Targeted delivery of doxorubicin-utilizing chitosan nanoparticles surface-functionalized with anti-Her2 trastuzumab

Parisa Yousefpour1
Fatemeh Atyabi2
Ebrahim Vasheghani-Farahani3
Ali-Akbar Mousavi Movahedi1
Rassoul Dinarvand2

1Department of Biotechnology, Faculty of Science, University of Tehran, 2Nanotechnology Research Centre, Faculty of Pharmacy, Tehran University of Medical Sciences, 3Biotechnology Group, Department of Chemical Engineering Faculty of Engineering, Tarbiat Modares University, Tehran, Iran

Background: Targeting drugs to their sites of action to overcome the systemic side effects associated with most antineoplastic agents is still a major challenge in pharmaceutical research. In this study, the monoclonal antibody, trastuzumab, was used as a targeting agent in nanoparticles carrying the antitumor drug, doxorubicin, specifically to its site of action.

Methods: Chitosan-doxorubicin conjugation was carried out using succinic anhydride as a crosslinker. Trastuzumab was conjugated to self-assembled chitosan-doxorubicin conjugate (CS-DOX) nanoparticles (particle size, 200 nm) via thiolation of lysine residues and subsequent linking of the resulted thiols to chitosan. Conjugation was confirmed by gel permeation chromatography, differential scanning calorimetry, Fourier transform infrared spectroscopy, and 1H nuclear magnetic resonance spectroscopy studies. Dynamic light scattering, transmission electron microscopy, and zeta potential determination were used to characterize the nanoparticles.

Results: CS-DOX conjugated nanoparticles had a spherical shape and smooth surface with a narrow size distribution and core-shell structure. Increasing the ratio of doxorubicin to chitosan in the conjugation reaction gave rise to a higher doxorubicin content but lower conjugation efficiency. Trastuzumab-decorated nanoparticles (CS-DOX-mAb) contained 47 μg/mg doxorubicin and 33.5 μg/mg trastuzumab. Binding of trastuzumab to the nanoparticles was further probed thermodynamically by isothermal titration calorimetry. Fluorescence microscopy demonstrated enhanced and selective uptake of CS-DOX-mAb by Her2+ cancer cells compared with nontargeted CS-DOX nanoparticles and free drug.

Conclusion: Antibody-conjugated nanoparticles were shown to discriminate between Her2+ and Her2− cells, and thus have the potential to be used in active targeted drug delivery, with reduction of drug side effects in Her2+ breast and ovarian cancers.

Keywords: chitosan, doxorubicin, self-assembled nanoparticles, active targeting, trastuzumab

Introduction

The main objective in anticancer drug development is to deliver therapeutic agents in a targeted and selective fashion to their site of action, and to decrease adverse effects and enhance efficacy. Over recent years, nanoparticulate carrier systems have aroused ever increasing interest in this area.1,2 These targeted nanosystems can deliver drugs in a passive or active way. Passive targeted drug delivery takes advantage of the poor lymphatic systems of tumor tissues and their leaky vasculature with pore sizes ranging from 100 to 780 nm.3-5 These characteristics enable what is called the “enhanced permeability and retention” effect, which allows enhanced deposition of delivery nanovehicles at the site of a solid tumor. Active targeted drug delivery on the other hand is achieved via covalent conjugation of targeting molecules on the nanoparticle
surface which can recognize and bind to specific ligands expressed specifically in cancer cells. One such ligand is human epidermal growth factor receptor 2 (Her2) or ErbB2 (Neu), the expression of which is amplified in about 30% of breast cancers and 20% of ovarian cancers, and this receptor is expressed weakly in normal adult tissues.5-8

Trastuzumab (Herceptin®) is a humanized monoclonal antibody directed against the Her2 receptor, and is the only Her2-targeted therapy approved by the US Food and Drug Administration for the treatment of advanced breast cancer. Combination of trastuzumab with conventional chemotherapy leads to increased response rates in comparison with trastuzumab alone.9-11 In addition, according to some clinical trials, anthracycline-based chemotherapy demonstrated more successful results in Her2+ women.12 However, trastuzumab has been shown to aggravate anthracycline-induced cardiotoxicity, and thus cannot be given concomitantly with anthracyclines, including doxorubicin.13,14 Conjugation of trastuzumab to doxorubicin-carrying nanoparticles allows transport of the chemotherapeutic agent specifically to tumor cells and reduced their adverse cardiotoxic effects. Furthermore, in such nanoparticulate formulations, trastuzumab is intended to act as a targeting ligand rather than as a therapeutic agent and thus its concentration is far below its therapeutic dose. Previous studies have shown promising results for either cancer therapy or imaging via trastuzumab decoration of such nanoparticles as dextran iron oxide nanoparticles,15 poly(d,l-lactide-co-glycolide)/montmorillonite nanoparticles,16 poly(d,l-lactic acid) nanoparticles,17 and human serum albumin nanoparticles.18-20

