

# Comparative study of proteasome inhibitory, synergistic antibacterial, synergistic anticandidal, and antioxidant activities of gold nanoparticles biosynthesized using fruit waste materials

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**Abstract:** The aim of this study was to compare the biological synthesis of gold nanoparticles (AuNPs) generated using the aqueous extracts of outer oriental melon peel (OMP) and peach. The synthesized OMP-AuNPs and peach extract (PE)-AuNPs were characterized by ultraviolet-visible spectroscopy, field emission scanning electron microscopy, energy dispersive X-ray analysis, X-ray powder diffraction, Fourier transform infrared spectroscopy, and thermogravimetric analysis. The surface plasmon resonance spectra were obtained at 545 nm and 540 nm for OMP-AuNPs and PE-AuNPs, respectively. The estimated absolute crystallite size of the synthesized AuNPs was calculated to be 78.11 nm for OMP-AuNPs and 39.90 nm for PE-AuNPs based on the Scherer equation of the X-ray powder diffraction peaks. Fourier transform infrared spectroscopy results revealed the involvement of bioactive compounds present in OMP and peach extracts in the synthesis and stabilization of synthesized AuNPs. Both the OMP-AuNPs and PE-AuNPs showed a strong antibacterial synergistic activity when combined with kanamycin (9.38–20.45 mm inhibition zones) and rifampicin (9.52–25.23 mm inhibition zones), and they also exerted a strong synergistic anticandidal activity (10.09–15.47 mm inhibition zones) when combined with amphotericin B against five pathogenic *Candida* species. Both the OMP-AuNPs and PE-AuNPs exhibited a strong antioxidant potential in terms of 1,1-diphenyl-2-picrylhydrazyl radical scavenging, nitric oxide scavenging, 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid) radical scavenging, and a reducing power, along with a strong proteasome inhibitory potential that could be useful in cancer drug delivery and cancer treatments. The PE-AuNPs showed comparatively higher activity than OMP-AuNPs, which could be attributed to the presence of rich bioactive compounds in the PE that acted as reducing and capping agents in the synthesis of PE-AuNPs. Overall, the results of the current investigation highlighted a novel green technology for the synthesis of AuNPs using food waste materials and their potential applications in the biomedical, pharmaceutical, and cosmetic industries.

**Keywords:** antibacterial, anticandidal, antioxidant, *Cucumis melo*, gold nanoparticles, proteasome inhibitor, *Prunus persica*

## Introduction

Nanotechnology is a field of applied science and technology that deals with materials in the nanoscale range with large surface areas.<sup>1</sup> These nanoscale particles, which are most commonly known as nanoparticles (NPs), have the ability to build micro-/macro-materials and products with atomic precision. NPs have recently drawn tremendous attention because of their potential applications in chemical, catalytic, electronic, optical,

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mechanical, magnetic, and medical fields.<sup>2-4</sup> NPs are usually synthesized by chemical or physical approaches because of their inherent advantage in producing well-defined NPs with fairly controllable shapes and sizes.<sup>5</sup> However, these methods involve tedious treatments, such as microemulsion, laser photolysis, hydrothermal induction, and toxic chemical reduction, along with expensive physical techniques and equipment, and are potentially toxic to the environment.<sup>6-8</sup> Thus, to avoid the adverse effects of NPs on the environment, the concept of green chemistry evolved. This method utilizes biological organisms for the fabrication of NPs.

In general, the synthesis of NPs using biological organisms such as microorganisms, plants or plant extracts, algae, and fungi is environmentally benign and cost-effective.<sup>9</sup> However, there are certain drawbacks in the synthesis of NPs using microorganisms because of difficulties in separation and purification procedures.<sup>10</sup> In contrast, plant biomass-mediated biosynthesis of NPs has recently attracted a great deal of attention because of their easy synthesis and extraction processes.<sup>11</sup> Naturally available biological, food, and agricultural waste materials have not been extensively investigated for the synthesis of different types of NPs.

Oriental melons and peaches are commonly consumed in East Asia. Basically, the juicy internal portions of both fruits are edible, while the outer peels are generally discarded and considered useless. The oriental melon (*Cucumis melo* L. var. *makuwa*) that belongs to the family Cucurbitaceae is primarily grown in South Korea, People's Republic of China, and Japan.<sup>12</sup> The word melon is derived from the Latin name "melo" meaning melon. The fruit is ~10 cm long and typically weighs slightly more than one pound (0.45 kg).<sup>12</sup> This fruit has been utilized in Korean folk medicine for treatment of acute gastritis, fever, mental disorders, dysuria, jaundice, alcoholism, and hyperesthesia/paralysis.<sup>13,14</sup> Peach (*Prunus persica* L. Batsch) that belongs to the family Rosaceae is a fruit of the deciduous tree native to Northwest China, between the Tarim basin and the northern slopes of the Kunlun Shan mountains.<sup>15</sup> This fruit is cultivated in many countries including the People's Republic of China, Japan, South Korea, and Vietnam. Peaches are considered as important economic crops with potential bioactive compounds and medicinal benefits.<sup>16,17</sup> The fruit is rich in a number of phenolic compounds, including chlorogenic acid, catechin, epicatechin, rutin, and cyanidin-3-glucoside,<sup>18</sup> and is utilized as a demulcent, antiscorbutic, and a stomachic agent.<sup>19</sup>

NPs, especially gold, have recently attracted a great deal of attention due to their nontoxic nature and extensive

applications in biomedical, chemistry, and electronics fields.<sup>20,21</sup> Gold NPs (AuNPs) can be easily synthesized and exhibit intense surface plasmon resonance (SPR) with high chemical and thermal stability.<sup>21</sup> Investigation of the morphological behavior of gold nanostructures is important because of their wide usage in catalysis, optics, optical electronics, microelectronics, biodiagnostics, imaging, and biological and chemical sensing techniques.<sup>21,22</sup> Biomedical applications of AuNPs such as drug and gene delivery, protein and pathogen detection, deoxyribonucleic acid labeling, fluorescent labeling, tissue engineering, photothermal ablation, and as contrast agents for magnetic resonance imaging and other imaging methods have become a highly active area of research during recent years.<sup>23,24</sup>

Many studies have investigated the biosynthesis of AuNPs using various plant extracts.<sup>25-28</sup> However, only a few studies have investigated the synthesis of AuNPs using food waste materials.<sup>29,30</sup> In the current study, the reducing ability of aqueous extracts of the outer peels of two common fruits, oriental melon and peach, were investigated to determine their usefulness for the synthesis of AuNPs. In addition, their antibacterial, anticandidal, antioxidant, and proteasome inhibitory potentials were evaluated.

## Materials and methods

### Preparation of aqueous extract of fruit waste materials

The fruits of oriental melon (*Cucumis melo* L. var. *makuwa*; Figure 1A) and peach (*Prunus persica* L.; Figure 1B) were purchased from a local market (Gyeongsan, South Korea). The fruits were washed thoroughly with double-distilled water and dried with tissue paper, and the outer nonedible portion of the fruits was peeled off using a peeler. The outer peels were subsequently cut into small pieces (~10 mm) with a knife, divided into 50 g aliquots, and immersed in 250 mL of deionized water in two separate 500 mL conical flasks.



**Figure 1** Fruit peels of oriental melon (*Cucumis melo*) (A) and peach (*Prunus persica*) (B) used for the synthesis of AuNPs.

**Abbreviation:** AuNPs, gold nanoparticles.

The mixtures were subsequently boiled for 15 minutes with continuous stirring using a magnetic stirrer, after which the aqueous oriental melon peel (OMP) extract and the aqueous peach extract (PE) were cooled to room temperature. Both the OMP and peach extracts were then filtered through Whatman No 1 filter paper, collected into separate sterilized bottles, and kept at 4°C until further use.

