

Ginger components as new leads for the design and development of novel multi-targeted anti-Alzheimer's drugs: a computational investigation

Faizul Azam^{1,2} Abdualrahman M Amer¹ Abdullah R Abulifa¹ Mustafa M Elzwawi¹

Faculty of Pharmacy, Misurata University, Misurata, Libya; ²Department of Pharmaceutical Chemistry, Nims Institute of Pharmacy, Nims University, Jaipur, Rajasthan, India

Abstract: Ginger (Zingiber officinale), despite being a common dietary adjunct that contributes to the taste and flavor of foods, is well known to contain a number of potentially bioactive phytochemicals having valuable medicinal properties. Although recent studies have emphasized their benefits in Alzheimer's disease, limited information is available on the possible mechanism by which it renders anti-Alzheimer activity. Therefore, the present study seeks to employ molecular docking studies to investigate the binding interactions between active ginger components and various anti-Alzheimer drug targets. Lamarckian genetic algorithm methodology was employed for docking of 12 ligands with 13 different target proteins using AutoDock 4.2 program. Docking protocol was validated by re-docking of all native co-crystallized ligands into their original binding cavities exhibiting a strong correlation coefficient value (r^2 =0.931) between experimentally reported and docking predicted activities. This value suggests that the approach could be a promising computational tool to aid optimization of lead compounds obtained from ginger. Analysis of binding energy, predicted inhibition constant, and hydrophobic/hydrophilic interactions of ligands with target receptors revealed acetylcholinesterase as most promising, while c-Jun N-terminal kinase was recognized as the least favorable anti-Alzheimer's drug target. Common structural requirements include hydrogen bond donor/ acceptor area, hydrophobic domain, carbon spacer, and distal hydrophobic domain flanked by hydrogen bond donor/acceptor moieties. In addition, drug-likeness score and molecular properties responsible for a good pharmacokinetic profile were calculated by Osiris property explorer and Molinspiration online toolkit, respectively. None of the compounds violated Lipinski's rule of five, making them potentially promising drug candidates for the treatment of Alzheimer's disease.

Keywords: Alzheimer's disease, ginger, molecular docking, structure–activity relationship, toxicity prediction

Introduction

Alzheimer's disease (AD) is a progressive neurodegenerative disorder of the central nervous system that accounts for 60%–70% of dementia cases in persons over 65 years of age worldwide. Neurodegeneration in AD, and other neurodegenerative diseases appear to be multifactorial, in that a complex set of deleterious reactions including glutamatergic neurotoxicity, increases in iron and nitric oxide, reduced expression of trophic factors, dysfunction of the ubiquitin-proteasome system, depletion of endogenous antioxidants, expression of proapoptotic proteins, and inflammation leads to the decease of neurons.^{2,3} With human life span increasing and with decreasing cognitive

Correspondence: Faizul Azam Faculty of Pharmacy, Misurata University, PO Box 2873, Misurata, Libya Tel +218 91 3300 265 Email faizulazam@gmail.com

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functions in elderly individuals with AD-related dementia, AD has become a major health problem in society.⁴

As of today, there are only few US Food and Drug Administration (FDA)-approved drugs in the market for treating AD patients. These include acetylcholinesterase (AChE) inhibitors (tacrine, donepezil, rivastigmine) and a noncompetitive inhibitor of N-methyl-D-aspartate (NMDA) receptors, memantine. All of these drugs improved the cognitive functions of AD patients symptomatically and have thus improved the quality of life for these patients; however, these drugs do not modify the disease progression in the long run.⁵ Additionally, they show limited clinical effects over the shorter term for some patients, mild to moderate cholinergic adverse effects in a minority of patients, and potentially distressing toxicity over the longer term.⁶

Furthermore, a great number of drugs with a variety of targets and clusters of mechanisms are currently in various stages of basic and clinical investigation. However, the development of therapies for this devastating disorder has been perturbing for physicians, researchers, and the pharmaceutical industry, with many drug candidates showing promise at one stage of clinical research only to fall at the next hurdle. Consequently, no ensuing experimental drug in development has been successful thus far; there has not been a new drug marketed for AD in a decade. The paucity of currently available drugs for treating AD, and their limited targets in AD pathology, as well as their proven side effects demand the development of a new generation of drugs that not only affect cholinergic functions associated with AD but also target other cellular pathways in AD pathogenesis.

Owing to the complexity of AD pathogenesis, the classic "one molecule, one target" solution may not be effective enough. Recently, the novel multi-target-directed strategy has received avid attention of researchers, since single molecules simultaneously interact with multiple targets in complex neurotoxic cascades, may achieve better efficacy in a complementary manner. Meanwhile, multi-targeted drugs would have a larger therapeutic window than those hitting a single target and would thus prove to be safer drugs. 9

Since ancient times, the beneficial effects of some natural compounds have been appreciated in preventing various age-related pathologic conditions, including brain aging as well as neurodegeneration, and have been invigorated by a plethora of experimental and epidemiological studies.² Ginger (*Zingiber officinale* [*Z. officinale*]) is a common dietary adjunct that contributes to the taste and flavor of foods. In addition to its flavor, ginger is known to contain a number of potentially bioactive phytochemicals, mainly

gingerols and their related dehydrating products, the shogaols (Figure 1) as well as volatile oils.¹⁰

Gingerols are pungent principles in the rhizome of ginger and possess the labile β-hydroxy keto functional group, which makes it susceptible to transformation to less-pungent compounds such as shogaols and zingerone by elevated temperature. Gingerols and shogaols have been reported to exhibit many interesting pharmacological and physiological functions, for example, antipyretic, cardiotonic, chemopreventive, anti-inflammatory, and antioxidant properties. 11-14 It has been reported that ginger extract inhibits the production of nitric oxide (NO) and proinflammatory cytokines in lipopolysaccharide (LPS)-stimulated BV-2 microglial cells via the NF-κB (nuclear factor kappa-light-chain-enhancer of activated B cells) pathway.¹⁵ An inhibitory effect by 6-gingerol was shown on the production of proinflammatory cytokines in murine peritoneal macrophages. ¹⁶ Likewise, 6-shogaol has been shown to inhibit LPS-induced inducible nitric oxide synthase (iNOS) and cyclooxygenase (COX) gene expression in macrophages. 14 Moreover, 6-shogaol showed significant neuroprotective effects in vivo in transient global ischemia via the inhibition of microglia. It suppressed the microglial activation induced by LPS both in primary cortical neuron-glia culture and in an in vivo neuroinflammatory model.¹⁷ In vitro data have shown that ginger's active principles protect nerve cells and may have potential in the treatment of AD.¹⁸

