Oral fondaparinux: use of lipid nanocapsules as nanocarriers and in vivo pharmacokinetic study

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Abstract: Oral anticoagulant therapy could be advanced using lipid-based nanoparticulate systems. This study examined lipid nanocapsules for their oral absorption potential as the first step in developing oral fondaparinux (Fp) novel carriers. Using phase inversion method and cationic surfactants such as hexadecyltrimethyl ammonium bromide (CTAB) or stearylamine (SA), cationic lipid nanocapsules (cLNCs), loaded with Fp on their surface, were prepared and characterized (zeta potential, size and Fp association efficiency and content). In vivo studies were conducted after single oral increasing doses of Fp-loaded cLNCs (0.5 to 5 mg/kg of Fp) in rats and the concentration of Fp in the plasma was measured by anti-factor Xa activity assay. The monodisperse, (~50 nm), positively charged Fp-cLNCs with high drug loadings demonstrated linear pharmacokinetic profiles of the drug with an increased oral absolute bioavailability (up to ~21%) compatible with therapeutic anticoagulant effect (>0.2 µg/mL).

Keywords: oral anticoagulant, fondaparinux, lipid nanocapsules, bioavailability, pharmacokinetics, rats

Introduction

Anticoagulants, being a mainstay of cardiovascular therapy, are widely used for prevention and treatment of venous thromboembolism, management of acute coronary syndromes and stroke prevention in patients with atrial fibrillation or mechanical heart valves.1,2 The traditional anticoagulants: unfractionated heparin (UFH); low molecular weight heparins (LMWH); and the synthetic pentasaccharide anti-factor Xa, fondaparinux (Fp), are administered only via intravenous (IV) or subcutaneous (SC) injection since they are not efficiently absorbed through the gastrointestinal tract when taken orally.3 Oral anticoagulation is still commonly achieved by administration of vitamin K antagonists (VKA) that are associated with untoward side effects due to their narrow therapeutic range and interaction potential.4 For the new oral direct factor Xa inhibitors and direct thrombin inhibitors,1,5 various challenges still remain such as decreasing the incidence of bleeding side effects.4,6,7 There is thus a rationale to develop an oral formulation of Fp.

Different oral strategies have proved to be effective as noninvasive delivery systems. The use of polymers of high molecular weight with mucoadhesion properties such as chitosan derivatives or thiolated polymers alone or with glutathione demonstrated a powerful capacity to increase UFH and recently Fp oral absorption.8-10 However, potentially, safety concerns have always been considered with these compounds.11 Polymeric drug delivery systems based on hydrogels, nanoparticles, microspheres,12-14 as well as lipid-based drug delivery systems (eg, microemulsions, liposomes, and solid...
lipid nanocapsules (LNCs) have been developed and employed for oral macromolecular drug delivery in general and specifically for UFH or LMWH delivery. They seem to provide high transfer of the drug across the epithelium mucosa and protect against enzymatic degradation. The role of lipid dosage forms in intestinal absorption was extensively discussed. One important issue with the lipid-based particles (conventional liposomes and microemulsions) is that, generally, they lack efficiency to entrap hydrophilic macromolecular drugs and have low stability in the GI tract. Lipid nanocapsules (LNCs), recently developed, combine the advantages of colloidal particles and lipid delivery systems. They are prepared by a solvent-free, simple phase inversion technique and are composed of rigid surfactant shell surrounding an oily liquid core. They have shown potential for the delivery of many drugs by different routes, mainly parenteral as well as pulmonary and more interestingly, the oral route. Their gastrointestinal stability and their P-gp inhibiting properties are attributed to SoluMelt (BASF, Ludwigshafen, Germany), which is a key excipient in their composition. Stability of nanocapsules and P-gp inhibition facilitate intracellular uptake of paclitaxel, an anticancer drug, resulting in its oral absorption proven to be mainly via active endocytic process. The safety of these new nanocapsules has been demonstrated.

