used as a delivery system for GA to treat breast and multidrug-resistant cancer.

Keywords: gambogic acid, Poloxamer 407, TPGS, P-glycoprotein, multidrug resistance, breast cancer

Introduction
Multidrug resistance (MDR) is the phenomenon of simultaneous resistance to many structurally unrelated cytotoxic agents, and is one of the most formidable challenges in the field of cancer therapy. A common mechanism of MDR is the active export of drugs from cancer cells by the overexpression of P-glycoprotein (P-gp) and other adenosine triphosphate (ATP)-binding cassette transporters. The use of Poloxamer block copolymers as a drug delivery vehicle to treat MDR cancers is a rapidly developing area for cancer chemotherapy. Poloxamers have been shown to reverse MDR by inhibiting the P-gp pump.
Gambogenic acid (GA) is the principal active ingredient of gamboge, the resin found in trees of the genus *Garcinia*. GA possesses diverse biological properties such as anti-inflammatory, antioxidant, antiviral, anti-infectious, and neuroprotective effects. Additionally, GA inhibits the growth of different types of cancer including gastric cancer, lung carcinoma, prostate cancer, melanoma, hepatic carcinoma, and breast cancer.

Several possible mechanisms are proposed for the anticancer effect of GA, such as induction of apoptosis, microtubule depolymerization, and downregulation of telomerase activity. Gu et al. showed that GA can induce apoptosis in MCF-7 cancer cells through upregulation of p53 and downregulation of Bcl-2 proteins. GA also depolymerized microtubules and elevated the phosphorylation levels of JNK1 and p38, which caused G2/M cell cycle arrest and apoptosis in MCF-7 cells. GA significantly inhibited the adhesion, migration, and invasion of the cancer cells in vitro. Wang et al. describe the potential role of GA to reverse docetaxel resistance though downregulation of survivin in gastric cancer cells. Survivin might play a key role in MDR in the presence of P-gp, and this might represent a novel strategy for modulating MDR in cancer cells. Recently, a selective apoptotic induction of GA on cancer cells compared with normal cells suggests GA might be an effective anticancer drug candidate with low toxicity to normal cells.

The poor water solubility of GA has limited its use. Several approaches have been used to improve the solubility of the GA, such as conjugation with L-arginine or use of polyoxylated castor oil (Cremophor EL®; BASF SE, Ludwigshafen, Germany), but lethal allergic reaction and ulceration was observed with L-arginine containing formulations. The Cremophor EL-containing formulation had several side effects such as cardiotoxicity, nephrotoxicity, neurotoxicity, and hypersensitivity reactions.

Polymeric micelle systems were recognized as one of the most promising strategies to deliver poorly soluble anticancer drugs. Polymeric micelles made of amphiphilic copolymers consist of a core shell structure formed by self-assembly of a hydrophobic block as an internal core and a hydrophilic block as a surrounding shell. Poorly soluble drugs can be made soluble within the hydrophobic inner core of a micelle and hydrophilic shell interface the biological media. As a result, micelles can substantially improve solubility and bioavailability of various hydrophobic drugs. The nanometer size (<100 nm) of micelles allows higher accumulation and delivery of encapsulated anticancer drugs, via the enhanced permeability and retention effect.

Among a number of natural and synthetic polymers used to form polymeric micelles, Poloxamer 407, a US Food and Drug Administration-approved polymer, is the most attractive due to its biocompatibility, biodegradability, and low toxicity. Poloxamer block copolymer consists of ethylene oxide (EO) and propylene oxide (PO) blocks that are arranged in a basic EO x–PO y–EO x structure. The PO core can serve as a “pool” and the hydrophobic drug can be incorporated into the hydrophobic PO core, while the hydrophilic corona maintains the dispersion stability of Poloxamer micelles. Incorporation of drugs into micelles leads to increased solubility, metabolic stability, and circulation time of the drug.

D-α-Tocopheryl polyethylene glycol 1000 succinate (TPGS), a derivative of natural vitamin E (α-tocopherol) and polyethylene glycol 1000, has been widely applied in the food and drug industry as solubilizer, absorption enhancer, and a vehicle for lipid-based drug delivery formulations. It has also been found that TPGS enhances the solubility, inhibits P-gp-mediated multidrug resistance, and increases the oral bioavailability of anticancer drugs. It was also reported that succinate esters of vitamin E behave as potent proapoptotic agents selective for cancer cells. Mixed micelles manifest synergistic properties, such as increased micelle stability and drug-loading efficiency, superior to those of the individual components.