Anti-Alzheimer potential of gingerol has been disclosed more precisely when it was reported that it attenuates β-amyloid-induced oxidative cell death via fortifying cellular antioxidant defense system. In the said study, [6]-gingerol pretreatment protected against $A\beta_{25-35}$ -induced cytotoxicity and apoptotic cell death such as DNA fragmentation, disruption of mitochondrial membrane potential, elevated Bax/Bcl-2 ratio, and activation of caspase-3.19 In addition, the potential of traditional Chinese medicinal ginger root extract has been assessed for its ability to prevent behavioral dysfunction in the AB-induced AD model in rats very recently. 20 Ginger's potential in AD treatment has been further ascertained when its extract inhibited AChE activity as well as lipid peroxidation in the brain in a dose-dependent manner.²¹ Various mechanisms by which ginger compounds elicit interesting pharmacological activities have been presented in Table 1.

Molecular docking is a computational method for finding out binding modes of ligands to their receptors rapidly, and is being applied consistently to drug design and discovery projects.^{22–25} Therefore, the present study seeks to employ this technique to investigate the binding interactions between active ginger components and various anti-Alzheimer drug targets. This will not only help in disclosing the interactions

Figure 1 Chemical structure of potential bioactive phytochemicals from *Zingiber officinale*. **Abbreviation:** Ac, acetyl.

of ginger components with multi-targets but will also play an important role in revealing the anti-Alzheimer's mechanisms as well as assist in lead optimization. In addition, drug-likeness score and molecular properties responsible for good pharmacokinetic profile were calculated by Osiris property explorer (www.organicchemistry.org/prog/peo/) and Molinspiration online toolkit (http://www.molinspiration.com/cgi-bin/properties) respectively.

Materials and methods

Molecular docking

Preparation of the protein receptor

The crystal structures of the protein-ligand complexes for the 13 AD-associated targets were used for the docking calculations (Table 2). They were downloaded from the Research Collaboratory for Structural Bioinformatics (RCSB) Protein Data Bank (PDB; http://www.rcsb.org/pdb/home/home.do). For each crystal structure, the crystallographic water molecules were removed, the missing hydrogen atoms were added, and the inhibitor from the crystal structure was used to define the active site.

Preparation of ligands

Structures of the 12 ginger compounds with proven in vitro and/or in vivo activity against various central nervous system insults were retrieved from the literature. The structures of the ligands were drawn in ChemDraw 8.0 (PerkinElmer Informatics, Waltham, MA, USA) and converted to their

Azam et al Dovepress

Table I Various mechanisms associated with ginger

Serial number	Ginger compound or extract	Mechanisms of action	Reference	
I	6-Shogaol	Inhibition of lipopolysaccharide (LPS)-induced inducible nitric oxide synthase (iNOS) and cyclooxygenase (COX) gene expression in macrophages	14	
2	6-Shogaol	Inhibition of nitric oxide (NO) and the expression of iNOS induced by LPS	17	
3	6-Shogaol	Neuroprotective effects in vivo in transient global ischemia via the inhibition of microglia	17	
4	6-Shogaol	Protects neurons by increasing acetyltransferase and choline transport expression through a brain-derived neurotrophic factor escalation	34	
5	6-Gingerol	Inhibits the production of proinflammatory cytokines in murine peritoneal macrophage	16	
6	[6]-Gingerol	Inhibits COX-2 expression by blocking the activation of p38 mitogen-activated protein (MAP) kinase and NF-κB in phorbol ester-stimulated mouse skin	40	
7	Gingerol	Attenuates β-amyloid-induced oxidative cell death	19	
8	Zerumbone	Inhibitors of acetylcholinesterase (anti-AChE)	33	
9	Zingerone	Acts as an antioxidant by inhibiting the formation of peroxynitrite (ONOO)	35	
10	Zingerone	Increases superoxide dismutase activity and scavenges superoxide radical	36	
П	Gingerols and diarylheptanoids	Inhibits prostaglandin and leukotriene biosynthesis	37	
12	Extract of Zingiber officinale	Inhibited the production of NO and proinflammatory cytokines in LPS-stimulated BV-2 microglial cells via the NF- κ B pathway	15	
13	Extract of Zingiber officinale	Improves memory impairment in focal cerebral ischemic rats and mitigates brain damage	38	
14	Extract of Zingiber officinale	Scavenges free radicals in quinic acid-induced lipid peroxidation	39	
15	Extract of Zingiber officinale	Ginger varieties inhibit acetylcholinesterase activities in vitro with the white ginger causing higher AChE inhibition than the red ginger	21	
16	Extract of Zingiber officinale	The authors indicated that ginger hexane extract significantly inhibited the excessive production of NO, prostaglandin E2, TNF- α , and IL-1 β in LPS-stimulated BV-2 cells	32	

 $\textbf{Abbreviation:} \ NF-\kappa B, \ nuclear \ factor \ kappa-light-chain-enhancer \ of \ activated \ B \ cells.$

Table 2 Protein targets with anti-Alzheimer's effect or target enzymes of drug design selected for docking studies

S. no	Name of the targets	PDB	Experimenta	ı	Docking predicted	
		code	K _i ^a	pK _i	K _i ^a	pK _i
Ī	Acetylcholinesterase (AChE)	4EY5 ⁵⁰	0.00851	7.62	0.0154	7.81
2	Butyrylcholinesterase (BuChE)	4B0P ⁵²	44 ⁵²	4.36	40.01	4.39
3	β -Site amyloid precursor protein cleaving enzyme (BACE-I)	4DJU ⁵³	3.653	5.44	2.49	5.60
4	Glycogen-synthase-kinase-3 β (GSK-3 β)	1Q5K ⁵⁴	0.7755	6.11	0.996	6.00
5	TNF- α converting enzyme (TACE)	2FV5 ⁵⁶	0.0005657	9.25	0.000448	9.35
6	c-Jun N-terminal kinase (JNK)	3G9N ⁵⁸	1.858	5.74	0.5929	6.23
7	Nitric oxide synthase (NOS)	IQWC ⁵⁹	7.360	5.14	3.43	5.46
8	Human carboxylesterase (hCE-1)	IMXI ⁶¹	10061	4.00	10.69	4.97
9	N-methyl-D-aspartate (NMDA)	IPBQ ⁶²	0.5462	6.27	3.69	5.43
10	Cyclooxygenase-I (COX-I)	IEQG ⁶³	964	5.05	6.76	5.17
П	Cyclooxygenase-2 (COX-2)	3QMO ⁶⁵	0.9266	6.04	0.2928	6.53
12	Phosphodiesterase-5 (PD-5)	IUDT ⁶⁷	0.001868	8.74	0.00192	8.72
13	Angiotensin converting enzyme (ACE)	3BKL ⁶⁹	0.67969	6.17	0.5312	6.27

Note: aData given in μM .