## Biosynthesis of AuNPs using OMP and peach extracts

Auric chloride ( $\text{AuCl}_3$ ; Sigma-Aldrich Co., St Louis, MO, USA) was used as the precursor element for the synthesis of AuNPs. Prior to the synthesis, 1 mM  $\text{AuCl}_3$  solution was prepared in double-distilled water in an amber bottle. Approximately 100 mL of  $\text{AuCl}_3$  solution was then placed into two separate 500 mL conical flasks, after which 10 mL of OMP extract was added to one flask and 10 mL of PE to another flask dropwise using a separating funnel. Samples were then incubated at room temperature with continuous stirring for 24 hours.<sup>31</sup> The bioreduction and color change of  $\text{AuCl}_3$  to AuNPs was then monitored with respect to the incubation time. After 24 hours, the synthesized AuNP solution (OMP-AuNPs and PE-AuNPs) was centrifuged at 12,000 rpm for 30 minutes in a high-speed centrifuge machine (Supra 22K; Hanil Science Industrial Co., Ltd., Daejeon, South Korea). The supernatant was then discarded, and the pellets were washed twice with distilled water and dried to powder using a vacuum dryer (LVS 201T; ILMVAC GmbH, Ilmenau, Germany).

## Characterization of the synthesized AuNPs

The synthesized OMP-AuNPs and PE-AuNPs were characterized by ultraviolet (UV)–visible spectroscopy, field emission scanning electron microscopy (FE-SEM), energy dispersive X-ray (EDX) analysis, X-ray powder diffraction (XRD), Fourier-transform infrared (FT-IR) spectroscopy, and thermogravimetric (TG) analysis.<sup>31,32</sup>

The bioreduction of Au ions to AuNPs in the reaction medium by the action of OMP and peach extracts was monitored by measuring the absorption spectra of the colloidal solution between 400 nm and 700 nm using a microplate reader (Infinite 200 PRO NanoQuant; Tecan, Mannedorf, Switzerland). The surface morphology of both OMP-AuNPs and PE-AuNPs was analyzed using an FE-SEM. Prior to the analysis, the OMP-AuNPs and PE-AuNPs were ground into fine powder using an agate mortar and pestle and then uniformly spread on the carbon tapes attached over

the sample holder. The samples were subsequently sputter coated with platinum in an ion coater for 120 seconds, after which they were observed using a FE-SEM (S-4200; Hitachi Ltd., Tokyo, Japan). The elemental composition of the OMP-AuNPs and PE-AuNPs was determined using an EDX detector (energy dispersive spectroscopy; EDAX Inc., Mahwah, NJ, USA) attached to the SEM machine. The nature of the powdered OMP-AuNPs and PE-AuNPs was determined using an XRD machine (X'Pert MRD model; PANalytical, Almelo, the Netherlands). The NPs were dried at 60°C in a vacuum oven and ground into fine powder using an agate mortar and pestle. XRD analysis was performed at 30 kV and 40 mA with  $\text{Cu K}\alpha$  radians ( $\lambda=1.5406 \text{ \AA}$ ) at an angle of  $2\theta$ , while scanning from 20° to 90°.<sup>31</sup> FT-IR analysis of OMP, PE, powdered OMP-AuNPs, and powdered PE-AuNPs was conducted using an FT-IR spectrophotometer (Jasco 5300; Jasco, Easton, MD, USA) in the wavelength range of 400–4,000  $\text{cm}^{-1}$ . Prior to use, the powdered OMP-AuNPs and PE-AuNPs were blended with powdered potassium bromide (KBr) in a 1:100 ratio, then compressed into a 2 mm semi-transparent disk to prepare a thin film of NPs. The OMP and peach extracts were used directly for FT-IR analysis. Both the NPs and OMP and peach extracts were analyzed for various modes of vibrations for the presence of different types of functional groups. The TG/derivative TG (DTG) analysis of the powdered OMP-AuNPs and PE-AuNPs was conducted using a TG analysis machine (SDT Q600; TA Instruments, New Castle, DE, USA). Prior to use, the powdered OMP-AuNPs and PE-AuNPs were placed in an alumina pan and heated from 20°C to 700°C at a ramping time of 10°C/min under nitrogen atmosphere.

## Synergistic antibacterial activity of synthesized AuNPs

The synergistic antibacterial potential of both OMP-AuNPs and PE-AuNPs was determined against five different foodborne bacteria, namely *Bacillus cereus* American Type Culture Collection (ATCC) 13061, *Listeria monocytogenes* ATCC 19115, *Staphylococcus aureus* ATCC 49444, *Escherichia coli* ATCC 43890, and *Salmonella typhimurium* ATCC 43174, using the standard disk diffusion method.<sup>33,34</sup> The pathogenic bacteria were obtained from the American Type Culture Collection (ATCC, Manassas, VA, USA) and maintained in nutrient broth (Difco; Becton, Dickinson and Company, Sparks Glencoe, MD, USA) medium at –80°C. Immediately before use, a colloidal solution of OMP-AuNPs and PE-AuNPs was prepared by dissolving 1,000  $\mu\text{g/mL}$  in 5% dimethyl sulphoxide (DMSO), then sonicated at 30°C

for 15 minutes. Filter paper disks containing 25 µg of the test compound per disk were used for the initial antibacterial activity test. OMP and peach extracts were taken as extract controls, and kanamycin and rifampicin at 5 µg/disk were taken as positive controls, while 5% DMSO was used as the negative control. To explore the synergistic effects of the AuNPs, 1,000 µg/mL of OMP-AuNPs or PE-AuNPs were mixed in a 1:1 ratio with 200 µg/mL of the two standard antibiotics separately and then sonicated for 15 minutes at room temperature. A total of 50 µL of the mixture of AuNP–antibiotics containing 25 µg AuNPs and 5 µg antibiotics were put on the filter paper disk, dried, and used for determination of the synergistic antibacterial effect. The diameter of zones of inhibition around each paper disk was measured after 24 hours of incubation at 37°C to determine the synergistic antibacterial effect of the mixture of AuNPs/antibiotics.

### Synergistic anticandidal activity of synthesized AuNPs

The synergistic anticandidal activity of OMP-AuNPs or PE-AuNPs with a standard anticandidal agent, amphotericin B, was determined against five different pathogenic *Candida* species, namely *Candida albicans* KACC 30003 and KACC 30062, *Candida glabrata* KBNO6P00368, *Candida geochares* KACC 30061, and *Candida saitoana* KACC 41238, using the disk diffusion method.<sup>35</sup> All *Candida* strains were obtained from the Korean Agricultural Culture Collection (KACC, Suwon, South Korea), except for *C. glabrata* KBNO6P00368, which was obtained from Chonbuk National University Hospital (Cheongju, South Korea). Filter paper disks containing 50 µg/disk of AuNPs were initially tested for their anticandidal activity. To explore the synergistic anticandidal activity, the OMP-AuNPs or PE-AuNPs (2,000 µg/mL) were mixed at a 1:1 ratio with amphotericin B (200 µg/mL) separately and then sonicated for 15 minutes at room temperature. Filter paper disks were prepared by adding 50 µL of the AgNPs/amphotericin B mixture solution to the sterile paper disks, which finally contained 50 µg AuNPs and 5 µg amphotericin B/disk. Plates of potato dextrose agar (Difco) were spread uniformly with the *Candida* species grown in potato dextrose broth, after which the filter paper disks were placed over it and incubated at 28°C for 48 hours. OMP and peach extracts were taken as extract controls, amphotericin B at 5 µg/disk were taken as positive control, while 5% DMSO was used as the negative control. Finally, the diameters of the zones of inhibition around each paper disk were measured to determine the synergistic anticandidal activity of the mixture of AuNPs/amphotericin B.

