

Potential Antioxidant and Wound Healing Effect of Nano-Liposol with High Loading Amount of Astaxanthin

This article was published in the following Dove Press journal: International Journal of Nanomedicine

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Background: Astaxanthin (ASTA), a carotenoid, is a strong antioxidant. However, its application in functional foods, pharmaceuticals, and cosmetics remains limited due to its low aqueous solubility and stability. Several different encapsulating materials have been used to improve the stability and bioavailability of ASTA; however, the currently investigated nano-carriers for ASTA require additional improvements with regard to their loading capacity and stability.

Methods: In this study, we developed lecithin nano-liposol (Lec NS) as a novel carrier of ASTA using a simple emulsion evaporation method. The physicochemical characteristics including hydrodynamic diameter, polydispersity index, surface charge and morphology were analyzed by DLS and TEM. The antioxidant activity of the ASTA-loaded Lec NS (ASTA@Lec NS) was evaluated using a DPPH radical scavenging assay and in vitro antioxidant assay. The study of in vitro wound healing efficacy was carried out to observe the beneficial effect of antioxidant activity of ASTA@Lec NS on cell migration.

Results: ASTA@Lec NS showed improved stability and efficacy owing to improved aqueous solubility of ASTA inside Lec NS. Both in situ and in vitro antioxidant activities of ASTA@Lec NS were higher than that of bare ASTA and Lec NS. It also exhibited strong wound healing efficacy by regulation of ROS level in in vitro cell model.

Conclusion: This study revealed that the encapsulation of ASTA into Lec NS using a wet phase transfer enhanced its physiological stability and bioavailability for effective scavenging of reactive oxygen species.

Keywords: optimization, lipid micelle, reactive oxygen species, ROS scavenging, cell proliferation

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Introduction

Astaxanthin (ASTA) is a natural carotenoid found mostly in yeast, algae, and other aquatic species. Like most carotenoids containing long chains of conjugated polyenes, ASTA is red in color and possesses high antioxidant activity.² Its strong antioxidative properties are attributed to the hydroxyl and keto groups at the ends of its molecular structure making it more antioxidative than other carotenoids or even the well-known lipophilic antioxidant α-tocopherol.³ Multiple studies have found ASTA to be potentially useful as a nutraceutical and cosmeceutical for treating various diseases mediated by oxidative stress.^{4–6} Oxidative stress is caused by an excess generation of reactive oxygen species (ROS) that damage biomolecules and lead to cell death. ROS are derived from oxygen as highly reactive forms of superoxide (O₂•), hydroxyl radical (OH•), hydrogen peroxide (H₂O₂), and hypochlorite (ClO⁻). It has been determined that ASTA scavenges ROS through several assays including 2,2-diphenyl-1-picrylhydrazyl (DPPH), superoxide anion, and hydroxyl radicals⁷ and thereby has a protective effect against various diseases.^{8–10} Regulation of optimal ROS levels also plays an important role in wound healing, while several studies have reported wound healing activity of ASTA.^{11–13} ASTA is involved in processes such as the expression of growth factors, cellular differentiation, angiogenesis, wound healing, and repair, and inflammation.^{14,15}

While it exhibits antioxidative activity, highly unsaturated ASTA is susceptible to heat, light, oxygen, and other external factors. ¹⁶ Therefore, ASTA is easily degraded during its preparation and storage. Furthermore, it has low bioavailability for in vivo application due to its lipophilicity. ¹⁷ To counteract these problems, ASTA has been encapsulated using various materials and is typically stored in powder form. It has been reported that complexation of ASTA improves its stability and shelf-life of bioactivity. ¹⁸ The use of polymers, ^{19,20} proteins, ²¹ and carbohydrates such as cyclodextrin^{22,23} have all been shown to possibly improve the stability, water-dispersibility, and bioavailability of ASTA. Amphiphilic liposomes are

especially capable of loading fat-soluble ASTA into their lipophilic cores, but have low loading capacities. ^{24,25} In addition, several studies have developed ASTA-encapsulated nano-carriers, but these have been primarily prepared using additional emulsifiers such as polysorbate 20, polysorbate 80, cholesterol, and others. ^{26,27} Despite the use of these carriers, the currently available delivery systems of ASTA still require improvement in both their loading capacity and stability.

