

Memecylon edule leaf extract mediated green synthesis of silver and gold nanoparticles

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Abstract: We used an aqueous leaf extract of *Memecylon edule* (Melastomataceae) to synthesize silver and gold nanoparticles. To our knowledge, this is the first report where *M. edule* leaf broth was found to be a suitable plant source for the green synthesis of silver and gold nanoparticles. On treatment of aqueous solutions of silver nitrate and chloroauric acid with *M. edule* leaf extract, stable silver and gold nanoparticles were rapidly formed. The gold nanoparticles were characterized by UV-visible spectroscopy, scanning electron microscopy (SEM), transmission electron microscopy (TEM), energy dispersive X-ray analysis (EDAX) and Fourier transform infra-red spectroscopy (FTIR). The kinetics of reduction of aqueous silver and gold ions during reaction with the *M. edule* leaf broth were easily analyzed by UV-visible spectroscopy. SEM analysis showed that aqueous gold ions, when exposed to *M. edule* leaf broth, were reduced and resulted in the biosynthesis of gold nanoparticles in the size range 20–50 nm. TEM analysis of gold nanoparticles showed formation of triangular, circular, and hexagonal shapes in the size range 10–45 nm. The resulting silver nanoparticles were predominantly square with uniform size range 50–90 nm. EDAX results confirmed the presence of triangular nanoparticles in the adsorption peak of 2.30 keV. Further FTIR analysis was also done to identify the functional groups in silver and gold nanoparticles. The characterized nanoparticles of *M. edule* have potential for various medical and industrial applications. Saponin presence in aqueous extract of *M. edule* is responsible for the mass production of silver and gold nanoparticles.

Keywords: *Memecylon edule*, nanoparticles, bioreduction, electron microscopy, FTIR

Introduction

Medicinal plants play a major role in the discovery of new therapeutic agents for drug development. *Memecylon edule* is a shrub or small tree with medicinal properties belonging to the family Melastomataceae.^{1–4} It is commonly known as iron wood tree (English) and puvai kaya (Tamil).⁵ The leaves of this plant have been employed in the dyeing industry to dye wool and silk.^{6,7} Medicinal substances found in plants are referred to as secondary metabolites or biologically active compounds. Some of the bioactive compounds present in *M. edule* include triterpenes, tannins, and flavonoids.⁸

The leaves of *M. edule* have been reported to possess strong astringent properties and have also been used to treat leucorrhoea and gonorrhoea in India. A lotion prepared from the leaves is used to treat bruises and cure eye ailments. This plant is also used to treat several conditions and diseases including wounds and diabetes.^{9,10} In the past decade, green nanoparticle synthesis has evolved into an important branch of nanotechnology because of its potential application in the biomedical, magnetics, energy science and aerospace industries. Large amounts of nanoparticles can be easily synthesized from plants and

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the majority of these are nontoxic.^{11–13} They have been used in pharmaceutical drug-based industries to treat B-chronic lymphocytic leukemia, for detecting DNA, to inhibit bacteria and fungi, and to prevent burns and open wound infections.^{14–16} Several techniques, such as chemical, electrochemical, radiation, photochemical, and biological, have been used for the biosynthesis of silver and gold nanoparticles.¹⁷

In many previous reports, green synthesis of nanoparticles was done in conical flasks utilizing H₂O as a solvent.¹⁸ The bioreduction of silver and gold nanoparticles from silver and chloroaurate ions due to their surface plasmon resonance (SPR) has been achieved using plants such as tamarind.¹⁹ *Helianthus annuus*, *Basella alba*, *Cinnamomum camphora*,^{20,21} alfalfa,²² *Capsicum annum*,²³ *Avena sativa*,²⁴ *Azadirachta indica*,²⁵ and *Pelargonium graveolens*,²⁶ have been reported for the presence of saponin. Due to the various medicinal properties of *M. edule*, thus commercial production of the saponin is quite high and thereby shows potential for the commercial synthesis of nanoparticles. In this study we describe for the first time, the *M. edule* leaf extract mediated biosynthesis of silver and gold nanoparticles. The silver and gold nanoparticles are of variable sizes, shapes, and chemical composition.^{27, 28} In this study, we show that an extract of *M. edule*, placed in a concentrated aqueous solution of AgNO₃, resulted in the reduction of the silver ions and formation of silver nanoparticles, and that an extract in aqueous solution of AuCl₄ resulted in the reduction of the gold ions and formation of gold nanoparticles. UV-visible (UV-vis) spectroscopy, transmission electron microscopy (TEM), scanning electron microscopy (SEM), energy dispersive X-ray analysis (EDAX), and Fourier transform infra-red spectroscopy (FTIR) were used to examine size and shape of gold and silver nanoparticles in aqueous extract of *M. edule*.

Materials and methods

Chemicals used for synthesis of silver and gold nanoparticles were chloroauric acid (HAuCl₄) and silver nitrate (AgNO₃) (Sigma-Aldrich, MO). Fresh *M. edule* leaves collected from the Potheri forest located at SRM University, (Kattankulathur, Tamil Nadu, India) were used as a plant source for green nanoparticle synthesis.

Equipment including an Orbitol shaker (Labquake shaker; Labindustries, Berkeley, CA) and a PerkinElmer spectrophotometer (Boston, MA) were used initially for nanoparticle synthesis. Nanoparticles size was confirmed by using TEM (1200 EX; JOEL USA, Inc, Peabody, MA) and SEM (FEI Quanta FEG 200; FEI Company, Hillsboro, OR) with high and low vacuum. The functional groups in the synthesized

nanoparticles were confirmed by using FTIR (6700 spectrum; Thermo Nicolet, Madison, WI).

Preparation of *M. edule* leaf extract

Leaves were thoroughly washed and dried completely in the shade. Leaf material was cut into small pieces and powdered in a mixer. Dried powder (20.0 g) of *M. edule* leaf was added to 100 mL of sterile distilled water in a 500 mL Erlenmeyer flask and then boiled for 5 minutes. The flasks were kept under continuous dark conditions at 30°C. The extract was filtered and stored at 4°C for further experiments.

Green synthesis of silver and gold nanoparticle using leaf extract

The aqueous leaf extract of *M. edule* was used for the bioreduction process. To synthesize nanoparticles from *M. edule* 5, 10, and 15 mL of aqueous leaf extract were carefully added to 10 mL of 1 mM aqueous AgNO₃ and HAuCl₄ solution in 250 mL Erlenmeyer flasks. The flasks containing extract were incubated in a shaker at 150 rpm in dark conditions.

UV-vis adsorbance spectroscopy analysis

The bioreduction (by AgNO₃ or HAuCl₄) of nanoparticles was monitored periodically by UV-vis spectroscopy. The samples used for analysis were diluted with 2 mL deionized water and subsequently measured by the UV-vis spectrum at regular different time intervals (Raut Rajesh et al).³⁰ A UV-vis spectrograph of the colloidal solution of silver and gold nanoparticles was recorded as a function of time by using a quartz cuvette with water as reference. The UV-vis spectrometric readings were recorded at a scanning speed of 200 to 800 nm.²⁹

TEM analysis of gold nanoparticles

The precipitate were settled down after the reaction time in the bottom of a conical flask. The suspension above the precipitate containing silver and gold nanoparticles of *M. edule* was sampled by TEM analysis. TEM samples were prepared by placing a drop of the suspension of silver and gold nanoparticle solutions on carbon-coated copper grids and allowing water to evaporate. The samples on the grids were allowed to dry for 4 minutes. The shape and size of gold nanoparticles from *M. edule* were determined from TEM micrographs.