### Antioxidant activity of synthesized AuNPs

The in vitro antioxidant potential of the AuNPs was determined by four antioxidant assays, including 1,1-diphenyl-2-picrylhydrazyl (DPPH) free radical scavenging, nitric oxide scavenging, 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid) (ABTS) free radical scavenging, and a reducing power assay. The OMP-AuNPs, PE-AuNPs, and butylated hydroxyl toluene (BHT), which was taken as the standard reference compound, were used at 20–100 µg/mL for all assays. The DPPH free radical scavenging potential of both OMP-AuNPs and PE-AuNPs was determined as described by Braca et al.<sup>36</sup> The absorbance of the reaction mixtures was recorded at 517 nm using a microplate reader, and the results in terms of the percentage scavenging effect were calculated using the following equation:

$$\% \text{scavenging} = \frac{\text{Abs}_c - \text{Abs}_t}{\text{Abs}_c} \times 100 \quad (1)$$

where Abs<sub>c</sub> is the absorbance of the control and Abs<sub>t</sub> is the absorbance of the treatment.

The nitric oxide radical scavenging potential of both OMP-AuNPs and PE-AuNPs was determined by the standard method as described by Makhija et al.<sup>37</sup> The absorbance of the reaction mixtures was measured at 546 nm using a microplate reader, and the results in terms of the percentage scavenging effect were calculated according to Equation 1. The ABTS radical scavenging activities of OMP-AuNPs and PE-AuNPs were determined as described by Thaipong et al.<sup>38</sup> The absorbance of the reaction mixtures was recorded at 750 nm using a microplate reader, and the results in terms of the percentage scavenging effect were calculated according to Equation 1. The reducing powers of the OMP-AuNPs and PE-AuNPs were determined using the procedure described by Patra and Baek.<sup>39</sup> The reducing power of AuNPs was determined in terms of the absorbance value at 700 nm.

### Proteasome inhibitory activity of synthesized AuNPs

The proteasome inhibitory potentials of OMP-AuNPs and PE-AuNPs were assayed using the 20S Proteasome Assay Kit for drug discovery (ENZO Life Sciences, Farmingdale, NY, USA) according to the manufacturer's protocols. Three different concentrations of both OMP-AuNPs and PE-AuNPs (1 µg/mL, 10 µg/mL, and 100 µg/mL) and epoxomicin, a standard reference compound (0.56 µg/mL, 2.78 µg/mL, and 5.56 µg/mL) were used for the assay. Suc-LLVY-AMC (37.5 mM) was used as the substrate to estimate the

chymotrypsin-like proteasomal activity of the AuNPs in a 96-well microplate from 1 minute to 12 minutes of incubation. The fluorescence intensities were measured at an excitation wavelength of 350 nm and emission wavelength of 440 nm at a 30°C reaction temperature using the microplate reader.

## Statistical analysis

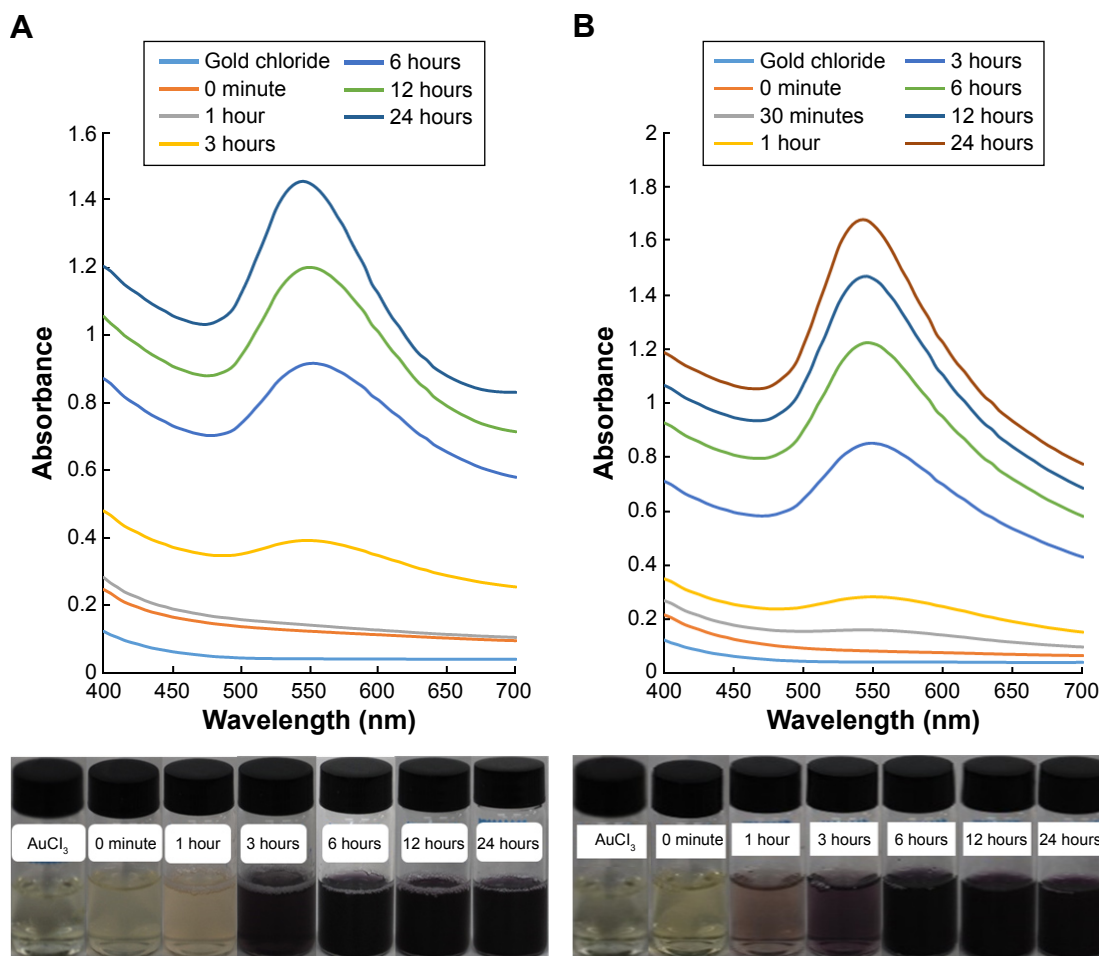
All data were calculated as mean  $\pm$  SD. Statistical analysis using one-way ANOVA followed by Duncan's test at the 5% level of significance ( $P < 0.05$ ) was conducted using the Statistical Analysis Software (SAS, Version: SAS 9.4; SAS Institute Inc., Cary, NC, USA).

## Results and discussion

### Synthesis and characterization of AuNPs

There have been many reports on the use of whole fruits in various biological activities and medicinal potentials; however, only a few studies have investigated the importance

of their peels. Fruit peels are rich in polyphenolic compounds, flavonoids, ascorbic acid, and a number of other biologically active components that have positive effects on health.<sup>40-43</sup> In the recent past, there have been reports on the pharmacological effects of some fruit peel extracts against diseases such as cardiovascular problems, thyroid abnormalities, and diabetes mellitus.<sup>29,43,44</sup> Thus, we investigated the synthesis of AuNPs utilizing the fruit peels of two fruits common in East Asia, oriental melon and peach, and then compared their potential applications. AuNPs were synthesized by taking AuCl<sub>3</sub> as the precursor element and both OMP and peach extracts as the reducing agents (Figure 1). The bioreduction of Au<sup>+</sup> to Au<sup>0</sup> (AuNPs) first started vigorously after 1 hour of incubation in the PE-mediated synthesis compared to the OMP-mediated AuNP synthesis, during which time there was a gradual change in the color of the reaction mixture from colorless to purple (Figure 2, inset). The solution was dark purple at 6 hours of incubation in both the cases; however, the



**Figure 2** UV-visible spectroscopy and change in color of the reaction mixture at different time intervals of the synthesized OMP-AuNPs and PE-AuNPs.

**Notes:** (A) OMP-AuNPs; (B) PE-AuNPs.

**Abbreviations:** AuNPs, gold nanoparticles; OMP, oriental melon peel; PE, peach extract; UV, ultraviolet.

reaction continued for up to 24 hours until the complete reduction of  $\text{Au}^+$  in the reaction mixture (Figure 2, inset).