Therefore, in this study, we developed a lecithin nanoliposol (Lec NS) carrier loaded with ASTA (ASTA@Lec NS) using a simple method of phase transfer in order to enhance ASTA stability and bioavailability (Figure 1A). Dispersion of ASTA in aqueous solution using the Lec NS improves its efficacy and possibly broaden its application in various fields. Through formulation optimization, the Lec NS exhibited a high loading capacity of ASTA. Physicochemical characteristics such as size, polydispersity, surface charges, and morphology of the nano-liposol were analyzed using dynamic light scattering (DLS), ultraviolet-visible (UV-Vis) spectroscopy, and transmission electron microscopy (TEM). The ASTA@Lec NS was highly stable at 37°C and showed enhanced antioxidant activity and wound healing efficacy in vitro.

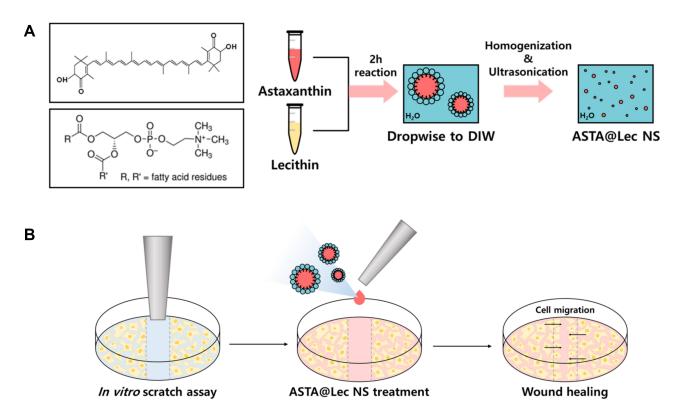


Figure I Development of astaxanthin-loaded lecithin nano-liposol (ASTA@Lec NS). (A) Preparation scheme of ASTA@Lec NS. (B) Scratch wound healing assay of ASTA@Lec NS following scratch wound made by a sterile p1000 micropipette tip across the NIH 3T3 fibroblast monolayer.

Abbreviations: DIW, deionized water; ASTA@Lec NS, astaxanthin-loaded lecithin nano-liposol.

Materials and Methods

Materials

ASTA was purchased from Sigma-Aldrich (St. Louis, MO, **GDE** Ltd. (Siheung, and Korea). phosphatidylcholine from soybean (lecithin), sucrose, and DPPH were purchased from Sigma-Aldrich. H₂O₂ (30%) was obtained from Junsei Chemical Co. (Tokyo, Japan). Chloroform was obtained from Wako Chemicals Ltd (Osaka, Japan) and methanol was purchased from SK Chemical (Seongnam, Korea). Deionized (DI) water and phosphate-buffered saline (PBS) were obtained from HyClone (Logan, UT, USA). The murine fibroblast NIH 3T3 cells (ATCC, MD, USA) used in the in vitro assays were cultured in Dulbecco's modified Eagle's medium (DMEM; Gibco, Grand Island, NY, USA) supplemented with fetal bovine serum (FBS) from Gibco and antibioticantimycotic (AA) from Thermo Fisher Scientific (Waltham, MA, USA). An in vitro indicator of ROS, 2',7'dichlorodihydrofluorescein diacetate (H2DCFDA), was purchased from Invitrogen (Carlsbad, CA, USA).

Preparation of Astaxanthin-Loaded Lecithin Nano-Liposol (ASTA@Lec NS)

Lec NS with various concentrations of ASTA loaded was simply prepared using an emulsion evaporation method with modification.²⁸ Briefly, 1 mL chloroform solution containing various concentrations of ASTA (0, 0.4, 1, 2, 3, 4, and 10 mg/mL) was added to 20 mg of lecithin. After 2 h of mixing using a rotatory shaker, each mixture was slowly added to 5 mL DI water with gentle stirring and homogenized for 5 min at 5000 rpm. The chloroform solvent was then removed by vacuum drying for 3 h and the remaining solution was subjected to ultrasonication (amplitude = 20%, on/off time = 5s/15s) for 20 min. Next, the ASTA@Lec NS was purified by centrifugation at 2000 rpm for 15 min using an Amicon Ultra-15 centrifugal filter (molecular weight cut-off 100 kDa; Merck Millipore, Burlington, MA, USA) to remove unloaded ASTA and then filtered through a sterile 0.45 µm syringe filter.