SEM analysis of silver and gold nanoparticles

The prepared silver and gold nanoparticles were characterized using high resolution SEM analysis. The samples were

prepared by simple drop coating of the suspension of silver and gold solutions onto an electric clean glass and allowing the solvent (water) to evaporate. The samples were left to dry completely at room temperature.

FTIR spectroscopy analysis of dried biomass after bioreduction

A carefully weighed quantity of the synthesized gold and silver nanoparticle powder was subjected to FTIR analysis. The bioreduced chlorauric and silver solution was centrifuged at 10,000 rpm for 15 minutes and the pellet was washed three times with 20 mL of deionized water. The resulting purified suspension was completely dried and ground with KBr pellets and analyzed by FTIR. In order to obtain good signal/noise ratio, 512 scans were recorded.

EDAX spectrum measurements

M. edule leaf extract–reduced gold and silver solutions were dried, drop coated on to carbon film, and tested using EDAX analysis (S-3400N; Hitachi, Tokyo, Japan).

Results and discussion

Green synthesis of silver nanoparticles by *M. edule* leaf

The aqueous extracts obtained from *M. edule* were used for preliminary phytochemical screening for secondary metabolites. The water extract of leaf of *M. edule* gave positive result for saponin and results suggested that saponin is the most favorable starting material to make nanoparticles. Saponin has several medicinal properties and pharmacological activities have been reported in saponins from many other plant species. Triterpene saponin from *Ganoderma atrum* has hypolipidemic, antiatherosclerotic, anticoagulant, antihypertensive, antimicrobial, hepatoprotective, and immunomodulation properties. Early studies demonstrated that the saponin-based nanoparticles extracted from aqueous extract of *Panax ginseng* has strong activity for upregulation of immune response in mice.³⁰

The kinetics of reduction of aqueous silver and chloraurate ions during reaction with the *M. edule* leaf extract were easily analyzed by UV-visible spectroscopy. The optical adsorption UV-v is spectra of silver nanoparticles produced by different amounts (5, 10, 15 mL) of the dried biomass of aqueous leaf extract of *M. edule*. The result shows the production of silver nanoparticles within 3 hours after silver ions came in contact with the biomass. The addition of *M. edule* leaf extract to the aqueous AgNO_3 solution resulted in the color of the solution changing to dark brown due to the SPR. The color changes

were noticed because of the production of silver nanoparticles. Comparative experiments were carried out to examine the effect of different amounts of *M. edule* leaf extract (5, 10, and 15 mL) on bioreduction of silver ions (Figure 1).

It is well-known that these results are supported in previous reports, where the addition of aqueous leaf extract of various plants to 10^{-3} M aqueous AgNO_3 solution started its color change due to the excitation of SPR in the production of silver nanoparticles.^{18,31} Figure 2 clearly shows the strong adsorption peak at 0.51 (AU) over increasing time of reaction by gently reducing AgNO_3 with 15 mL of extract at around 475 nm. A weak adsorbance peak was observed at 0.29 AU in 5 mL of extract and mild reduction of adsorption of silver ions at 0.48 AU with 10 mL of extract. The 15 mL of extract was checked for bioreduction at time intervals over 3 to 18 hours. The evolution of UV-vis spectrum of synthesized silver nanoparticles have an increasingly sharp adsorbance peak identified at around 475 nm in the adsorption spectra of silver nanoparticles. Adsorption peak was not clear in the 5 mL and 10 mL extracts, but a sharp peak was observed at the higher dosage level 15 mL and peak resolution was also very clear. The bioreduction of silver ions was increased at the end of 10 hours of reaction time, as illustrated in Figure 2. Reductive biomolecules for reduction of gold ions were responsible for the formation of the few nanoparticles, as the maximum adsorbance is much lower than that in 10 mL and 15 mL extracts.

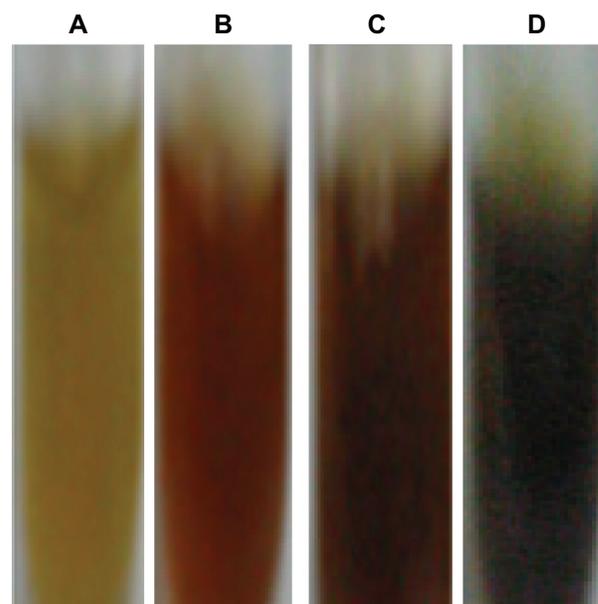


Figure 1 Optical photograph of the colloidal solution of silver nanoparticles reduced by *Memecylon edule* leaf extract at (5, 10 and 15 mL). Color change occurred due to surface plasmon resonance (A) *M. edule* leaf extract (B) 5 ml of extract with Ag^+ solution (C) 10 mL of extract with Ag^+ solution (D) 15 mL of extract with Ag^+ solution after the process of Ag^+ to Ag nanoparticles.

Abbreviations: AgNPs, silver nanoparticles; SPR, surface plasmon resonance.