Fondaparinux (Arixtra® GlaxoSmithKline, London, UK) shares all the pharmacological and biological advantages of LMWHs over UFH. Being an entirely synthetic molecule, with high affinity to antithrombin (AT), Fp catalyzes AT-mediated inactivation of factor Xa. It displays a rapid onset of action and long terminal half-life that permits once-daily administration. It lacks antifactor IIa activity, which has been associated with bleeding in Hp and other derivative molecular weight of 364.45 and 269.52 g/mol respectively, were added from the beginning to the other components surrounding a cohesive interface and dispersed in an aqueous medium. The oily phase is composed of 5.11% w/w of the lipophilic Labrafil® WL 1349 (caprylic-capric acid triglycerides; European Pharmacopeia, IVth, 2002, Gattefossé S.A., Saint-Priest, France). The aqueous phase consists of 0.42% w/w NaCl (Prolabo; Fontenay-sous-Bois, France) and 14.04% w/w water (Milli RO® System, Millipore, Paris, France). Two surfactants are used for LNC preparation: 0.39%–0.7% w/w Lipoïd® S75-3 (a soybean lecithin at 69% of phosphatidylcholine, Lipoïd GMBH Ludwigshafen, Germany) and 4.72% w/w SoluMelt® HS 15 (a mixture of free polyethylene glycol 660 and polyethylene glycol 660 hydroxystearate, European Pharmacopeia, IVth, 2002, BASF, Ludwigshafen, Germany). Briefly, the emulsion is first prepared by mixing all the components under magnetic stirring at room temperature, with a progressive heating to 85°C followed by a progressive cooling from 85°C to 60°C at a rate of 4°C/minute. Three temperature cycles are applied to obtain the inversion process defined by a temperature range which depends on the composition of the mixture. An irreversible shock is then induced by dilution with cold deionized water (2°C) added to the mixture (3.5-fold by dilution) leading to the formation of stable nanocapsules. Afterwards slow magnetic stirring is applied for 5 minutes, followed by filtration (0.2 μm syringe filter, Sartorius, Gottingen, Germany).

Cationic surfactants (hexadecyltrimethyl ammonium bromide (CTAB) or stearylamine (SA) (Sigma Aldrich, St Quentin-Falavier, and Yvelines, France)) having a molecular weight of 364.45 and 269.52 g/mol respectively, were added from the beginning to the other components in different concentrations (0.125% to 1% w/v of the final volume). The concentration of the cationic surfactant was chosen based on an adsorption study and on the desired amount of drug to be associated with the nanocarriers; higher cationic surfactant concentrations were used for higher drug loadings. Lipoïd® concentration was also changed in all formulations, and its ratio with the cationic surfactants differed being the least with SA (1:1.5) (Table 1). Different volumes

### Material and methods

#### Preparation of lipid nanocapsules

The cLNCs were prepared by the phase inversion method previously described but modified to incorporate cationic surfactants: They are composed of a liquid oily core surrounded by a cohesive interface and dispersed in an aqueous medium. The oily phase is composed of 5.11% w/w of the lipophilic Labrafil® WL 1349 (caprylic-capric acid triglycerides; European Pharmacopeia, IVth, 2002, Gattefossé S.A., Saint-Priest, France). The aqueous phase consists of 0.42% w/w NaCl (Prolabo; Fontenay-sous-Bois, France) and 14.04% w/w water (Milli RO® System, Millipore, Paris, France). Two surfactants are used for LNC preparation: 0.39%–0.7% w/w Lipoïd® S75-3 (a soybean lecithin at 69% of phosphatidylcholine, Lipoïd GMBH Ludwigshafen, Germany) and 4.72% w/w SoluMelt® HS 15 (a mixture of free polyethylene glycol 660 and polyethylene glycol 660 hydroxystearate, European Pharmacopeia, IVth, 2002, BASF, Ludwigshafen, Germany). Briefly, the emulsion is first prepared by mixing all the components under magnetic stirring at room temperature, with a progressive heating to 85°C followed by a progressive cooling from 85°C to 60°C at a rate of 4°C/minute. Three temperature cycles are applied to obtain the inversion process defined by a temperature range which depends on the composition of the mixture. An irreversible shock is then induced by dilution with cold deionized water (2°C) added to the mixture (3.5-fold by dilution) leading to the formation of stable nanocapsules. Afterwards slow magnetic stirring is applied for 5 minutes, followed by filtration (0.2 μm syringe filter, Sartorius, Gottingen, Germany).

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Table 1 Variation in composition of cationic lipid nanocapsule formulations

<table>
<thead>
<tr>
<th>Formulations</th>
<th>Cationic surfactant concentration (% w/v)</th>
<th>Lipoid concentration (% w/w)</th>
<th>Fp total concentration (μg/mL)</th>
</tr>
</thead>
<tbody>
<tr>
<td>CTAB</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>cLNC-C0</td>
<td>0.25</td>
<td>0.68</td>
<td>0</td>
</tr>
<tr>
<td>Fp-cLNC-C1</td>
<td>0.25</td>
<td>0.68</td>
<td>95</td>
</tr>
<tr>
<td>Fp-cLNC-C2</td>
<td>0.25</td>
<td>0.68</td>
<td>400</td>
</tr>
<tr>
<td>Fp-cLNC-C3</td>
<td>0.5</td>
<td>0.7</td>
<td>900</td>
</tr>
<tr>
<td>Fp-cLNC-C4</td>
<td>0.5</td>
<td>0.7</td>
<td>1666</td>
</tr>
<tr>
<td>SA</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>cLNC-S0</td>
<td>0.25</td>
<td>0.68</td>
<td>0</td>
</tr>
<tr>
<td>Fp-cLNC-S1</td>
<td>0.25</td>
<td>0.68</td>
<td>95</td>
</tr>
<tr>
<td>Fp-cLNC-S2</td>
<td>0.375</td>
<td>0.48</td>
<td>400</td>
</tr>
<tr>
<td>Fp-cLNC-S3</td>
<td>0.99</td>
<td>0.63</td>
<td>900</td>
</tr>
<tr>
<td>Fp-cLNC-S4</td>
<td>0.83</td>
<td>0.45</td>
<td>1100</td>
</tr>
</tbody>
</table>