Characterization of ASTA@Lec NS

The hydrodynamic diameter, polydispersity index (PDI), and zeta potential of ASTA@Lec NSs with different loaded ASTA content were characterized by DLS measurements using a Zetasizer instrument (ELSZ-2000,

Otsuka, Osaka, Japan). Loading content (LC) of ASTA was determined using the following equation:²⁹

$$LC(wt\%) = \begin{bmatrix} \frac{\text{weight of feeding}}{\text{ASTA}} & \frac{\text{weight of}}{\text{unloaded ASTA}} \\ \\ \hline & \frac{\text{weight of LecNS}} \end{bmatrix}$$
(1)

Loading efficiency (LE) of ASTA into the Lec NS was calculated using the following equation:

$$LE(\%) = \begin{bmatrix} \frac{\Delta A_{490} \text{ of feeding } -\Delta A_{490} \text{ of unloaded } \\ \frac{\Delta STA}{\Delta A_{490} \text{ of feeding } \\ \Delta STA} \end{bmatrix}$$

$$ASTA$$
(2)

where ΔA_{490} of feeding ASTA and unloaded ASTA were the UV-Vis absorbance of the initial ASTA and remaining ASTA after encapsulation at a wavelength of 490 nm. ³⁰ The amount of ASTA in the Lec NS was quantified via a chemical colorimetric technique using a UV-Vis spectrophotometer (Mega900, Scinco, Seoul, Korea). Morphology of the ASTA@Lec NS was analyzed by TEM by evaluating 20 μ L of sample on a copper grid with 200 mesh carbon film using a JEM-2100Plus HR (JEOL, Tokyo, Japan).

In situ Antioxidant Activity of ASTA@Lec NS

Antioxidant activities of ASTA and ASTA@Lec NS were evaluated using the stable-free radical DPPH. ³¹ Bare ASTA and ASTA@Lec NS (LC 15 wt%) were dissolved in DI water at various concentrations. DPPH solution (0.5 mM) was prepared in methanol. Then, 150 μ L of each sample solution and 150 μ L of DPPH solution were combined in a 96-well plate. To prepare the negative control, 150 μ L of DI water was added to the DPPH solution instead of sample solution. Following a 30 min incubation with minimal exposure of light, absorbance of the reaction mixtures was measured at a wavelength of 515 nm using a microplate reader (VICTOR X5, PerkinElmer, Singapore, Republic of Singapore). Antioxidant activity was calculated using the following equation:

Antioxidant activity(%)
$$= \left[\frac{(\Delta A_{515} \text{ of control} - \Delta A_{515} \text{ of sample})}{\Delta A_{515} \text{ of control}} \right]$$
(3)

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As an overlap of absorbance at 515 nm with the sample absorbance may lead to inaccurate results,³² background absorbance was subtracted from the absorbance of the test samples.

Stability of ASTA@Lec NS

To increase the storage stability of ASTA@Lec NS, the Lec NS was freeze-dried with different amounts of sucrose as a cryoprotectant (0, 1, 5, 10, and 15 wt%) for 3 d.³³ The appearance after lyophilization with sucrose was observed and stability of the physicochemical characteristics of the ASTA@Lec NS was also analyzed. Lyophilized nanoliposols were dissolved in biological buffer (PBS) and then kept at room temperature or in a shaking incubator (100 rpm) at 37°C. The samples were monitored for 4 weeks for any change in dispersity or hydrodynamic diameter.

In vitro Cytotoxicity of ASTA@Lec NS

For in vitro assays, NIH 3T3 cells were cultured in DMEM containing 10% FBS and 1% AA. To evaluate cytotoxicity of Lec NS and ASTA@Lec NS, cells were seeded into a 96-well plate at a density of 10,000 cells per well and incubated in a humidified atmosphere of 5% CO₂ at 37°C.^{34,35} After 12 h incubation, different concentrations of the nano-liposols were added to the cells and incubation continued for 24 h. Then, Cell-Counting Kit-8 (CCK-8) solution (Dojindo Laboratories, Japan) diluted to 10% in DMEM was added to each well for 1 h. Viable cells produced orange-colored formazan at an absorbance of 450 nm. The absorbance was measured using a microplate reader and cell viability calculated using the following equation:

$$\mbox{Cell viability(\%)} \ = \ \left(\frac{\Delta A_{450} \ of \ test \ group}{\Delta A_{450} \ of \ control \ group} \right) \ \times \ 100$$