Abbreviations: PDB, protein data bank; K, inhibition constant; pK, negative logarithm of inhibition constant.

three-dimensional structures in Chem3D 8.0 (PerkinElmer Informatics). Geometry optimization was done using PM3 method by MOPAC program (http://OpenMOPAC.net). Finally, all the compounds were saved in pdb format for further docking studies.

Docking simulation

Lamarckian genetic algorithm methodology was employed for docking simulations implemented in AutoDock 4.2 (The Scripps Research Institute, La Jolla, CA, USA).²⁷ The standard docking procedure was used for a rigid protein and a flexible ligand whose torsion angles were identified (for ten independent runs per ligand). A grid of 60, 60, and 60 points in x, y, and z directions was built with a grid spacing of 0.375 Å and a distance-dependent function of the dielectric constant was used for the calculation of the energetic map. The default settings were used for all other parameters.

Analysis and visualization of docking simulation results

At the end of docking, the best poses were analyzed for hydrogen bonding or π interactions and root mean square deviation (RMSD) calculations using Discovery Studio Visualizer 2.5 (Accelrys Software Inc., San Diego, CA, USA) and PyMol version 1.3 (The PyMOL Molecular Graphics System; Schrödinger, LLC, New York, NY, USA) programs. From the estimated free energy of ligand binding ($\Delta G_{\text{binding}}$, kcal/mol), the inhibition constant (K_i) for each ligand was calculated (Table 3).

Calculation of pharmacokinetic parameters

Molinspiration online property calculation toolkit and Osiris property explorer were used to check the pharmaceutical fidelity of the drug candidates. Molecular descriptors, such as miLogP, the number of hydrogen bond donors, the number of hydrogen bond acceptors, the molecular mass of the compounds, topological polar surface area (TPSA), number of rotatable bonds, and violations of Lipinski's rule of five²⁸ were calculated using Molinspiration online property calculation toolkit. Percentage of absorption (%ABS) was calculated by: %ABS =109–[0.345× TPSA] according to the method of Zhao et al.²⁹

Calculation of toxicity potential

Osiris Property Explorer was used to analyze various attributes of the drugs, such as toxicity, drug-likeness, and drug score.

Results and discussion

Validation of docking protocol

Initially, all the 13 native co-crystallized ligands were extracted from the receptor and re-docked within the inhibitor binding cavity of respective receptors in order to validate the docking calculations, reliability, and reproducibility of the docking parameters for the study. It was evident that the docked conformation of the ligands was almost superimposed with that of the respective co-crystallized ligands (Figure 2). As a general rule, if the best-docked conformation of a ligand resembles the bound native ligand in the experimental crystal structure, the used scoring function is said to be successful. According to the method of validation cited in the literature, 30 the successful scoring function is the one in which the RMSD of the best-docked conformation is ≤ 2.0 Å from the experimental one. In this study, RMSD values of all docked targets were within 2.0 Å (Figure 2), indicating that our docking protocol is valid for the given structures and AutoDock 4.2, therefore deemed reliable for docking ginger components into the inhibitor binding cavity of multiple targets implicated in the pathogenesis of AD.

In addition, all of the experimentally reported and docking predicted binding affinity data were converted to their logarithmic scale and plotted to obtain a correlation coefficient r^2 of 0.931 (Figure 3). It means that the docking methodology employed in the present study could be a promising computational tool to aid optimization of lead compounds obtained from ginger. In addition, the docked conformation of each ligand was compared with the respective crystal structure conformation by calculating RMSD values and satisfactory results were achieved (Figure 2).

Interaction of the ginger compounds with potential protein targets

Thirteen potential protein receptors, which are either conventional clinical targets with anti-Alzheimer's effects or target enzymes of drug design, were selected for docking studies. Docking predicted as well as experimentally reported K_i values for native co-crystallized ligands, PDB codes and related citation is presented in Table 2. For every target protein, a total of ten poses were visualized for each of the 12 ginger compounds to identify the model with minimum binding energy and estimated K_i values as well as best ligand–receptor interaction. The results in terms of binding free energy, predicted K_i , RMSD, number of hydrophobic as well as hydrophilic interaction are presented in Table 3. Since the values of docking predicted K_i , belong to different scales, they were converted

Dovepress Azam et al

Table 3 Results obtained after docking of ginger compounds (I-I2) with various protein targets