Note: *The total concentration of Fp (associated and free) in cLNCs dispersion was calculated from the initial weight of Fp (μg) and the final volume of the dispersion (mL).

Abbreviations: CTAB, hexadecyltrimethyl ammonium bromide; Fp, fondaparinux; cNLC, cationic lipid nanocapsules; SA, stearylamine; µg, microgram; mL, milliliter; w/v, weight/volume; w/w, weight/weight.

of Fp solution (Arixtra® 10 mg/0.8 mL, GlaxoSmithKline, Evreux, France) were rapidly injected at the phase inversion zone at a temperature close to 70°C, at the third cycle before the final dilution with cold water. A final Fp concentration ranging from 90 to 1517 μg/mL was obtained in cationic lipid nanocapsules dispersions (Fp-cLNCs). Blank cLNCs with CTAB and SA were also prepared as controls.

In vitro characterisation

Particle size and zeta potential measurements

The cLNCs were analyzed for their average particle size (PS) and the polydispersity index (PI) by photon correlation spectroscopy (PCS) using a Malvern Zetasizer® Nano ZS Series DTS 1060 (Malvern Instruments S.A., Worcestershire, UK) at a fixed angle (173°) at 25°C furnished by a 4 mW He–Ne laser at 633 nm. The polydispersity index (PDI) was used as a measurement of the size distribution (a value of PDI ≤ 0.2 indicates a unimodal size distribution). A 1:60 v/v dilution of the nanoparticle suspensions in deionized water was achieved in order to enable measurements.

PCS data are presented as the mean (volume) intensity-particle size distribution, and standard deviation (SD) of the mean of particle size measurements performed in triplicate was calculated. The samples were kept in polystyrene vials. The shelf stability of Fp-cLNCs at 4°C with respect to particle size was investigated for up to 4 months.

The zeta (ζ) potential and electrophoretic mobility measurements of the different nanoparticles were performed using the same Malvern instrument equipped with an AZ-4 cell and were based on the laser Doppler effect. Zeta potential measurements were made also after similar dilutions (1:60 v/v) in deionized water at 25°C, with a dielectric constant of 79, refractive index of 1.33, viscosity of 0.89 cP, cell voltage 150 V and current of 5 mA. Data from the analysis of ζ potential are presented as mean and SD of triplicate runs.

Drug association efficiency and drug payload

The amount of Fp unentrapped within the cLNCs was determined with a modified Azure A colorimetric method.37 The amount of free dye in the external LNC suspension was indirectly proportional to the amount of unentrapped drug. Typically, aliquots (500 μL) of cLNCs samples were reacted with 1.5 mL of Azure A (Sigma Aldrich, St Quentin-Fallavier, Yvelines, France) solution (0.04 mg/mL) at room temperature followed by ultrafiltration at 4000 rpm for 30 minutes using (Millipore® Centricon YM-100 filters, Millipore Corporation, Billerica, MA) to separate the purple complex of Azure with free Fp as well as the Fp-cLNCs. The error of determination of Fp does not exceed 0.04 μg/mL, and the detection limit is 4 μg/mL. The drug association efficiency was expressed as the percentage of Fp associated with respect to the total drug content.

Prior to using the indirect technique for routine assay purposes, the correlation of our results with the biological anti-Xa activity assay was done to prove that association with cLNCs did not affect its anticoagulant effect.