We hypothesize that GA-loaded mixed micelles made of Poloxamer 407 and TPGS may allow a better nanoformulation, more stability, and higher anticancer efficiency in resistant breast cancer. GA-loaded mixed Poloxamer 407/TPGS micelles were prepared and their physical properties were characterized. The in vitro cellular uptake and cytotoxicity against breast cancer cells, MCF-7, and multidrug resistant NCI/RES-ADR cells were studied.

**Methods and materials**

Poloxamer 407 was obtained as a gift from BASF (Tarrytown, NY), TPGS was purchased from Sigma-Aldrich (St Louis, MO). GA was purchased from 2A PharmaChem (Lisle, IL). All other reagents and buffer solutions were of analytical grade. Distilled and deionized (DI) water was used in all experiments.

**Preparation of micelles**

GA-loaded Poloxamer 407/TPGS mixed micelles were prepared by thin film hydration method as described earlier in Zhang et al. Briefly, GA (9 mg), Poloxamer 407 (68 mg),
and TPGS (23 mg) were dissolved in 3 mL acetonitrile in a round bottom flask. The solvent was evaporated by rotary evaporation at 40°C for 1 hour to obtain a solid GA/copolymer matrix. Residual acetonitrile remaining in the film was removed under vacuum overnight at room temperature. The resultant thin film was hydrated with 4 mL water at 40°C for 30 minutes to obtain a micelle solution, which was then filtrated through 0.2 µm filter membrane to remove the unincorporated GA aggregates, followed by lyophilization. Empty Poloxamer 407/TPGS mixed micelles were also prepared as described above, excepting the addition of GA.

Critical micelle concentration (CMC) determination

To determine the CMC of binary mixture of Poloxamer 407/TPGS in DI water, an ultraviolet-visible (UV-Vis) spectroscopy method was used, using iodine as a hydrophobic probe.30 The KI/I₂ standard solution was prepared by dissolving 0.5 g of iodine and 1 g of potassium iodide in 50 mL DI water. Samples of polymer solution with concentrations ranging from 0.00001% to 0.1% were prepared. To each of the Poloxamer 407/TPGS binary mixture (molar ratio 3:7) solutions, 25 µL KI/I₂ standard solution was added. The mixtures were incubated for 12 hours in a dark room at room temperature before measurement. The ultraviolet absorbance value of varying polymer concentrations at 366 nm was measured using a UV-Vis spectrometer (DU 800; Beckman Coulter, Indianapolis, IN). Experiments were performed in triplicate. The absorption intensity was plotted against the logarithm of polymer concentration. The CMC values correspond to the concentration of the polymer at which the sharp increase in absorbance is observed.

Particle size and zeta potential analysis

Particle mean size and size distribution were measured by the dynamic light scattering method using a Brookhaven Particle Size/Zeta Potential analyzer (ZetaPALS, Brookhaven Instruments, Holtsville, NY). The analyses were performed with 5 mW He-Ne laser (632.8 nm) at a scattering angle of 90° at 25°C. Each freeze dried or freshly prepared sample was diluted to the appropriate concentration using DI water. The reported experimental result of each sample was expressed as a mean size ± standard deviation (SD) for three separate experiments.

Drug loading and encapsulation efficiency

Drug loading (DL%) and encapsulation efficiency (EE%) were determined by UV-Vis-spectrophotometry at maximum absorbance of 361 nm. The micellar solution was suitably diluted with acetonitrile prior to determination. DL% and EE% were calculated using following equations:26

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DL\% = \frac{\text{Weight of the drug in micelles}}{\text{Weight of the feeding polymer and drug}} \times 100
\]

(1)

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EE\% = \frac{\text{Weight of the drug in micelles}}{\text{Weight of the feeding drug}} \times 100
\]

(2)

Differential scanning calorimetry (DSC)

The physical state of GA encapsulated in the mixed micelles and the free GA were examined by DSC using DSC Q200 (TA Instruments, New Castle, DE) under liquid nitrogen at a flow rate of 50 mL per minute. Then, 5 mg of the sample was heated from 0°C to 300°C at a temperature-rising speed of 10°C per minute.

X-ray powder diffraction analysis (XRPD)

XRPD experiments were performed from 2° to 40° (2θ angle) using a D8 X-Ray diffractometer system (Bruker AXS, Madison, WI) with a copper anode source (Kα, λ = 1.5406 Å) with a filtered flat lithium fluoride secondary beam monochromator. The divergence, receiver, and detector slit widths were 2 mm; the scatter slit width was 0.6 mm.