PDB	Docking results	Ginger	compou	nds									
code		I	2	3	4	5	6	7	8	9	10	П	12
4EY5	$\Delta G_{\rm b}^{\ a}$	-8.5 I	-9.25	-6.21	-8.14	-9.07	-9.92	-9.66	-10.89	-8.67	-9.48	-9.21	-8.68
	K_i^b	0.58	0.17	28.19	1.08	0.23	53.35 ^f	82.36 ^f	10.38 ^f	0.44	0.11	0.18	0.44
	RMSD ^c	2.33	4.22	1.01	1.89	4.04	4.22	4.71	4.07	4.23	3.58	1.69	2.25
	HB⁴	I	I	I	3	2	2	3	5	2	4	2	5
25)/5	π-int ^e	-	-	-	2	2	-	2	2	5	5	2	-
2FV5	ΔG_{b}^{a}	-8.60 0.40	-8.85 0.33	-6.80	-8.39	-8.66 0.45	-9.30	-9.47	-8.92 0.30	-8.63	-9.68	-9.84	-9.20
	<i>K</i> ₁⁰ RMSD°	0.49 0.67	0.33 2.25	10.43 4.73	0.71 1.56	0.45 0.98	0.15 1.79	0.11 1.69	0.29 2.33	0.47 2.32	80.25 ^f 2.35	61.76 ^f 1.85	0.18 2.26
	HB ^d	2	2.23	2	2	3	3	2	4	4	5	6	4
	π -int $^{ m e}$	_	1	_	1	_	_	_	_	_	_	_	_
4B0P	$\Delta G_{\rm b}^{\ a}$	-7.78	-8.36	-6.15	-7.55	-7.82	-8.38	-8.81	-8.10	-8.26	-8.73	-9.72	-9.70
	К , ь в	1.99	0.74	31.08	2.93	1.85	0.72	0.35	1.15	0.88	0.40	74.44 ^f	77.61
	RMSD°	3.12	3.61	5.36	2.87	3.37	1.41	2.22	1.78	1.63	1.67	1.54	1.24
	HB⁴	-	1	I	1	-	2	_	-	I	I	I	-
	π -int $^{ m e}$	-	1	-	_	-	2	_	I	2	1	2	3
IQWC	$\Delta G_{\rm b}^{\ a}$	-7.99	-8.12	-6.36	-7.42	-6.89	-7.59	-8.11	-7.92	-7.82	-6.98	-9.61	-9.02
	K _i ^b	1.39	1.12	21.93	3.66	8.88	2.73	1.14	1.57	1.86	7.63	90.09 ^f	0.25
	RMSD° HB⁴	7.83 I	8.22	8.41 I	7.23 I	6.67	6.60	6.08	5.80 I	6.85 2	7.28	7.21 3	7.62
	π-int ^e	2	_ 3	2	2	_ 3	1 3	 -	2	3	6 3	_	2 5
IUDT	$\Delta G_{\rm b}^{\ a}$	_7.79	_7.97	-5.81	_7.13	-6.94	_7.54	-8.40	-9.21	-7.61	_7.27	-9.12	_8.60
.05.	K_i^b	1.96	1.45	55.27	5.98	8.13	2.99	0.69	0.18	2.64	4.69	0.21	0.49
	RMSD ^c	3.19	4.17	5.51	3.01	1.33	1.65	5.58	2.14	5.76	5.56	2.48	1.38
	HB⁴	2	1	I	2	I	2	I	4	4	5	3	4
	π -int $^{\mathrm{e}}$	1	_	-	1	-	_	_	1	1	-	-	-
IPBQ	$\Delta G_{_{ m b}}{}^{ m a}$	-7.25	-7.15	-5.5 I	-6.85	-7.30	-8.30	-7.94	-7.25	-7.77	-7.97	-9.13	-9.21
	K_i^b	4.85	5.77	90.94	9.52	4.45	0.83	1.51	4.82	2.01	1.43	0.20	0.18
	RMSD ^c	4.69	5.14	0.63	4.13	4.50	3.87	4.30	2.39	2.90	3.69	3.93	3.34
	HB⁴	-	_	3	_	-	_	-	_	I	-	2	3
2040	π-int ^e	-	-	I	- 7.70	l	I	I	2	-	-	-	1
3QMO	ΔG_{b}^{a}	-8.38 0.72	-8.13 1.09	-6.02 38.96	-7.79 1.95	-8.34 0.77	-8.02 1.31	-8.13 1.10	-6.69 12.54	-7.57 2.83	–7.86 1.72	–7.85 1.76	-9.43
	<i>K</i> ₁ʰ RMSD°	2.80	2.32	2.55	2.10	2.11	2.33	4.25	1.64	2.63	1.72	0.74	0.12 2.07
	HB⁴	1	1	1	2	1	2.33	2	5	4	4	4	4
	π -int $^{ m e}$	_	_	_	_	_	_	_	_	_	_	I	ı
IMXI	$\Delta G_{\!\scriptscriptstyle b}^{^{\;a}}$	-7.12	-6.53	-5.30	-6.20	-6.89	-8.00	-8.13	-6.91	-7.77	-8.03	-8.34	-7.72
	K _i ^b	6.00	16.31	0.13^g	28.43	8.89	1.37	1.10	8.59	2.00	1.31	0.77	2.20
	RMSD ^c	1.76	3.45	3.57	1.92	2.35	6.05	1.71	3.50	2.05	5.32	3.63	1.76
	HB⁴	I	I	I	4	4	3	2	_	3	4	3	2
	π -int $^{ m e}$	-	-	-	-	-	_	-	I	-	-	-	-
IEQG	ΔG_{b}^{a}	-7.3 l	-7.55	-5.83	-7.39	-8.23	-7.70	-7.97	-8.07	-7.22	-7.37	-8.36	-8.80
	K _i ^b	4.35	2.92	53.68	3.80	0.93	2.29	1.44	1.22	5.08	3.94	0.75	0.35
	RMSD° HB⁴	1.83 I	2.25 I	3.96 I	0.85 5	3.64 2	1.12 2	1.99 I	1.95 2	3.73 5	1.73 2	1.31 5	1.57 3
	π -int $^{\mathrm{e}}$	_	_	_	_	_	_	_	_	I	_	_	_
IQ5K	$\Delta G_{\rm b}^{a}$	-7.14	-6.33	-5.50	-6.17	-5.52	-7.63	-7.28	-6.36	-6.43	-6.43	-7.26	-6.77
•	<i>K</i> _i ^b	5.82	23.01	93.21	30.23	89.54	2.56	4.63	21.80	19.19	19.51	4.79	10.91
	R ['] MSD ^c	1.92	1.27	3.15	0.90	2.39	1.40	1.77	3.40	2.21	1.10	1.24	3.05
	HB^d	2	-	I	1	2	3	2	4	5	2	4	4
	$\pi ext{-int}^{ ext{e}}$	-	I	I	1	_	_	1	-	_	1	I	-
4DJU	$\Delta G_{b}^{\;\;a}$	-6.39	-6.26	-5.09	-5.72	-5.75	-7.49	-7.49	-6.06	-5.73	-6.44	-6.86	-7.35
	K _i ^b	20.88	25.58	0.19g	64.39	60.84	3.23	3.24	35.84	62.83	18.91	9.35	4.12
	RMSD ^c	3.18	2.39	5.18	3.92	1.75	1.77	1.36	0.76	4.40	1.66	3.35	4.83
	HB⁴	2	I	2	2	2	2	4	2	5	3	5	4

(Continued)

Table 3 (Continued)