In vivo experiments

Experiments were carried out according to the French legislation on animal experiments. Male Wistar rats (mean weight 325.8 ± 14.1 g) were fasted for 12 hours. They were anesthetized with isoflurane (0.5%) and divided into...
4 groups: two groups received Fp (200 µg/kg) in isotonic solution by IV bolus injection via the tail vein or by SC injection (dorsal), the two remaining groups received orally Fp solution or Fp-loaded cLNCs suspensions as a single dose with different Fp concentrations. Oral administration was through an oral gavage that was carefully passed down the esophagus into the stomach using a stainless steel catheter with a blunt end to avoid lesions. The Fp control solution and Fp-loaded cLNCs suspensions were prepared in a sodium bicarbonate buffer (1.5 g/100 mL, pH 8.2) to neutralize the gastric acidity as described elsewhere.38 The total orally administered volume of the Fp-loaded cLNCs suspensions was 1.5 to 2.5 mL. The dose amounts were 0.5, 2 and 5 mg/kg. Five rats were included in each group at the beginning of the study: some died during the study due to the sampling procedure (n final = 3–5). Blood (950 µL) was collected serially by capillary intracardiac sampling into pediatric citrated tubes containing a constant volume of sodium citrate (0.109 M). Blood sampling was started at 2 minutes in the case of IV or SC injection and after 15 minutes in the case of oral administration followed by sampling at 30 minutes, 1 hour, 2, 4, 8 and 24 hours. Plasma was collected after centrifugation (4000 rpm, 15 minutes, 4°C) and stored at −20°C until analysis using the chromogenic assay.

Measurement of anti-Xa activity (biological assay)

The anti-Xa chromogenic assay was performed using the Biophen Heparin 6 kit (HYPHEN BioMed, Neuville-sur-Oise, France) on an automated analyzer (STA-R®, STAGO, Asnières-sur-Seine, France) according to the protocol supplied by the manufacturer. In case of results over range, samples were diluted with human normal plasma to obtain Fp concentrations below 0.7 µg/mL.

The anti-Xa activity was measured from the calibration curve of absorbance versus Fp standard concentrations (µg/mL) with STA-fondaparinux calibrator and STA-fondaparinux control kits (Diagnostica Stago, Asnières-sur-Seine, France) according to the protocol supplied by the manufacturer. This assay had a coefficient of variation <7% and a limit of quantification of 0.01 µg/mL. The fondaparinux concentration was directly deduced from a calibration curve.

Determination of pharmacokinetic parameters (PK)

Noncompartmental analysis (EquivTest™, Statistical Solutions Ltd, Saugus, MA) was performed for Fp absorption profiles. The area under the plasma concentration versus time curve (AUC_{0→∞}) was calculated by the trapezoidal method. The absolute bioavailability BA (%) of oral Fp was calculated as the ratio of the AUC after oral and after IV administration with a correction for the difference in dose. Using the plasma concentration time curves after oral and IV administration, relevant PK parameters including mean residence time (MRT), half-life, volume of distribution, total clearance were calculated using the appropriate equations derived from noncompartmental analysis.

Statistical analysis

All studies were repeated a minimum of three times and parameters determined at least in triplicate. Results were reported as means ± SD (SD = standard deviation). Statistical significance was analyzed using the Student’s t-test or Anova (Excel®, Microsoft Corporation, Redmond WA,). Differences between experimental groups were considered significant when the P-value was less than 0.05.

Results

In vitro characterization

Particle size, zeta potential measurements and association efficiency

The mean diameter of Fp-loaded cLNCs prepared with different cationic surfactant ranged from 45.2 ± 2.4 nm to 60.1 ± 2.3 nm with a mean of 53.2 ± 8.9 nm and 47.0 ± 0.8 nm for Fp-CTAB-LNCs and Fp-SA-LNCs respectively, whereas that for the empty cLNCs was 49.8 ± 2.6 nm. A low PDI was shown for all the formulations (<0.22) (Table 2). The particle size increased with the increase in drug loading showing the highest (60.1 ± 2.3 nm) in case of Fp-cLNCs-C_4 but remained unchanged with Fp-SA-LNCs formulations. CTAB concentration range was (0.12%–0.36% w/w) of total cLNCs weight. Higher CTAB and SA concentrations were needed for higher drug concentrations as shown in Table 2. The cLNCs sizes remained almost unchanged after 4 months storage at 4°C. Ratios of initial size versus size after storage for Fp-CTAB-LNCs and Fp-SA-LNCs were as follows: 0.72 ± 0.02 and 0.80 ± 0.06, respectively.

The overall charge of the cLNCs was positive as shown from ζ-potential data, ranging from (3.9 ± 0.4 to 42.9 ± 0.5 mV), showing the highest positive value with Fp-cLNCs-C_4 and the lowest values shown with SA formulations.