(4)

In vitro Antioxidant Activity of ASTA@Lec NS

The in vitro ability to scavenge ROS generated from NIH 3T3 cells stimulated with the oxidative stress agent $\rm H_2O_2$ was analyzed according to a method previously described. Briefly, the NIH 3T3 fibroblast cells were seeded into a 96-well plate (10,000 cells/well). Following an overnight incubation, cells were treated with Lec NS or ASTA@Lec NS (10 μ g/mL) and then subsequently treated with $\rm H_2O_2$ (5 μ M) for 8 h, and were then washed with PBS to remove the remaining media. H2DCFDA solution (10

μM) was added to each well for 90 min under dark conditions. In the presence of ROS, the detection reagent was oxidized to fluorescent dichlorofluorescein (DCF). Thus, the in vitro antioxidant activity of ASTA@Lec NS could be measured using a microplate reader according to the decrease in fluorescence at an excitation wavelength of 485 nm and an emission wavelength of 535 nm.

In vitro Wound Healing Effect of ASTA@Lec NS

A scratch wound-healing assay was performed as previously reported to evaluate enhanced migration of fibroblast cells following treatment of ASTA@Lec NS (Figure 1B). First, cells were seeded into a 24-well plate (200,000 cells/well) and grown to confluency. A scratch wound was then made using a sterile P1000 micropipette tip. Cell debris was removed by washing with DMEM and different concentrations of ASTA@Lec NS were used to treat the cells. The control group received only DMEM without FBS. Wound closure was evaluated at each desired time point over a 24-h incubation period using a DMi1 light microscope (Leica, Wetzlar, Germany) at 4× magnification. The wound gap distance was calculated by using ImageJ software 1.8.0 (NIH, Bethesda, MD, USA).

Statistical Analysis

All experiments were performed in triplicate and the resulting data are shown as the mean \pm standard deviation. Experimental groups were statistically compared using Student's *t*-test. Differences were considered to be statistically significant or highly significant at p-values <0.05 or <0.001, respectively.

Results and Discussion

Preparation and Characterization of ASTA@Lec NS

ASTA@Lec NS containing various concentrations of ASTA were prepared through reactions using ASTA and lecithin in the chloroform solvent. Oil-in-water ASTA emulsions were formed by adding the mixtures to the water phase and the nanoscale preparations generated by homogenization and ultrasonication following the removal of organic solvent. As the ASTA@Lec NS developed, the nano-liposols exhibited red color with increasing loading contents of ASTA (Figure 2A). The hydrodynamic diameter prior to loading was 140 ± 4 nm, which was consistently maintained, even after loading up to 15 wt%

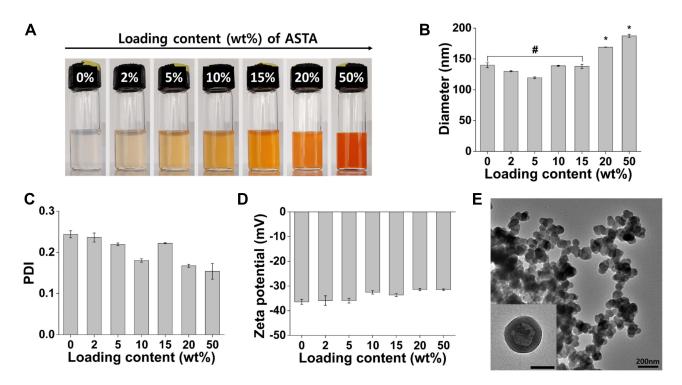


Figure 2 Characterization of astaxanthin-loaded lecithin nano-liposol (ASTA@Lec NS). (A) Loading content; (B) hydrodynamic diameters; (C) polydispersity index (PDI); (D) zeta potential of ASTA@Lec NS with different loading contents of astaxanthin (ASTA). (E) Transmission electron microscopy (TEM) image of ASTA@Lec NS with a loading content of 15 wt% (scale bar = 200 nm).

Notes: # or * refers p-value above 0.05 or below 0.05, respectively.

Abbreviations: ASTA, astaxanthin; wt%, weight percent; PDI, polydispersity index.