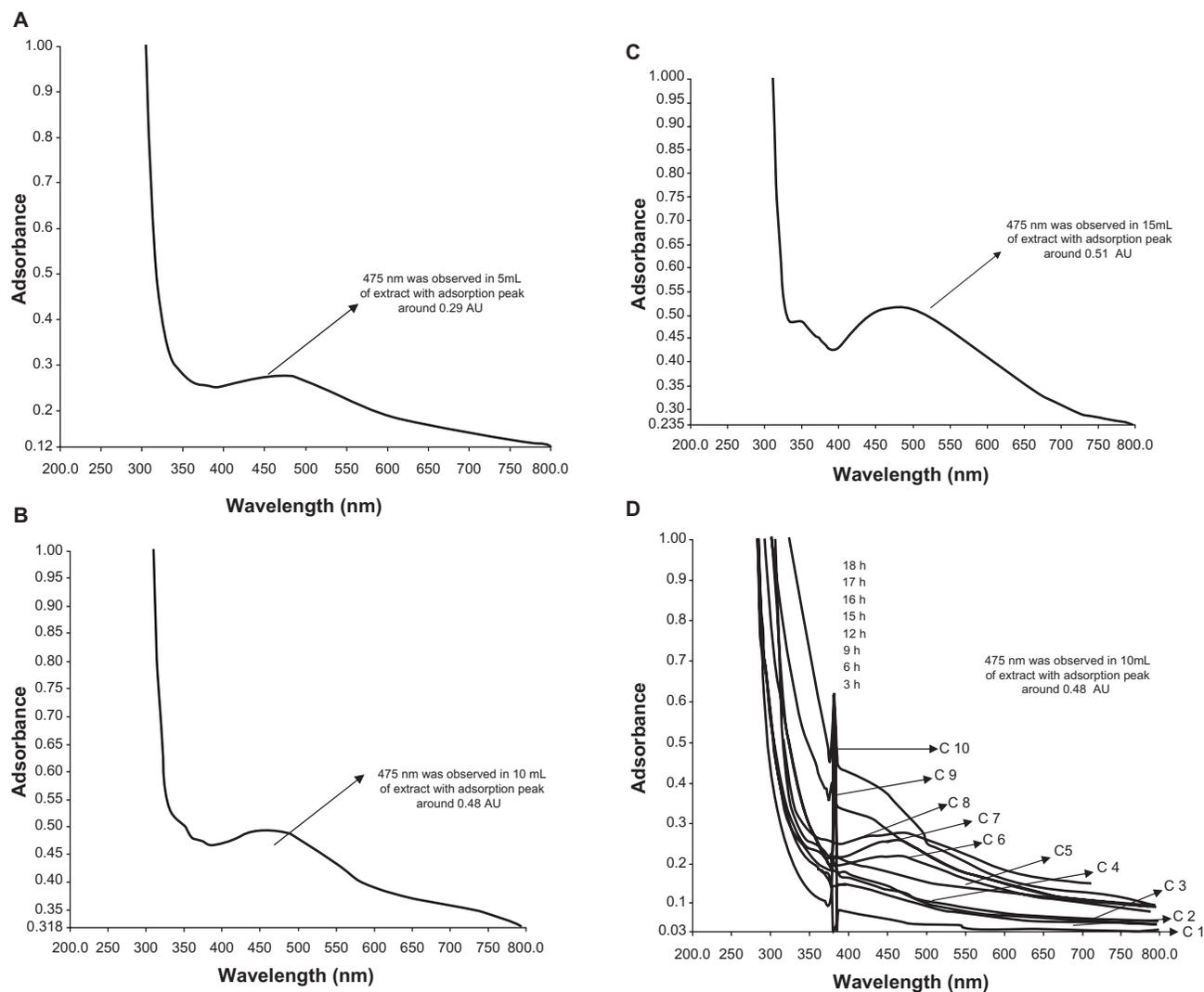


Figure 2 UV-vis adsorption spectra of silver nanoparticles after bioreduction kinetics of the reaction of *M. edule* leaf extract with aqueous silver ions at 5 mL (A), 10 mL (B), 15 mL (C) in the concentration range of 200–800 nm with different time intervals. A) 5 mL of silver and leaf aqueous solution (B) 10 mL of silver with leaf aqueous solution and (C) 15 mL of silver with leaf aqueous solution. D) Curves 1–10 correspond to 3, 6, 9, 12, 15, 16, 17, 18, 21, 24 hours of reaction time, respectively, in 15 mL of leaf extract with silver aqueous solution.

Abbreviation: C, curves.

Previous reports from Huang et al on *C. camphora* show that silver nanoparticles may grow in a process involving rapid bioreduction and that they strongly influence the SPR in the water extract.²² This is accordance with the results obtained from bioreduction of silver nanoparticles using *Spirulina Platensis*, which showed that a SPR silver band occurred at 400–480 nm.^{21,32}

Aqueous extracts of *M. edule* possess remarkable anti-oxidant activity and they should be considered important nutraceuticals. We are studying other biological activities of gold and silver nanoparticles of *M. edule* in vitro and in vivo properties, which we will communicate shortly. A phytochemical study of this genus (*Memecylon*) showed

the presence of triterpenes, flavonoids, and tannins. The first systematic phytochemical survey was conducted in Malaysia by Said et al, of Melastomataceae and some other families for alkaloids, triterpenoids, steroid compounds, saponins, essential oils, glycosides, and hydrogen cyanide.⁹ They identified saponin as a major compound in the leaves of *Memecylon* sp and, further, that this compound is involved in taste properties (bitterness), flavor, and pigmentation.^{8,33} Aqueous extracts of *Teucrium polium*, *Angelica sinensis*, *Lycium barbarum*, and *Lycium cocos* have strong antioxidant activity. Free radicals are known to be the main cause of aging, coronary heart disease, stroke,³⁴ diabetes mellitus, rheumatism, and cancer.^{35,36} The antioxidant activity

of nanoparticulate saponin is potentially therapeutic for a variety of diseases including cancer and aging, and nanoparticulate saponin is also used to treat hypersensitivity.³⁷ Our preliminary work here has confirmed the nanosaponin present in *M. edule* could be a good remedial plant source for many diseases.

Figure 3 shows representative SEM images recorded at different magnifications from drop-coated films of the silver nanoparticles synthesized by treating AgNO₃ solution with *M. edule* leaf broth. The resulting silver nanoparticles were predominantly square and of uniform size. Higher magnification showed the average diameter of these square nanoparticles to be about 50 to 90 nm. SEM images of biologically synthesized typical silver nano irregular particles (the shape of the nanoparticles were not clearly predicted) were obtained from 15 mL of leaf extract. The resulting nanoparticles showed that a small percentage of spherical particles were in the size range 10–45 nm. SEM images of the nanotriangles in the same suspension are depicted in Figure 3. Silver ions were not well separated from each other in the nanotriangles and dimensions in the size range 45–60 nm are capped with smaller particles due to the presence of small crystal and hexagonal particles on the triangular surface 10–25 nm in diameter.

The SEM analysis of silver nanoparticles of *M. edule* support the results of Govindaraju et al in *S. platensis*.^{38,39} In addition, rapid biosynthesis of silver nanoparticles in different shapes (twinning, mixed structures, dislocations) was observed and sizes of nanoparticles were increased by high concentrations of leaf extract of *M. edule*. It is interesting that the size of the triangular silver nanoparticles shown in Figure 3 increased to about 60 nm in size as the dosage of *M. edule* leaf extract increased to 15 mL. The size of silver nanoparticles obtained in the present investigation is supported by previous studies.²² According to nanoparticle size of *M. edule*, results obtained was comparable with silver nanoparticles synthesized by using *C. camphora* size ranging mostly from 55 to 80 nm,²² and 31 to 40 on *C. zeylanicum*.^{40,41}

Figure 3 shows the selected-area electron diffraction (SAED) pattern recorded from this triangular- and irregular-shaped single crystalline could be indexed based on the structure of silver nanoparticles. The square-shaped nature of the diffraction crystals are a clear indication.

To summarize these results, the size of the silver nanoparticles obtained from *M. edule* can be increased when adding a dosage of 15 mL to silver solution. As a result, researches in the field of nanoparticle synthesis described that these silver nanoparticles obtained from plant extracts can be used for

medical purposes. The EDAX technique is used to further identify the silver nanoparticles of *M. edule*.

Together with TEM images it is interesting that the EDAX profile of silver nanoparticles showed strong silver signals for silver atoms as shown in Figure 3. The EDAX pattern thus clearly shows that the silver nanoparticles are crystalline in nature by the reduction of silver ions by using *M. edule* leaf broth. The EDAX analysis obtained in the present study (shown in Figure 3F) confirmed that the presence of silver nanoparticles of *M. edule* and mostly showed strong signal energy peaks for silver atoms in the range 2–4 keV. In an earlier study, Gardea-Torresdey et al obtained formation of individual spherical-shaped silver nanoparticles in the range 2.5–4 keV by using *Alfalfa*.²³

The size distribution and characterization of silver nanoparticles could be further verified by FTIR in *M. edule* to synthesis nanoparticles. Figure 4 shows the adsorption band at about 1028.14, 1161.79, 1261.90, 1384.80, 1495.40, 1631.54, 1730.82 cm⁻¹. FTIR measurements were carried out to identify the silver nanoparticles in the *M. edule* leaf broth. Among them, observed adsorbance spectra band at 1028.14, 1384.80, and 1631.54 cm⁻¹ are more characteristic and mainly responsible for the bioreduction of silver ions.

Among them strong silver adsorbance band at 1384.80 was associated with the stretch vibration of functional groups such as –C–O–C–, –C–O–, –C=C, C=O. FTIR spectrum of the *M. edule* showed peaks indicating the presence of saponin (an important water soluble component) and this overall observation confirms the presence of saponin.