Association efficiency for the anionic drug was high for all cLNCs, ranging from 80% to 100% with a mean of 93.5% ± 6.0% and the highest values (100%) with Fp-cLNCs-C_4 and Fp-cLNCs-C_5.
Table 2 In vitro characterization of the cLNC

<table>
<thead>
<tr>
<th>Formulation</th>
<th>Size ± SD (nm)</th>
<th>PDI ± SD</th>
<th>Zeta ± SD (mV)</th>
<th>Fp content by Triton® method (µg/mL)</th>
<th>% AE ± SD*</th>
<th>Drug payload† (µg/g)</th>
</tr>
</thead>
<tbody>
<tr>
<td>LNC (blank)</td>
<td>49.8 ± 2.7</td>
<td>0.12 ± 0.01</td>
<td>−1.2 ± 1.1</td>
<td></td>
<td></td>
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</tr>
<tr>
<td>cLNC-C0 (blank)</td>
<td>45.2 ± 2.4</td>
<td>0.09 ± 0.01</td>
<td>33.0 ± 2.1</td>
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<td></td>
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</tr>
<tr>
<td>Fp-cLNC-C1</td>
<td>55.0 ± 2.5</td>
<td>0.16 ± 0.01</td>
<td>30.3 ± 1.3</td>
<td>90.94 ± 4.72</td>
<td>100 ± 0</td>
<td>0.68</td>
</tr>
<tr>
<td>Fp-cLNC-C2</td>
<td>58.1 ± 0.6</td>
<td>0.08 ± 0.02</td>
<td>33.2 ± 2.3</td>
<td>397.62 ± 20.90</td>
<td>98.18 ± 1.89</td>
<td>2.79</td>
</tr>
<tr>
<td>Fp-cLNC-C3</td>
<td>58.2 ± 1.4</td>
<td>0.14 ± 0.00</td>
<td>42.9 ± 0.5</td>
<td>916.29 ± 56.75</td>
<td>100 ± 0</td>
<td>7.20</td>
</tr>
<tr>
<td>Fp-cLNC-C4</td>
<td>60.1 ± 2.3</td>
<td>0.22 ± 0.01</td>
<td>18.2 ± 1.1</td>
<td>1517.74 ± 74.2</td>
<td>96.48 ± 0.45</td>
<td>13.84</td>
</tr>
<tr>
<td>cLNC-S0 (blank)</td>
<td>46.0 ± 2.4</td>
<td>0.07 ± 0.02</td>
<td>11.9 ± 1.6</td>
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<tr>
<td>Fp-cLNC-S1</td>
<td>46.5 ± 2.1</td>
<td>0.12 ± 0.00</td>
<td>12.5 ± 2.1</td>
<td>95.00 ± 0.01</td>
<td>90.00 ± 0.01</td>
<td>0.67</td>
</tr>
<tr>
<td>Fp-cLNC-S2</td>
<td>46.8 ± 0.6</td>
<td>0.07 ± 0.03</td>
<td>11.4 ± 1.2</td>
<td>395.00 ± 6.85</td>
<td>95.59 ± 0.16</td>
<td>3.06</td>
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<tr>
<td>Fp-cLNC-S3</td>
<td>48.1 ± 2.4</td>
<td>0.15 ± 0.01</td>
<td>4.0 ± 0.4</td>
<td>912.56 ± 34.65</td>
<td>80.20 ± 0.48</td>
<td>5.15</td>
</tr>
<tr>
<td>Fp-cLNC-S4</td>
<td>47.5 ± 2.1</td>
<td>0.06 ± 0.13</td>
<td>6.3 ± 1.4</td>
<td>ND</td>
<td>91.00 ± 0.01††</td>
<td>5.43</td>
</tr>
</tbody>
</table>

Notes: *AE% (% association efficiency) calculation was based on free drug determined by Azure A method and on total drug content (Triton® method); †The drug payload was calculated for each formula from the weight of Fp associated (total – free) to the total dry weight of the cLNCs; ††AE% calculated using the total drug content calculated from weight of Fp in final volume of dispersion as Fp content could not be determined by triton method.

Abbreviations: PDI, polydispersity index; Fp, fondaparinux; LNC, lipid nanocapsules; cLNC, cationic lipid nanocapsules; SD, standard deviation; AE, association efficiency.

In vivo study

Pharmacokinetics of Fp-loaded lipid nanocapsules

Fondaparinux plasma concentration after intravenous administration of a 200 µg/kg dose is shown in Figure 1. A SC injection of a dose equal to the IV dose resulted in an anti-Xa level of 0.836 µg/mL within 60 minutes and 115% ± 17% bioavailability (Table 3). Fp is totally absorbed by the SC route.29 The maximum plasma concentration (Cmax) is reached between 1 and 3 hours.

Oral formulations with CTAB-LNCs had a fast onset of action (30–60 minutes). On the contrary, SA-LNCs showed a much slower action rate (Tmax 90–210 minutes, P<0.05) with the anticoagulant effect lasting up to 16 hours (Figure 2).