PDB code	Docking	Ginger	Ginger compounds											
	results	I	2	3	4	5	6	7	8	9	10	П	12	
3BKL	$\Delta G_{\rm b}^{\ a}$	-5.68	-6.94	-4.7 I	-5.42	-5.82	-6.56	-7.04	-5.63	-6.30	-6.78	-7.47	-7.56	
	K,b	68.88	8.20	0.35^g	0.118	54.29	15.65	6.96	75.02	24.15	10.70	3.37	2.85	
	RMSD ^c	1.83	1.79	4.83	2.09	1.15	1.63	4.29	3.44	3.70	2.78	5.35	5.01	
	HB⁴	1	3	2	2	3	2	3	4	3	5	6	8	
	π -int $^{ ext{e}}$	_	_	_	_	1	_	-	2	-	-	-	_	
3G9N	$\Delta G_{\rm b}^{\ a}$	-6.82	-6.73	-5.6 l	-6.56	-5.94	-6.78	-6.86	-6.40	-6.46	-6.79	-6.97	-7.33	
	K,b	9.99	11.65	77.76	15.56	44.44	10.72	9.31	20.32	18.44	10.53	7.78	4.25	
	RMSD ^c	0.79	2.10	4.07	2.30	0.42	1.61	0.52	2.35	3.07	0.93	3.40	2.07	
	HB⁴	1	I	I	2	I	2	3	2	3	3	3	3	
	$\pi ext{-int}^{ ext{e}}$	_	_	_	_	_	_	_	_	_	_	_	-	

Notes: *Binding free energy (kcal/mol); *docking predicted inhibition constant in μM ; *root mean square deviation in Å; *dnumber of hydrogen bond interactions; *number of π -interactions; *data presented in mM; *data presented in mM.

Abbreviations: PDB, protein data bank; RMSD, root mean square deviation; HB, hydrogen bond; K, inhibition constant.

to pK_i (negative logarithm of K_i) values in order to have uniform data for a comparative study. A comparison of docking predicted activities of ligands against respective targets demonstrated in Figure 4 implicates that all of the docked ginger compounds possess the properties of promiscuous drugs.

Natural products are a rich source of lead compounds. Many of today's medicines are either obtained directly from a natural source or were developed from a lead compound originally obtained from a natural source. In initial stage, the level of activity associated with lead compound may not be prodigious and there may be undesirable side effects,

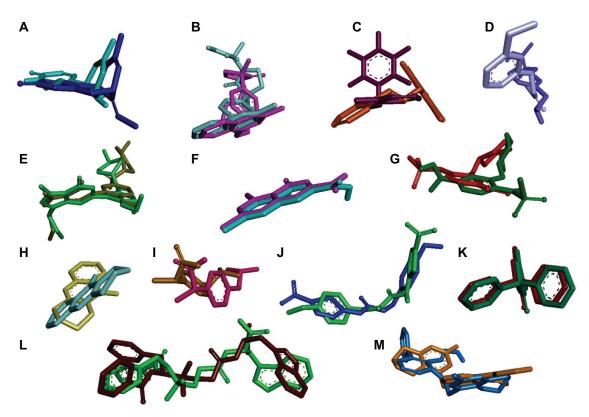


Figure 2 The validation of accuracy and performance of AutoDock 4.2.

Notes: The native and docked-ligands of AChE, blue and sky blue (A); TACE, pink and sky blue (B); BuChE, dark magenta and dark orange (C); NOS, light steel blue and purple (D); PDE-5, gold and pale green (E); NMDA, magenta and turquoise (F); COX-2, red and green (G); hCE-1, medium turquoise and khaki (H); COX-1, brown and fuchsia (I); GSK-3, blue and green (J); BACE-1, red and medium sea green (K); ACE, maroon and green (L); JNK, orange and sky blue (M) respectively.

Abbreviations: AChE, acetylcholinesterase; TACE, TNF- α converting enzyme; BuChE, butyrylcholinesterase; NOS, nitric oxide synthase; PDE-5, phosphodiesterase-5; NMDA, N-methyl-D-aspartate; COX, cyclooxygenase; hCE-1, human carboxylesterase-1; GSK, glycogen-synthase-kinase-3 β ; BACE, β -site amyloid precursor protein cleaving enzyme; ACE, angiotensin-converting enzyme; JNK, c-Jun N-terminal kinase.

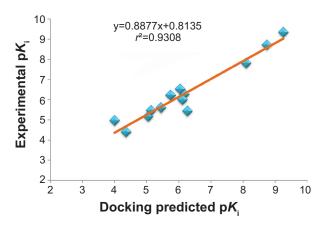


Figure 3 Plot between experimentally reported and docking predicted activities of native co-crystallized ligands of all 13 targets. **Abbreviation:** pK, negative logarithm of inhibition constant.

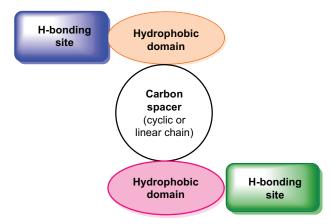


Figure 5 Suggested pharmacophore model of ginger compounds for eliciting anti-Alzheimer's effects.

but the lead compound provides a start for the drug design and development process. Drugs interacting with multiple targets might have a better chance of affecting the complex equilibrium of whole cellular networks than drugs that act on a single target. Ginger compounds were capable of interacting with all docked targets showing variable affinities, which indicated that these ligands have broad-spectrum structural features that make them proficient for recognizing numerous significant target proteins. Figure 5 reveals common structural features essential for activity.

It is evident from the plot presented in Figure 4 that AChE is the most promising and JNK (c-Jun N-terminal kinase) is the least favorable anti-Alzheimer's drug target. In addition, butyrylcholinesterase (BuChE), TNF-α converting enzyme, COX-2, NOS, and NMDA are proposed as best putative targets for ginger's bioactive phytochemicals. Docked

ligand–protein complexes of most promising compounds in various potential targets are presented in Figure 6.