This present result obtained from *M. edule* agrees with those reported previously for *C. camphora*, *S. platensis*,^{21,37,17} Among them, the adsorbed peaks of silver nanoparticles in *M. edule* at 1631 cm⁻¹ were associated with the stretch vibration of –C=C, and the adsorbance peak as located at around 1028 cm⁻¹ can be assigned as the C–N stretching vibrations of amine. Similarly bioreduction of AgNO₃ by *Azadirachta indica* led to the formation of silver nanoparticles in the adsorption peak range observed at 1608 cm⁻¹, 1384 cm⁻¹, and 1076 cm⁻¹. Moreover, the above observed peaks of *M. edule* are more characteristic of terpenoid and flavonoid group components in having the ability to adsorb on the surface of silver nanoparticles. It plays a major role in the reduction of nanoparticles by interaction through carbonyl groups and in the oxidation of aldehyde groups in carboxylic acids.⁴² Our observation lends support to a previous study in which formation of spherical silver and triangular gold nanoparticles was reported by using various plant extracts. Further, the FTIR patterns of *M. edule* prove to be

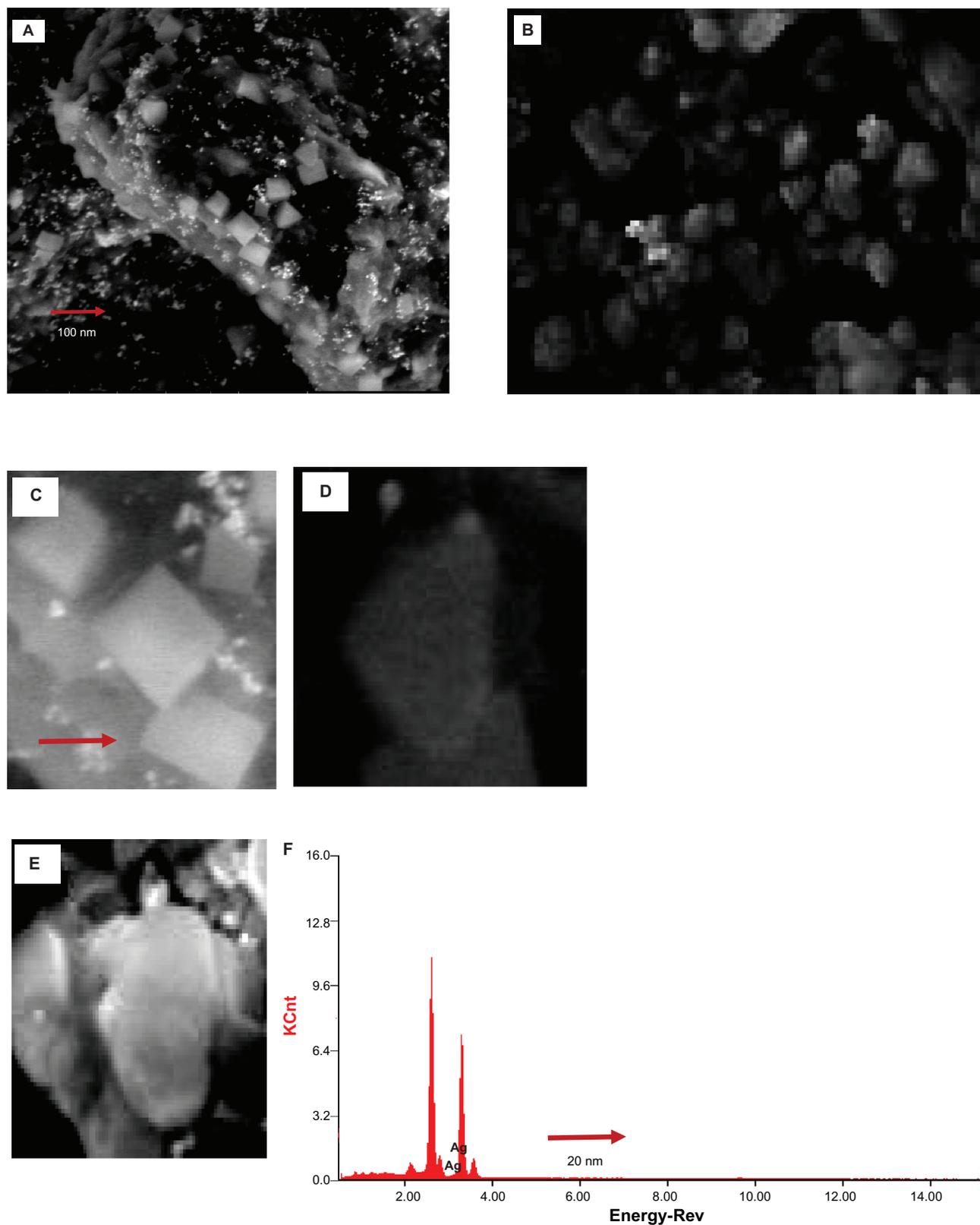


Figure 3 SEM images of the silver nanoparticles after bioreduction of AgNO_3 with *Memecylon edule* leaf extract. 100 nm (A), 50 nm (B), SEM images of silver nanoparticles with corresponding (SAED) selected area electron diffraction patterns showing square-shaped crystals (C), irregular-shaped particle (D), and triangular-shaped particle in 20 nm (E). EDAX profile of silver nanoparticles resulting from the experiment by using the 15 mL of *M. edule* extract. (F).

Abbreviations: SAED, selected area electron diffraction; EDAX, energy dispersive X-ray.

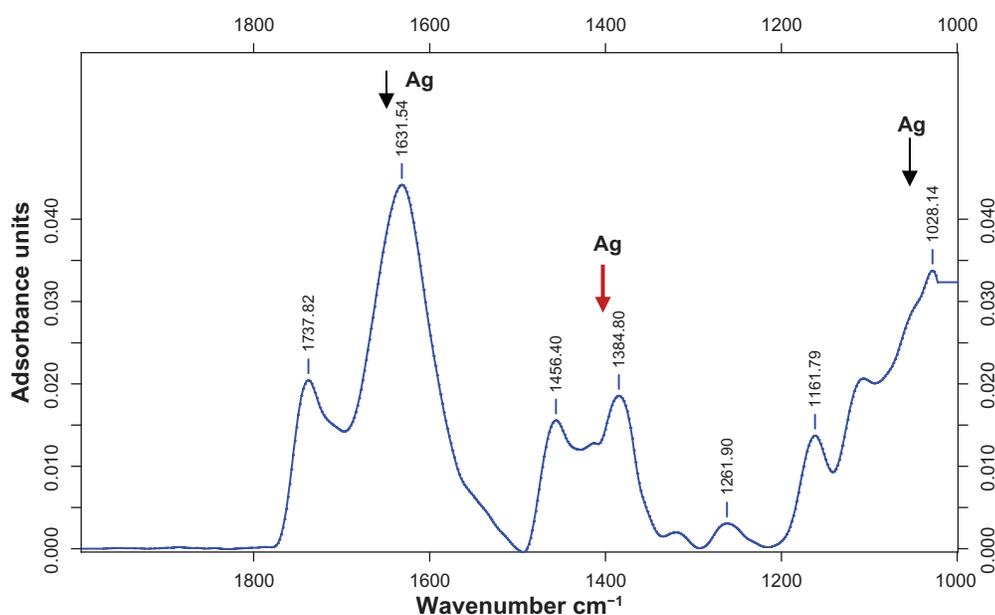


Figure 4 Typical FTIR adsorption spectra of silver nanoparticles synthesized by *M. edule* leaf extract after bioreduction of silver ions.
Abbreviation: FTIR, Fourier transform infra-red spectroscopy.

powerful evidence in favor of the UV-vis spectra and SEM images for the presence of silver nanoparticles. Although, as mentioned earlier, it has also been found to be responsible for the reduction of silver ions when using *Gliricidia sepium* for the synthesis of silver nanoparticles.²⁹

Green synthesis of gold nanoparticles

The reduction of chloroaurate ions was much more rapid than that of silver ions. On addition of *M. edule* biomass to 10^{-3} M aqueous HAuCl_4 , the color of the shaken solution changed from pale yellow to pink ruby red color, indicating formation of gold nanoparticles almost at the beginning of the reaction. These color changes arose due to excitation of SPR with the gold nanoparticles.