ASTA (Figure 2B). However, the ASTA@Lec NSs with loading contents of 20 and 50 wt% had larger diameters of 169 ± 1 nm and 187 ± 2 nm, respectively, indicating that the ASTA was overloaded (p < 0.05). There was no significant change in PDI or zeta potential (Figure 2C and D). The nano-liposols were well dispersed with PDI values below 0.3 while their surfaces were negatively charged with zeta potentials of approximately -33.9 ± 0.9 mV. The morphology of the nano-liposols was evaluated by TEM (Figure 2E). As expected from the DLS analysis data, the ASTA@Lec NS (15 wt%) were spherical in shape of similar size (approximately 150 nm). Thus, it was determined that the optimal loading capacity of ASTA@Lec NS was 15 wt% with an entrapment efficiency of 98.8%. Subsequent experiments were performed using ASTA@Lec NS containing 15 wt% ASTA.

In situ Antioxidant Activity of ASTA@Lec NS

Antioxidant activity of ASTA@Lec NS was evaluated according to its ability to serve as an electron donor where deep violet DPPH radicals are decolorized and become yellow. Consistent with Li et al, ³⁸ ASTA was more effective at radical scavenging

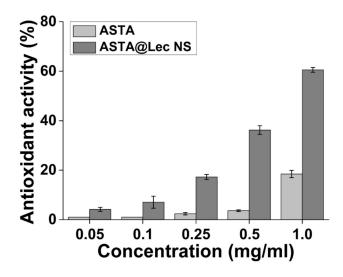


Figure 3 In situ antioxidant activity analysis of astaxanthin (ASTA) and astaxanthin-loaded lecithin nano-liposol (ASTA@Lec NS; 15 wt%) at various concentrations of ASTA according to a 2.2-diphenyl-1-picrylhydrazyl-free radical scavenging assay. **Abbreviations:** ASTA, astaxanthin; ASTA@Lec NS, astaxanthin-loaded lecithin nano-liposol.

at higher concentrations (Figure 3). However, much higher concentrations of free ASTA were required to reach approximately 20% antioxidant activity as an aqueous solution of ASTA was used for the DPPH radical scavenging assays in

this study instead of an organic solvent. This suggests that the solubilization of ASTA is important for its radical scavenging activity. For this reason, the antioxidant activity of ASTA improved significantly following encapsulation in Lec NS. At the same concentration of 1 mg/mL, the antioxidant activity of ASTA@Lec NS was approximately 60%, which was much higher than that of bare ASTA. This result was consistent with a previous report. Furthermore, it is worth noting that the use of high amounts of encapsulating material decreased the antioxidant activity of the complexes by reducing the accessibility of ASTA to radicals, which was previously reported by Yuan et al. 23

Stability Testing of ASTA@Lec NS

The ASTA@Lec NS was prepared as a powder by lyophilization to assure extended maintenance of ASTA bioactivity. To prevent agglomeration of the nano-liposols during lyophilization, the amount of sucrose used as a cryoprotectant was optimized. Without sucrose, the diameter of the nano-liposols increased to 644 ± 16 nm after freeze drying (Figure 4A). The addition of more than 5 wt% of sucrose kept the diameter of the nano-liposols similar to that prior to freeze drying. With increasing amounts of sucrose, the PDI values decreased below 0.3 and the zeta potential of the nano-liposols became less negative (Figure 4B and C). Furthermore, it was easier to

obtain a fine powder of ASTA@Lec NS with more than 10 wt% sucrose, suggesting that 10 wt% sucrose was optimal for freeze drying ASTA@Lec NS (Figure 4D). This was consistent with a previous report that sucrose enhances storage stability post lyophilization by lowering residual water content. Therefore, samples lyophilized with insufficient amounts of sucrose (<5 wt%) were too sticky to be ground and considered unsuitable for subsequent applications.

The stability of ASTA@Lec NS in solution was also evaluated for 4 weeks at 25°C and 37°C (Figure 5). The diameters and PDIs of ASTA@Lec NS barely changed, even after prolonged storage at 25°C and 37°C, indicating that the nano-liposols could maintain their nanostructure for sufficiently long periods of time at both room temperature and body temperature.