The oral administration of Fp solution alone (0.5, 2 mg/kg) did not affect the anti-Xa level in rat plasma (all samples were found to be below the limit of quantification). All failed to attain the therapeutic39 anti-factor Xa levels in rats (0.2 µg/mL).40

![Figure 1](https://www.dovepress.com/)

**Figure 1** Anti Xa activity in plasma versus time of fondaparinux in rats, after IV bolus injection and SC injection of Fp (200 µg/kg, n = 4). **Abbreviations:** MEC, minimum effective therapeutic concentration; IV, intravenous; SC, subcutaneous; µg, microgram; mL, milliliter.
Oral administration of Fp-loaded cLNCs at a 2 mg/kg dose produced a slight increase in the peak plasma concentration of Fp (0.18 μg/mL) with both formulations and reached higher values with higher dose (5 mg/kg), up to 0.98 ± 0.33 μg/mL with CTAB-LNCs and more significantly with SA-LNCs (1.31 ± 0.37 μg/mL, P < 0.05).

The area under the anti-Xa activity in rat plasma versus time curve (AUC_{0→h}) at the dose of 2 mg/kg with CTAB-LNCs and SA-LNCs formula was 78.7 μg·minutes/mL and 42.75 μg·minutes/mL respectively. The lowest dose (0.5 mg/kg) did not affect AUC_{0→h} but the effect was significantly higher with the highest doses especially with SA-LNCs (5 mg/kg) compared to 138.21 μg·minutes/mL for the IV dose (200 μg/kg). The extent of increase of the absolute bioavailability is shown in Figure 3.

In summary, cLNCs significantly increased oral bioavailability of Fp and improved its PK profile in a dose-dependent fashion. It is obvious from the above results that the Fp-loaded cLNCs increased the oral bioavailability of the pentasaccharide significantly with a 5 mg/kg oral dose CTAB formulation (9.75% ± 5.4%, P < 0.05) and more with SA-LNCs (21.18% ± 6.2%, P < 0.05) but non significantly with lower oral doses (2 mg/kg) of the same formulations. Furthermore, the MRT and half-life increased up to three times after oral administration of the different Fp-cLNCs formulations compared to 70.49 minutes and 108.17 minutes respectively after IV administration of Fp solution (Table 3).

The computation of PK parameters with a non-compartmental model for the different groups showed high variations in each group from the mean, especially at the highest doses of Fp-loaded CTAB-LNCs and SA-LNCs (5 mg/kg). Data from SA-LNCs (5 mg/kg) were shown to be less variable with SEM <30% and n = 5. Variability is well recognized in a number of studies conducted in vivo.23,41

**Discussion**

Heparin derivatives (UFH, LMWH, Fp) cannot be administered by the oral route due to a very low bioavailability. New oral direct factor Xa inhibitors have reached the market but do not display the same safety profile as heparin derivatives, especially on the liver. New lipid nanocapsules (LNCs) have shown a great potential to improve the bioavailability of drugs such as paclitaxel23 or Sn38,42 but these drug delivery systems easily entrap only lipophilic and amphiphilic drugs.43 In the present work we studied the possibility of using LNCs to improve the bioavailability of oral Fp, which is a negatively charged hydrophilic molecule. The challenge was to find a way to associate Fp to LNCs and then to obtain therapeutic anti-Xa activity after oral administration of the new formulations.

<table>
<thead>
<tr>
<th>Study (Unit/dose per os)</th>
<th>Cmax (μg/mL)</th>
<th>Tmax (minutes)</th>
<th>AUC_{0→∞} (μg·minute/mL)</th>
<th>T_{1/2} (minutes)</th>
<th>MRT (minutes)</th>
<th>Clearance (μL/minute)</th>
<th>Absolute bioavailability (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>CTAB formulations</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Fp-cLNC-C1</td>
<td>2 mg/kg</td>
<td>0.18 ± 0.10</td>
<td>60</td>
<td>78.72 ± 74.41</td>
<td>248.25 ± 165.67</td>
<td>349.23 ± 210.21</td>
<td>467.63 ± 10.17</td>
</tr>
<tr>
<td>Fp-cLNC-C1</td>
<td>5 mg/kg</td>
<td>0.98 ± 0.33</td>
<td>30</td>
<td>336.48 ± 187.32</td>
<td>177.10 ± 82.62</td>
<td>255.53 ± 124.22</td>
<td>450.72 ± 7.09</td>
</tr>
<tr>
<td>SA formulations</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Fp-cLNC-S1</td>
<td>2 mg/kg</td>
<td>0.18 ± 0.96</td>
<td>90</td>
<td>42.75 ± 28.47</td>
<td>204.13 ± 149.41</td>
<td>239.80 ± 112.41</td>
<td>465.94 ± 5.85</td>
</tr>
<tr>
<td>Fp-cLNC-S2</td>
<td>5 mg/kg</td>
<td>1.31 ± 0.37</td>
<td>210</td>
<td>730.89 ± 214.89</td>
<td>188.54 ± 9.69</td>
<td>338.44 ± 69.44</td>
<td>485.86 ± 13.76</td>
</tr>
<tr>
<td>Control oral Fp solution</td>
<td>5 mg/kg</td>
<td>0.06 ± 0.02</td>
<td>120</td>
<td>48.70 ± 37.16</td>
<td>51.7 ± 36.43</td>
<td>81.39 ± 58.69</td>
<td>455.43 ± 14.51</td>
</tr>
<tr>
<td>IV Fp solution</td>
<td>200 μg/kg</td>
<td>1.42 ± 0.12</td>
<td>2</td>
<td>138.21 ± 46.32</td>
<td>70.49 ± 17.63</td>
<td>108.17 ± 26.07</td>
<td>490.52 ± 14.74</td>
</tr>
<tr>
<td>SC Fp solution</td>
<td>200 μg/kg</td>
<td>0.836 ± 0.19</td>
<td>60</td>
<td>154.01 ± 23.92</td>
<td>88.46 ± 83.94</td>
<td>232.21 ± 78.66</td>
<td>493.89 ± 5.478</td>
</tr>
</tbody>
</table>