Figure 5 shows the UV-vis spectra recorded from the aqueous HAuCl_4 with *M. edule* reaction medium, as a function of time of reaction. The SPR band occurred at 555 nm within 1 hour of reaction time. The strong gold resonance peak occurred in 530 nm and increased in intensity as a function of time. The color did not change further after 18 hours of reaction time.

The bioreduction rate of the gold ions occurs quickly; more than 90% of the reduction of gold ions is complete in 18 to 21 hours after addition of the *M. edule* leaf broth to the gold solution. In earlier studies on the synthesis of gold nanoparticles using bacteria and fungi, the time required for complete reduction ranged from 24 to 120 hours. It is

generally accepted that UV-vis spectroscopy measurements could be used to check the reduction of the silver and gold nanoparticles. In earlier studies by Shankar et al and Dubay et al, it was reported that the gold nanoparticles obtained (after 13 hours) after the addition of tamarind leaf broth to the AgNO_3 , a visible color change was noticed, from colorless to yellowish-brown to ruby red, due to the excitation of SPR in gold nanoparticles.^{42,43} Figure 6 displays the SEM images of typical nanoparticles, produced by 15 mL of *M. edule* leaf extract with gold ions recorded at different magnifications. The resulting nanoparticles were polydisperse. Triangular or circular gold nanoparticles with an irregular edge can occur at 500, 200, 100, and 50 nm. Large uniform square-shaped particles produced by fast reduction of gold ions with 15 mL of *M. edule* extract in 200 nm are shown in Figure 6. An individual gold nanohexagonal tilting against the carbon-coated copper grid with an average edge length was measured to be about 85 nm. All the nanohexagonal shapes under low resolution were thin, flat, and sharp edged. According to size, distribution of the gold nanoparticles ranged from 20 to 35 nm with high frequency and possessed a square-shaped, plate-like morphology.

The sintering of some gold nanoparticles formed in *M. edule* were predominantly nanotetrahedral. Under careful observation, it was noted that the gold nanoparticles were surrounded by procircular shapes on the surface of the tetrahedral region. However, gold nanoparticles synthesized

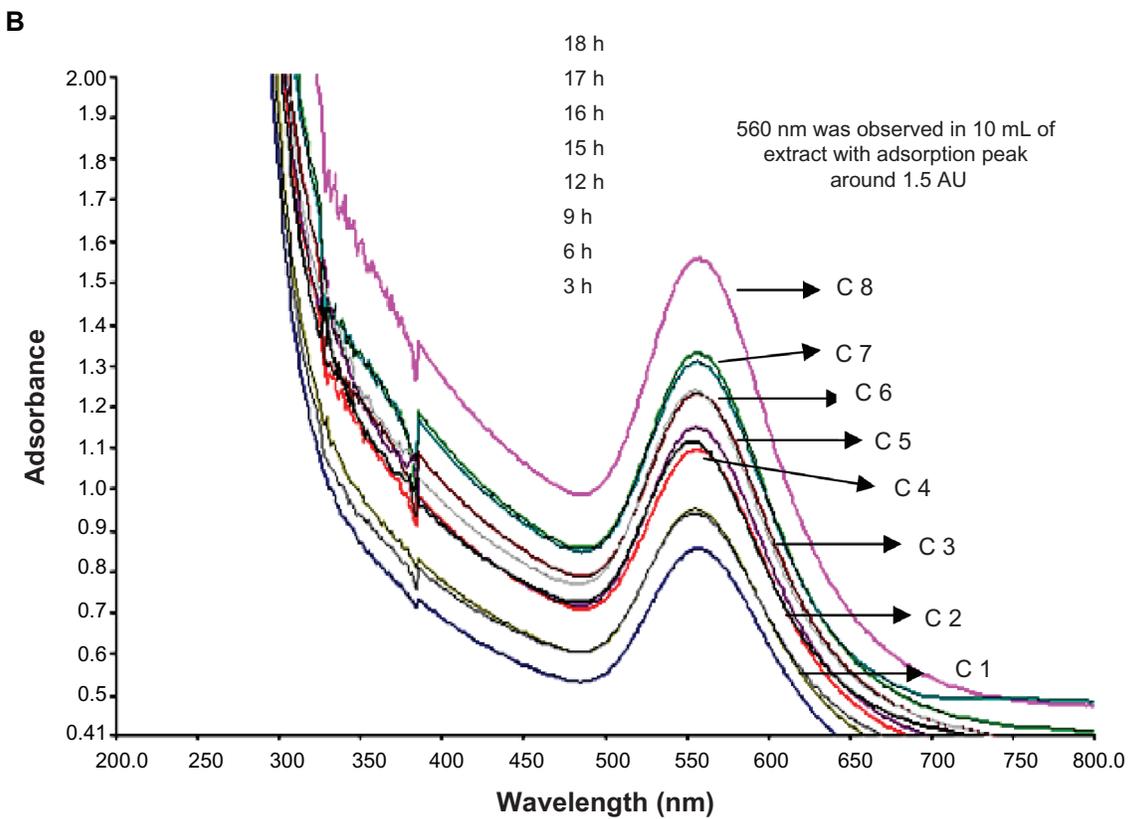
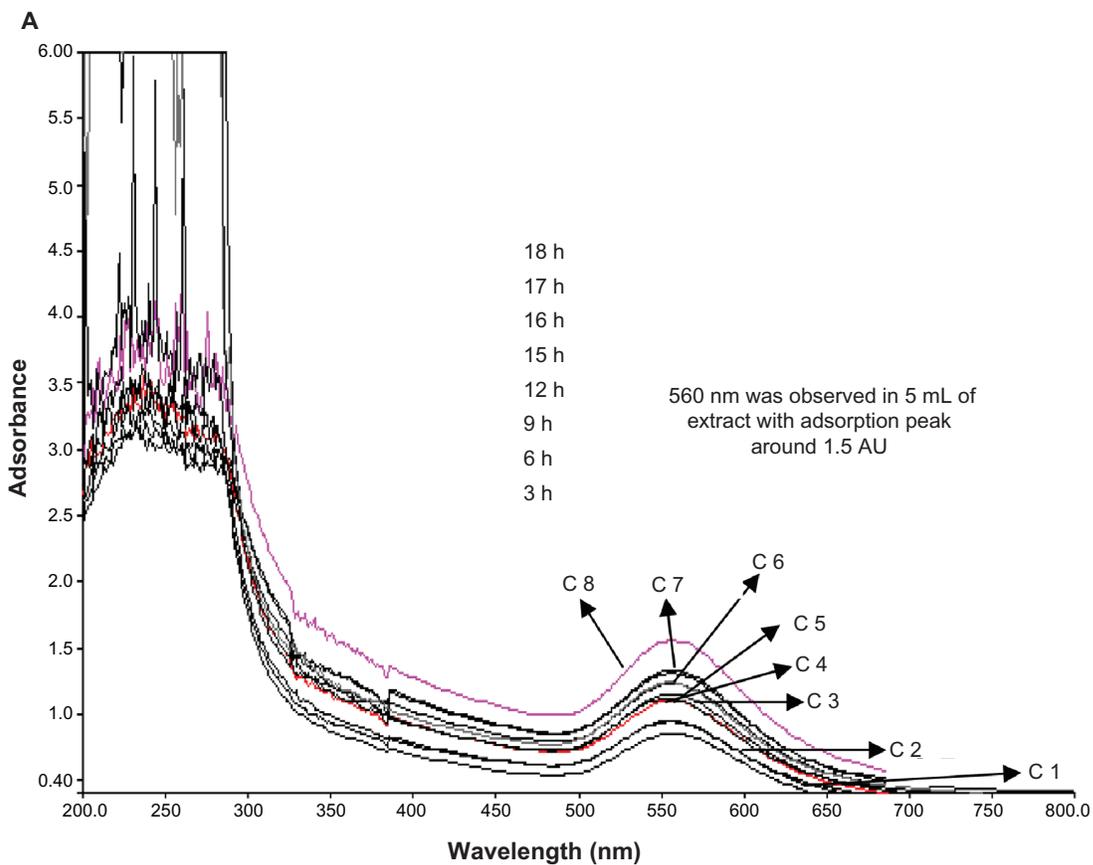