In vitro drug release

To create a pseudo-sink condition, the in vitro release behaviors of GA from mixed micelles were monitored in phosphate-buffered saline (PBS) (pH 7.4) with 0.5% Tween® 80 (ICI Americas Inc, Wilmington, DE).26 Briefly, 10 mg of GA-loaded mixed micelles was introduced into a dialysis membrane bag (MWCO 6000–8000 Da), and the sealed dialysis bag was incubated in 20 mL of release media into an orbital shaker (VWR International, LLC, Suwanee, GA) at 37°C and 100 rpm. At predetermined time intervals, a sample was withdrawn and complete release media was replaced with fresh release media. The concentration of GA in a sample was measured by the DU 800 UV-Vis spectrophotometer at maximum absorbance of 360 nm and the cumulative release percentage was calculated. For comparison, the release of GA from dimethyl sulfoxide solution was conducted under the same conditions.31

Cell culture

Breast cancer MCF-7 cells were obtained from the American Type Culture Collection (ATCC; Manassas, VA) and
multidrug-resistant NCI/ADR-RES cells were obtained from the National Cancer Institute (Bethesda, MD). Cells were cultured in RPMI 1640 media supplemented with 10% fetal bovine serum and 1% penicillin/streptomycin (all from Mediatech, Manassas, VA).

**Cellular internalization study**

MCF-7 and NCI/ADR-RES cells grown on two-well LabTek chambered glass slides (Nalge Nunc International, Rochester, NY) at a density of $5 \times 10^4$ cells/well and incubated for 24 hours. One milliliter of media containing free GA or GA-loaded mixed micelles equivalent to 1 µM of GA was added to the cells. After incubation for 4 hours, cells were washed three times with ice cold PBS, fixed with 4% paraformaldehyde. A cover slip was mounted on the slide and the slide was photographed using a Nikon Eclipse Ti (Nikon Instruments Inc, Melvill, NY) confocal laser scanning microscope (CLSM).

**In vitro cytotoxicity study**

In vitro cytotoxicity of GA-loaded mixed micelles was determined using 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay. Briefly, MCF-7 and NCI/ADR-RES cells were transferred to 96-well tissue culture plates at a density of $2 \times 10^3$ cells per well and $5 \times 10^3$ cells per well, respectively, 24 hours prior to treatment. The medium was then replaced with fresh medium containing free GA, GA-loaded micelles, and blank micelles at different concentrations. The culture medium without any drug formulation was used as the control. After 72 hours of incubation at 37°C and 5% CO₂, media was removed and cells were washed twice with sterile PBS. Then 50 µL of MTT solution (0.5 µg per mL) was added to each well and further incubated for 4 hours at 37°C. Medium was removed and 100 µL dimethyl sulfoxide was added to each well to dissolve the purple formazan crystal converted from MTT. Cell viability was assessed by absorbance at 570 nm measured on a NOVOstar micro plate reader (BMG Labtech Inc, Cary, NC) and the concentration at which 50% of growth is inhibited (IC₅₀) was calculated by GraphPad Prism 5.0 (GraphPad Software, La Jolla, CA).

**Statistical analysis**

Data are represented as means ± SD from at least three separate experiments. Statistical significance was assessed using Student’s t-test. Differences were considered significant at $P < 0.05$.

**Results and discussion**

**CMC determination**

The CMC of the micelles influences their in vitro and in vivo stability. Low CMC values of Poloxamer 407/TPGS binary mixture underlie high stability of Poloxamer 407/TPGS mixed micelles in solutions upon dilution. In this study, the formation of micelles was monitored by using iodine (I) as a hydrophobic probe. Solubilized I₂ prefers to participate in the hydrophobic microenvironment of Poloxamer copolymer, causing the conversion of I₃⁻ to I₂ from the excess potassium iodide (KI) in the solution. The absorbance intensity of I₂ has been plotted as a function of polymer concentration in Figure 1 and the CMC of Poloxamer 407/TPGS binary mixture was found to be 0.0013% w/v. The low CMC of the micelles suggests high stability and ability to maintain integrity even upon dilution in the blood circulation.