Chitosan is a carbohydrate polymer with the desirable properties of biodegradability and biocompatibility that have made it a candidate polymer for preparation of drug delivery carriers.21-29 Several methods have been developed for the preparation of doxorubicin delivery systems based on chitosan,30 which include dextran sulfate-chitosan hydrogel nanoparticles, glycol-chitosan nanoaggregates, oleoyl-chitosan nanoparticles, chitosan-poly(acrylic acid) hollow nanospheres, and stearic acid-grafted chitosan oligosaccharide micelles.

However, in targeted delivery systems, covalent conjugation of drug to its carrier is more advantageous than drug encapsulation because it prevents premature drug release into the blood circulation before its delivery to the target site. In this study, chitosan-doxorubicin conjugate (CS-DOX) nanoaggregates were prepared via covalent conjugation of doxorubicin to chitosan. Trastuzumab was conjugated to the nanoaggregates, and the efficacy of the resulting actively targeted nanocarriers was studied in vitro.

Methods and materials

Materials

Doxorubicin (purity approximately 98.5%) was purchased from RPG Life Sciences Ltd (Ankleshwar, India). Chitosan, with a medium molecular weight and deacetylation of about 96%, was supplied by Fluka, Germany. Sodium nitrite, hydrochloric acid, glacial acetic acid, sodium hydroxide, succinic anhydride, 1-ethyl-3-(3-dimethyl amino-propyl) carbodiimide hydrochloride (EDC), N-hydroxysuccinimide (NHS), acetonitrile, triethylamine, and ethyl acetate and chloroform (analytical grade) were obtained from Merck, Darmstadt, Germany. Total protein kit (product code TP0200) and sulfosuccinimidyl 4-(N-maleimidomethyl) cyclohexane-1-carboxylate were purchased from Sigma (St Louis, MO). Trastuzumab was purchased from Roche, Mannheim, Germany. Cell lines were provided by the Pasteur Institute, Tehran, Iran. All other chemicals were of analytical grade. Deionized water was used throughout.

Conjugation of doxorubicin to chitosan

CS-DOX conjugates were synthesized using succinic anhydride as a spacer. Succinic anhydride was employed to react with and convert the amine group of doxorubicin to carboxylic residues, ie, succinic acid residues. The resulting succinyl doxorubicin (SDOX) was then introduced to chitosan via amide bond formation mediated by EDC and NHS.

At first, 120 kDa chitosan was prepared according to the depolymerization method described by Peniston and Johnson.31 Briefly, 10 mL nitrite sodium solution (1 mg/mL) was added to 100 mL 2% (w/v) chitosan solution in 6% (v/v) acetic acid. The depolymerization reaction was allowed to proceed for 1 hour while stirring and was then stopped by raising the pH to 9 using 5 N NaOH. The precipitated white-yellowish chitosan was then filtered and washed thoroughly with acetone. The filtrate was redissolved in a minimum volume of acetic acid 0.1 N and was dialyzed against deionized water (2 × 2 L for 90 minutes and 1 × 2 L overnight). The dialyzed product was lyophilized at 50°C and 0.01 mbar (LyoTrap plus, LTE Scientific, Chorley, UK).

To prepare SDOX, 40 mg doxorubicin HCl was dissolved in dry acetonitrile to which 70 µL triethylamine and 690 mg succinic anhydride in 3 mL dry acetonitrile was added. The reaction was allowed to complete by 15 hours stirring at 4°C in the dark. Afterwards, the solution was distributed between 10 mL sodium bicarbonate 5% (w/v) solution and 40 mL chloroform. The chloroform phase was decanted and the residual solution was extracted by ethyl acetate after lowering the pH using 1 M HCl. Ethyl acetate was evaporated by a rotary evaporator to obtain SDOX.
For covalent conjugation of SDOX to chitosan, 200 mg chitosan was hydrated in 2 mL of 1 M HCl. Deionized water was added to give a final chitosan concentration of 1% (w/v). SDOX in 2% sodium bicarbonate was added to obtain an SDOX/CS (w/w) ratio of 20%, 10%, 5%, 2.5%, and 1%, followed by addition of EDC and NHS (5 molar equivalents with respect to SDOX). pH was adjusted to 6.6 using 5 N NaOH and the reaction was allowed to stir for 36 hours at room temperature. Afterwards, unreacted components were removed by extensive dialysis against deionized water. CS-DOX was concentrated via centrifugation with 30 kDa cutoff centrifugal ultrafilters (Millipore Corporation, Billerica, MA) at 4000 g and 10°C for 15 minutes. The CS-DOX were stored at 4°C until further use (Figure 1).