Figure 5 (Continued)

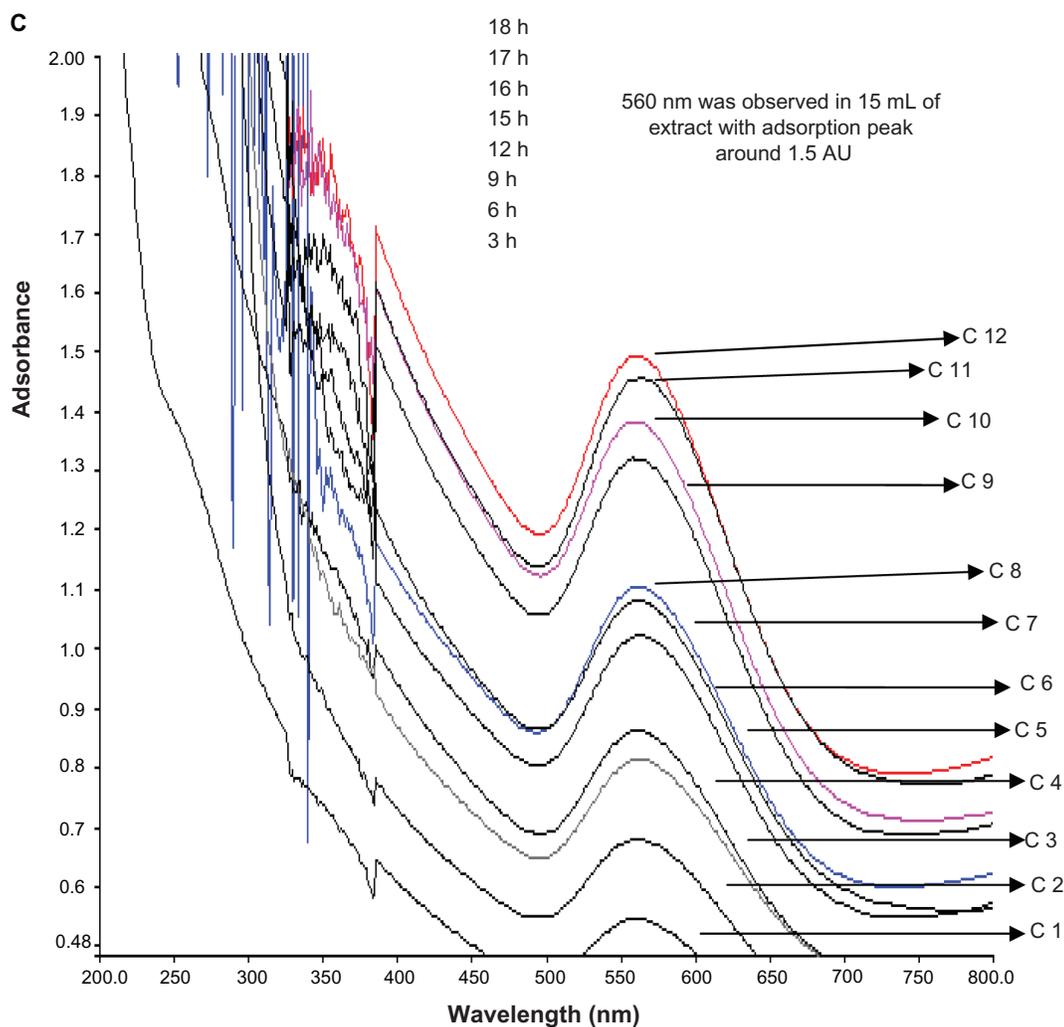


Figure 5 UV-vis bioreduction kinetics of the reaction of *M. edule* leaf extract with aqueous chloroaurate ions at 5, 10, and 15 mL concentration in the range of 200–800 nm with different time intervals. 5 mL of gold and leaf aqueous solution (A), 10 mL of gold and leaf aqueous solution (B), and 15 mL of gold and leaf aqueous solution (C). Curves 1–12 correspond to 3, 6, 9, 12, 15, 16, 17, 18, 21, 24, 27 hours of reaction time, respectively.

Abbreviation: C, curves.

using *M. edule* appear to have sharp angled edges on the surface of the particles. It is generally accepted that the shape of gold nanoparticles affects their optical, electronic, and catalytic properties.

From Rubio et al,⁴⁴ an experimental example from the solution growth of gold nanoparticles studied by scanning electron microscopy is presented in Figure 6, where icosahedral multiple twin-shaped gold nanoparticles were observed at a particle size of about 50–80 nm in the leaf broth of *M. edule*. Large numbers of nano spherical particles were observed, ranging in size from 20 to 50 nm, with an average particle size of 25.2 nm. The size distribution of the gold nanoparticles was narrow, mostly ranging from 55 to 80 nm.

Rejinolda et al reported that saponin-loaded nanoparticles were observed in the SEM with sizes of about 65–72 nm which are therapeutic for cancer. In the current study, experimental

evidence was obtained using SEM analysis, which clearly shows the strong gold nanoparticles in the above size range in the *M. edule* leaf extract.³⁸

Gold nanospindle-shaped particles were observed at higher adsorbance. As discussed above, the rate silver nanoparticle formation is relatively slow when compared with that of gold nanoparticles. Gold nanoparticles were formed on very small pieces of leaf of *M. edule* by a process involving rapid reduction. A low magnification TEM image recorded from the biologically synthesized gold nanoparticles at the end of the reaction with *M. edule* leaf extract showed that the gold nanoparticles were predominantly triangular, circular, and hexagonal in morphology. A large quantity of circular gold nanoparticles with thin, smooth ends on the exterior of the nanoparticles was seen in the TEM micrographs.