In vitro Cytotoxicity and Antioxidant Activity of ASTA@Lec NS

CCK assays were used to evaluate the viability of murine dermal fibroblasts following treatment with Lec NS and ASTA@Lec NS (Figure 6A). Both nano-liposols with and without ASTA demonstrated no cytotoxicity, resulting in >95% viability at a concentration of 1000 µg/mL. In addition, Y. Chen et al found using MTT assays that ASTA does not affect the proliferation of human dermal fibroblast.⁸ This

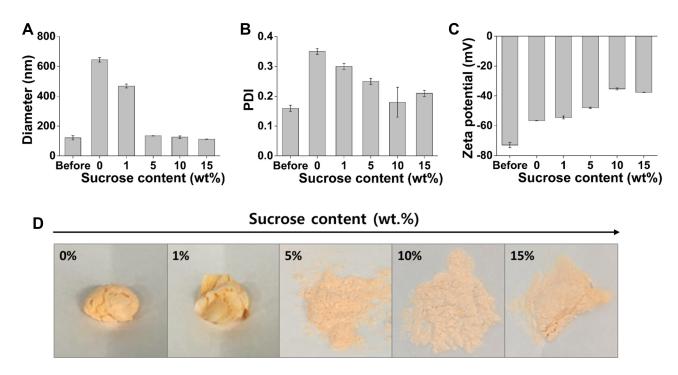


Figure 4 Stability of astaxanthin-loaded lecithin nano-liposol (ASTA@Lec NS; 15 wt%) following lyophilization using sucrose as a cryoprotectant. (A) Hydrodynamic diameter; (B) polydispersity index (PDI); (C) zeta potential; (D) photographs of ASTA@Lec NS (15 wt%) following lyophilization using different amounts of sucrose. Abbreviations: wt%, weight percent; PDI, polydispersity index.

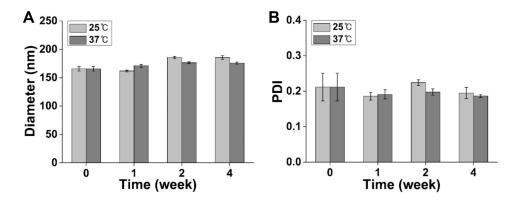


Figure 5 Stability of physicochemical properties of astaxanthin-loaded lecithin nano-liposol (ASTA@Lec NS; 15 wt%) after 0–4 weeks of storage at room temperature and body temperature. (A) Hydrodynamic diameters; (B) polydispersity index (PDI) of ASTA@Lec NS (15 wt%) after 0–4 weeks of storage.

Abbreviation: PDI, polydispersity index.

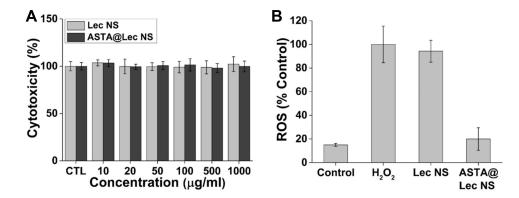


Figure 6 In vitro assays using CCK-8 and H2DCFDA assay kit. (A) Cytotoxicity analysis of lecithin nano-liposol (Lec NS) and astaxanthin-loaded lecithin nano-liposol (ASTA@Lec NS; 15 wt%). (B) Antioxidant activity of Lec NS and ASTA@Lec NS (15 wt%), while the control and H₂O₂ samples indicate the lowest and highest ROS level. Abbreviations: Lec NS, lecithin nano-liposol; ASTA@Lec NS, astaxanthin-loaded lecithin nano-liposol; ROS, reactive oxygen species; H₂O₂, hydrogen peroxide.

indicates that ASTA@Lec NS is biocompatible and does not cause any harmful effects.

ASTA@Lec NS exhibited significant antioxidant activity at low concentrations of 10 μg/mL (Figure 6B). Following treatment with H₂O₂ as an oxidative stress agent, DCF fluorescence was analyzed to quantify in vitro level of ROS. NIH 3T3 cells treated with only H₂O₂ was normalized as 100% ROS levels. The encapsulating material (Lec NS) alone, barely reduced ROS levels with fluorescence being approximately 94%. In contrast, ASTA@Lec NS effectively scavenged cellular ROS to a level of approximately 20%, which similar to that of the negative control at approximately 15%. This proved that ASTA@Lec NS possessed very high antioxidant activity, both in situ and in vitro.