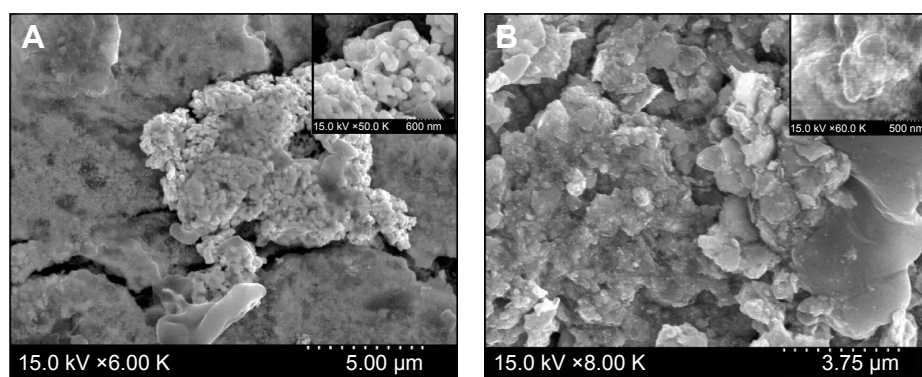
The dark purple color unveiled by the metallic colloidal AuNPs was due to the articulate excitation of all of the free electrons within the conduction band that led to the in-phase oscillation referred to as the SPR.<sup>45</sup> The bioactive compounds present in both fruit peels might have been responsible for the bioreduction of  $\text{Au}^+$  to AuNPs. The oriental melon is a rich source of vitamin A, vitamin C, 5-hydroxy-6,7,8-trimethoxy-2,3-dimethyl-4*H*-chromen-4-one, and anhalinine;<sup>46,47</sup> therefore, it has potential for use as a reducing agent to synthesize OMP-AuNPs. Similarly, peach is rich in phenolic compounds, including chlorogenic acid, catechin, epicatechin, rutin, and cyanidin-3-glucoside;<sup>18</sup> therefore, it might be useful as a reducing agent to synthesize PE-AuNPs. Following synthesis, the OMP-AuNPs and PE-AuNPs are subjected to characterization of their structure, morphology, and chemical properties based on UV-visible spectroscopy, FE-SEM, EDX, XRD, FT-IR, and TG/DTG analysis.

UV-visible spectroscopy is a preliminary step in the confirmation of the synthesis and stability of AuNPs with dark purple coloration in aqueous reaction medium. The UV-visible spectra of both OMP-AuNPs and PE-AuNPs were recorded with an interval of 3 hours for 24 hours from 400 nm to 700 nm, and the results are presented in Figure 2A and B. The PE-AuNPs displayed sharp peaks with the SPR absorbance maxima at 540 nm (Figure 2B), whereas the OMP-AuNPs displayed a small flat peak with the SPR absorbance maxima at 545 nm (Figure 2A). Both sets of spectral data confirmed the synthesis of AuNPs, which were concordant with the results of earlier studies.<sup>48</sup> In general, AuNPs have been reported to exhibit a dark purple color in aqueous medium due to the intensity and size of the NPs owing to its SPR.<sup>48,49</sup> The SPR absorbance is exceptionally

sensitive to the nature, size, and shapes of the NPs formed, dielectric constant of the medium and temperature, and their interparticle distances.<sup>50–52</sup>

The surface morphology of the synthesized AuNPs was characterized by FE-SEM after 1 month of synthesis following completion of all reactions. Both the OMP-AuNPs and PE-AuNPs displayed slightly agglomerated distinct spherical structures within the nanoscale range that confirmed the synthesis of NPs (Figure 3A and B). Similar observations were reported in an earlier study.<sup>28,53</sup> The elemental composition of the synthesized NP was confirmed by EDX analysis (Figure 4A–F). The results showed that OMP-AuNPs and PE-AuNPs contained 73.87% and 64.50% gold composition, respectively (Figure 4B and E). In addition, the OMP-AuNPs contained carbon (14.46%), oxygen (8.36%), chlorine (1.78%), and potassium (1.53%; Figure 4B and C), while the PE-AuNPs contained carbon (23.48%) and oxygen (12.02%; Figure 4E and F), possibly due to the presence of different types of secondary metabolites such as flavonoids and phenolic compounds in OMP and peach extracts. These secondary metabolites might have acted as the reducing and stabilizing agents during the synthesis of AuNPs.<sup>53,54</sup>

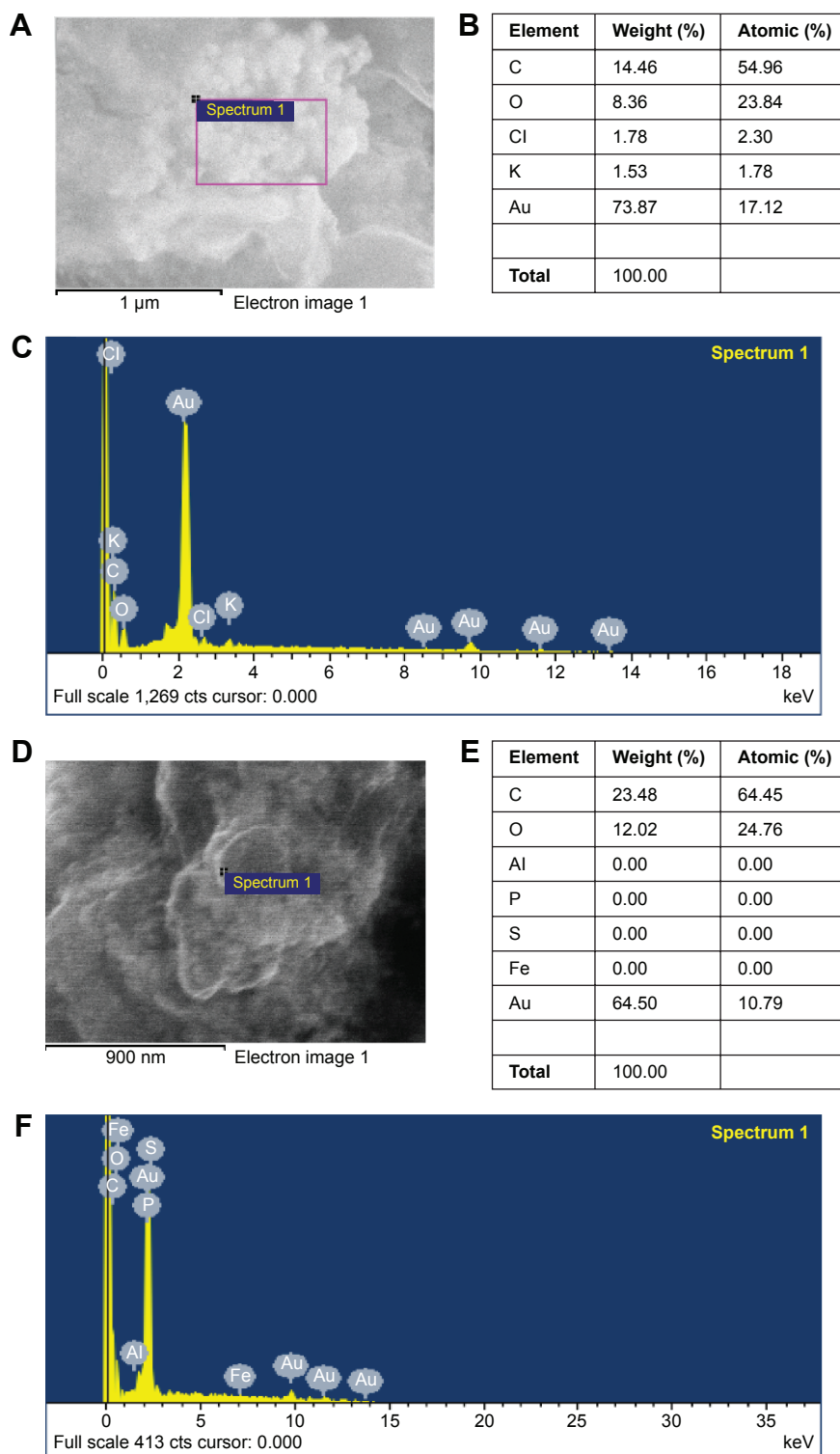
The synthesized AuNPs were further characterized for their structural information and crystallinity nature by XRD analysis (Figure 5A and B). Both OMP-AuNPs and PE-AuNPs displayed five distinct diffraction peaks in the  $2\theta$  range of  $10^\circ$ – $90^\circ$  that indexed the planes (111), (200), (220), (311), and (222) of the face-centered cubic (fcc) gold particles (Figure 5A and B). These peaks were located at  $37.96^\circ$ ,  $44.18^\circ$ ,  $64.51^\circ$ ,  $77.48^\circ$ , and  $81.74^\circ$ , for OMP-AuNPs (Figure 5A) and  $38.35^\circ$ ,  $44.52^\circ$ ,  $64.82^\circ$ ,  $77.80^\circ$ , and  $82.00^\circ$  for PE-AuNPs (Figure 5B). The XRD analysis showed predominant peaks at (111) and (200), indicating the presence of crystallite particles displaying fcc