Various experimental reports have evidenced that ginger extract is capable of inhibiting targets such as AChE, BuChE, ³¹ COX-1, COX-2, JNK, and NOS. ³² Table 1 demonstrates the biological activities of ginger compounds or extracts that have been reported in the literature. ^{14–17,19,21,32–40} Due to the molecular complexity of AD, multi-targeted therapies are becoming increasingly important as, in the long-term, they maximize the therapeutic effect and overcome the adverse effects associated with combination therapy. ⁹ Thus, the potential of multi-targeted therapies that have been identified in ginger compounds may be a key explanation for why ginger extract is effective as an anti-Alzheimer's treatment. ^{14–17,19,21,32–40} In addition, it has been recognized that ginger in folklore medicinal practice possesses various pharmacological properties

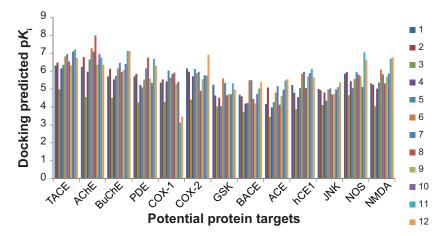


Figure 4 Plot between docked targets and negative logarithmic values of docking predicted K_i of ginger compounds. **Abbreviations:** AChE, acetylcholinesterase; TACE, TNF-α converting enzyme; BuChE, butyrylcholinesterase; NOS, nitric oxide synthase; PDE-5, phosphodiesterase-5; NMDA, N-methyl-D-aspartate; COX, cyclooxygenase; GSK, glycogen-synthase-kinase-3 β ; BACE, β -site amyloid precursor protein cleaving enzyme; ACE, angiotensin converting enzyme; JNK, c-Jun N-terminal kinase; hCE-1, human carboxylesterase-1; pK₁, negative logarithm of inhibition constant.

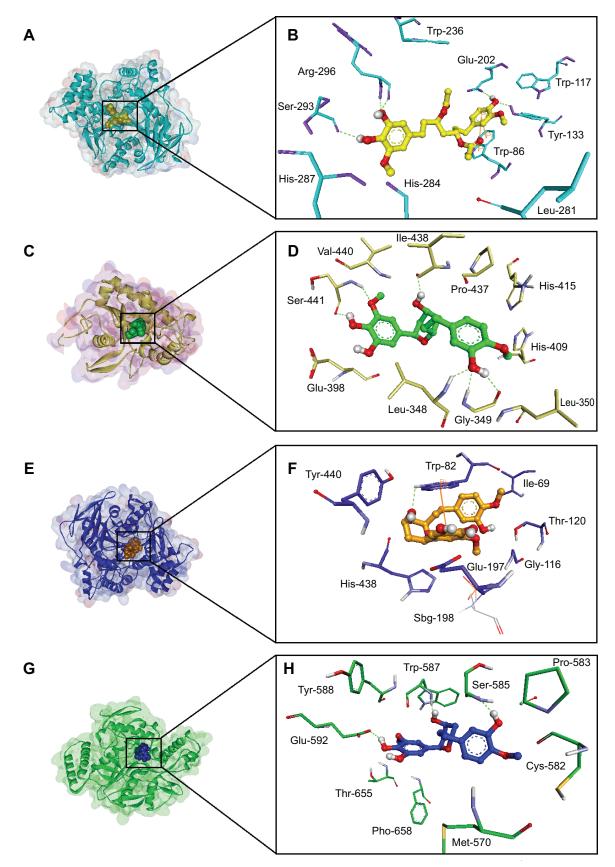


Figure 6 3D structures of proteins showing the binding sites (left), and main residues involved in the ligand–protein (right) interaction of compound 8 and AChE (**A**, **B**), compound 11 and TACE (**C**, **D**), compound 11 and BuChE (**E**, **F**), and compound 11 and NOS (**G**, **H**). **Abbreviations:** AChE, acetylcholinesterase; TACE, TNF-α converting enzyme; BuChE, butyrylcholinesterase; NOS, nitric oxide synthase.

due to the different components attacking various targets or different steps in the pathologic process of AD.²¹ Therefore, the components of ginger, which have different mechanisms of anti-Alzheimer action, interact primarily in an additive or synergistic manner. A single drug should be more economical and lead to fewer adverse effects than a combination with each drug targeting a different protein.

Structure–activity relationship study presented in Figure 7 identifies the importance of various functionalities for ligandreceptor interactions: 1) replacement of distal hydrophobic domain (phenyl ring) with methyl group (compound 3) is detrimental for activity in all targets; 2) in hydrogen-bonding area, p-OH and m-OCH, groups are important for all targets; additional m-OH is beneficial for most targets except glycogen-synthase-kinase-3β and β-site amyloid precursor protein cleaving enzyme; 3) double bond between C₁ and C₂ in compound 2 appreciates the activity at COX-1, COX-2, NOS, and angiotensin converting enzyme (ACE); however, double bond between C₄ and C₅ in compound 1 improves the activity at COX-2, glycogen-synthase-kinase- 3β and JNK; 4) C₃ and/or C₅ must be substituted with C=O, OH, and OAc groups for optimum activity at all targets; 5) carbons C₁ and C₅ may be cyclized to form tetrahydropyran, which is essential for the activity of compounds 11 and 12 at NMDA, BuChE, ACE, JNK, and NOS; 6) for maximum activity, R should be aromatic ring substituted with H-bond donor/acceptor group at para and/or meta positions for all targets; 7) compounds having linear chain as carbon spacer are moderate or poor in activity; however, compounds 1 and 5 have exhibited appreciable activity at COX.

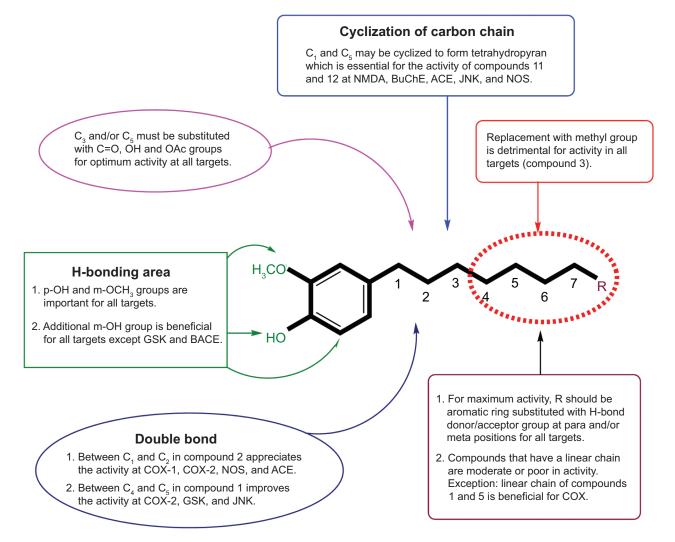


Figure 7 An overview of the structural requirements of ginger compounds for their interaction with different target receptors associated with Alzheimer's disease. Abbreviations: BuChE, butyrylcholinesterase; NOS, nitric oxide synthase; NMDA, N-methyl-D-aspartate; COX, cyclooxygenase; GSK, glycogen-synthase-kinase-3β; BACE, β-site amyloid precursor protein cleaving enzyme; ACE, angiotensin converting enzyme; JNK, c-Jun N-terminal kinase; Ac, acetyl.