Conjugates with an initial CS-DOX (w/w) ratio of 20%, 10%, 5%, 2.5%, and 1% were named as CS-DOX-1, CS-DOX-2, CS-DOX-3, CS-DOX-4, and CS-DOX-5, respectively.

**Gel permeation chromatography**

The molecular weight of the depolymerized chitosan was determined by gel permeation chromatography. A PL Aquagel-OH mixed gel filtration column (300 mm × 7.5 mm internal diameter, pore size 8 µm) from Agilent Technologies, Santa Clara, CA, was used. All chromatograms were generated on an Agilent 1100 liquid chromatographer (Agilent Technologies), and the eluting fraction was monitored using a refractive index signal detector. The lyophilized powder of depolymerized chitosan was dissolved in 300 mM acetate buffer, pH 4.5, with a final

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**Figure 1** Schematic illustration of the procedures for the synthesis of chitosan-doxorubicin conjugate.

**Abbreviations:** EDC, 1-ethyl-3-(3-dimethyl amino-propyl) carbodiimide hydrochloride; NHS, N-hydroxsuccinimide.
concentration of 3 mg/mL, and was chromatographed at a flow rate of 5 mL/min. Chromatograms were generated and analyzed by EZChrom Elite software (Version 3.1, Scientific Software, Pleasanton, CA) using the narrow method. Gel permeation chromatography analysis was also carried out to examine covalent conjugation of doxorubicin to chitosan.

Fourier transform infrared spectroscopy
To study the chemical reactions between doxorubicin and succinic anhydride, doxorubicin and rotary-dried SDOX were pressed into KBr discs and examined by Fourier transform infrared spectroscopy using a Nicolet spectrometer (Nicolet Magna IR 550, Madison, WI).

^1H nuclear magnetic resonance spectroscopy
^1H nuclear magnetic resonance (NMR) spectra were recorded on a Bruker FT-80 spectrometer (Bruker, Rheinstetten, Germany) and chemical shifts (δ) are in parts per million (ppm) relative to internal tetramethylsilane. To this end, CDCl₃ was used as the solvent for SDOX while doxorubicin, chitosan, and CS-DOX were dissolved in D₂O.

Differential scanning calorimetry
The thermal behavior of CS-DOX was studied by differential scanning calorimetry using a 822e differential scanning calorimeter (Mettler-Toledo, Greifensee, Switzerland). Samples were purged with dry nitrogen at a flow rate of 10 mL/min and were heated in the temperature range 25°C–450°C at a heating rate of 10°C/min.

Preparation of self-assembled nanoparticles
CS-DOX conjugates were dissolved in deionized water at a concentration of 5 mg/mL, and were gently stirred for one hour at room temperature. In addition, effect of concentration of conjugate was investigated by preparation of self-assembled nanoparticles using different concentrations of CS-DOX-2 (0.25, 1, 5, 10, and 20 mg/mL).

Determination of conjugation efficiency and drug content
Five hundred microliters of concentrated conjugates were placed on cover slips and dried in an oven (Memmert INB 400, Schwabach, Germany) at 50°C for three hours. The cover slips were weighed afterwards, and knowing the initial weight of the cover slip, the mg of the conjugates was calculated. Quantification of doxorubicin was achieved via spectrophotometry at 480 nm (Scinco S-3100, Seoul, Korea). The conjugation efficiency and doxorubicin content of the CS-DOX conjugates was determined applying the following equations:

Conjugation efficiency (%) = \( \frac{\text{weight of drug found conjugated}}{\text{weight of total drug used}} \times 100 \)

Drug content (% w/w) = \( \frac{\text{weight of drug found conjugated}}{\text{weight of conjugate}} \times 100 \)

Measurement of particle size, polydispersity, and zeta potential
The hydrodynamic mean diameter, polydispersity index, and zeta potential of the nanoparticles were measured by dynamic light scattering and laser Doppler electrophoresis using a Zetasizer (Nano-ZS, Malvern, Worcestershire, UK). All dynamic light scattering measurements were performed at a wavelength of 633 nm at 25°C with an angle detection of 90°. All measurements were done in triplicate.