**Figure 3** SEM image of synthesized OMP-AuNPs and PE-AuNPs.

**Notes:** (A) OMP-AuNPs; (B) PE-AuNPs.

**Abbreviations:** AuNPs, gold nanoparticles; OMP, oriental melon peel; PE, peach extract; SEM, scanning electron microscopy.



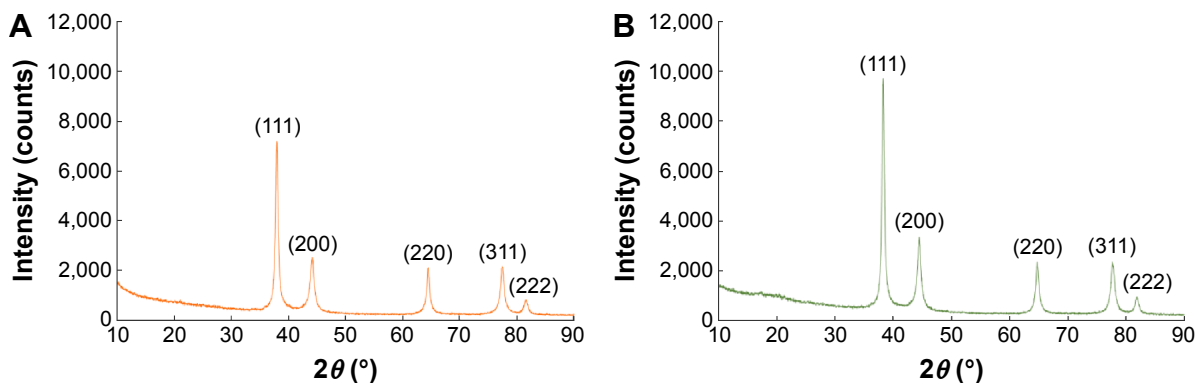
**Figure 4** EDX analysis of synthesized OMP-AuNPs and PE-AuNPs.

**Notes:** OMP-AuNPs (A–C) and PE-AuNPs (D–F).

**Abbreviations:** AuNPs, gold nanoparticles; EDX, energy dispersive X-ray; OMP, oriental melon peel; PE, peach extract.

lattice structures. The fcc structures of gold matched those in the database of the Joint Committee on Powder Diffraction Standards, USA (JCPDS No 00-004-0784), confirming that the synthesized AuNPs were composed of pure crystallite

gold particles. The estimated absolute crystallite size of the synthesized AuNPs was calculated to be 78.11 nm for OMP-AuNPs and 39.90 nm for PE-AuNPs based on the Scherer equation.<sup>55</sup> The XRD patterns of both OMP-AuNPs



**Figure 5** XRD spectra of synthesized OMP-AuNPs and PE-AuNPs.

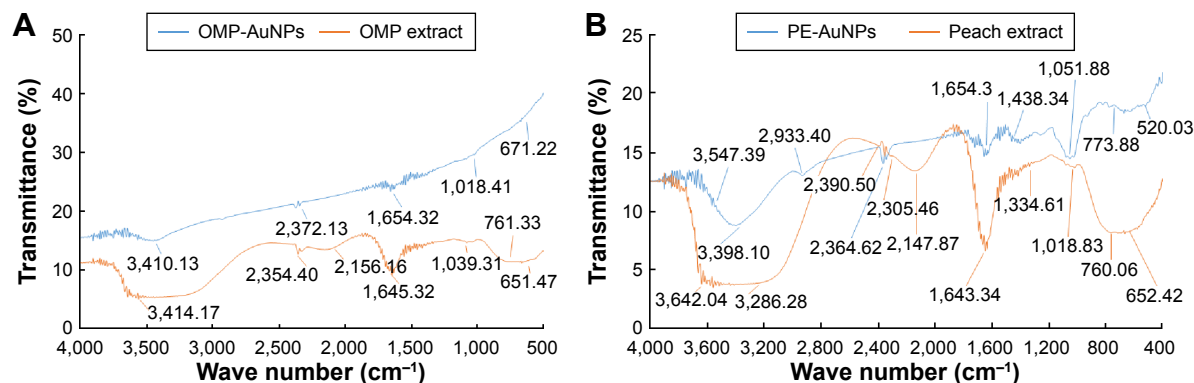
**Notes:** (A) OMP-AuNPs; (B) PE-AuNPs.

**Abbreviations:** AuNPs, gold nanoparticles; OMP, oriental melon peel; PE, peach extract; XRD, X-ray powder diffraction.

and PE-AuNPs were consistent with the results of earlier studies of AuNPs.<sup>56–58</sup>

FT-IR analysis was conducted to identify possible functional groups responsible for the reduction of AuCl<sub>3</sub> to AuNPs and their stabilization. FT-IR analysis of OMP, PE, OMP-AuNPs, and PE-AuNPs was performed in the range of 400–4,000 cm<sup>-1</sup>. FT-IR signals of OMP extract were observed at 3,414 cm<sup>-1</sup>, 2,350 cm<sup>-1</sup>, 2,156 cm<sup>-1</sup>, 1,645 cm<sup>-1</sup>, 1,039 cm<sup>-1</sup>, 761 cm<sup>-1</sup>, and 564 cm<sup>-1</sup>, while the FT-IR signals of OMP-AuNPs were observed at 3,410 cm<sup>-1</sup>, 2,372 cm<sup>-1</sup>, 1,654 cm<sup>-1</sup>, 1,018 cm<sup>-1</sup>, and 671 cm<sup>-1</sup> (Figure 6A). Similarly, the FT-IR signals of PE were observed at 3,642 cm<sup>-1</sup>, 3,286 cm<sup>-1</sup>, 2,390 cm<sup>-1</sup>, 2,305 cm<sup>-1</sup>, 2,147 cm<sup>-1</sup>, 1,643 cm<sup>-1</sup>, 1,334 cm<sup>-1</sup>, 1,018 cm<sup>-1</sup>, 760 cm<sup>-1</sup>, and 652 cm<sup>-1</sup>, while the FT-IR signals of PE-AuNPs were observed at 3,547 cm<sup>-1</sup>, 2,933 cm<sup>-1</sup>, 2,364 cm<sup>-1</sup>, 1,654 cm<sup>-1</sup>, 1,438 cm<sup>-1</sup>, 1,051 cm<sup>-1</sup>, 773 cm<sup>-1</sup>, and 520 cm<sup>-1</sup> (Figure 6B). In the case of the OMP extract, the strong peak band at 3,414 cm<sup>-1</sup> corresponded to the O–H stretch vibration of alcohols and phenols that was shifted toward 3,410 cm<sup>-1</sup> in the case of OMP-AuNPs.<sup>59</sup>