Binding interactions between ginger components and most promising target, AChE

Comparative analysis of the docking results revealed that AChE is the most favorable target for interaction of ginger components (Figure 4). The neuropathology of AD is characterized by early loss of basal forebrain cholinergic neurons, leading to decreased cholinergic transmission, which can be improved with AChE inhibitors or by modulation of muscarinic and nicotinic acetylcholine receptors. In folklore medicine, ginger has been reportedly used for the treatment of AD as ginger extract, ginger tea, or as inclusion in food formulations and preparation.²¹ Inhibition of AChE activity by extract of *Z. officinale* has been documented in an in vitro study where white ginger caused higher AChE inhibition than the red ginger.²¹

Due to differences in substrate specificity and susceptibility to various kinds of inhibitors, cholinesterases were divided into two types: AChE, which hydrolyzes acetylcholine, and BuChE, which is able to hydrolyze larger molecules, such as butyrylcholine. 41,42 AChE is the main enzyme metabolizing acetylcholine. It is also responsible for cerebral blood flow modulation, β -amyloid aggregation, activation and expression of APP95 protein, τ protein phosphorylation and has an influence on inflammatory processes. It interacts

with β -amyloid, leading to creation of stable complexes and formation of senile plaques.^{43,44} The other enzyme, BuChE, is also important, because of its ability to hydrolyze acetylcholine and other choline esters. It was observed that the BuChE level increases in AD patients. Its role is not fully understood, but some studies suggested that it could promote amyloid plaque formation, and therefore, the search for inhibitors of both enzymes has been undertaken for the treatment of AD.^{45,46}

Native co-crystallized ligand, Huperzine A is surrounded by residues Trp-86, Tyr-119, Gly-120, Gly-121, Gly-122, Tyr-124, Ser-125, Gly-126, Tyr-133, Glu-202, Ser-203, Tyr-337, and His 447, constituting active site of AChE enzyme (Figure 8). All of the docked compounds utilized the same amino acids of inhibitor binding pocket for polar as well as nonpolar interactions. The active site of AChE is composed of a catalytic triad (Ser-203, Glu-334, and His-447) that sits at the bottom of a narrow ~20 Å deep gorge. Just at the mouth of the gorge is the peripheral anionic binding site, which is composed of Tyr-72, Asp-74, Tyr-124, Glu-285, Trp-286, and Tyr-341 (Figure 8). Compounds 4, 9-10, and 12 utilized Tyr-72 and Tyr-124 for hydrophilic interaction while hydrophobic π – π interaction was noted with Trp-286 and Tyr-341 in the peripheral anionic-binding site. None of the compounds interacted with the residues of the catalytic

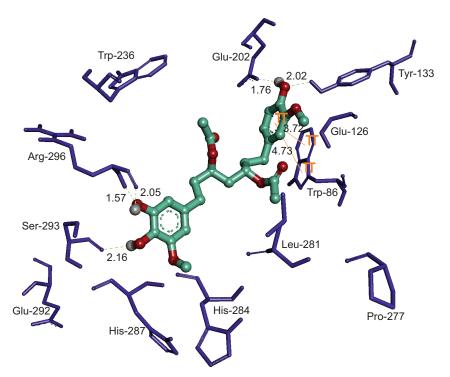


Figure 8 The lowest energy configuration of docking result of compound 7 with binding pocket of acetylcholinesterase.

Notes: The amino acids (blue) are shown as stick while compound 7 is presented as ball and stick style in olive green color. Dashed lines in green indicate H-bonds while π - π interaction is shown as orange line. Oxygens are in red and polar hydrogens in light gray color.

triad except compound 10, which used phenyl ring for sharing π - π interaction with His-447.

The oxy-anionic hole comprises Gly-120, Gly-121, and Ala-204. Compound 4 and 12 shared H-bond with Gly-120 of oxy-anionic hole. The choline-binding site is defined mostly by Trp-86. Compounds 1-4, and 6 used Trp-86 for hydrophobic interaction with choline binding site. Compound 7 appears to interact with AChE most persuasively (Figure 9), conferring minimum binding energy among the docked compounds. The binding interactions are dominated by polar interactions involving hydroxyl group of Tyr-133, and carboxylic groups of Glu-202, Ser-293, and Arg-296. van der Waals contacts in terms of π - π interactions were noted with Trp-86. This compound has exploited the residues of ligand-binding area (Glu-202 and Trp-86) to demonstrate its maximum potency (Figure 8). In general, the docking energies are lower for all ligands in BuChE in comparison to AChE (Table 3).

Prediction of pharmacokinetic properties

A successful oral drug is one that is promptly and completely absorbed from the gastrointestinal tract, distributed specifically to its site of action in the body, metabolized in a way that does not instantly remove its activity, and eliminated in a suitable manner, without causing any harm to the organs. It is a well-known fact that approximately half of all drugs in development fail to make it to the market because of poor pharmacokinetics (PK).²⁸ The PK properties depend on the chemical descriptors of the molecule. Computational prediction of PK properties such as absorption, distribution, metabolism, excretion, and toxicity (ADMET) have become

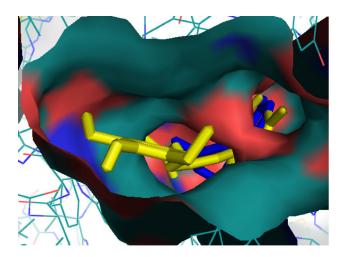


Figure 9 The docked compound 7 (yellow) and native Huperzine A (blue) in the binding pocket of human acetylcholinesterase.

progressively important in drug selection and promotion process and are promising tools to determine the success of the compound for human therapeutic use.⁴⁷ Therefore, early prediction of ADMET properties has been done with the objective of increasing the success rate of the ginger compounds in future development processes.

Molinspiration online property calculation toolkit was utilized to screen the ginger compounds as drug candidate based on Lipinski's rule of five and the results are presented in Table 4. This rule is based on the surveillance that most orally administered drugs have a molecular weight of ≤ 500 , a LogP (logarithm of partition coefficient) ≤ 5 , five or fewer hydrogen bond donor sites, and ten or fewer hydrogen bond acceptor sites. Molecules violating more than one of these rules may have problems with bioavailability. Fortunately, none of the compounds under study has violated these criterions.