Scanning electron microscopy
The surface morphology of the nanoparticles was imaged using a scanning electron microscope (XL 30, Philips, Eindhoven, the Netherlands). One drop of the nanoparticle preparation was layered on the scanning electron microscope stub and allowed to dry in air at room temperature. The dried nanoparticles were then coated with gold metal using a sputter coater (SCD 005, Bal-Tec, Balzers, Switzerland).

Transmission electron microscopy
A transmission electron microscope (CEM 902A, Zeiss, Oberkochen, Germany) was used to get a better idea of the inner structure of the self-assembled nanoparticles. The nanoparticle suspension was deposited on copper grids, subsequently negatively stained using 0.5% phosphotungstic acid, air dried, and viewed under the transmission electron microscope.

Thiolation of trastuzumab and quantification of immobilized thiol groups
Trastuzumab thiolation was carried out using a 100-fold molar excess of 2-iminothiolane (Traut's reagent). The trastuzumab monoclonal antibody was prepared in 50 mM phosphate buffer (pH 8) containing NaCl 0.15 M. To each 5 mL of trastuzumab solution (2 mg/mL), 458 µL Traut’s reagent (2 mg/mL) was added. The thiolation reaction was stirred gently for 2 hours at room temperature. Unreacted Traut’s reagent was excluded by centrifugation three times with 30 kDa cutoff centrifugal
ultrafilters (Millipore Corporation) at 4000 g and 10°C for 15 minutes.

The number of thiol groups added to the monoclonal antibody was subsequently dosed by Ellman’s test as follows. The Ellman reagent, with a final concentration of 10 mM, was dissolved in 0.1 M phosphate buffer (pH 8). Ellman’s reagent 100 µL was added to 400 µL of 2 mg/mL monoclonal antibody solution in phosphate buffer (0.1 M, pH 8). After 15 minutes of incubation, the absorbance was measured at 450 nm using a spectrophotometer (Scinco S-3100). A standard calibration curve was obtained from 10–200 µM N-acetyl L-cysteine.

Preparation of trastuzumab-modified nanoparticles

CS-DOX-2 nanoparticles were dispersed in phosphate buffer (0.1 M, pH 7.2) at a concentration of 5 mg/mL and were activated using 2 mg of the heterobifunctional crosslinker, sulfo succinimidyl 4-(N-maleimidomethyl) cyclohexane-1-carboxylate. After 3 hours of incubation at room temperature, sulphydryl-reactive nanoparticles were purified by centrifugation (4000 g, 10°C, 15 minutes) using 100 kDa cutoff centrifugal ultrafilters (Millipore Corporation). The coupling reaction was achieved by adding 500 µL of thiolated trastuzumab 2 mg/mL to 4 mL nanoparticle suspension 5 mg/mL, followed by three hours of incubation at room temperature. The trastuzumab-modified nanoparticles (CS-DOX-mAb) were purified by centrifugation (26,000 rpm, 10°C, 30 minutes) and nonconjugated antibody in supernatant was quantified with using the Total Protein Kit (Micro Lowry, Onishi and Barr modification, Sigma).

Isothermal titration calorimetry study

Isothermal titration calorimetry experiments were conducted using an isothermal titration calorimetry instrument (VP-ITC, MicroCal, Northampton, MA). The titration vessel was made from stainless steel. Trastuzumab solution (39 nM) was injected using a Hamilton syringe into the calorimetric vessel which contained 1.8 mL protein A 8.86 nM. The coupling reaction was achieved by adding 500 µL of thiolated trastuzumab 2 mg/mL to 4 mL nanoparticle suspension 5 mg/mL, followed by three hours of incubation at room temperature. The trastuzumab-modified nanoparticles (CS-DOX-mAb) were purified by centrifugation (26,000 rpm, 10°C, 30 minutes) and nonconjugated antibody in supernatant was quantified with using the Total Protein Kit (Micro Lowry, Onishi and Barr modification, Sigma).

Cell lines and culture conditions

The Her2+ human breast cancer cell line (MCF7) and Her2+ ovarian carcinoma cell line (SKOV-3) were obtained from National Cell Bank of Iran (Pasteur Institute). Cells were maintained in RPMI-1640 culture medium supplemented with fetal bovine serum 10%, penicillin 100 units/mL, and streptomycin 100 ng/mL at 37°C in a 5% CO₂ humidified incubator (New Brunswick Scientific, Edison, NJ).