Similarly, the strong peak band at 3,642 cm<sup>-1</sup> in the peach extract corresponded to the O–H stretch vibration and free hydroxyl groups of the alcohols and phenols that was shifted toward 3,547 cm<sup>-1</sup> in the case of the PE-AuNPs.<sup>60,61</sup> The bands at 1,645 cm<sup>-1</sup> and 1,643 cm<sup>-1</sup> in OMP and peach extracts, respectively, were due to –C=C– stretching of carbonyl amide I vibrations that were shifted toward 1,654 cm<sup>-1</sup> in both OMP-AuNPs and PE-AuNPs. The presence of this particular band in the AuNPs might have been due to the proteins/peptides or amino acid groups of OMP and peach extracts, which may have acted as the reducing and capping agent for the synthesis of NPs.<sup>32,62</sup> The peak bands at 1,039 cm<sup>-1</sup> and 1,018 cm<sup>-1</sup> in both OMP and peach extracts corresponded to the C–N stretching of aliphatic amine groups and were shifted toward 1,018 cm<sup>-1</sup> and 1,051 cm<sup>-1</sup> in OMP-AuNPs and PE-AuNPs, respectively.<sup>31</sup> The peak bands at 651 cm<sup>-1</sup> and 652 cm<sup>-1</sup> in both OMP and peach extracts corresponded to the C–Br stretching of alkyl halide groups and were shifted toward 671 cm<sup>-1</sup> and 520 cm<sup>-1</sup> in OMP-AuNPs and PE-AuNPs, respectively.<sup>63,64</sup> It is speculated that the components



**Figure 6** FT-IR analysis of synthesized OMP-AuNPs and PE-AuNPs.

**Notes:** (A) OMP-AuNPs; (B) PE-AuNPs.

**Abbreviations:** AuNPs, gold nanoparticles; FT-IR, Fourier transform infrared spectroscopy; OMP, oriental melon peel; PE, peach extract.

of OMP and peach extracts containing O–H, –C=C–, C–N, or C–Br functional groups might have been responsible for the reduction of Au<sup>+</sup> to Au<sup>0</sup>.

The TG/DTG analysis of both the OMP-AuNPs and PE-AuNPs at 20°C–700°C is presented in Figure 7A and B. The TG/DTG analysis of OMP-AuNPs exhibited a total weight loss of 32.75% in three phases, 20°C–210°C with 9.96% weight loss that is mainly due to the water molecules attached to the particles, then 210°C–440°C with 19.62% weight loss that corresponds to the loss of organic materials, and a third phase of weight loss of 3.17% from 440°C to 690°C due to residual compounds. Similarly, TG/DTG analysis of PE-AuNPs exhibited a total weight loss of 40.74% in three phases, 20°C–200°C, 200°C–420°C, and 420°C–690°C, during which there were losses of 12.32%, 24.88%, and 3.54%, respectively. The reduction in weight of the synthesized AuNPs confirmed the involvement of bioactive organic compounds from OMP and peach extracts in the reduction and stabilization of NPs.<sup>31,62</sup>

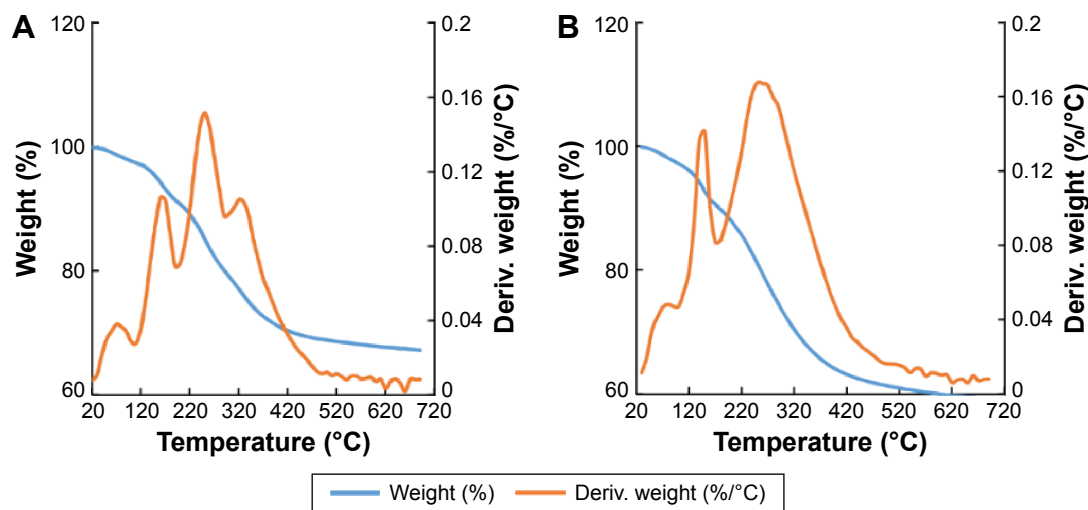
After characterization of both synthesized OMP-AuNPs and PE-AuNPs, comparative investigations of their biological potential in terms of their antibacterial, anticandidal, antioxidant, and proteasome inhibitory activities were conducted.

## Antibacterial activity

Neither the OMP-AuNPs nor PE-AuNPs, nor the OMP and peach extracts (taken as extract control), exerted any antibacterial activity against the five tested foodborne pathogenic bacteria at 25 µg/disk (Table 1). Similarly, the positive controls, kanamycin and rifampicin, at 5 µg/disk and negative

control, DMSO, did not exhibit any effects on the five tested foodborne pathogenic bacteria (Table 1). When the NPs and antibiotics were combined and the mixtures were tested for antibacterial effects, they exhibited positive synergistic responses against all the tested foodborne pathogenic bacteria (Table 1). Among the two mixtures of NPs and kanamycin, PE-AuNPs + kanamycin displayed higher synergistic activity against the four foodborne pathogenic bacteria than OMP-AuNPs + kanamycin. When the antibacterial activity of mixtures of NPs and rifampicin were compared, both exerted antibacterial activity against only three pathogenic bacteria, *B. cereus* ATCC 13061, *E. coli* ATCC 43890, and *S. aureus* ATCC 49444. Of the two mixtures of NPs/antibiotics, PE-AuNPs with kanamycin displayed stronger synergistic antibacterial activity than OMP-AuNPs. All the mixtures of NPs/antibiotics displayed strong antibacterial synergistic activity against *S. aureus*, with inhibition zones of 13.26–25.23 mm (Table 1).

The antibacterial activities of different types of NPs have been widely investigated using different pathogenic bacteria in recent years.<sup>65–67</sup> Earlier reports have suggested an increase in the antibacterial activity of AuNPs in response to the addition of antibiotics,<sup>68,69</sup> which was also evident in the present investigation. It has been reported that NPs could easily penetrate the bacterial cell membrane due to their small size, causing numerous holes in the bacterial cell wall and leakage of essential elements from the cell. When applied in combination with antibiotics, NPs could influence the bacterial nucleus and DNA, causing cell death.<sup>66,67</sup> The antimicrobial potentials of different parts of *Cucumis melo*



**Figure 7** TG/DTG analysis of synthesized OMP-AuNPs and PE-AuNPs.

**Notes:** (A) OMP-AuNPs; (B) PE-AuNPs.

**Abbreviations:** AuNPs, gold nanoparticles; Deriv., derivative; DTG, derivative thermogravimetric; OMP, oriental melon peel; PE, peach extract; TG, thermogravimetric.