**Particle size distribution**

To achieve longevity during systemic circulation, the micelles must be small enough to evade detection and destruction by the reticulo-endothelial system (RES). The average micelle size and the unimodal size distribution of both empty and GA-loaded Poloxamer 407/TPGS mixed micelles are illustrated in Figure 2A and B, indicating the formation of nanosized mixed micelles. The mean diameter of blank micelles as well as GA-loaded Poloxamer 407/TPGS mixed micelles was close to 16.6 ± 0.1 nm and 17.4 ± 0.5 nm respectively, with an acceptable polydispersity index between 0.1–0.2. Encapsulating the GA in the micelles

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**Abbreviation:** TPGS, D-α-Tocopheryl polyethylene glycol 1000 succinate.
did not increase the micelle average size and size distribution. It was found that there was no increase in the particle size of empty and GA-loaded Poloxamer 407/TPGS mixed micelles after freeze drying (Figure 2C and D). After 6 months in storage at 4°C, the drug content of freeze dried GA-loaded Poloxamer 407/TPGS mixed micelles did not change, indicating that lyophilization is a good choice for storage of mixed micelles.

Zeta potential
Both empty and GA-loaded Poloxamer 407/TPGS mixed micelles were negatively charged with zeta potential of about −8.52 mV and −13.57 mV, respectively. In the structure of Poloxamer 407 both the polypropylene oxide and polyethylene oxide segments were nonionic, so the change of the surface charge of the micelles should result from addition of GA and/or TPGS. The free carboxylic group of GA can dissociate and result in the anionic property for GA. Therefore, the absolute value of zeta potential was increased by addition of drug. The increased negative surface charge of GA might be attributed to TPGS, which enhanced the solubility of GA and more anionic GA was present in solution.

Drug loading and encapsulation efficiency
DL% of the GA-loaded Poloxamer 407/TPGS mixed micelles was about 9.38% ± 0.29% and the EE% was 93.1% ± 0.5%. The encapsulation of GA in the mixed micelle system is high at this drug loading.

DSC and XRPD analysis
The physical status of GA inside the micelles was examined by DSC and XRPD. DSC thermogram (Figure 3A) revealed endothermic peaks of GA and Poloxamer 407/TPGS. Thermogram for physical mixture shows all the endothermic peaks of GA and Poloxamer/TPGS while the thermogram for blank Poloxamer 407/TPGS mixed micelles only shows endothermic peaks of polymers. The thermogram of GA-loaded Poloxamer 407/TPGS micelles shows all endothermic peaks of polymers as shown in the blank micelles thermogram, but the characteristic peaks (at 140°C) of GA was absent, which confirmed the encapsulation of GA inside the Poloxamer 407/TPGS micelles.

XRPD diffraction spectra (Figure 3B) shows characteristic diffraction peaks of GA, which were visible in the pattern obtained for GA as well as for the physical mixture of GA, Poloxamer 407, and TPGS, but disappeared in the pattern of GA-loaded Poloxamer 407/TPGS, especially around 4°C–10°C. This indicates that GA was encapsulated in the polymeric micelles in molecular or amorphous state and there was no free drug on the surface of micelles.
In vitro drug release

The in vitro release of GA from mixed micellar formulation under sink conditions was investigated by dialysis method with 0.5% Tween 80 solution as release medium. As shown in Figure 4, about 30% of GA was released from mixed micelles within the first 4 hours, while almost 100% of GA was released from the GA solution during the same time period. After 4 days approximately 90% of the GA was released from GA-loaded Poloxamer 407/TPGS mixed micelles. The micellar carrier can not only solubilized the poorly soluble GA, but also sustained the GA release for more than 4 days. Drug diffusion, polymer erosion or swelling might be possible mechanisms for release of GA from micelles.32

Cellular uptake study

To evaluate the cellular internalization of GA-loaded Poloxamer 407/TPGS mixed micelles, the cellular uptake by MCF-7 and NCI/ADR-RES cells were investigated. Figure 5A1–3 and B1–3 shows the CLSM images of free GA and GA-loaded Poloxamer 407/TPGS mixed micelles respectively, in MCF-7 cells. GA itself was fluorescent under the CLSM and the images show that, similar to free GA, GA-loaded Poloxamer 407/TPGS mixed micelles also distributed throughout the cytoplasm. This indicates that the mixed micelles have been taken up by the cells and are located inside the cells. There is equal or higher uptake of GA-loaded Poloxamer 407/TPGS mixed micelles in MCF-7. The uptake of free GA may be due to the binding affinity of GA for transferrin receptors33 while the micelles are internalized by endocytosis. Figure 5C1–3 and D1–3 show the CLSM images of free GA and GA-loaded Poloxamer 407/TPGS mixed micelles respectively, in NCI/ADR-RES cells. It was clear that the intensity of the green fluorescence inside the cells treated with GA-loaded Poloxamer 407/TPGS mixed micelles (Figure 5D) was much greater than that of the free GA (Figure 5C). This indicates an enhanced uptake of GA-loaded Poloxamer 407/TPGS mixed micelles in NCI/ADR-RES cells as compared to free GA. NCI/ADR-RES cells expressing the P-gp may have higher efflux of GA from the cells, but mixed micelles taken up by cells via endocytosis are not subject to efflux by P-gp. Also, Poloxamer 407/TPGS mixed micelles may inhibit the P-gp efflux pump.34,35