In vitro cytotoxicity assay

The cytotoxicity of CS-DOX-2, CS-DOX-mAb, and free doxorubicin against SKOV-3 cells was evaluated using the tetrazolium-based colorimetric (MTT) assay. The tested concentrations were 8, 40, and 200 nM doxorubicin equivalents. Cytotoxicity was then measured after 48 hours of exposure using the MTT method.

Specificity of trastuzumab-modified nanoparticles

The specificity of the trastuzumab-modified nanoparticles was investigated by comparing cellular uptake of CS-DOX-mAb into Her2-overexpressing SKOV-3 cells and Her2+ MCF-7 cells. Cells were seeded in eight-well chamber slides and treated with CS-DOX-2, CS-DOX-mAb, and free doxorubicin at a concentration of 200 nM doxorubicin equivalents. After 24 hours of incubation at 37°C in a humidified 5% CO₂ atmosphere, the medium was removed, the cells were fixed with methanol/acetone (v/v 9:1) at 4°C for 30 minutes, rinsed in phosphate-buffered saline, and observed under a fluorescence microscope with a green filter (λex 490 nm and λem 520 nm) and 1000 msec exposure time (BX40, Olympus, Tokyo, Japan).

Results

Drug content and conjugation efficiency

Doxorubicin was conjugated to chitosan using a succinic anhydride crosslinker. The drug content and conjugation efficiency of the prepared conjugates are shown in Table 1.

Fourier transform infrared and 1H NMR spectroscopic study

Fourier transform infrared and 1H NMR spectra of doxorubicin, SDOX, and CS-DOX conjugates are depicted in Figure 2A, 2B, and 2C. Observed signals for 1H-NMR of doxorubicin were:

1H NMR (DMSO-d6): 7.90- 7.92 (M, 2H, H₃, H₄), 7.64 (d, J = 7.2, 1H, H₂), 5.30 (S,1H, H₁), 4.93 (bs, 1H, H₁), 4.58 (S, 2H, CO- CH₂- OH), 4.17 (m, 1H), 3.93 (S, 3H, OCH₃), 2.32 (1H), 2.92 (m, 2H), 2.14 (m, 2H), 1.92 (m, 1H), 1.69 (m, 1H), 1.18 (d, 3H, CH₃).
Calorimetric study of CS-DOX conjugation

Thermograms of doxorubicin, chitosan, and CS-DOX are presented in Figure 3. A sharp endothermic peak located at 100°C was seen in all differential scanning calorimetry diagrams. The differential scanning calorimetry thermogram of chitosan showed another endothermic peak at 180°C and an exothermic peak at 370°C. Doxorubicin was characterized by a relatively sharp endothermic peak at 200°C followed by two other endothermic peaks at higher temperatures. The differential scanning calorimetry curve for the CS-DOX conjugates exhibited an endotherm at 160°C and an exotherm at 340°C.

Gel permeation chromatograms

Gel permeation chromatograms of chitosan and CS-DOX-2 are shown in Figure 4. Analysis of the resulting chromatograms using a narrow method revealed the molecular weights of chitosan and CS-DOX-2 to be 115.644 and 119.937 kDa, respectively.

Size, zeta potential, and polydispersity index of CS-DOX nanoparticles

All the prepared CS-DOX conjugated but CS-DOX-1 self-assembled into particles, the size, zeta potential, and
polydispersity indexes for which are given in Table 2. In addition, the effect of conjugate concentration on size of the prepared nanoparticles was investigated for CS-DOX-2, and the resulting data are shown in Figure 5.

Surface and inner morphology of nanoparticles

Scanning electron microscopic images as depicted in Figure 6A revealed spherical and smooth-surfaced nanoparticles with a uniform size distribution. In addition, according to the transmission electron microscopic images, the inner morphology of CS-DOX self-assembled nanoparticles consists of a core-shell structure (Figure 6B).

Trastuzumab coupling to CS-DOX nanoparticles

Thiol groups required for the coupling reaction were added to trastuzumab by use of 2-iminothiolane. According to the Ellman’s test, 12.33 reactive thiol groups were linked to the lysine residues of the monoclonal antibody, which is equal to 13.7% of the lysine residues available. Furthermore, under the experimental conditions used, 67% of the monoclonal antibody molecules were attached to CS-DOX-2 nanoparticles. Each gram of the CS-DOX-mAb formulation obtained thus consisted of 47 µg doxorubicin and 33.5 µg trastuzumab.