**Table 1** Comparative synergistic antibacterial activity of OMP-AuNPs (25 µg) and PE-AuNPs (25 µg) with standard antibiotics, kanamycin (5 µg) or rifampicin (5 µg), against foodborne pathogenic bacteria

Bacteria	Antibacterial activity (control samples)												
	OMP extract					PE							
	OMP extract	PE	OMP-AuNPs	PE-AuNPs	DMSO	Rifampicin	Kanamycin	Rifampicin	DMSO	AuNPs with rifampicin			
<i>B. cereus</i> ATCC 13061	0	0	0	0	0	0	0	0	0	11.42±0.74 <sup>a*</sup>	10.47±0.50 <sup>cd</sup>	10.04±0.19 <sup>c</sup>	9.52±0.14 <sup>d</sup>
<i>E. coli</i> ATCC 43890	0	0	0	0	0	0	0	0	0	11.03±0.52 <sup>c</sup>	20.45±0.51 <sup>a</sup>	9.78±0.38 <sup>cd</sup>	10.01±0.21 <sup>cd</sup>
<i>L. monocytogenes</i> ATCC 19115	0	0	0	0	0	0	0	0	0	9.38±0.27 <sup>d</sup>	13.48±0.56 <sup>b</sup>	0.00±0.00 <sup>e</sup>	0.00±0.00 <sup>e</sup>
<i>S. aureus</i> ATCC 49444	0	0	0	0	0	0	0	0	0	13.26±0.90 <sup>b</sup>	13.98±0.07 <sup>b</sup>	22.39±0.53 <sup>b</sup>	25.23±0.90 <sup>a</sup>
<i>S. Typhimurium</i> ATCC 43174	0	0	0	0	0	0	0	0	0	11.23±0.28 <sup>c</sup>	13.07±0.21 <sup>b</sup>	0.00±0.00 <sup>e</sup>	0.00±0.00 <sup>e</sup>

Notes: \*Data are expressed as the mean zone of inhibition in millimeter ± SD. Values with different superscript letters in the columns with kanamycin and in the columns of AuNPs with rifampicin are significantly different at  $P < 0.05$ .  
 Abbreviations: AuNPs, gold nanoparticles; ATCC, American Type Culture Collection; *B. cereus*, *Bacillus cereus*; DMSO, dimethyl sulphoxide; *E. coli*, *Escherichia coli*; *L. monocytogenes*, *Listeria monocytogenes*; OMP, oriental melon peel; PE, peach extract; *S. typhimurium*, *Salmonella typhimurium*; *S. aureus*, *Staphylococcus aureus*.

and *Prunus persica* have been reported.<sup>19,70–75</sup> The bioactive compounds present in peel extracts, which have been used for AuNPs synthesis and stabilization, might be responsible for the antibacterial activity of the AuNPs. The synergistic potential of AuNPs with antibiotics could be useful in minimizing the uncontrolled use of antibiotics, which would have resulted in the development of many antibiotic resistant strains.

### Anticandidal activity

The synergistic anticandidal activities of both OMP-AuNPs and PE-AuNPs mixed with amphotericin B were evaluated against five different *Candida* species (Table 2). Initial testing of both OMP-AuNPs and PE-AuNPs at 50 µg/disk and amphotericin B at 5 µg/disk for their anticandidal activities revealed no inhibitory activity against any of the tested pathogenic *Candida* species (Table 2). However, these compounds were subsequently mixed together at the same concentrations, after which their synergistic anticandidal potentials were tested (Table 2). The extract control, positive control, and negative control did not display any inhibition activity against any of the pathogenic *Candida* species (Table 2). Both the OMP-AuNPs and PE-AuNPs combined with amphotericin B displayed a strong anticandidal activity against all the five tested *Candida* species. The AuNPs were more active against *C. albicans* KACC 30062 and *C. geochares* KACC 30061, as indicated by a 13.27–15.47 mm inhibition zone (Table 2). Among the two AuNPs, the PE-AuNPs + amphotericin B mixture exhibited a stronger positive activity than the OMP-AuNPs + amphotericin B. The comparative high anticandidal activity of PE-AuNPs could be correlated with their smaller particle size (average size = 39.90 nm), which could have facilitated the easy penetration of the amphotericin B/PE-AuNPs into *Candida* cells, resulting in cellular lysis. The use of less amphotericin B and AuNPs using the mixtures could be beneficial to clinical applications of the mixture solution, avoiding the adverse effects caused by the use of high doses of drugs.<sup>76</sup>

### Antioxidant activity

The potential application of both OMP-AuNPs and PE-AuNPs was evaluated in terms of its antioxidant activity by a number of in vitro antioxidant assays. The DPPH free radical scavenging potential of both OMP-AuNPs and PE-AuNPs are presented in Figure 8A. Among the two AuNPs, PE-AuNPs exhibited the stronger DPPH radical scavenging activity of 56.92% at 100 µg/mL, while OMP-AuNPs and BHT (the reference standard compound) showed activities of 26.21% and 34.82%, respectively (Figure 8A). The DPPH scavenging activity of AuNPs

**Table 2** Comparative synergistic anticandidal activities of OMP-AuNPs (50 µg) and PE-AuNPs (50 µg) mixed with amphotericin B (5 µg), a standard antifungal agent against pathogenic *Candida* species

<i>Candida</i> species	Anticandidal activity (control samples)					Synergistic anticandidal activity		
	OMP extract	PE	Amphotericin B	DMSO	OMP-AuNPs	PE-AuNPs	OMP-AuNPs + amphotericin B	PE-AuNPs + amphotericin B
<i>Candida albicans</i> KACC 30003	0	0	0	0	0	0	10.96±0.93 <sup>e*</sup>	10.97±0.41 <sup>de</sup>
<i>Candida albicans</i> KACC 30062	0	0	0	0	0	0	14.45±0.53 <sup>b</sup>	15.47±0.27 <sup>a</sup>
<i>Candida glabrata</i> KBNO6P00368	0	0	0	0	0	0	10.66±0.47 <sup>de</sup>	11.09±0.66 <sup>d</sup>
<i>Candida geochares</i> KACC 30061	0	0	0	0	0	0	13.27±0.28 <sup>c</sup>	14.25±0.22 <sup>b</sup>
<i>Candida saitoana</i> KACC 41238	0	0	0	0	0	0	10.09±0.30 <sup>e</sup>	11.08±0.28 <sup>de</sup>

**Notes:** \*Data are expressed as the mean zone of inhibition in millimeter ± SD. Values with different superscript letters are significantly different at  $P < 0.05$ .

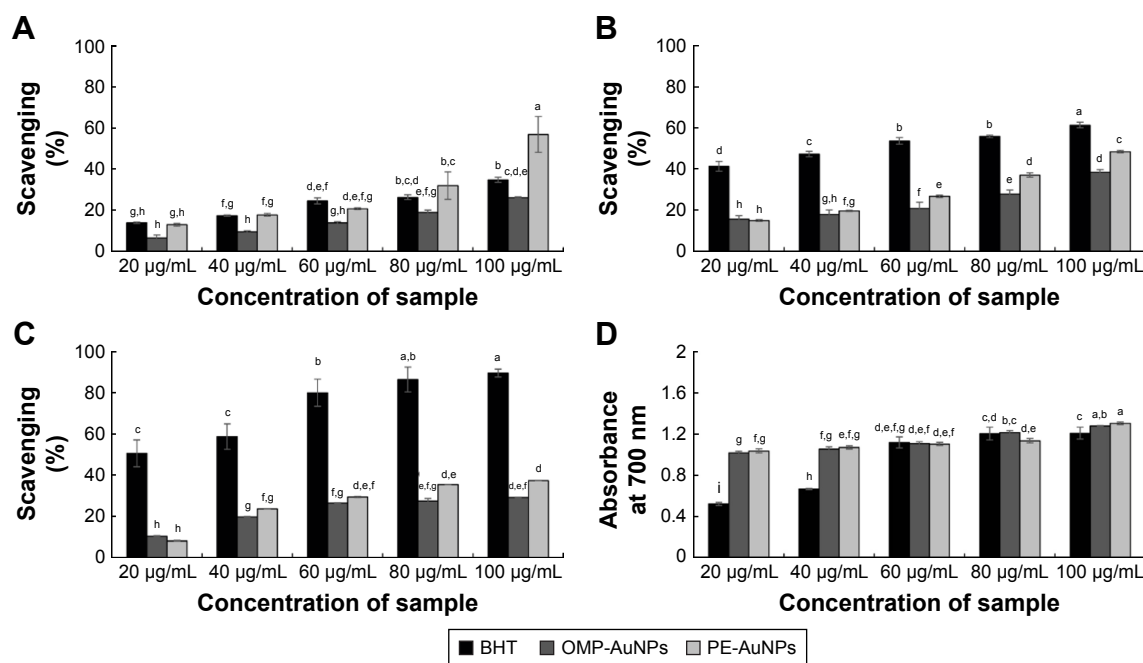
**Abbreviations:** AuNPs, gold nanoparticles; DMSO, dimethyl sulphoxide; OMP, oriental melon peel; PE, peach extract; SD, standard deviation.

might be attributed to the presence of a number of secondary metabolites, mostly phenolic compounds, in the fruit peel extracts that were used for the biosynthesis of AuNPs. The potential effects of AuNPs might occur via its inhibitory effects on the peroxy radicals, which are responsible for lipid peroxidation.<sup>77</sup>

The nitric oxide scavenging potentials of AuNPs are presented in Figure 8B. Both OMP-AuNPs and PE-AuNPs displayed nitric oxide scavenging activity of 38.57% and 48.46%, respectively, at 100 µg/mL; however, the potentials were less than that of BHT, the standard reference compound, which showed 61.47% scavenging activity (Figure 8B). Nitric oxide is a very unstable compound responsible for various diseases, including cancer and inflammatory diseases;<sup>78,79</sup>

thus, the strong nitric oxide scavenging potentials of both PE-AuNPs and OMP-AuNPs could be beneficial for the biomedical and cosmetic industries.