In vitro Wound Healing Activity of ASTA@Lec NS

Elimination of excessive ROS prevents endothelial dysfunction, damage to vessel walls, neurological diseases, and other negative outcomes. 40,41 In particular, the scavenging activity and bio-signaling induced by ASTA may play a key role in wound healing. Scratch wound-healing assays are widely accepted as a direct and inexpensive in vitro wound healing test, alternatives to testing on animals. 15 After forming a scratch in the cell monolayer, cells around the wound attempt to restore the damaged area through cell proliferation and migration. We analyzed the wound healing activity of ASTA@Lec NS by evaluating its effect on cell growth (Figure 7). Following 24 h of treatment, the scratch distance of the control group without ASTA@Lec NS exhibited almost no change from that of the initial distance of 321 \pm 7 μm (Figure 7A). However, the scratch distance of the ASTA@Lec NS-treated cells decreased from $324 \pm 6 \mu m$ to $227 \pm 19 \mu m$ after 24 h of treatment. The effect of wound healing increased relative to ASTA concentration (Figure 7B); specifically, the scratch distances of cells treated with ASTA@Lec NS at different concentrations (10, 100, 500, and 1000 µg/mL) were approximately 328 μm, 291 μm, 254 μm, and 227 μm,

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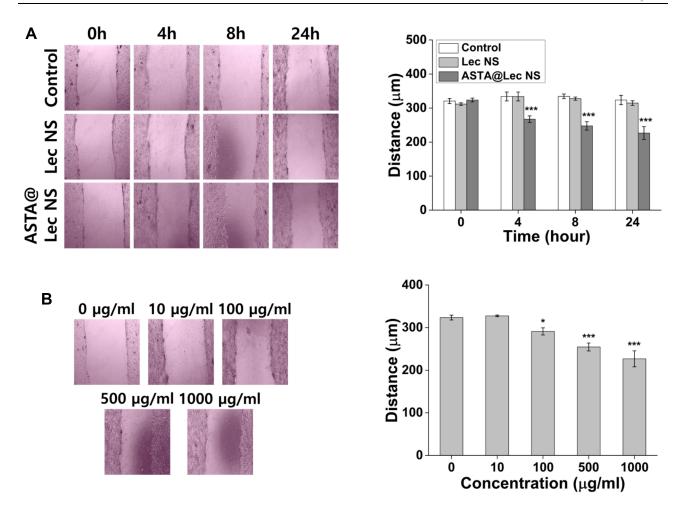


Figure 7 In vitro wound healing activity of astaxanthin-loaded lecithin nano-liposol (ASTA@Lec NS; 15 wt%). (A) Wound closure of NIH 3T3 fibroblast cells after treatment of lecithin nano-liposol (Lec NS) and ASTA@Lec NS for I d. (B) Wound healing activity of ASTA@Lec NS at different concentrations after 24 h of treatment.

Notes: The p-value below 0.05 or 0.001 was denoted by * or ****, respectively.

Abbreviations: Lec NS, lecithin nano-liposol; ASTA@Lec NS, astaxanthin-loaded lecithin nano-liposol.

respectively. Wound closure following a treatment of 1000 μ g/mL of ASTA@Lec NS was statistically greater than that of the control group (p < 0.001). This is consistent with the report by Chou et al, that ASTA has a positive effect on skin wound healing by promoting collagen generation in human dermal fibroblasts and inhibiting expression of matrix metalloproteinases MMP-1 and MMP-3.

Conclusion

In this study, ASTA@Lec NS was successfully developed using a simple emulsion evaporation method. Hydrodynamic diameter, surface charge, and morphology were characterized for ASTA@Lec with different LC. Without any change in its physicochemical properties, it could load high amount of ASTA with high efficiency. Furthermore, ASTA@Lec NS was stable for a sufficiently long period and readily lyophilized to powder state when an appropriate amount of sucrose

was used as a cryopreservative. It suggests a feasibility of storage and transport for its various applications. In addition, the enhanced aqueous solubility of ASTA@Lec largely improved the bioavailability of ASTA. As a result, it could remove ROS and promote wound healing of NIH 3T3 cells more effectively without causing any cytotoxicity. Therefore, ASTA@Lec NS represents an outstanding delivery system to overcome limitations of ASTA for its applications in functional foods, medicines, cosmetics, and many others.

Acknowledgment

This research was supported by the National Research Foundation of Korea (NRF) grant No. NRF-2018 R1D1A1B07043620 (W.I.C.) funded by the Korea government (MSIT) and the Soonchunhyang University Research Fund (J.-M.L.).

Disclosure

The authors report no conflicts of interest in this work.

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