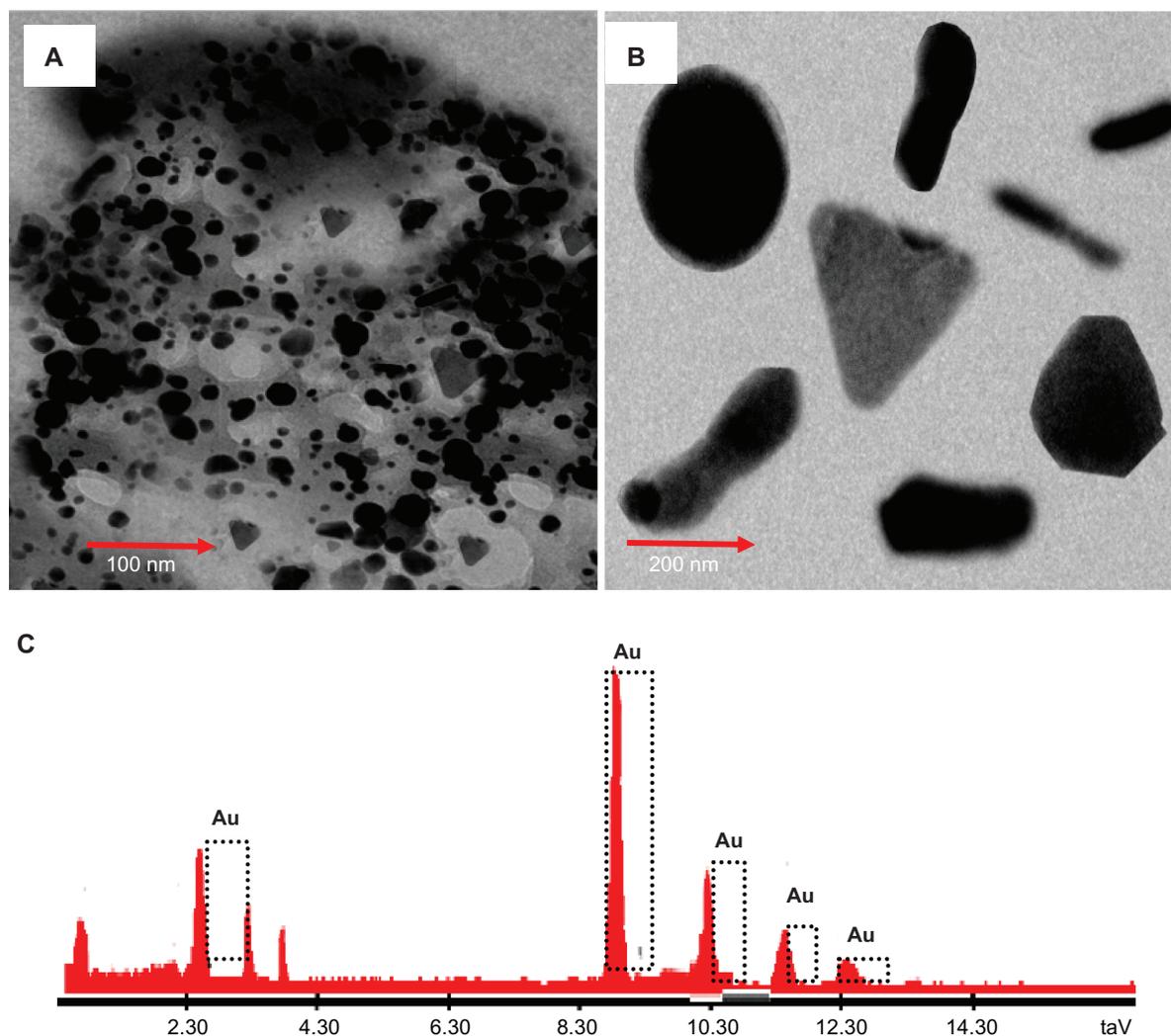


Figure 6 TEM images and EDAX spectrum of gold nanoparticles. Representative TEM image obtained by the reduction of aqueous AuCl_4^- ions using *Memecylon edule* leaf extract. Scale bars: 200 nm (A); high magnification TEM micrograph of a single gold nanotriangle, circular, spindle, hexagon (B); energy dispersive X-ray spectrum of gold nanoparticles resulting from the experiment using 15 mL of *M. edule* leaf extract (C).

The biologically obtained nanotriangles consisted of a mixture of triangles and truncated triangles. The overall morphology of the gold nanoparticles of the *M. edule* sample produced by bioreduction of gold ions with 15 mL of extract dosage is shown in Figure 7. Most of the gold nanoparticles ranged in size from 10 to 45 nm. At higher magnification, the average diameter of these nanoparticles was found to be approximately 24.3 nm.

Similar results have been reported for gold nanoparticles, although Shankar et al reported a nanoparticle size range of 50–100 nm and many of the observed structures of nanoparticles were spherical when *A. indica* leaf extract was used as a biomaterial.⁴² Most of the nanoparticles formed tended to be triangular or spherical in the observed hydrosol results at around 570 and 1000 nm adsorbance. These above

structures were similar to that of gold nanoparticles produced from the leaf extract of *C. camphora*.²¹

The spherical gold nanoparticles were further confirmed by using EDAX spectrometry, which confirmed the presence of gold with no other contaminants. The optical adsorption peak was observed at approximately 2.30 keV, which is typical for the adsorption of gold nanocrystallites due to SPR.

The current profile of EDAX of gold nanoparticles of *M. edule* showed strong gold atom signals around 2.30, 9, 10.30, 11.30, and 12.30 keV. The EDAX of gold nanoparticles showed strong signals for gold atoms. All the gold nanoparticles under EDAX observation of *M. edule* show similar results to other plants including coriander and *C. camphora*.^{20–22}