Notes: AUC_{0→∞} = AUC_{0→t}; the bioavailability was calculated from the dose corrected areas under the curves for oral or subcutaneous versus IV administration (EquivTest/ PK software); n = 3–5 (the final number of surviving animals).

Abbreviations: CTAB, hexadecyltrimethyl ammonium bromide; Fp, fondaparinux; cLNC, cationic lipid nanocapsule; IV, intravenous; SC, subcutaneous; SA, stearylamine; os, MRT, mean residence time; AUC, area under curve; μg, microgram; mL, milliliter; µL, microliter; mg, milligram; kg, kilogram.
Figure 2 Anti-Xa activity in plasma versus time of fondaparinux in rats after oral administration of LNCs formulations (A) = CTAB-LNCs, (B) = SA-LNCs, different Fp doses (0.5, 2 or 5 mg/kg).

Notes: Data are shown as the mean concentration, and error bars represent the SEM (n = 3–5).

Abbreviations: IU, international unit; mL, milliliter; mg, milligram; kg, kilogram; CTAB, hexadecyltrimethyl ammonium bromide; LNC, lipid nanocapsule; SA, stearylamine; Fp, fondaparinux; SEM, standard error mean.
To successfully entrap Fp, introduction of a positively charged molecule into the structure of LNCs was conducted in order to form ion pairs with the polyanionic Fp at the water/oil interface. Using CTAB or SA in the primary emulsion did not prevent the phase inversion phenomenon, which is a key step in producing LNCs. In another parallel study, the LNC surface was positively charged using a layer-by-layer approach with lipochitosan as the cationic polymer.44 Other authors reported similar strategies for other anionic molecules.45

Blank LNCs have an average diameter (~50 nm) based on the respective proportions of their components.35 CTAB-LNCs average particle diameter was increased slightly (~60 nm) with an increase in drug loading (Table 2). This could be attributed to the association of the drug on the surface of the cLNCs. The positively charged quaternary ammonium groups (QA) of CTAB could be mainly directed toward the external aqueous phase, thus leading to the increase in the mean diameter, with the increase in drug loading. In SA formulations, the particle size remained more or less constant (47 nm) although the proportions of total surfactants were higher than that in CTAB formulations (5.88%–7.27% w/w compared to 5.41% w/w respectively) and even more if molecular weight is taken into account. This could be due to reduced interaction of Fp with the primary amine groups of SA as reflected by lower association efficiency (84%–90%).

The ζ potential was changed for the cLNCs when compared to blank LNCs (~1 mV), being more positive in the case of CTAB-LNCs (+32.0 ± 1.7 mV) than in the case of SA-LNCs (+11.9 ± 0.5 mV). This is due to the presence of the cationic agent in its ionized form at the oil/water interface.46 The magnitude of the ζ potential decreased in a Fp concentration-dependent manner due to neutralization. Whether Fp was located on the surface of the cLNCs or in the core has to be further investigated. The steric effect of SA or CTAB or Fp chains at the surface of the complex can significantly improve stability and prevent aggregation of the cLNCs even when the ζ potential is less than 10 mV (eg, SA-LNCs). Similar phenomena were also observed by other groups.47

The toxicity of CTAB and SA is well defined (rat LD50 oral 410 and 1395 mg/kg respectively) and their concentration can be kept to a minimum in the formulations (0.45 and 0.92 mg/kg LNC) respectively, corresponding to doses of 30 and 90 mg/kg in rats. This is lower than a tenth of the LD50. Moreover, although CTAB was chosen as the initial surfactant, the work with SA clearly shows that CTAB can be replaced, if necessary, by safer single-tailed cationic lipid moieties without reducing the potency of the delivery system. However, the toxicity of these new cationic nanocapsules remains to be determined, as it was previously assessed with standard LNCs containing only Solutol HS15 as a surfactant.27