In addition, the bioavailability of ginger compounds was judged through TPSA analysis. This descriptor has been reported to correlate with passive molecular transport through membranes and therefore, allows prediction of transport properties of drugs and has been linked to drug bioavailability. As per the Veber's rule for good oral bioavailability, the number of rotatable bond must be ≤10, and TPSA values ≤140 Ų.⁴8 The number of rotatable bonds has been shown to be a very good descriptor of oral bioavailability of drugs. Rotatable bond is defined as any single non-ring bond, bounded to non-terminal heavy (ie, non-hydrogen) atom. Amide C−N bonds are not considered because of their high rotational energy barrier. With exception of compounds 5 and 8, number of rotatable bonds was found to be appropriate in all ginger compounds.

Percentage of absorption was estimated using the equation: %ABS =109–0.345× TPSA, according to Zhao et al.²⁹ TPSA was also calculated using Molinspiration online property calculation toolkit according to the fragment-based method of Ertl et al.⁴⁹ Generally, it has been seen that passively absorbed molecules with a TPSA >140 Å² are thought to have low oral bioavailability. According to the above criterions, calculated percentages of absorption for ginger compounds 1–12, ranged between 68.82% and 92.95%.

Toxicity risks and drug score assessment

Nowadays, it is much more convenient to predict the toxicity risks of compounds through reliable bioinformatics tools. In the present study, Osiris property explorer was used to

Table 4 Physicochemical parameters for good oral bioavailability of ginger compounds (I-I2)

Compound	%ABS ^a	TPSA (Ų)b	MW ^c	miLog P d	HBD°	HBAf	n-ROTB ^g	Violation of Lipinski's rule
Rule	_	_	<500	≤5	<5	<10	≤10	≤
I	92.95	46.53	276.37	4.34	I	3	9	0
2	87.06	63.60	290.35	3.06	1	4	9	0
3	92.95	46.53	194.23	1.51	I	3	4	0
4	85.97	66.76	294.39	3.21	2	4	10	0
5	85.99	66.76	350.49	5.23	2	4	14	0
6	76.89	93.06	372.41	2.24	2	6	10	1
7	82.78	75.99	356.41	3.32	2	5	9	0
8	75.32	97.62	448.55	4.30	3	7	14	0
9	74.71	99.38	376.44	2.61	4	6	10	0
10	68.82	116.45	390.43	2.13	4	7	10	0
11	71.53	108.61	390.43	2.33	4	7	6	0
12	74.71	99.38	360.40	2.51	4	6	5	0

Notes: *Percentage of absorption (%ABS); *topological polar surface area (TPSA); *molecular weight (MVV); *dlogarithm of partition coefficient between *n*-octanol and water (miLogP); *number of hydrogen bond donors (HBD); *fnumber of hydrogen bond acceptors (HBA); *snumber of rotatable bonds (*n*-ROTB).

calculate toxicity risks parameters such as mutagenicity, tumorigenicity, irritation, and reproductive or developmental toxicity of all the ginger compounds, 1–12 (Table 5). The predictions are based on the functional group similarity for the query molecule with the in vitro and in vivo validated compounds present in the database of this online program. The toxicity risk predictor locates fragments within a molecule, which indicate a potential toxicity risk. The results can be visualized using color codes; green color shows low toxic tendency, yellow shows the mediocre, and red color shows high tendency of toxicity. Toxicity screening results presented in Figure 9 showed that none of the compounds (1-12) pose the risk of tumorogenicity and reproductive toxicity; however, compound 1 indicated high risk of mutagenicity. On the other hand, compounds 2 and 3 indicated high risk of irritation.

To assess the ginger compound's overall potential to qualify for a drug, overall drug score was calculated, which combines drug-likeness, hydrophobicity (LogP), aqueous solubility (LogS), MW, and toxicity risk parameters. The hydrophobicity of drugs could be inferred from LogP value. LogP values are directly proportional to the oral hydrophobicity of the drug. The more hydrophobic the drug, higher is the ability of the drug to circulate longer in our body. It would not be easy to excrete such a drug. In the present investigation, the miLogP values of the drug molecules were observed to be in the range of 1.51–5.23 (Table 4).

Conclusion

In present study, a comparative molecular docking approach using AutoDock was taken to identify the potential anti-Alzheimer receptors for ginger bioactive phytochemicals such as gingerols, shogaols, zingerone, and related compounds. The results show that: 1) several targets such as AChE, BuChE, TNF- α converting enzyme, NOS, COX-2, and NMDA, identified in this study are proposed as best putative targets

Table 5 Drug-likeness/scores and toxicity calculations of ginger compounds based on Osiris property explorer

Compound	Solubility	Drug-likeness	Drug score	Mutagenic	Tumorigenic	Irritant	Reproductive effect
I	-3.42	-14.48	0.22	Red	Green	Green	Green
2	-3.44	-16.52	0.24	Green	Green	Red	Green
3	-2.03	-2.22	0.31	Green	Green	Red	Green
4	-3.25	-7.78	0.4	Green	Green	Green	Green
5	-4.33	-15.79	0.27	Green	Green	Green	Green
6	-3.35	-6.82	0.39	Green	Green	Green	Green
7	-3.47	-3.48	0.38	Green	Green	Green	Green
8	-3.81	-3.6	0.3	Green	Green	Green	Green
9	-3.25	0.23	0.61	Green	Green	Green	Green
10	-3.0 l	2.51	0.77	Green	Green	Green	Green
H	-2.89	2.26	0.77	Green	Green	Green	Green
12	-2.87	1.27	0.74	Green	Green	Green	Green

for ginger phytochemicals; 2) a number of targets identified by docking, such as AChE, BuChE, COX-1, COX-2, JNK, and NOS, have already been verified by experiments for their inhibition by ginger extracts; 3) it is being proposed that 1,3-diacetoxy derivative (compound 7) binds to the AChE active site with certain orientation and conformation so that it may act as an inhibitor of that enzyme. Structure analysis shows that electrostatic interaction and hydrogen bonding play an important role in their binding process. The study provides important information for optimizing lead compounds of ginger for the treatment of AD.

Disclosure

The authors have no conflicts of interest to disclose.

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