In vitro cytotoxicity study

Figure 6A shows the in vitro cytotoxic effect of free GA and GA-loaded Poloxamer 407/TPGS mixed micelles in MCF-7, and Figure 6B shows it in NCI/ADR-RES cells. The results indicate that GA formulated in the polymeric micelles has
shown advantages in achieving higher cytotoxicity than the free GA (Figure 7). The IC$_{50}$ of GA-loaded Poloxamer 407/TPGS mixed micelles (0.44 ± 0.02 μM) in MCF-7 cells which is about 1.6 fold less than the free GA (0.71 ± 0.06 μM) ($P < 0.05$) and the IC$_{50}$ of GA-loaded Poloxamer 407/TPGS mixed micelles (0.86 ± 0.01 μM) in NCI/ADR-RES cells was found to be 2.9 fold less than the free GA (2.5 ± 0.08 μM) ($P < 0.01$). The lower cytotoxicity of unencapsulated GA may be due to its efflux by P-gp pumps. On the other hand, GA-loaded Poloxamer 407/TPGS mixed micelles are taken up by cells via endocytosis, and are not subject to efflux by P-gp. This allows sustaining the drug concentration inside the cells which results in a higher cytotoxic effect, which is also reported in the literature.³⁰,³⁶,³⁷ Another explanation for improved cytotoxicity of GA-loaded mixed micelles was the effect of Poloxamer 407 and TPGS, which are well known P-gp inhibitors and also reduce drug efflux.³⁴,³⁵ Empty Poloxamer 407/TPGS mixed micelles had no significant effects on the viability of MCF-7 and NCI/ADR-RES cells at concentrations up to 17.4 μg/mL (Figure 6C), which is much higher than the concentration needed for IC$_{50}$ of GA-loaded mixed micelles. The anticancer activity of TPGS is reported to be related to its unique apoptosis-inducing properties via the generation of reactive oxygen species. This might be the reason for toxicity of empty mixed micelles to cancer cells at a higher concentration.³⁸,³⁹ The P-gp inhibition by the Poloxamer 407/TPGS mixed micelles makes these micelles a better nanocarrier than other micelles such as PEG-PCL⁴⁰ and chitosan micelles⁴¹ to treat multi-drug resistant cancer.

**Conclusion**

The purpose of this study was to design mixed micelles composed of Poloxamer 407 and TPGS loaded with the poorly soluble natural anticancer agent, GA. The micelles have a low CMC, small particle size and high encapsulation efficiency. GA release was sustained as a result of encapsulation into the inner core of the micelles. GA encapsulation by Poloxamer 407/TPGS mixed micelles also demonstrated an equal or increased cellular uptake and increased in vitro cytotoxicity compared to free GA in breast cancer cells MCF-7 as well as in multidrug resistant NCI/ADR-RES cells. Results suggested that GA encapsulated Poloxamer 407/TPGS mixed micelles will be a good nanoformulation to treat breast and multidrug-resistant cancer. Future studies will be focus on in vivo evaluation of GA-loaded Poloxamer 407/TPGS mixed micelles.

**Disclosure**

The authors report no conflicts of interest in this work.

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**Figure 6** In vitro cytotoxicity of free gambogic acid (GA), GA-loaded Poloxamer 407/TPGS mixed micelles, blank poloxamer/TPGS mixed micelles in (A) breast cancer MCF-7 cells and (B) multidrug resistant NCI/ADR-RES cells after 72 hours of incubation. (C) Cytotoxicity of different concentrations of blank poloxamer/TPGS mixed micelles in breast cancer MCF-7 cells and multidrug-resistant NCI/ADR-RES cells after 72 hours of incubation.

**Notes:** Mean ± SD, n = 3.

**Abbreviations:** TPGS, D-α-Tocopherol polyethylene glycol 1000 succinate; SD, standard deviation.
References