PE-AuNPs and OMP-AuNPs also displayed moderate ABTS radical scavenging activities of 29.11% and 37.36%, respectively, while that of the standard reference compound BHT was 89.61% (Figure 8C). Moreover, they exhibited a strong reducing power of 1.28 and 1.31 absorbance at 700 nm, while the absorbance of BHT was 1.20 (Figure 8D). The strong reducing power of both OMP-AuNPs and PE-AuNPs might be attributed to the presence of phenolic compounds in the outer peel extracts. Among the AuNPs, PE-AuNPs exhibited the highest antioxidant potentials as indicated by all of the in vitro assays when compared to

**Figure 8** Antioxidant potential of synthesized OMP-AuNPs and PE-AuNPs.

**Notes:** (A) DPPH free radical scavenging potential; (B) nitric oxide scavenging potential; (C) ABTS radical scavenging potential; and (D) reducing power potential. Values with different superscript letters are significantly different at  $P < 0.05$ .

**Abbreviations:** ABTS, 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid); AuNPs, gold nanoparticles; BHT, butylated hydroxyl toluene; DPPH, 1,1-diphenyl-2-picrylhydrazyl; OMP, oriental melon peel; PE, peach extract.

the OMP-AuNPs. The high antioxidant potentials might be attributed to the presence of a number of phenolic compounds in the outer peel of *Prunus persica*, which was used for the green synthesis of PE-AuNPs.<sup>29</sup>

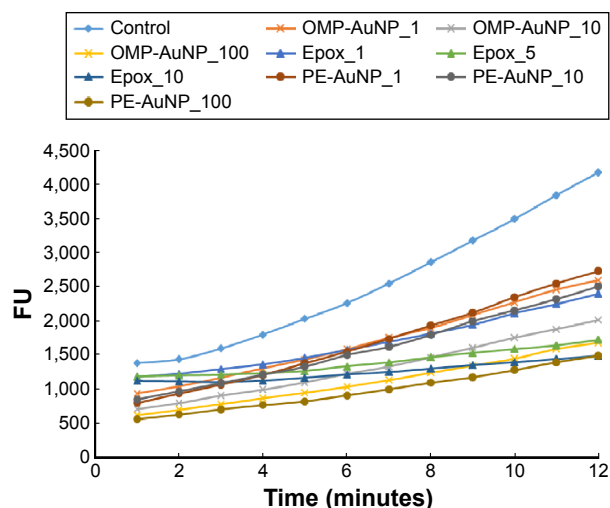
### Proteasome inhibitory activity

The proteasome inhibitory potential of both OMP-AuNPs and PE-AuNPs is presented in Figure 9. The results showed that the values of the fluorescence unit decreased in all three PE-AuNPs, OMP-AuNPs and epoxomicin as the concentration increased, confirming their proteasome inhibitory potential (Figure 9). PE-AuNPs displayed proteasome inhibitory activities of 33.77%, 36.04%, and 60.46% at 1  $\mu\text{g/mL}$ , 10  $\mu\text{g/mL}$ , and 100  $\mu\text{g/mL}$ , respectively, while the OMP-AuNPs displayed activities of 31.74%, 47.67%, and 55.35% at 1  $\mu\text{g/mL}$ , 10  $\mu\text{g/mL}$ , and 100  $\mu\text{g/mL}$ , respectively. Epoxomicin, the reference standard, displayed 30.28%, 40.06%, and 45.67% activity at concentrations of 0.56  $\mu\text{g/mL}$ , 2.78  $\mu\text{g/mL}$ , and 5.56  $\mu\text{g/mL}$ , respectively. The results confirmed that PE-AuNPs displayed a stronger proteasome inhibitory activity than OMP-AuNPs and epoxomicin, which might have been due to the presence of a number of compounds, including chlorogenic acid, catechin, epicatechin, rutin, and cyanidin-3-glucoside in the outer peel of *Prunus persica* that was used for the biosynthesis of PE-AuNPs. It is well known that cancer cells have defective cell cycle checkpoints and DNA repair control mechanisms, while proteasomes are responsible

for the degradation of many cytosolic proteins, including cyclins, misfolded, or denatured proteins, and transcription factors in cells.<sup>80,81</sup> If these cancer cells are placed under stress conditions by the use of proteasome inhibitors such as AuNPs, they will then face tremendous problems and not be able to correct the cell cycle transition blockade. As a result, they may enter the apoptosis phase without affecting the normal cell.<sup>82</sup> Both PE-AuNPs and OMP-AuNPs with potent proteasome inhibitory potential could serve as potential candidates for targeted drug delivery and drug conjugation with efficient anticancer drugs in cancer treatment.

### Conclusion

In the current study, the biosynthesis of AuNPs was attempted using the outer peel extracts of *Cucumis melo* and *Prunus persica* fruits as the sole reducing agents. To the best of our knowledge, the use of naturally available fruit waste materials has not been investigated for such applications to date. The NPs were synthesized utilizing green technology, which is a nontoxic and environmental friendly procedure and is better than the conventional physical and chemical methods that give rise to highly toxic products. The synthesized NPs were subjected to a number of characterization techniques, including UV-visible spectroscopy, FE-SEM, EDX, XRD, FT-IR, and TG/DTG analysis, and confirmed to be AuNPs. The estimated absolute crystallite size of the synthesized AuNPs was calculated to be 78.11 nm for OMP-AuNPs and 39.90 nm for PE-AuNPs based on the Scherer equation. Both synthesized AuNPs were evaluated for antibacterial, anticandidal, antioxidant, and proteasome inhibitory potential. Among the synthesized NPs, the PE-AuNPs were found to be the most promising NPs, with strong antibacterial, anticandidal, antioxidant, and proteasome inhibitory activities. Bioactive compounds such as chlorogenic acid, catechin, epicatechin, rutin, and cyanidin-3-glucoside, which are the major constituents of peach fruits as evident from previous published literatures, might be responsible for the potential activity of PE-AuNPs. The significant biological properties of both synthesized ecofriendly NPs could make them promising candidates in biomedical, pharmaceutical, food, and cosmetic industries.



**Figure 9** Proteasome inhibitory potential of both OMP-AuNPs and PE-AuNPs and standard reference compound, Epox.

**Notes:** 100  $\mu\text{g/mL}$  (OMP-AuNP\_100, PE-AuNP\_100); 10  $\mu\text{g/mL}$  (OMP-AuNP\_10, PE-AuNP\_10); 1  $\mu\text{g/mL}$  (OMP-AuNP\_1, PE-AuNP\_1) and 5.56  $\mu\text{g/mL}$  (Epox\_10); 2.78  $\mu\text{g/mL}$  (Epox\_5) and 0.56  $\mu\text{g/mL}$  (Epox\_1).

**Abbreviations:** AuNPs, gold nanoparticles; FU, fluorescence units; Epox, epoxomicin; OMP, oriental melon peel; PE, peach extract.

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## Disclosure

The authors report no conflicts of interest in this work.

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