Interaction of nanoparticle-conjugated trastuzumab with protein A

Conjugation of the monoclonal antibody to the nanoparticles was analyzed by isothermal titration calorimetry taking advantage of the ability of the Fc fragment of an immunoglobulin molecule to interact with protein A. Isothermal titration calorimetry thermograms of protein A with free antibody and nanoparticle-conjugated monoclonal antibody are shown in Figure 7.

According to data analyzed by MicroCal Origin 7.0 software, the free monoclonal antibody-protein A interaction fitted well into a one-binding site model, while the conjugated monoclonal antibody-protein A interaction resulted in a minimum Chi-square value when the two-binding site model was applied. Thermodynamic parameters derived from corresponding models have been summarized in Table 3. The

<table>
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<tr>
<th>PDI</th>
<th>Zeta potential (mV)</th>
<th>Size (nm)</th>
<th>Drug content</th>
<th>Conjugate</th>
</tr>
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<tbody>
<tr>
<td>0.108 ± 0.054</td>
<td>16.5 ± 3</td>
<td>228 ± 22</td>
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<td>CS-DOX-2</td>
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<tr>
<td>0.173 ± 0.041</td>
<td>17.5 ± 4</td>
<td>346 ± 37</td>
<td>2.37%</td>
<td>CS-DOX-3</td>
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<tr>
<td>0.083 ± 0.047</td>
<td>14.3 ± 1</td>
<td>493 ± 26</td>
<td>1.55%</td>
<td>CS-DOX-4</td>
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<tr>
<td>0.253 ± 0.023</td>
<td>18.1 ± 4</td>
<td>671 ± 18</td>
<td>0.69%</td>
<td>CS-DOX-5</td>
</tr>
</tbody>
</table>

Abbreviations: CS-DOX, chitosan-doxorubicin conjugate; PDI, polydispersity index.

Table 2 Size, zeta potential, and polydispersity index of nanoaggregates prepared from different chitosan-doxorubicin conjugates
affinity constant ($K_a$), enthalpy changes ($\Delta H$), and binding stoichiometry ($n$) of the interaction were used to determine the change in Gibb's free energy ($\Delta G$) and change in entropy ($\Delta S$) for each interaction event.

**Cytotoxicity assay**
The results of the cell viability MTT assay as presented graphically in histograms demonstrated increased cytotoxic activity of CS-DOX-mAb compared with CS-DOX-2. However, the growth inhibitory activity of doxorubicin in the CS-DOX-mAb formulation was reduced with respect to the equivalent concentration of free drug. Such differences were slightly more apparent for higher doses of drug (Figure 8).

**Cellular uptake**
Selective uptake of trastuzumab-modified nanoparticles by Her2$^+$ cells was investigated by fluorescence microscopy utilizing the red fluorescence of doxorubicin. CS-DOX-mAb showed significantly higher entry into the Her2-overexpressing cell line, SKOV3, compared with the control MCF-7 cell line. Nontargeted nanoparticles, on the other hand, showed no obvious difference in this regard (Figure 9).

**Discussion**
This study aimed to develop a targeted delivery system for delivery of the chemotherapeutic drug, doxorubicin, to Her2-overexpressing cancer cells. CS-DOX conjugates were synthesized via carboxylation of doxorubicin and subsequent amidation of SDOX with amine groups on chitosan.
Table 3 Interaction thermodynamic parameters of protein A with free and attached antibody

<table>
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<tr>
<th>Nanoparticle-conjugated monoclonal antibody</th>
<th>Free monoclonal antibody</th>
<th>Parameter</th>
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<tr>
<td>Second binding site</td>
<td>First binding site</td>
<td></td>
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<td>1.07 ± 0.261</td>
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<td>2.78 ± 0.469</td>
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<td>203.6</td>
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<td>+68.1</td>
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<td>-8.664</td>
<td>-10.162</td>
<td>-6.96</td>
</tr>
<tr>
<td>+0.318</td>
<td>-0.281</td>
<td>-0.330</td>
</tr>
</tbody>
</table>

Notes: ^binding stoichiometry; ^affinity constant; ^molar enthalpy change for the binding; ^molar free energy change; ^molar entropy change.