Fondaparinux, like Hp, is presumed unstable under acidic enzymatic conditions.48 Few studies support the observations that Hps are absorbed by the gastrointestinal tract but they do not reach therapeutic concentration.49 Active transport is involved and is pH-dependent, preferably around pH = 7.50 Some in vivo studies of UFH and LMWHs were carried out using in situ intestinal administration.51,52 Subsequently, most were carried out by oral gavage where promising results were obtained53 and some even extended to clinical trials.54 In the present study, cLNCs have been prepared using bicarbonate buffer, resulting in dispersions having a final mean pH of 8.5 ± 0.5 for CTAB and SA formulations. Bicarbonate was also added to the control formulation (free Fp). They were administered in doses that are somewhat higher and much higher than the IV or SC doses for Fp, to compensate for the detrimental effects of the oral route (eg, degradation and dilution).55 Fp reached therapeutic concentration in plasma only if it was associated to LNCs (Figure 2A and B), but not if administered as free Fp at a dose of 5 mg/kg, thus highlighting the role of cLNCs in facilitating Fp absorption and stability.

Linear PKs were reported in the literature for Hp, LMWH and Fp after SC injection.59 Fp-CTAB-LNCs and Fp-SA-LNCs gave the same responses for low oral doses of 2 mg/kg: Cmax and percentage bioavailability did not differ (P > 0.05) although CTAB formulations were more positive (30 mV) compared to SA formulations (11 mV).
In the high dose group (5 mg/kg), a small positive charge on the SA-LNCs formulation (3 mV) gave a significantly better pharmacokinetics profile than that of CTAB-LNCs formulations (42 mV) (Table 3), and further resulted in a significant increase in AUC (17-fold, \( P < 0.05 \)). Bioavailability also significantly increased (up to 21%) (Figure 3). These results suggest that the influence of surface charge on bioavailability is complex for Fp-LNCs. In fact, a high positive charge is often claimed to improve nanoparticle absorption but could on the other hand limit mucus diffusion.\(^5\)

The pharmacokinetic parameters, such as half-life and mean residence time (MRT), were longer for Fp-cLNCs formulations after oral administration compared to Fp control solutions (oral and IV) but clearance was not significantly different (Table 3). These results suggest a long circulating time in vivo with sustained release of the drug resulting in better bioavailability while the elimination of the drug remained constant whatever the formulation or route of administration used. This could be helpful in providing a sustained anti-Xa effect. However, further studies are necessary to explore any possible permeation of intact LNCs across the GI membrane or its distribution to lipid tissues acting as storage compartment in order to explain the observed PK parameters.

In a recent study using another oral delivery system as a permeation enhancer for fondaparinux, based on thiolated polycarbophil (PCP-Cys) and glutathione (GSH) combined with sodium decanoate, only 4.4% bioavailability was reported using tablets and/or gel formulations containing doses of 5 mg/kg Fp.\(^10\) The bioavailability found with cLNCs was higher (up to 21%).

Is oral Fp better than the new generation of oral anti-factor Xa and IIa inhibitors such as dabigatran etexilate (DTI), a promising novel, small molecule that specifically and reversibly inhibits thrombin?\(^57\) Interestingly, Fp-loaded cLNCs gave an anti-factor Xa effect (Cmax >0.1 \( \mu \)g/mL) within 2 hours (Tmax) with sustained release properties. Could oral Fp compete with these agents, having the same advantages and including: no requirement for anticoagulant monitoring; a low drug–drug interaction potential; and the possibility of use in both acute and chronic settings. Further studies and comparisons should be carried out to provide a better therapeutic outcome.

**Conclusion**

This study is considered an extension of the applicability of lipid nanocapsules as an oral delivery system. In the present case, lipid nanocapsule technology was used for encapsulation of a high therapeutic value drug, fondaparinux. Fp-loaded cLNCs demonstrated potential for a formulation increasing both the oral bioavailability and residence time of the indirect anti-factor Xa inhibitor. Oral Fp could thus be envisioned as an alternative to currently used subcutaneous delivery. This first proof of concept of enhanced oral bioavailability of fondaparinux by nanocarriers may be of great clinical use. This new simple formulation strategy for an oral treatment may replace the currently used injectable anticoagulants, to improve patient acceptability and therapeutic effects, facilitating the development of chronic treatment schedules without the bleeding side effects and the coagulation monitoring required with vitamin K antagonist therapy.

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**Disclosure**

The authors have no other relevant affiliations or financial involvement with any organization or entity with a financial interest in or financial conflict with the subject matter or materials discussed in the manuscript apart from those disclosed.

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