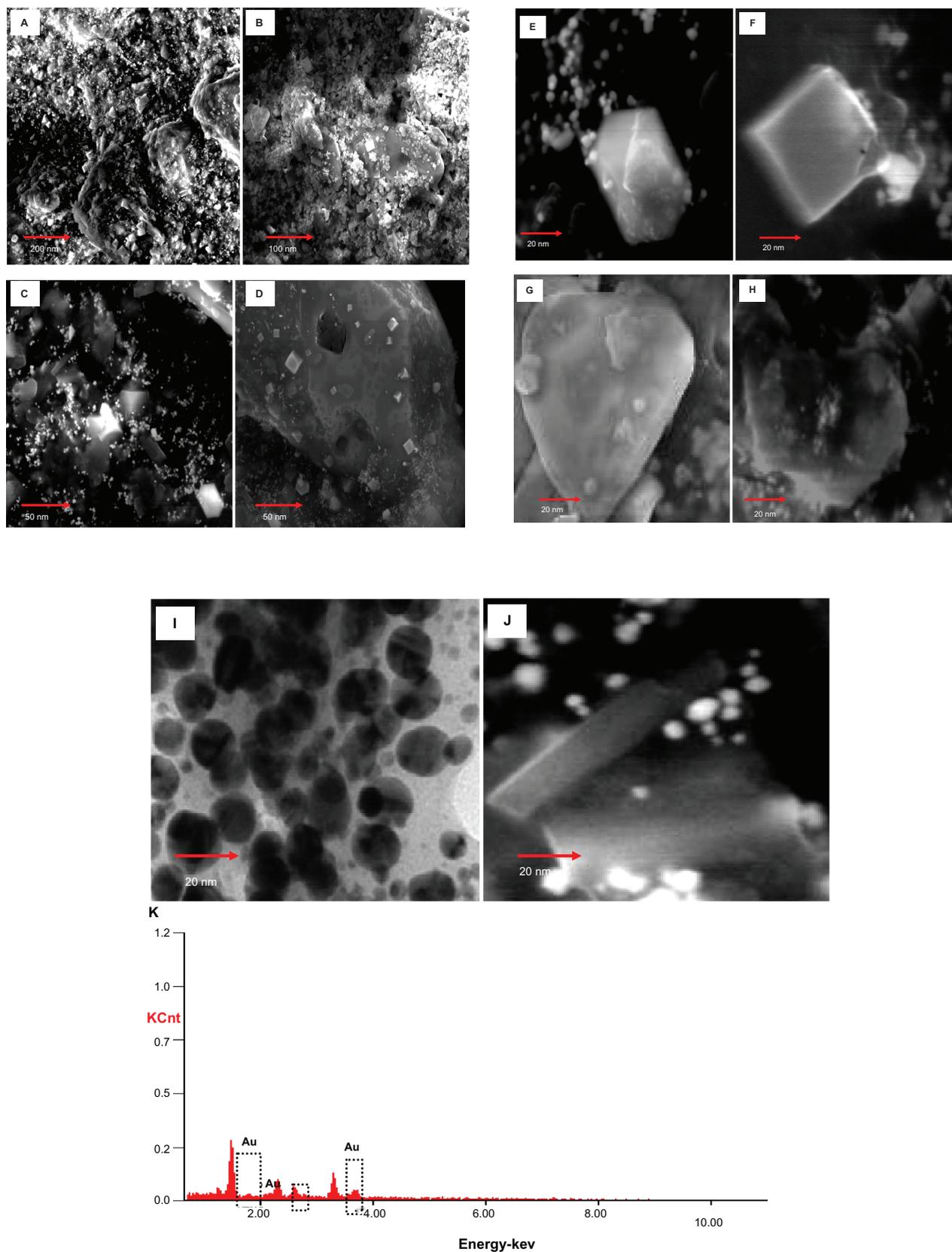


Figure 7 SEM micrographs of the different kinds of gold nanoparticles observed from the reaction of Au^{3+} cations with *Memecylon edule* leaf extract: scale bar 500 nm (A), 200 nm (B), 100 nm (C), 50 nm (D). SEM images of gold nanoparticles with corresponding selected area electron diffraction (SAED) patterns of the nanocrystalline *M. edule* samples at 20 nm. Hexagonal (E), square (F), tetrahedral (G), icosahedral multiple twinned (H), circular (I) spindle (J), energy dispersive X-ray spectrum equipped with SEM of gold nanoparticles resulting from the experiment using 15 mL of *M. edule* leaf extract (K).

Most of the previous research has demonstrated that the reduction of chloroaurate ions was much faster than that of silver ions and arrangement of atoms inside the gold nanoparticles was disordered in the size range 25–85 nm. FTIR clearly showed that more functional groups could be available for a large number of Au(III) complexes with the plant extract simultaneously. The lower wavenumbers, followed by the disappearance of the gold resonance, may be due to its binding with the gold nanoparticle surface.^{40,44}

Figure 8 shows the FTIR measurements carried out to identify the functional groups involved in the bioreduction of gold nanoparticles in the *M. edule* leaf broth. Representative FTIR spectra of the obtained nanoparticles are also shown in this figure. The spectrum reduced gold with the adsorption peak located at about 1618.93, 1452.80, 1378.82, 1384.80, 1319.69, 1166.01, 1112.15, 1056.43 cm^{-1} , in the region of 1000 to 1800 cm^{-1} . These bands are characteristic of carbonyl stretching vibration in ketones, aldehydes, and carboxylic acids. Three adsorption peaks located at around 1618 cm^{-1} (C=C or aromatic groups), 1452 cm^{-1} (geminal methyls), and 1056 cm^{-1} (aromatic C–C) were observed for gold nanoparticles. In addition, there were some strong adsorption spectra in the region 1600–1700 cm^{-1} .

In earlier studies, Govindaraju stated that FTIR spectra of the synthesized gold nanoparticles of *S. platensis* showed gold ion peaks around 1653, 1541, and 1242 cm^{-1} .³⁷ For gold

nanoparticles, the adsorption at about 1725, 1615, 1401, 1228, 1140, and 1076 cm^{-1} has been found to be responsible for the reduction of gold ions when using *A. indica*.⁴¹ Present FTIR observation of *M. edule* lends support to a previous study on the reduction of gold nanoparticles, using *C. camphora*, in the gold ion peak region at 1038, 1387, and 3418 cm^{-1} , corresponding to C–N stretching vibrations of amine, C–N stretching of aromatic amine group, and NH stretching of amide (II) band.^{22,43,44} The result obtained in the present study is supported by previous reports.^{45,46} FTIR spectrum of tamarind leaf broth-reduced gold nanoparticles shows the adsorption bands at 1744 cm^{-1} and 1616 cm^{-1} .⁴⁰ The same observed gold peaks of *M. edule* were more characteristic of terpenoids.^{45–49}

Conclusion

We have demonstrated the *M. edule* could be an excellent bioreductant and easily available plant source for green synthesis of silver and gold nanoparticles. The *M. edule* aqueous leaf extract appears to be environmentally friendly and therefore this protocol could be used for the rapid production of silver nanoparticles. The size of silver and gold particles could be easily adjusted by using different amounts of leaf extract.

The successful synthesis of gold and silver nanoparticles by reducing silver and gold ions using an aqueous extract of

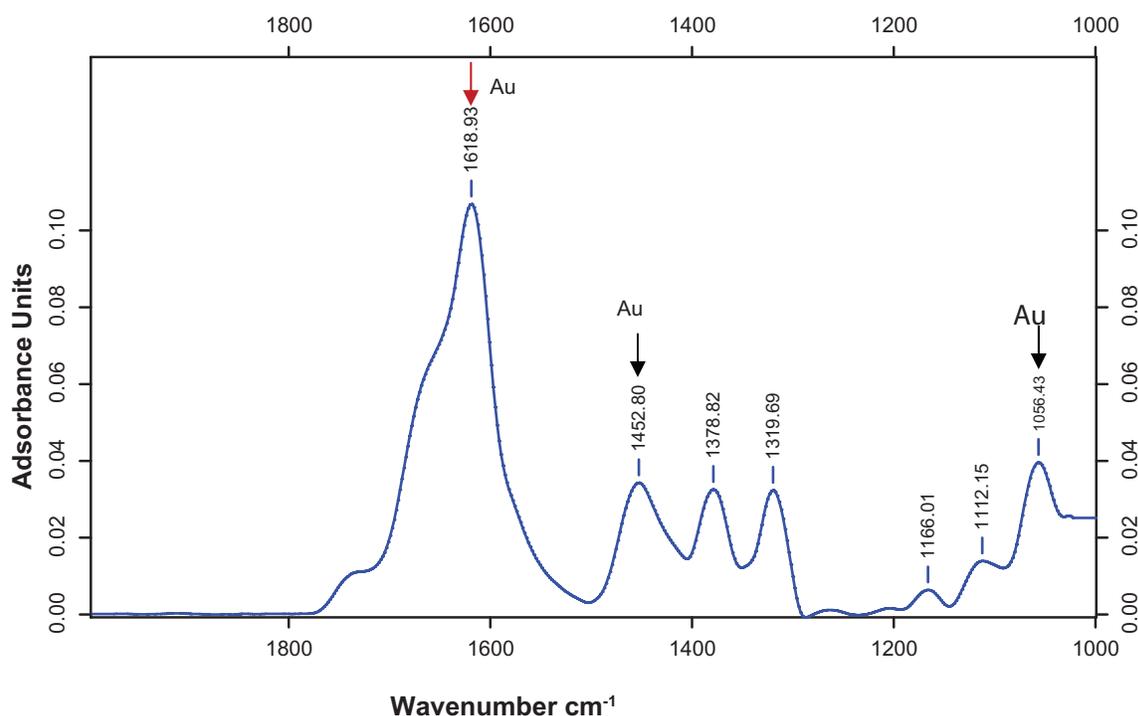


Figure 8 Typical FTIR adsorption spectra of the nanoparticles synthesized by bioreduction of AuCl_4^- ions by *M. edule* leaf broth.

M. edule leaves, showed that the reduction rate of silver ions is much faster than for gold. Water soluble compounds present in the *M. edule*, like saponins, were mostly responsible for the reduction of silver and gold ions to nanosized silver and gold particles. The reduction rate of HAuCl₄ also varied and color intensity was changed due to the formation of gold nanoparticles. The reaction of the ingredients present in the plant leaf extract analyzed by UV-vis spectroscopy revealed that silver nanoparticles in the solution may be correlated with the UV-vis spectra. This change of color indicates the formation of silver nanoparticles.

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Disclosure

The authors declare no conflicts of interest in this work.

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