The appearance of peaks at 2.9 ppm and 3.3 ppm in the 1H NMR spectrum for SDOX indicates the presence of a -CH2-CH2 group in addition to peaks belonging to the protons of doxorubicin (Figure 2B). Furthermore, appearance of new bands at around 1689 cm^-1 which related to a carbonyl group (C=O) compared with that of the carbonyl group in doxorubicin (1728 cm^-1) is attributed to the carbonyl group in succinate (Figure 2A).

Amide binding of SDOX to chitosan was carried out using EDC/NHS reagents. Conjugation was confirmed by gel permeation chromatography, differential scanning calorimetry, Fourier transform infrared spectroscopy, and 1H NMR studies. Gel permeation chromatography as presented in Figure 4 shows the increase in molecular weight of chitosan upon conjugation of doxorubicin moieties, and thus verifies the chemical nature of CS-DOX binding. In addition to peaks related to aliphatic protons in chitosan, the 1H NMR spectrum of CS-DOX conjugate showed the appearance of aromatic protons of doxorubicin at 5.4 ppm and 8.3 ppm plus its methyl group at 1.17 ppm. Differential scanning calorimetry analysis of chitosan, doxorubicin, and CS-DOX conjugates was also carried out (Figure 3). A sharp peak at 100°C present in all three thermograms is attributed to the evaporation of humidity absorbed by the specimen. The differential scanning calorimetry thermogram of chitosan showed an endothermic peak at 180°C and an exothermic peak at 370°C which are ascribed, respectively, to hydrogen-bonding dissociation and degradation of this polymer. The differential scanning calorimetry plot obtained for doxorubicin revealed an endothermic melting peak at 200°C with endothermic peaks of degradation appearing at higher temperatures. Conjugation of doxorubicin to chitosan led to a shift in hydrogen-bonding dissociation and degradation peaks of chitosan, while no trace of the doxorubicin peaks was observed in the differential scanning calorimetry thermogram of CS-DOX. The differential scanning calorimetry experiments thus corroborate conjugation of CS and DOX. In fact, addition of hydrophobic and voluminous doxorubicin molecules to the glucosamine units of chitosan weakens...
the hydrogen binding between these units and renders the dissociation temperature lower. The absence of differential scanning calorimetry peaks for doxorubicin in the thermogram of CS-DOX confirms that no free drug exists in the CS-DOX conjugate.

Increasing the ratio of doxorubicin to chitosan in the conjugation reaction gave rise to a higher doxorubicin content but lower conjugation efficiency. All in all, CS-DOX conjugates with drug contents of 0.7%, 1.6%, 2.7%, 4.1%, and 6.7% were synthesized. The amphiphilic property of CS-DOX conjugates in which the hydrophobic doxorubicin molecule is attached to the hydrophilic chitosan chain enables their self-assembly into nanoparticles. Such nanoaggregation behavior has been previously reported by Son et al and Hyung Park et al for glycol chitosan-doxorubicin conjugates.34,35 The exception is CS-DOX-1, the high doxorubicin content of which confers too high a degree of hydrophobicity, which leads to its precipitation in aqueous medium. Figure 10 is a schematic illustration of the self-assembly of CS-DOX conjugates, which is in accordance with the core-shell structures observed in transmission electron micrographs (Figure 6B).
Scanning electron microscopic imaging also revealed smooth-surfaced and spherical nanoparticles with a narrow size distribution. The polydispersity index was determined to be satisfactorily low in nanoaggregates of all CS-DOX conjugates. The zeta potential showed no significant difference between the different types of conjugates, while the hydrodynamic diameter of the nanoparticles was measured to be larger in conjugates with higher drug content. In fact, hydrophobic interactions between the drug-bearing parts of the conjugates and hydrogen bonding between the bare carbohydrate skeleton and surrounding water molecules are the main contributing forces involved in self-assembly of CS-DOX conjugates. With higher drug contents, more portions of the conjugate chain take part in inner hydrophobic interactions and cause more compact aggregation of conjugates and consequently a smaller nanoaggregate size. In addition, the concentration of chitosan-drug conjugate showed no significant effect on size of the nanoaggregates. This indicates the micelle-like nature of these aggregates as the thermodynamics governing the packing of micellar structures have been shown to make the size of spherical micelles independent of concentration.

Size of the drug carriers plays a key role in their pharmacokinetics; the larger the nanoparticles, the higher the probability of their scavenging and clearance by macrophages. Nanoparticles larger than 300 nm activate the complement system and are cleared from the blood. Taking this into consideration, small-sized CS-DOX-2 nanoaggregates were used for the conjugation of trastuzumab, the targeting ligand.

Trastuzumab was conjugated to self-assembled CS-DOX-2 nanoparticles via thiolation of lysine residues by 2-iminothiolane and subsequent linking of the resulting thiols to amine groups on chitosan. The targeted nanoparticles obtained contained 47 μg/mg doxorubicin and 33.5 μg/mg trastuzumab. Trastuzumab decoration led to no significant change in size or zeta potential of the nanoparticles, and the attached trastuzumab is embedded among the dynamic conjugate chains and so does not affect nanoparticle size.

Binding of trastuzumab to the nanoparticles was further probed thermodynamically by an isothermal titration calorimetry study of the interaction of protein A with free and attached trastuzumab. Protein A is a 40–60 kDa surface protein originally found in the cell wall of *Staphylococcus aureus*,

Figure 10 Schematic illustration of self-assembly of chitosan-doxorubicin conjugates into nanoparticles.
and binds with the Fc region of antibodies. This polypeptide chain consists of five homologous IgG-binding domains (from the N-terminus; E, D, A, B, and C). However, steric hindrance restricts concomitant access of all five binding sites to the antibodies. The isothermal titration calorimetry study gave a binding ratio of 3 mol antibody/mol protein A. The trastuzumab attached generated an isothermal titration calorimetry profile different from that of the free antibody, which confirms covalent conjugation of trastuzumab to nanoparticles. The isothermal titration calorimetry profile for the free trastuzumab-protein A interaction was composed of negative exothermic peaks. However, the interaction between the attached trastuzumab and protein A revealed initial exothermic peaks followed by exothermic peaks. A similar isothermal titration calorimetry thermogram was observed by Coles et al for the interaction of dendrimers with DNA. According to this group, the endothermic part of the thermogram is related to the precipitation of the dendrimer-DNA complex. Here, interaction of protein A with the trastuzumab attached led to an opaque appearance in the solution which clearly demonstrated the occurrence of aggregation and precipitation in the system.

Attachment of several trastuzumab monoclonal antibodies to each nanoparticle along with the capacity of each protein A to bind to more than one monoclonal antibody brings the nanoparticles in close proximity to each other and leads to their aggregation into a macrostructure which precipitates (Figure 11).

Furthermore, the initial endothermic peaks demonstrate that the FC fragment of the attached trastuzumab monoclonal antibody has enough freedom to interact with protein A. This property may be taken advantage of for purification of such targeted delivery systems, using protein A affinity chromatography or for determination of monoclonal antibody on the surface of nanoparticles by techniques like surface plasmon resonance.

In vitro cytotoxicity evaluation showed reduced efficacy of the CS-DOX nanoparticles with respect to free drug. This may be attributed to covalent and irreversible conjugation of doxorubicin to chitosan via a succinic anhydride crosslinker, as stated elsewhere. Free doxorubicin can penetrate the nuclear envelope, localize in nucleus and exert its cytotoxicity through its primary mechanism of action, ie, inhibition of topoisomerase II. However, when conjugated to chitosan, doxorubicin cannot access its target and consequently its cytotoxic efficiency is hampered.

On the other hand, fluorescence microscopy displayed enhanced and selective uptake of CS-DOX-mAb by Her2+ cancer cells comparing with CS-DOX nontargeted nanoparticles. The enhanced cellular uptake partially compensated for the diminished effect of attached doxorubicin and improved the cytotoxicity of this doxorubicin nanodelivery system.

**Conclusion**

A novel, chitosan-based targeted drug delivery system has been developed. The anti-HER2 humanized antibody, trastuzumab, was used as the targeting moiety. The resulting amphiphilic polymer-drug conjugates with a doxorubicin content of 0.69%–4.1% (w/w) were demonstrated to self-assemble into nanoparticles ranging from 200–700 nm in size depending on the doxorubicin content of the conjugate. Although covalent conjugation of doxorubicin to chitosan weakens its anticancer activity, antibody-conjugated nanoparticles were shown to discriminate between Her2+ and Her2− cells and thus have potential applications in active targeted drug delivery and reduction of drug side effects in Her2+ breast and ovarian cancers. However, substitution of succinic anhydride with a pH-sensitive crosslinker like cisaconitryl anhydride or hydrazone bonds may be taken into consideration in future studies to achieve an actively targeted vector releasing its cargo drug at its site of action.
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Disclosure

The authors report no conflicts of interest in this work.

References


